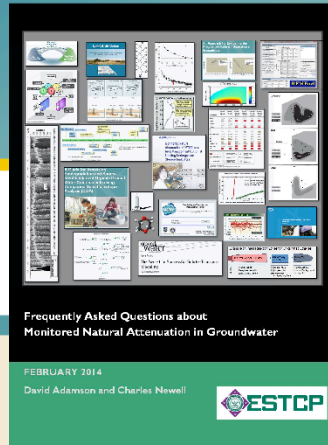


# FREQUENTLY-ASKED QUESTIONS ABOUT MONITORED NATURAL ATTENUATION IN GROUNDWATER



*Charles Newell, Ph.D., P.E.,  
David Adamson, Ph.D., P.E.*



## PROJECT TEAM



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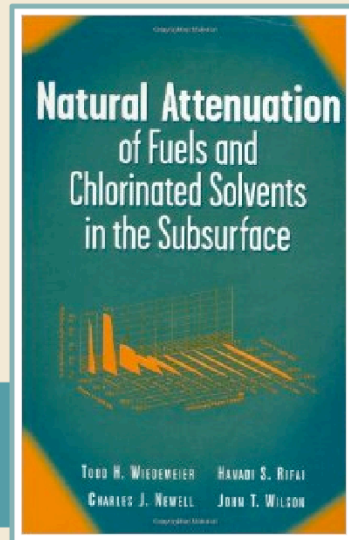
## PRESENTATION OUTLINE

1. Motivation
2. What's an "FAQ" document?
3. MNA themes – Part I
4. Interview w/ John Wilson (followed by Initial Q&A)
5. MNA themes – Part II (including poll questions)
6. Tour through the interactive version
7. Final Q&A

## MOTIVATION

- MNA in 2014: multiple sources of info, different vintages, different focus, need a library
- Example: 1999 MNA book
  - Great info, but we've learned a lot in the last 15 years!

**SOLUTION:** Frequently-Asked Questions document about MNA to easily communicate MNA issues and alternatives

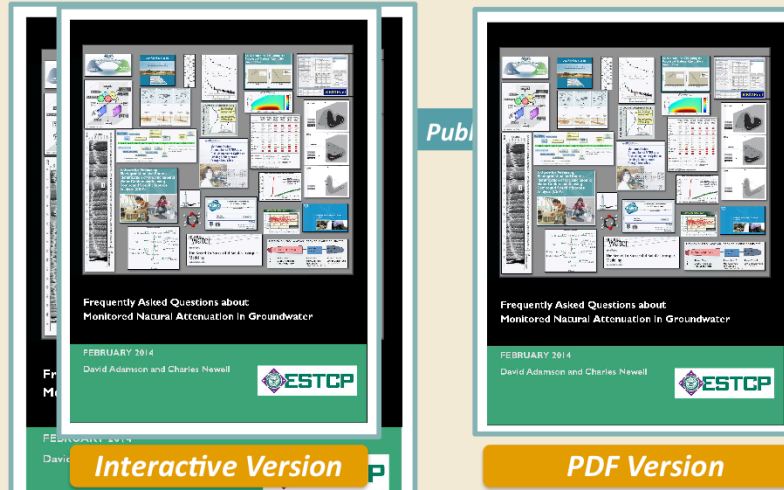






Answers to Frequently Asked Questions  
About Managing Risk at LNAPL Sites

## WHAT'S AN "FAQ" DOCUMENT?



<http://serdp-estcp.org/Program-Areas/Environmental-Restoration/Contaminated-Groundwater/Persistent-Contamination/ER-201211/>

## MNA THEMES

*MNA AS AN EVOLVING TECHNOLOGY:*

*THE BASIS FOR MNA:*

*NEW CONTAMINANTS FOR THE MNA LINEUP:*

*NEW TECHNIQUES, NEAT TOOLS:*

*EMERGING ISSUES FOR EVALUTING MNA AS A REMEDY:*

*IMPLEMENTATION AND SITE CLOSURE:*

## MNA THEMES

### *MNA AS AN EVOLVING TECHNOLOGY:*

1. Do we still need MNA?  
*THE BASIS FOR MNA.*
2. What key protocols explain how to implement MNA?
3. What are the most important new MNA developments?  
*NEW CONTAMINANTS FOR THE MNA LINEUP:*

### *NEW TECHNIQUES, NEAT TOOLS:*

### *EMERGING ISSUES FOR EVALUTING MNA AS A REMEDY:*

### *IMPLEMENTATION AND SITE CLOSURE:*

## FAQ 1: *Do we still need MNA?*

### *MNA from a Technical-Regulatory-Scientific Perspective:*

#### 3 Views on How MNA fits in (quotes from recent ITRC docs)

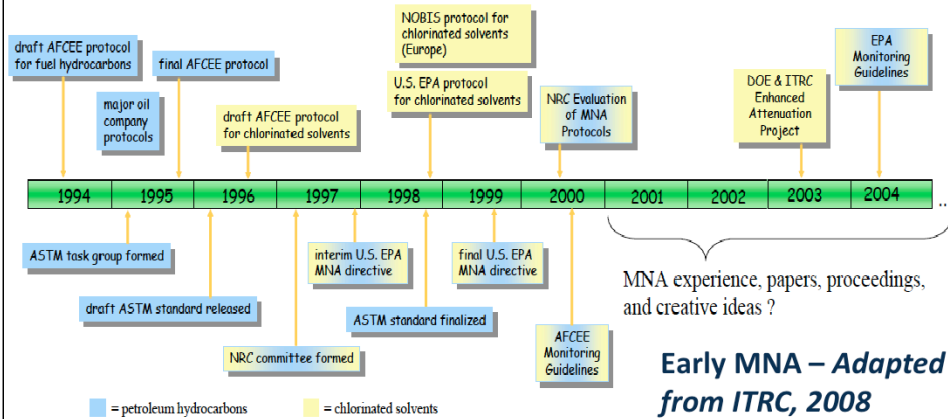
- 1.** *Active remediation technologies rarely achieve complete remediation of all contaminant mass; thus, in effect, MNA is typically a component of every chlorinated-solvent site remedy.*
- 2.** *(Natural Source Zone Depletion) is also of significance because engineered remedial actions typically do not always completely remediate soils and NSZD may be useful to address the residual hydrocarbon.*
- 3.** *Monitored natural attenuation (MNA) may also be a viable remedial alternative for situations in which the potential for adverse impacts to public health or sensitive environmental receptors is very low.*

## FAQ 2: *How has MNA changed over time?*

*New technologies, concepts, and increasing use*

Through 2004

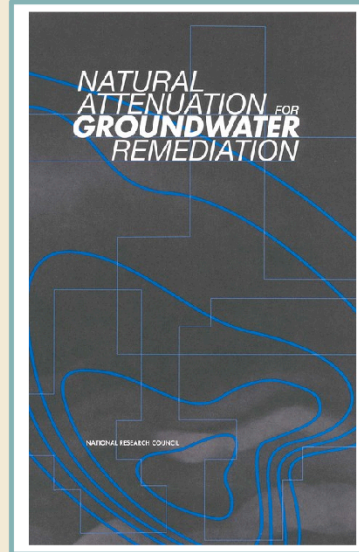
### Timeline – Natural Attenuation of hydrocarbons and chlorinated solvents



## FAQ 2: *How has MNA changed over time?*

### KEY POINTS:

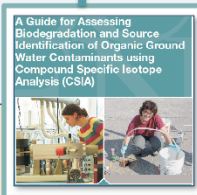
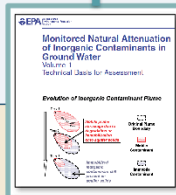
- Move away from scoring systems following National Research Council review of early protocols
- Better understanding of attenuation processes
- More contaminants, settings
- Incorporate new generation of tools



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### FAQ 3: *What are the most important new MNA developments?*

| Year      | New Contaminant | New Measurement   | New Process  | New Tools   |
|-----------|-----------------|---|--|---|
| 2000-2005 | MTBE-TBA        | Two types of rates  | Source attenuation of hydrocarbon sites  | BIOChlor<br>MAROS<br>NAS<br>SourceDK  |
| 2005-2010 | Metals-Rads     | Compound-Specific Isotopes<br>Molecular<br>Biological Tools | Biogeochemical/abiotic trans. of chlor. solvents<br>Matrix diffusion<br>Oxidation of chlor. solvents at low DO | REMChlor<br>Mass flux toolkit<br>BIOBALANCE<br>Scenarios for chlor. solvents<br>MNA<br>Sustainability |

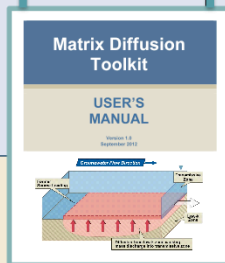


Probably the most important "recent" development?



### FAQ 3: *What are the most important new MNA developments?*

| Year         | New Contaminant         | New Measurement                | New Process   | New Tools   |
|--------------|-------------------------|--------------------------------|---|---|
| 2010-present | "Emerging Contaminants" | CO <sub>2</sub> traps for NSZD | Natural source zone depletion (NSZD)<br>Source attenuation of chlorinated solvent sites<br>Attenuation in low-k zones | PREMChlor<br>Matrix Diffusion Toolkit<br>Scenarios for metals/rads<br>Source History Tool |



**Coming Attractions:**  
ESTCP ER-201129

*Development and Validation of a Quantitative Framework and Management Expectation Tool for the Selection of Bioremediation Approaches (MNA, Biostimulation and/or Bioaugmentation) at Chlorinated Solvent Sites*

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## MNA THEMES

### *MNA AS AN EVOLVING TECHNOLOGY:*

4. What is the philosophy behind MNA?

5. What evidence is needed for MNA?

6. MNA lines of evidence—how have they changed?

7. MNA for the source zone—how does this work?

### *NEW TECHNIQUES, NEAT TOOLS:*

### *EMERGING ISSUES FOR EVALUTING MNA AS A REMEDY:*

### *IMPLEMENTATION AND SITE CLOSURE:*

## What is the philosophy behind MNA?

**Nature can help!**

*It is harder and more expensive to clean these sites up than first thought.*

*Nature is amazing and seems to be degrading or sequestering some of these chemicals.*

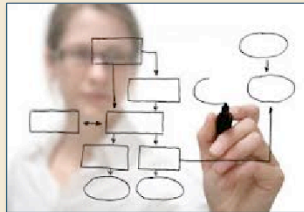
*Let's let nature do the job.*

*But you have to do three things:*

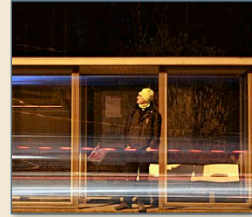
**Protect**



**Understand**



**Watch**



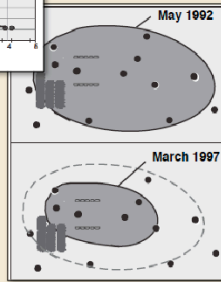
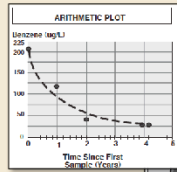
## FAQ 5: *What evidence is needed for MNA?*

### Typically Multiple Lines of Evidence

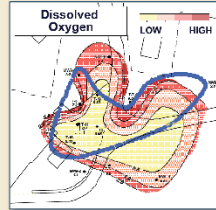
| <i>Historical (based on USEPA, 1999)</i>                                       | <i>WHY</i>                                   | <i>WHAT</i>  | <i>WHEN</i>                            |
|--|--|--|--|
| <b>LOE 1:</b> Historical contaminant mass reduction - monitoring data vs. time | "I shrink, therefore I am"                   | Direct method to demonstrate decreasing trend              | Always                                 |
| <b>LOE 2:</b> Hydrogeologic or geochemical data                                | Need to know more than just it is decreasing | Indirect method to demonstrate degradation process or rate | Most of the time                       |
| <b>LOE 3:</b> Field or microcosm studies                                       | Need to know for than LOE 1 or 2             | Direct method to demonstrate particular knowledge          | Rarely; used to prove specific process |

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## FAQ 5: *What evidence is needed for MNA?* *New Trends in LOEs*



**LOE 1: Historical  
contaminant mass  
reduction**



**LOE 2: Hydrogeologic  
or geochemical data**



**LOE 3:  
Microcosm or  
Field data**

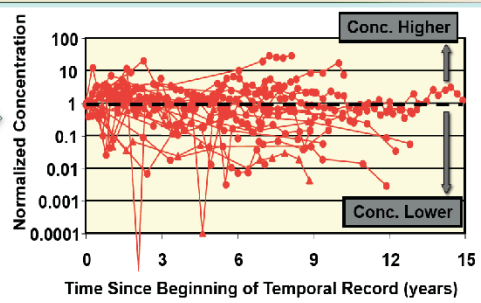
LOE: "Lines of Evidence"

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## FAQ 7: *MNA for the source zone too?*

**An emerging consensus: YES**

1. Long-term monitoring data showing source attenuation
2. Source attenuation modeling tools (BIOChlor, REMChlor, NAS)



Long-term temporal records from 22 monitoring wells at 13 Untreated TCE Sites (Newell et al., 2006)



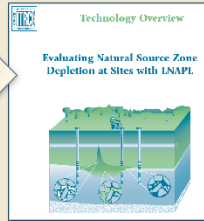
2004 survey of 191 plumes - MNA was the sole remedy (no active source remediation) at 21% of sites with CVOC concentrations > 10 mg/L (McGuire et al., 2004)

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## FAQ 7: *MNA for the source zone too?*

An emerging consensus: YES

3. Source attenuation protocols
4. Field studies demonstrating Source Zone Natural Attenuation (SZNA)



ITRC, 2009

*"[Natural source zone depletion (NSZD)] is of significance because engineered remedial actions typical do not always completely remediate soils and NSZD may be useful to address the residual hydrocarbon"*

Methodology developed at Arizona St. Univ.

| Site                    | Contaminant         | Hydrogeology               | Loss Rate from Dissolution (kg/year) | Loss Rate from Dissolved Transport Related Biodegradation (kg/year) | Loss Rate Estimate Due to Vadose Transport* (kg/year) |
|-------------------------|---------------------|----------------------------|--------------------------------------|---|---|
| Guadalupe Diluent Tanks | Hydrocarbon         | Sand dune                  | 500 - 1600                           | 600 - 1600  | 140,000 (O <sub>2</sub> Flux)                         |
| Guadalupe Compressor    | Hydrocarbon         | Sand dune                  | 300 - 500                            | 0   | 16,000 (O <sub>2</sub> Flux)                          |
| NAS Jacksonville        | Chlorinated Solvent | Silty sand, clay           | 12 - 32                              | -   | 0.27 - 0.79 (Vapor Flux)                              |
| Parris Island           | Chlorinated Solvent | Silty clay, sand stringers | 1.8 - 5                              | -   | 0.22 - 0.68 (Vapor Flux)                              |

Selected results from NSZA studies at several sites (Ekre et al., ESTCP ER-0705 Short Course, 2011)

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## MNA THEMES

### *MNA AS AN EVOLVING TECHNOLOGY:*

8. Can I apply MNA to metals, inorganics, and radionuclides?  
*THE BASIS FOR MNA:*

9. I can apply MNA to BTEX, but how about oxygenates?  
*NEW CONTAMINANTS FOR THE MNA LINEUP:*

10. Which emerging contaminants are MNA candidates?

### *NEW TECHNIQUES, NEAT TOOLS:*

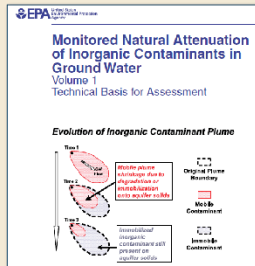
### *EMERGING ISSUES FOR EVALUATING MNA AS A REMEDY:*

### *IMPLEMENTATION AND SITE CLOSURE:*

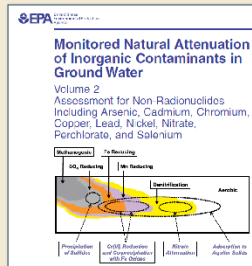


## FAQ 8: Can I apply MNA to metals, inorganics, and radionuclides?

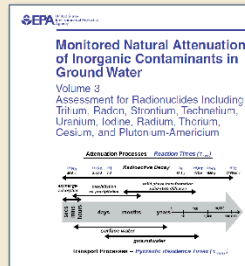
**YES, says USEPA**



2007



2008



2010

**Tiered Lines-of-Evidence Approach**  
(similar to protocols for organics)

1. Plume is not expanding and sorption is occurring
2. ID the attenuation mechanism and estimate rate
3. Determine capacity and sustainability
4. Develop monitoring and contingency measures

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## FAQ 8: *Can I apply MNA to metals, inorganics, and radionuclides?*

### KEY POINTS:

- Primary attenuation pathway for many inorganics is transformation to less mobile forms through co-precipitation or sorption
- Reactions are generally more complex and highly influenced by geochemical conditions

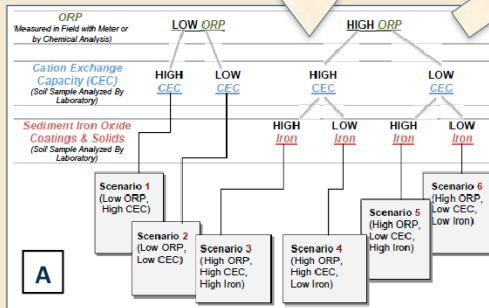
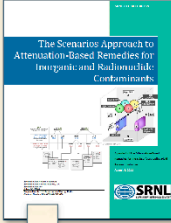
| Contaminant  | Biological Reaction                 |                                       | Abiotic Reaction                    | Sequestration                    |
|--|-------------------------------------|---------------------------------------|-------------------------------------|----------------------------------|
|  | Anaerobic                           | Aerobic                               |                                     |                                  |
| Nitrate  | Yes, degradation                    | No                                    | Yes (reactive iron)                 | No                               |
| Perchlorate  | Yes, degradation                    | No                                    | Conflicting Data                    | No                               |
| Chromium (Cr), Selenium (Se), Copper (Cu), Cadmium (Ca), Lead (Pb), Nickel (Ni), Zinc (Zn), Beryllium (Be), Arsenic (As) (metalloid) | Valence change, generally favorable | Valence change, generally unfavorable | Valence change, generally favorable | Yes (sorption, co-precipitation) |
| Uranium, Technetium, Strontium, Cesium, Radium, Iodine   | Valence change, generally favorable | Valence change, generally unfavorable | Valence change, generally favorable | Yes (sorption, co-precipitation) |

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# FAQ 8: Can I apply MNA to metals, inorganics, and radionuclides?

**ADDITIONAL GUIDANCE:**

**"SCENARIOS"  
DOE, 2011**



| Contaminant   | Scenarios                          |                                   |   |  |  |   |
|---|------------------------------------|-----------------------------------|---|--|--|---|
|   | Scenario 1<br>Low ORP,<br>High CEC | Scenario 2<br>Low ORP,<br>Low CEC | Scenario 3<br>High ORP,<br>High CEC,<br>High Iron | Scenario 4<br>High ORP,<br>High CEC,<br>Low Iron | Scenario 5<br>High ORP,<br>Low CEC,<br>High Iron | Scenario 6<br>High ORP,<br>Low CEC,<br>Low Iron |
| Cr(III)   | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| Cr(VI)  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| <sup>99</sup> Tc(V)                                 | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| <sup>99</sup> Tc(VII)                               | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| Pu  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| U   | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| Cd, Cu, Pb, Zn                                      | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| Ni  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| As  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| Se  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| <sup>90</sup> Sr, Cs <sup>2</sup> , Ra <sup>3</sup> | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| NO <sub>3</sub> , ClO <sub>4</sub>                  | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |
| •••   | LOW                                | LOW                               | HIGH  | HIGH   | HIGH   | HIGH  |

|                 |  |                                   |
|-----------------|--|-----------------------------------|
| HIGH Mobility   | Mobility increases above and below pH=7                | Increasing pH decreases mobility  |
| MEDIUM Mobility | Mobility increases above pH=7                          | Increasing TDS increases mobility |
| LOW Mobility    | Mobility decreases above pH=7 and increases below pH=7 | Transforms to other valence state |

## FAQ 9: *I can apply MNA to BTEX, but how about oxygenates?*

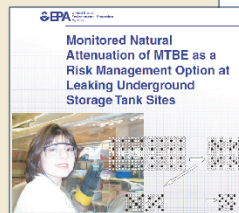
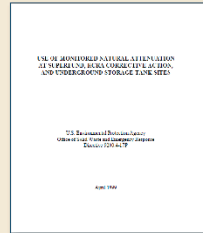
### Originally No, But Now YES

- Not promising in early protocols

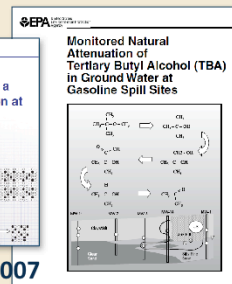
MTBE had been found to “...migrate large distances and threaten downgradient water supplies at the same sites where the BTEX component of a plume has either stabilized or diminished due to natural attenuation” and included MTBE among compounds “...that tend not to degrade readily in the subsurface”.

- Lots of research and field work in the following 5-10 years, and we ended up with a completely different story!

### USEPA MNA Directive, 1999



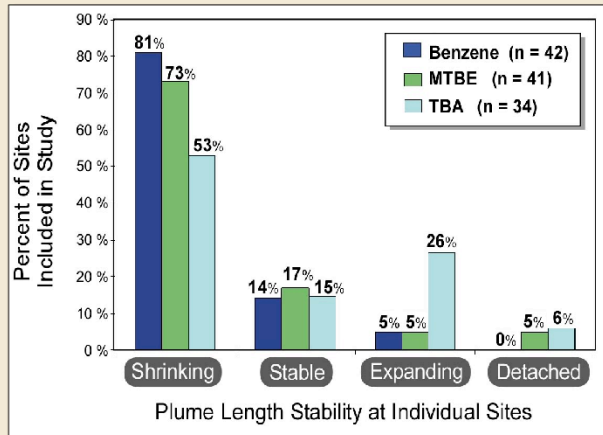
USEPA, 2005 & 2007



## FAQ 9: *I can apply MNA to BTEX, but how about oxygenates?*

*What do the data say?*

Several plume studies have documented that majority of MTBE/TBA plumes are relatively short and/or attenuating



### Kamath et al., 2012

- MTBE exhibits similar characteristics to BTEX in terms of median plume size (142 ft) and attenuation rate ( $-0.63 \text{ yr}^{-1}$ )
- TBA also similar to BTEX but fewer are stable/shrinking

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## FAQ 9: *I can apply MNA to BTEX but how about oxygenates?*

| Contaminant Class | Biological Degradation   |   |
|-------------------|--|---|
|                   | Anaerobic  | Aerobic   |
| MTBE              | <b>YES</b> (may require acclimation period)  | <b>YES</b>  |
| TBA               | <b>YES</b> if electron acceptors are readily available (may be limited in methanogenic conditions) | <b>YES</b><br>(generally faster than in anaerobic conditions) |
| Ethanol           | <b>YES</b> (tends to be preferentially degraded over BTEX)   | <b>YES</b> (tends to be preferentially degraded over BTEX)    |

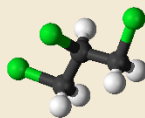
## FAQ 10: Which emerging contaminants are MNA candidates?

### Dioxane, TCP, NDMA, Phthalates, and Maybe Others?

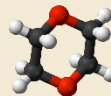
- DoD general goal for emerging contaminants:

*"Identify chemicals or materials that either lack human health standards or have an evolving science and regulatory status."*

- Other problems
  - Prevalence at individual sites is largely unknown
  - Absence of well-established treatment technologies
  - Absence of tools for establishing MNA (e.g., CSIA, MBTs)



1,2,3-  
Trichloropropane



1,4-Dioxane

## FAQ 10: Which emerging contaminants are MNA candidates?

| Emerging Contaminant          | Biological Degradation                         |   | Abiotic Degradation   | Sequestration  |
|-------------------------------|--|---|---|--|
|                               | Anaerobic                                      | Aerobic   |   |  |
| 1,4-Dioxane                   | Limited  | <b>YES</b><br>(mostly lab studies; can be cometabolic or used as a carbon source) | Not documented  | <b>No</b><br>(poor sorption)   |
| Perfluorinated Compounds      | Very limited (incomplete pathway)              | Very limited (incomplete pathway)   | Limited (a reliable light +Fe(III) reaction has been established) | Moderate (primarily electrostatic sorption to ferric iron minerals; limited organic carbon sorption) |
| N-Nitrosodimethylamine (NDMA) | <b>YES</b>                                     | <b>YES</b><br>(cometabolic)   | <b>No</b> (several ex situ methods, including UV photolysis)      | <b>No</b> (poor sorption)  |
| 1,2,3-Trichloropropane        | <b>YES</b><br>(slow, often incomplete pathway) | <b>YES</b><br>(slow, incomplete pathway)  | Very limited (reactive iron, base hydrolysis)                     | Limited (moderate sorption)  |

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## MNA THEMES

### MNA AS AN EVOLVING TECHNOLOGY:

11. Can isotopes prove contaminants are being destroyed?

### THE BASIS FOR MNA:

12. How can molecular biological tools help me with MNA?

13. How can you show attenuation that occurred before the start of your monitoring program?

### NEW CONTAMINANTS FOR THE MNA LINEUP:

14. Can DO measurements be a problem for MNA studies?

15. What are CO<sub>2</sub> traps and how do they help me show attenuation?

### NEW TECHNIQUES, NEAT TOOLS:

16. How do I estimate rates and timeframes for MNA?

17. What is required for MNA monitoring?

### EMERGING ISSUES FOR EVALUATING MNA AS A REMEDY:

18. What is the new thinking about monitoring frequency?

19. Statistics for 2 questions: how far and how long?

### IMPLEMENTATION AND SITE CLOSURE:

20. Which computer models work best for MNA?

21. Are MNA reactions sustainable? Is MNA sustainable?

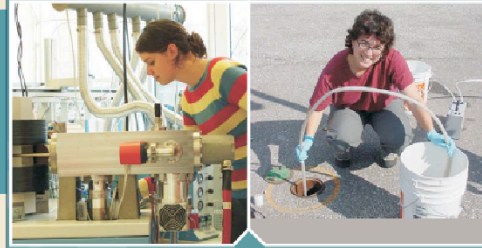
## FAQ 11: *Can isotopes prove contaminants are being destroyed?*

*Yes, and more*

**A Guide for Assessing Biodegradation and Source Identification of Organic Ground Water Contaminants using Compound Specific Isotope Analysis (CSIA)**

*“Stable isotope analyses can provide unequivocal documentation that biodegradation or abiotic transformation processes actually destroyed the contaminant.”*

USEPA, 2008



## FAQ 11: *Can isotopes prove contaminants are being destroyed?*

### KEY PRINCIPLE:

**Isotopic Fractionation** =

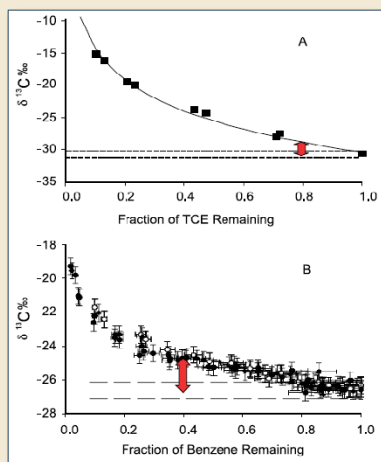
Lighter isotopes are used preferentially during degradation, such that the remaining non-degraded compound becomes enriched in the heavier isotope

carbon ( $^{13}\text{C}/^{12}\text{C}$ )  
oxygen ( $^{18}\text{O}/^{16}\text{O}$ )  
nitrogen ( $^{15}\text{N}/^{14}\text{N}$ )  
chlorine ( $^{37}\text{Cl}/^{35}\text{Cl}$ )  
hydrogen ( $^2\text{H}/^1\text{H}$ )

### KEY BENEFITS (continued next page):

- Demonstrating that parent compound is being degraded
- Estimating the extent of degradation

$\delta^{13}\text{C}$  data from several studies as compiled in Hunkeler et al., USEPA, 2008

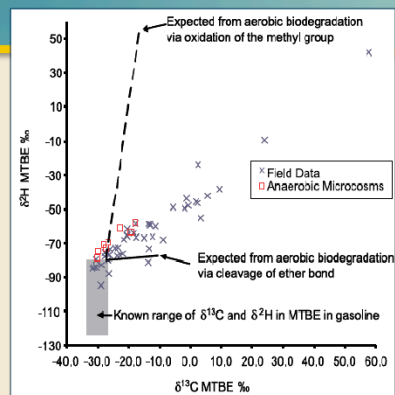
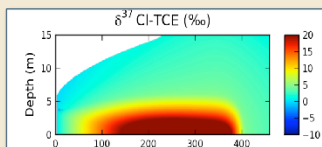


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## FAQ 11: *Can isotopes prove contaminants are being destroyed?*

### KEY BENEFITS (continued):

- Differentiating between destructive and non-destructive pathways
- Differentiating between various destructive pathways
- Demonstrating that complete degradation has occurred
- Estimating rate of degradation
- Source identification and differentiation
- Can be incorporated into reactive transport modeling



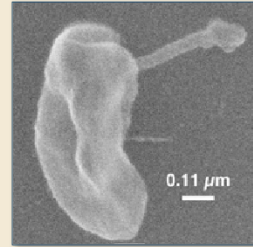
2-D plot of  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$  in MTBE in groundwater to associate natural biodegradation of MTBE in groundwater with an anaerobic process (from Hunkeler et al., USEPA, 2008)

Example of CSIA-based reactive transport modeling from ESTCP ER-201029 (see Van Brueken et al., 2008 for method description)

## FAQ 12: *How can molecular biological tools help me with MNA?*

*MBTs provide strong, but not definitive evidence of MNA*

1. Show that key organisms are present (e.g., *Dehalococcoides*, *Dehalobacter*)
2. Show that key enzymes are present (e.g., *vcrA*, oxygenase-encoding genes)
3. Establish relative abundance of key microbial populations



*Our friend,  
Dehalococcoides  
(Apkarian and Taylor)*

**KEY ISSUE:** Most tests focus on presence, not activity!

## FAQ 12: *How can molecular biological tools help me with MNA?*

### MOST POPULAR?

Evaluating chlorinated solvent degradation using **PCR-based** methods for tracking *Dehalococcoides (Dhc)*

| Tools  | MNA Application  | MNA Limitations  |
|--|--|--|
| <ul style="list-style-type: none"><li>• PCR / qPCR</li></ul> | <ul style="list-style-type: none"><li>• Identify if key organisms / enzymes</li><li>• Determine if abundance of key biomarkers is increasing</li></ul> | <ul style="list-style-type: none"><li>• Many techniques cannot differentiate between live and inactive cells</li><li>• Attempts to correlate in situ activity and gene expression still in infancy</li><li>• Target mostly well-known pathways (others in development)</li></ul> |

### **Others:**


*Stable Isotope Probing (SIP), microbial fingerprinting, microarrays, enzyme activity probes*

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## FAQ 12: *How can molecular biological tools help me with MNA?*

### HOW TO COLLECT AND USE THE DATA?

- **Groundwater or Soil using established procedures**
  - starting at about \$200 per sample/target)
- **Quantitative Rules for MNA.**
  - Specific recommendations for MNA
  - Lu et al., 2006: “generally useful” attenuation rates of *cis*-1,2-DCE and VC (> 0.3/yr) were associated with sites where *Dhc* was detected, while no attenuation was observed at sites where it was absent



**GUIDANCE PROTOCOL**

Environmental Restoration Project ER-0518

Application of Nucleic Acid Based Tools for Monitoring Monitored Natural Attenuation (MNA), Biostimulation, and Bioaugmentation at Chlorinated Solvent Sites

January 2011

- *Dhc* at  $10^4$  to  $10^6$  gene copies/L can support MNA
- *Dhc* at  $> 10^6$  gene copies/L is the target threshold for ensuring ethene production

## INTERVIEW: *John Wilson*

1. What are the most important new developments you've seen since the original protocols were developed in the 1990s?
2. How did you get involved in isotopes? What is the key message in the the document you wrote with Hunkeler and others in 2008?



## INTERVIEW: *John Wilson (continued)*

3. You just completed a Short Course at Battelle on calculating and applying rates for MNA processes. What is the key message from your short course?
4. You got involved in MNA statistics with your 2011 document: “An Approach for Evaluating the Progress of MNA in Groundwater”. Why did you write this one? Are you a pretty good statistician? Any thoughts on sampling frequency?
5. What is the big question to answer for MNA?

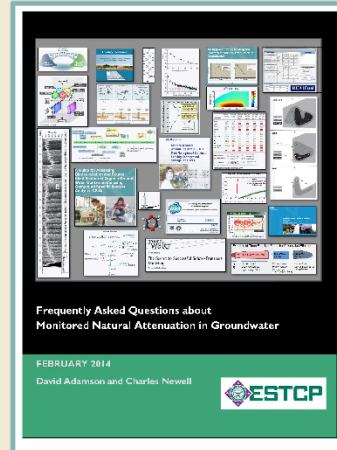
## BREAK FOR INITIAL Q&A

**Go to:**

[http://serdp-estcp.org/Program-Areas/  
Environmental-  
Restoration/Contaminated-Groundwater/  
Persistent-Contamination/ER-201211/](http://serdp-estcp.org/Program-Areas/Environmental-Restoration/Contaminated-Groundwater/Persistent-Contamination/ER-201211/)

**Or Google:**

**"MNA" + "FAQ" + "ESTCP"**



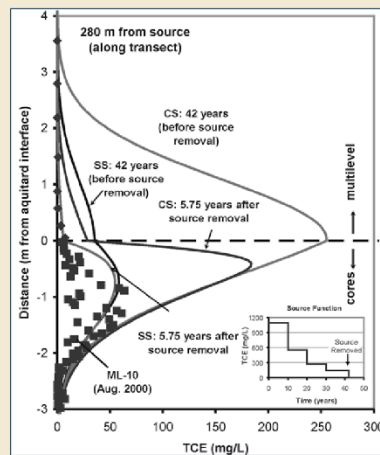
## FAQ 13: *How can you show attenuation that occurred before the start of monitoring?*

*By Using Clay Diffusion Profiles or Tree Rings*

**“Source History” using soil data from Low Permeability Zone (e.g., clay)**

- Combo of high-res characterization and inverse modeling
- Core acts like “tree ring” to provide information on the source concentration over time in adjacent aquifer

New FREE software tool developed for applying this method: <http://www.serdp.org/Program-Areas/Environmental-Restoration/Contaminated-Groundwater/Persistent-Contamination/ER-201032>



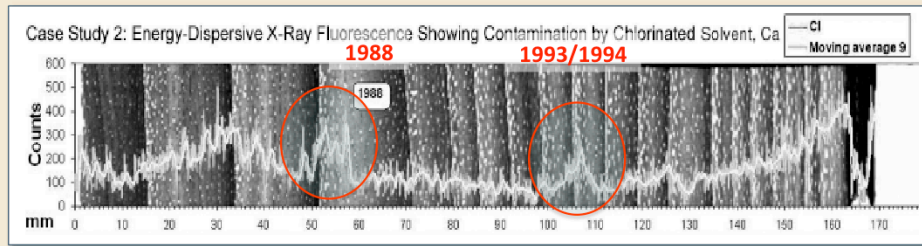
Example of “Source History” estimate from Chapman and Parker, 2005 – best fit with Stepped Source (SS) (see inset)

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## FAQ 13: *How can you show attenuation that occurred before the start of monitoring?*

ALTERNATIVE: Use actual tree rings as part of a “phytoforensic” approach

- Assay cores for **Cl** for chlorinated solvents, **S** for petroleum hydrocarbons
- Applied at > 20 sites



Cl<sup>-</sup> patterns (shown on y-axis as x-ray fluorescence counts) over time in tree core (shown on x-axis in mm of core). Pattern identifies potential exposure events (releases) in 1988 and 1993/1994, along with continuing impact at date when core was collected (far right hand side)

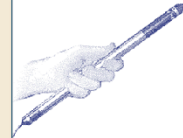
(from Balouet et al., 2007)

## FAQ 14: *Can DO measurements be a problem for MNA studies?*

**YES, if not carefully done**

**Some early conclusions of ESTCP 2011-29 project:**

- *Great care should be taken to minimize oxygen introduction by measuring at well head using a flow-through cell*
- *Erroneous DO measurements are often recorded and reported which can cause misinterpretation of subsurface conditions*
- *Measured and actual DO concentrations rarely coincide*



## FAQ 15: *What are CO<sub>2</sub> traps and how do they help me show attenuation?*

### *A new tool for measuring natural source zone attenuation at LNAPL sites*

- **What's the principle?**
  - LNAPL is subject to degradation following release, with essentially all being converted to CO<sub>2</sub>
- **How do they work?**
  - Collect CO<sub>2</sub> that has migrated to surface using a passive adsorption device installed at grade
  - Correct for background/non-fossil fuel-related CO<sub>2</sub>
  - **Convert CO<sub>2</sub> to LNAPL attenuation rate (gallons/acre/yr)**



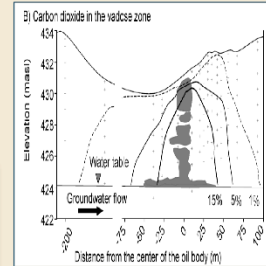
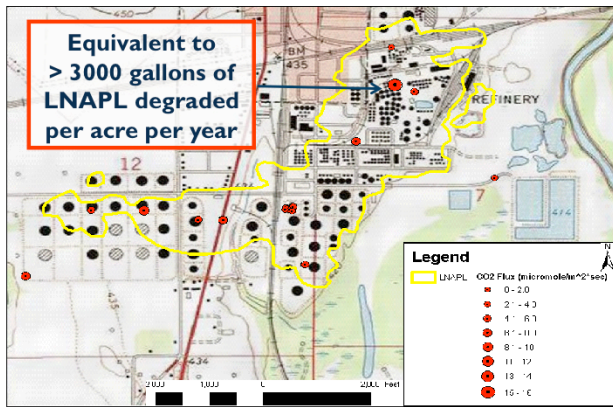
*Example of Colorado State University CO<sub>2</sub> trap installation; other approaches have been developed by Ariz. St. Univ. and Univ. of Brit. Columbia*

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## FAQ 15: *What are CO<sub>2</sub> traps and how do they help me show attenuation?*

*What do the data say?*

Traps have been deployed at multiple sites and have documented large CO<sub>2</sub> fluxes equivalent to 100s to 1000s of gallons of LNAPL per acre per year



*Shotra et al., 2011*

*Example of field site where CO<sub>2</sub> traps were used to delineate LNAPL natural loss rates.*

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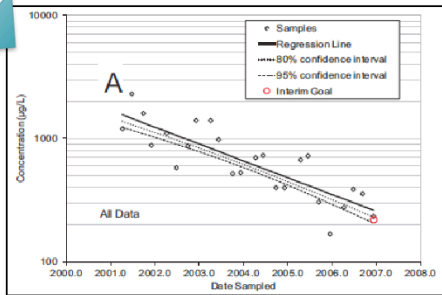
## BREAK FOR POLL QUESTIONS



## FAQ 16: *How do I estimate rates and timeframes for MNA?*

*By extrapolating monitoring well data and the right rate...*

| Rate Constant  | Method of Analysis | Significance   | Use of Rate Constant |               |                 |
|--|--------------------|--|----------------------|---------------|-----------------|
|  |                    |  | Plume Attenuation    | Plume Trends? | Plume Duration? |
| Point Attenuation Rate (Fig. 1)<br>( $k_{\text{point}}$ , time per year) | C vs. T Plot       | Reduction in contaminant concentration over time at a single point | NO*                  | NO*           | YES             |
| Bulk Attenuation Rate (Fig. 2)<br>( $k$ , time per year)                 | C vs. T Plot       | Reduction in dissolved contaminant concentration with...           | YES                  | NO*           | NO              |



from Wilson, USEPA, 2011

Example of extrapolation method to estimate MNA remediation timeframes with confidence limits.<sup>12</sup>

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## FAQ 16: *How do I estimate rates and timeframes for MNA?*

*By using mass balance methods in simple models...*

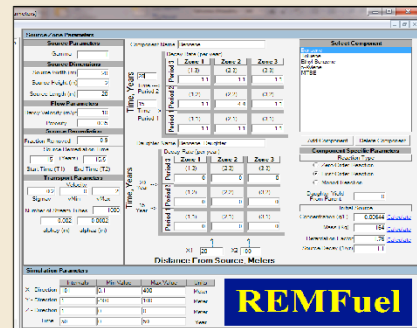
Simple mass balance model for first order decay in source concentration

$$k_{\text{source}} \text{ (per year)} = \frac{\text{Mass discharge (kg/yr)}}{\text{Mass (kg)}}$$

REMChlor Model  
(Falta et al., 2007)

and

NAS Software  
(Widdowson et al, 2005;  
Chapelle et al., 2007)

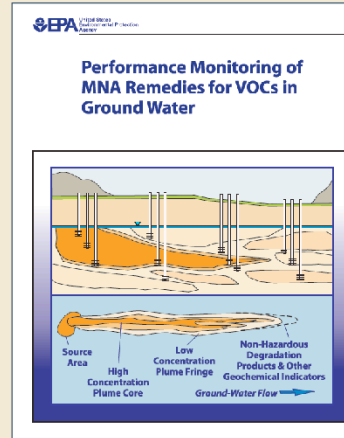


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## FAQ 17: *What is required for MNA monitoring?*

### Follow and understand USEPA Guidance

| From USEPA Performance Monitoring of MNA                                   | Our Comments  |
|--|---|
| Sample both contaminated and uncontaminated areas.                         | Source zones can change the geochemical conditions  |
| Monitor areas supporting site hydrogeology.                                | Use piezometers to better understand groundwater flow   |
| Match screen length to stratigraphic unit or contaminant loading interval. | it is critical to get flow-weighted concentrations from transmissive zones that would be utilized by a receptor. This may mean long well screens. |
| Be aware of changing groundwater flow directions.                          | See graphics in USEPA 2004 doc  |



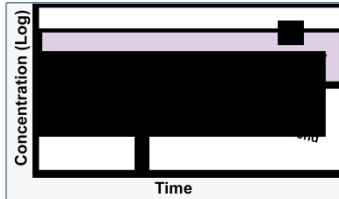
USEPA, 2004

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## FAQ 18: *What is the new thinking about monitoring frequency?*

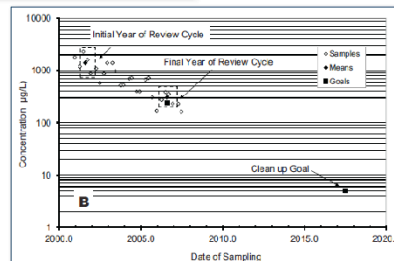
### **KEY POINT:**

Groundwater monitoring data can be highly variable, so it may take many years to establish attenuation trends



*SERDP ER-1705  
Monitoring Variability  
Study: Short-term  
variability masks  
long-term trend*

*Wilson, USEPA  
Guidance, 2011:  
Focus on longer  
review cycles to  
answer these  
questions*



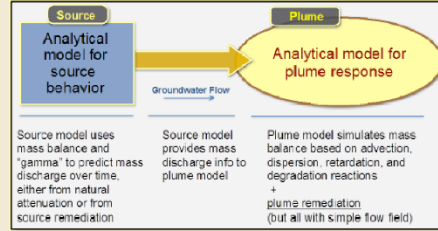
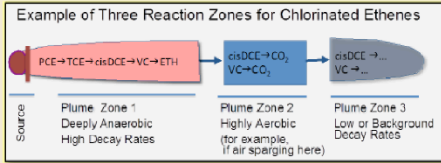
**FAQ 19: Statistics for two questions:  
How far and how long?**



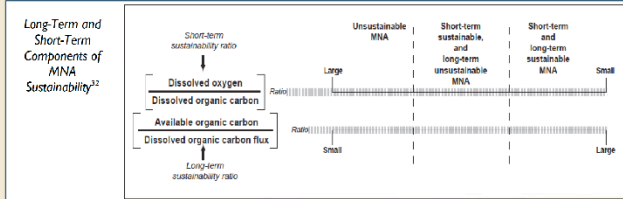
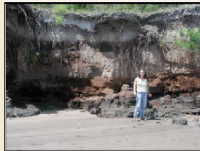
|                        |       | Confidence Factor |                  |  |
|------------------------|-------|-------------------|------------------|--|
|                        |       | CF > 95%          | 90% < CF < 95%   | CF < 90%   |
| Mann-Kendall Statistic | S < 0 | Decreasing        | Prob. Decreasing | If COV < 1,<br>Stable<br><br>If COV > 1,<br>No Trend |
|                        | S > 0 | Increasing        | Prob. Increasing | No Trend   |

# FAQ 20, 21: Which computer models work best for MNA? Are MNA Reactions Sustainable?

**REMChlor and REMFuel**  
(Falta et al., 2007; USEPA, 2012)



**NAS Software**  
(Widdowson et al, 2005;  
Chapelle et al., 2007)



## MNA THEMES

### *MNA AS AN EVOLVING TECHNOLOGY:*

22. What is matrix diffusion? How does it affect MNA?

*THE BASIS FOR MNA:*

23. Do contaminants in low-permeability units attenuate?

24. Why are interfaces important for MNA?

*NEW CONTAMINANTS FOR THE MNA LINEUP:*

25. How do reactive mineral species contribute to attenuation?

*NEW TECHNIQUES, NEAT TOOLS:*

26. What is a low-risk site? How is MNA involved?

### *EMERGING ISSUES FOR EVALUATING MNA AS A REMEDY:*

### *IMPLEMENTATION AND SITE CLOSURE:*

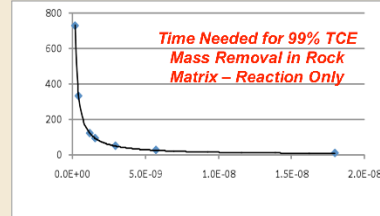




## FAQ 23: *Do contaminants in low-permeability units attenuate?*

### Sometimes, we think

- Some evidence based on lab/field studies, but simply not investigated much yet
- Even slow attenuation could be significant



Impact of Abiotic Degradation Rates in Rock Matrix on Mass Reduction Timeframes (data described in Schaefer et al., 2013)

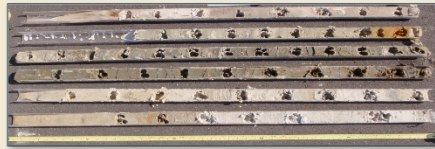
| Characteristics That Favor Attenuation   | Characteristics That Hinder Attenuation  |
|--|--|
| Long retention times (little advection/flushing)   | Smaller pore throat size restricts migration of microbes, influx of nutrients/carbon sources, and growth density |
| Reducing conditions are common; favorable for biological and biogeochemical reductive dechlorination | Salinity can be high and may limit microbial activity  |
| Potentially large reservoir of organic carbon (silts/organo-clays)                                   | Limited bioavailability of organic carbon  |
| Potentially large reservoir of reactive mineral species  | Reactivity of mineral species may be limited due to dependence on microbial activity (e.g., iron reduction)      |

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## FAQ 23: *Do contaminants in low-permeability units attenuate?*

### Potential lines of evidence for low k zone attenuation to support MNA

- Molecular biological data confirming presence and/or activity of degraders within low k zone
- Daughter product distribution suggests greater extent of degradation in low k zone relative to adjacent transmissive zone
- Favorable geochemical conditions within low k zone
- CSIA data showing higher fractionation within low k zone
- Mineralogical analysis of low k zone soil samples that show minerals capable of abiotic degradation of contaminants

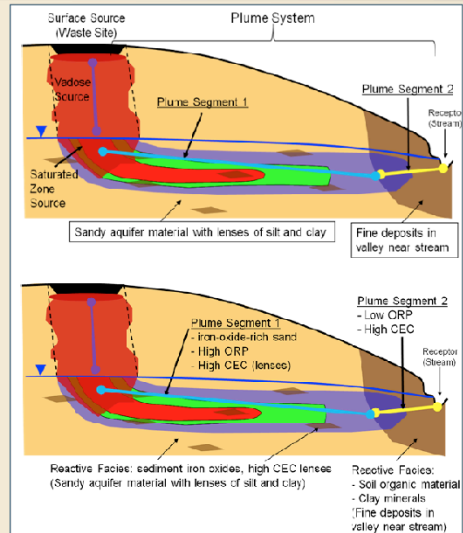


Example of high-resolution sub-sampling of soil cores for assessing attenuation in low-k zones

## FAQ 24: *Why are interfaces important for MNA?*

*Interfaces are where the action is at many sites!*

- Interfaces between groundwater and surface water – hyporheic zone
- Transitions between aerobic zones and more reducing conditions
- Biogeochemical gradients at the interface between different geologic conditions
- Interface between vadose zone and saturated zone



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## FAQ 24: *Why are interfaces important for MNA?*

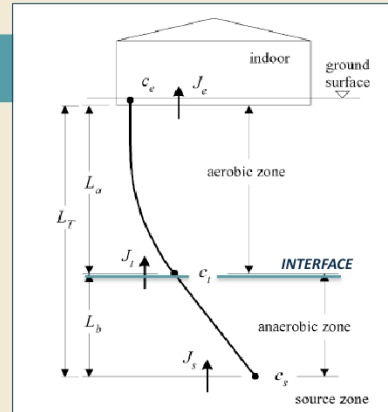
**KEY POINT:** Big changes that occur across short distances can drive attenuation

- **Hyporheic zone example (USGS study by Landmeyer et al., 2010)**

*"MTBE, TBA, and TAME concentrations in groundwater discharge in a 5-foot (1.5-m) thick section of the hyporheic zone were attenuated between 34% and 95%, in contrast to immeasurable attenuation in the shallow aquifer during contaminant transport between 0.1 and 1.5 miles (0.1 to 2.4 km)."*

- **Vadose zone example (Kurt, Shin, & Spain, 2012)**

*Thin (2-3 mm) anaerobic/aerobic interface was found to have a "remarkable capacity" to degrade chlorobenzene (2000 to 4200 milligrams per meter squared per day) and nitrobenzene.*



**Conceptual model for vadose zone degradation (from Devaull, 2007)**

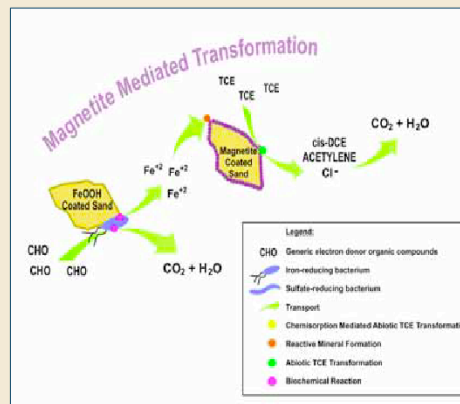
## FAQ 25: *How do reactive mineral species contribute to attenuation?*

**Naturally-occurring minerals can degrade contaminants**

- CVOC degradation that is abiotically-mediated by a number of reactive mineral species

Iron(II) Sulfide (FeS)  
 Mackinawite ( $-(Fe_{1+x}S)$ )  
 Pyrite (FeS<sub>2</sub>)  
 Magnetite (Fe<sub>3</sub>O<sub>4</sub>)  
 Goethite ( $\alpha$ -FeO(OH))  
 Hematite (Fe<sub>2</sub>O<sub>3</sub>)  
 Lepidocrocite ( $\gamma$ -FeO(OH))  
 Green Rust--( $Fe^{2+}$  and  $Fe^{3+}$  cations,  $O^{2-}$  and  $OH^-$  anions, with loosely bound  $[CO_3]^{2-}$  groups and  $H_2O$  molecules between the layers)

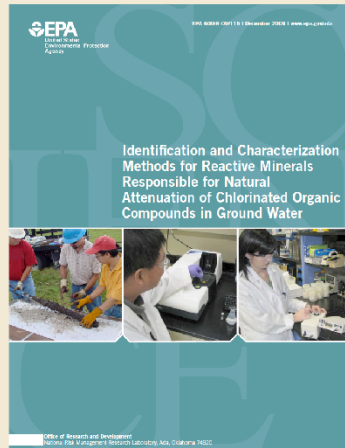
- Basis for ZVI and other PRB designs, but significant evidence of natural attenuation in anaerobic environments



Example of abiotic TCE degradation by magnetite (from ESTCP/AFCEE/NAVFAC, 2007)

## FAQ 25: *How do reactive mineral species contribute to attenuation?*

- While contaminants are degraded abiotically, formation of reactive minerals typically have a biological component
  - e.g., biological iron reduction and/or sulfate reduction to produce iron sulfide
- Methods for assessing abiotic degradation capacity are available and/or being developed
  - E.g., magnetite in sediments via magnetic susceptibility testing
- Current research suggests slow but sustainable attenuation rates



EPA, 2009 –  
detailed descriptions of  
important methods

## FAQ 26: *What is a low-risk site? How is MNA involved?*

*Low-Risk means MNA the rest of the way*

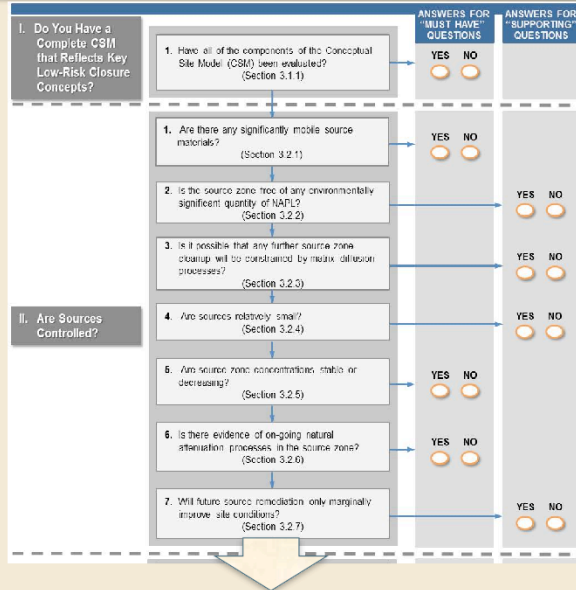
- Recognition that at many sites complete closure is difficult and/or unattainable
- Concentrations may be very low and pose no significant risk
- Being adopted as part of state regulatory programs (e.g., California)



Low-Risk Site Closure Guide for Air Force (Farhat et al., 2012)

## FAQ 26: *What is a low-risk site? How is MNA involved?*

**Air Force LoRSC Manual Methodology to Determine Low Risk Sites – Parts I and II**





## MNA THEMES

### *MNA AS AN EVOLVING TECHNOLOGY:*

**27. Can MNA be a stand-alone remedy? When should you transition a site to MNA?**

**28. How can stakeholders considering MNA make better decisions?**

### *NEW TECHNIQUES, NEAT TOOLS:*

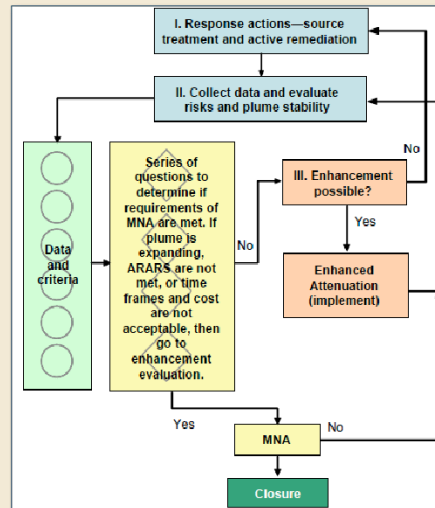
### *EMERGING ISSUES FOR EVALUTING MNA AS A REMEDY:*

### *IMPLEMENTATION AND SITE CLOSURE:*

## FAQ 27: *Can MNA be a stand-alone remedy?* *When should you transition to MNA?*

### *It depends*

- **1999 EPA Directive:**  
MNA should not be considered a default or presumptive remedy, and that it should be applied *“very cautiously as the sole remedy”* and that *“source control will be fundamental components of any MNA remedy.”*
- **Lots of data suggesting MNA being used extensively**
  - Sole remedy
  - Some source treatment with MNA as sole groundwater remedy
- **Many states have specific risk-based criteria for MNA as default remedy (e.g., Florida)**



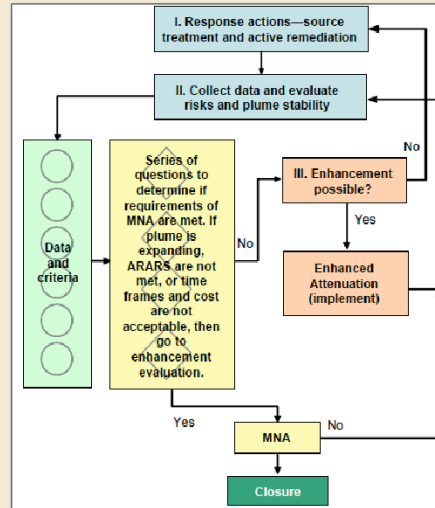
## FAQ 27: *Can MNA be a stand-alone remedy? When should you transition to MNA?*

### MNA Transition Flowchart, ITRC 2008

*Are the risks acceptable?  
Is the plume stable or shrinking?  
Are conditions sustainable?  
Is the remediation timeframe acceptable?  
Are the cost-benefits acceptable?*



**Enhanced Attenuation  
(instead of MNA)**

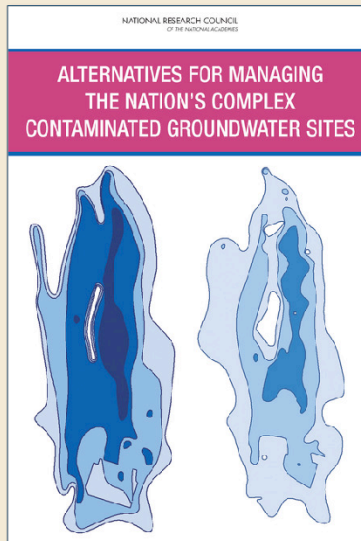


## FAQ 27: *Can MNA be a stand-alone remedy? When should you transition to MNA?*

### NRC "Complex Sites" Report, 2012

#### Transition Assessment Concept

*At many complex sites, contaminant concentrations in the plume remain stalled at levels above cleanup goals despite continued operation of remedial systems. There is no clear path forward to a final end state embodied in the current cleanup programs, such that money continues to be spent, with no concomitant reduction in risks. **If the effectiveness of site remediation reaches a point of diminishing returns prior to reaching cleanup goals and optimization has been exhausted, the transition to monitored natural attenuation or some other active or passive management should be considered using a formal evaluation.** This transition assessment would determine whether a new remedy is warranted at the site or whether long-term management is appropriate.*



## FAQ 28: *How can stakeholders considering MNA make better decisions?*

### Use SMART Objectives

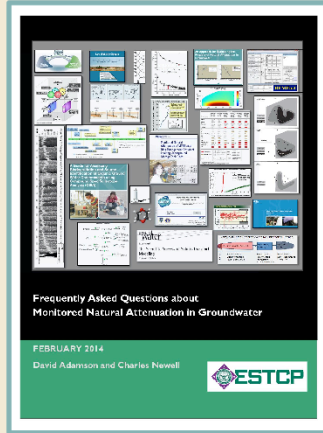
“**Absolute** (objectives based on broad social values, such as protection of public health) or **functional** (steps or activities taken to achieve absolute objectives, such as supplying bottled water to affected residents). Functional objectives are established to demonstrate attainment of absolute objectives and have often been missing, difficult to measure, or unattainable”

- Functional Objectives must be:**
- Specific
  - Measurable
  - Attainable
  - Relevant
  - Time-Bound

from ITRC Integrated DNAPL Site Strategy (IDSS), 2012

- S Specific**—The objectives should specify what is to be achieved through a remedial action. They should be concrete, detailed, and well defined.
- Diagnostic questions:
    - What exactly are we going to do?
    - Is the objective well understood?
    - Will this objective lead to a desired result?
  - *S does not mean “shifting”!*
- M Measurable**—Managers should be able to measure whether or not the objectives are being met. Numbers, quantities, or comparisons should be specified, and the uncertainty in key measurements should be understood.
- Diagnostic questions:
    - How will we know that the change has occurred?
    - Can these measurements be obtained?
  - *M does not mean “magical”!*
- A Attainable**—Objectives should be realistic, given the proposed time frame, political climate, and/or the amount of money available.
- Diagnostic questions:
    - Can we get this done in the proposed time frame?
    - Do we understand the limitations and constraints?
    - Can we do this with the resources we have?
    - Is this possible?
    - Has anyone else done this successfully?
  - *A does not simply mean “ambitious”!*
- R Relevant**—The objective should have a value and represent a realistic expectation.
- Diagnostic questions:
    - Does the outcome of the objective directly support achievement of the absolute objective?
    - Do we have the resources available to achieve this objective?
  - *R does not mean “remarkable”!*
- T Time-bound**—The time allotted for achieving the objective should be clearly defined and short enough to ensure accountability.
- Diagnostic questions:
    - When will this objective be completed?
    - Is someone still going to be accountable for meeting the time frame?
  - *T does not mean “timeless”!*

## INTERACTIVE “FAQ”



**Interactive Version**

### **Advantages:**

- Easy to navigate
- Hosted on ESTCP and GSI websites
- Does not require download

**LET'S TAKE A  
QUICK TOUR!**

<http://serdp-estcp.org/Program-Areas/Environmental-Restoration/Contaminated-Groundwater/Persistent-Contamination/ER-201211/>

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## INTERACTIVE “FAQ”

**LET'S TAKE A  
QUICK TOUR!**

*5 minute run-through of  
interactive version*

## FINAL Q&A

**Go to:**

[http://serdp-estcp.org/Program-Areas/  
Environmental-  
Restoration/Contaminated-Groundwater/  
Persistent-Contamination/ER-201211/](http://serdp-estcp.org/Program-Areas/Environmental-Restoration/Contaminated-Groundwater/Persistent-Contamination/ER-201211/)

**Or Google:**

**“MNA” + “FAQ” + “ESTCP”**

