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# Using Ecological-Based Tools and Approaches to Assess Bioavailability

Sponsored by:National Institute of Environmental Health Sciences, Superfund Research Program

Delivered: June 30, 2010, 1:30 PM - 3:30 PM, EDT (17:30-19:30 GMT)

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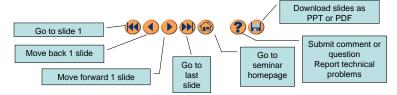
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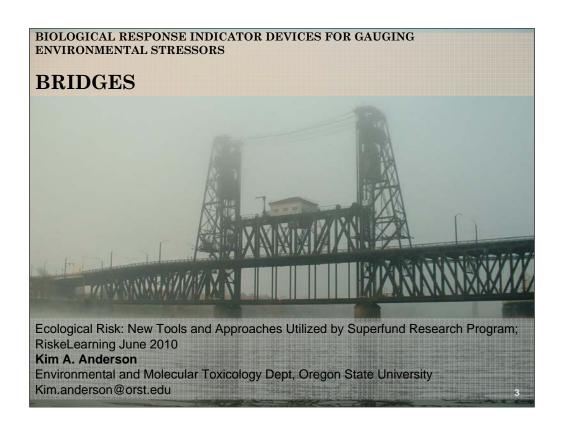
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- Q&A {indicate if there are breaks, or ask whenever, mention? Submission button/form}
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- Move through slides using # links on left or buttons



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### WHY BIOAVAILABLE ?

- Estimating exposure concentration
- Predicting environmental fate
- o Understanding Environmental Factors on Diseases...
  - MUST develop new bio-analytical tools to measure exposure
     L.S. Birnbaum, EHP, 2010



# REAL WORLD-DISSOLVED CONCENTRATIONS NOT SIMPLE $\mathbf{S}_{w}$ FROM THE LABORATORY, ENVIRONMENTAL CONDITIONS EFFECT SPECIATION, BIOAVAILABILITY

Chlordane Solubility ( $S_w$ )- **56 ppb** (ug/L)

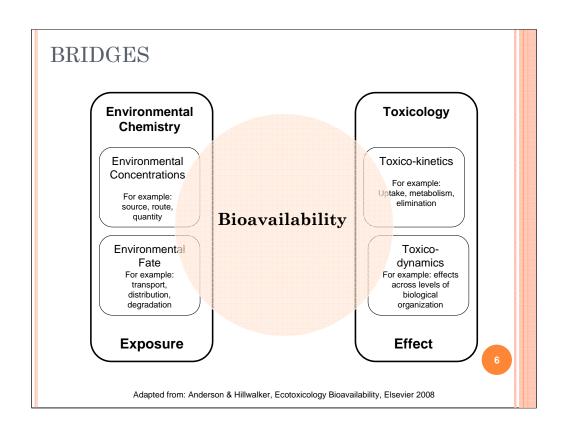
Chlordane Rainbow trout LC50 90 ppb (ug/L)

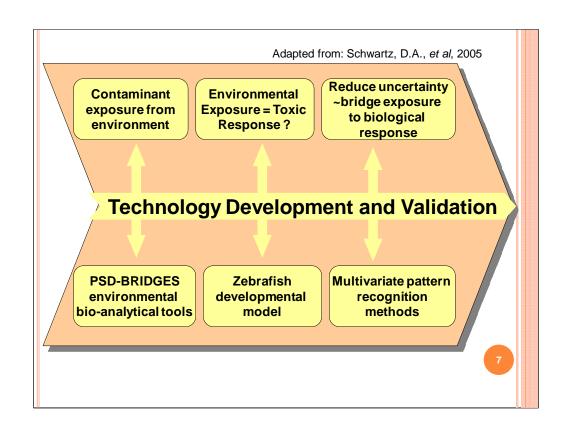
However when more REAL world conditions used

Chlordane Solubility\* (S<sub>w</sub>)- 28,000 ppb (ug/L)

\*Water containing 34 mg/L dissolved organic carbon

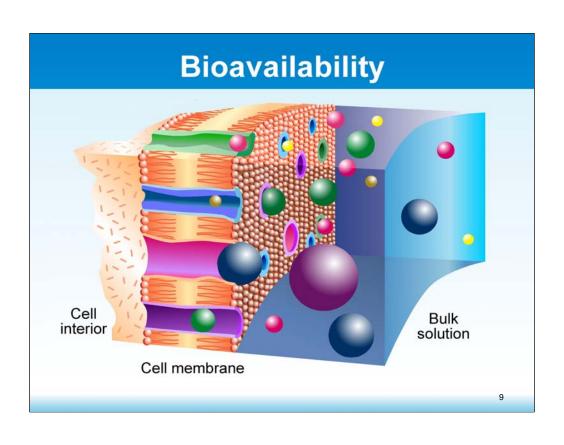
Casarett & Doull's Toxicology, 6th Ed, 2001, pg 1016-1020





#### **BEYOND** CHEMICAL ANALYSIS...

- Exposure dose (exposure concentration)
  - Chemical mixtures
  - Predicting environmental fate
- Advantage not to "lump" bioavailable processes
  - Insight will depend on *isolating* processes
  - For example FIRST step in aquatic food chain
- Integration of space-time into health risk framework
  - Ambient rough estimate
  - Bio-monitoring transient estimate
- o Multi-media

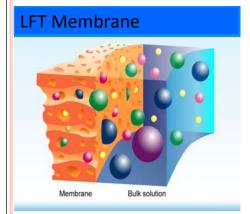


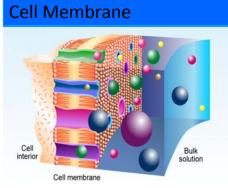
#### PASSIVE SAMPLING DEVICES FOR EXAMPLE: LIPID FREE TUBE (LFT)

- Bioavailability processes include
  - **chemical** (cell membrane lipophilic character) and
  - **physical** (pore size ~9.5A) control of contaminant uptake.
- Passive sampling devices (such as LFT) mimic both chemical and physical processes
  - lipophilic membrane character, and
  - pore size ~10A

- o Other types of membranes:
  - Polymer film or tubes
    - o low density polyethylene
    - silicone or silastic
    - polypropylene
- Södergren, 1987 dialysis bag filled with hexane
- Huckins et al, 1992 developed semipermeable membrane device (SPMD) tube filled with a triolein
- Reviewed in: Namiesnik, J.; Zabiegalal, B.; Kot-Wasik, A.; Partyka, M.; Wasik, A., Passive sampling and/or extraction techniques in environmental analysis: a review. Anal. Bioanal. Chem. 2005, 381, 279-301





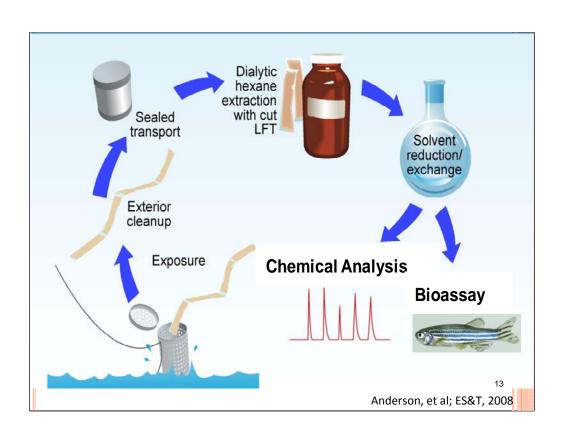


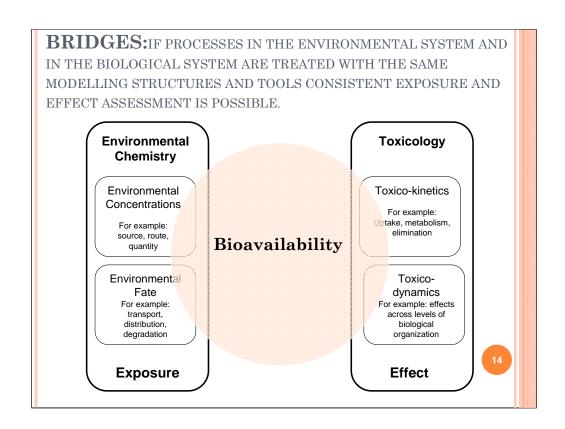
LFTs are polyethylene membranes that, similar to cell membranes, passively uptake freely dissolved (bioavailable) hydrophobic compounds.

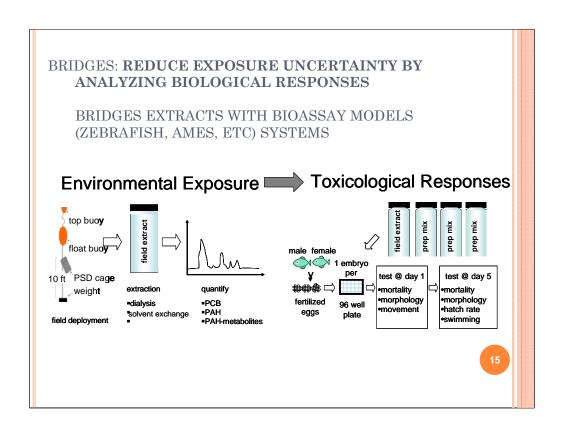
## PASSIVE SAMPLING DEVICES DEPLOYED AT SUPERFUND SITES

• PSD theory: our LFTs are constructed of sealed polyethylene lay flat tubes which represent an organic lipid membrane. Like a membrane, LFTs discriminate against particulate bound material. As *in situ* time integrative passive samplers, LFTs may be deployed for extended periods of time to sequester contaminants. This overcomes potential issues such as detection limits, bioavailable fraction collection and fluctuating contaminant concentrations.









#### BRIDGES BIO-ANALYTIC TOOL

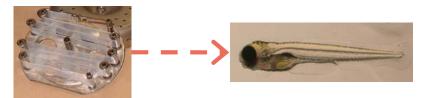
Biological Response Indicator Devices Gauging Environmental Stressors

• Developed in response to the need to link environmental exposure to biological responses.



#### BRIDGES BIO-ANALYTIC TOOL

 Combines Lipid-Free Tubing (LFT) passive sampling devices with the embryonic zebrafish model.

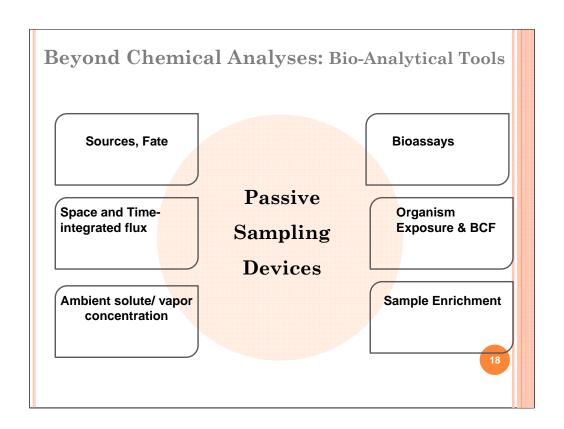


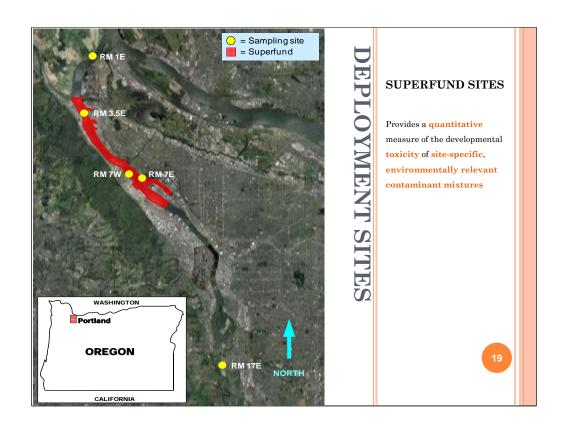
LFTs passively concentrate the **bioavailable** fraction of chemicals from the environment:

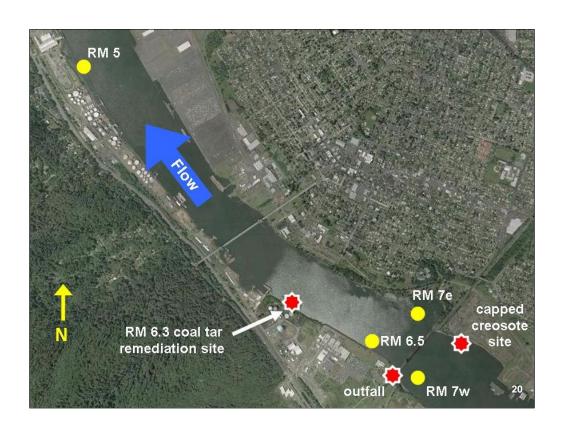
• time integrated, biologically relevant chemical concentrations

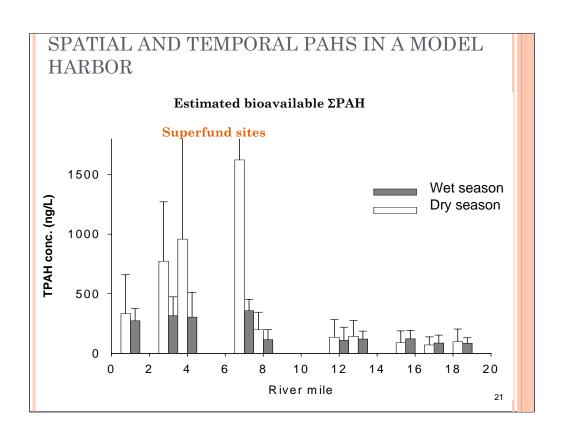
The embryonic zebrafish developmental model is a **high throughput**, **whole organism**, **vertebrate** bioassay widely used for toxicity assessments.

• Provides a quantitative measure of the developmental toxicity of site-specific, environmentally relevant contaminant mixtures.



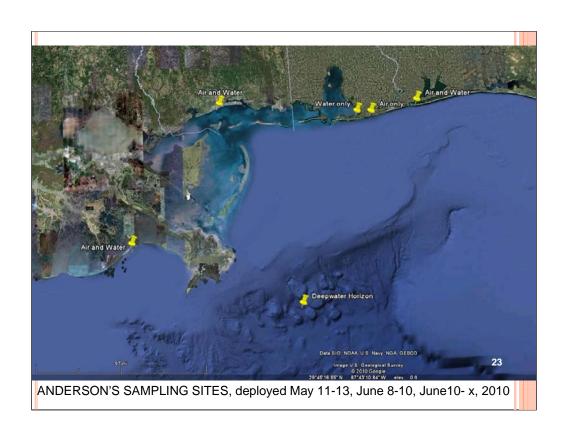






## SOURCES OF PAHS IN THE ENVIRONMENT

- Biogenic (minor)
- Petrogenic
  - Generated by geological processes
    - NATURAL- seeps, coal outcrops
    - ANTHROPOGENIC –fossil fuel release (0.2-7% PAHs)
- Pyrogenic
  - Generated by high temperature combustion of organic matter
    - NATURAL –forest fires, piare fires
    - ANTHROPOGENIC- wood stoves, car exhaust, coal tar



#### **EXAMPLES:**

## PASSIVE SAMPLING DEVICES IN GULF OF MEXICO

PRE-, POST-IMPACT DEEPWATER HORIZON

**BLOWOUT** 

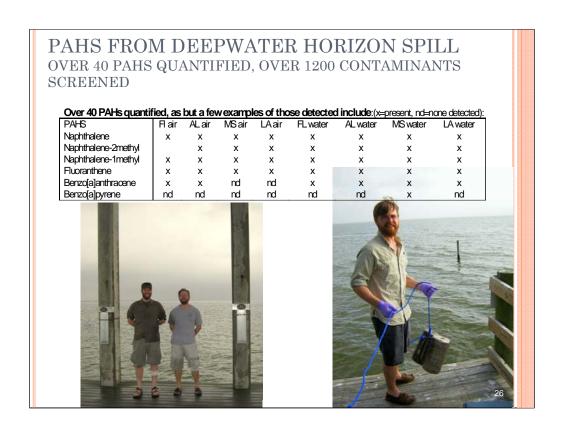
- o 4 sites FL, AL, MS, LA
- o Early, May, Early June
- Paired Air and Water
- o Paired Chem- and Bio-
- Grande Isle hit by oil

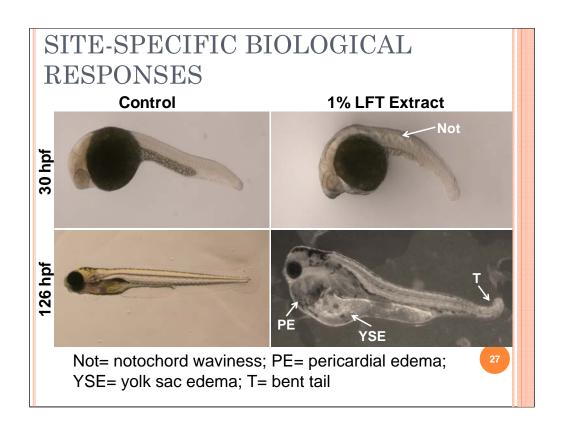


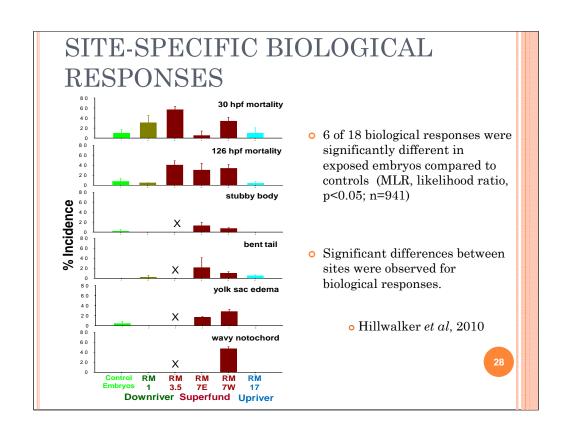
#### **BEYOND, NEXT** CHEMISTRY

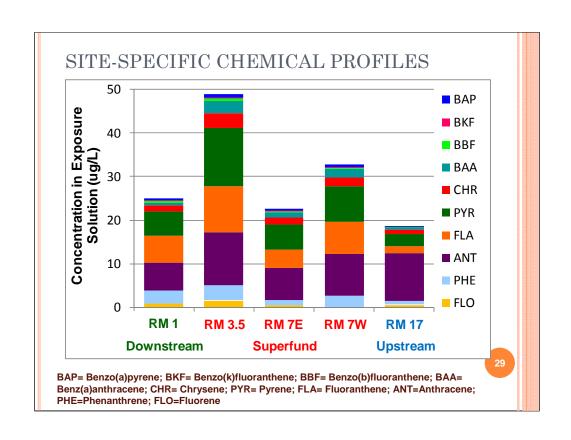
- Expanded PAHs
  - 302 isomers
  - alkylated
- Oxygenated PAHs
  - facilitated by dispersant
  - more bioavailable
  - as toxic as PAHs
- Screening method imperative for mixture assessment
  - **1,200+** analytes





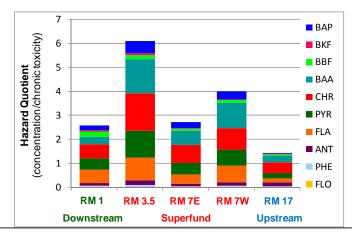


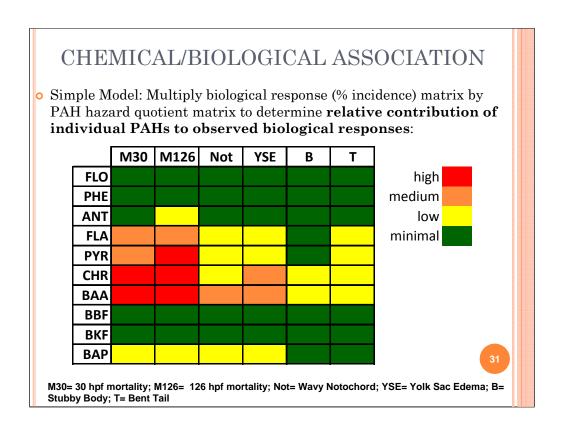




# SITE-SPECIFIC HAZARD QUOTIENT PROFILES

- Divide the concentration by the chronic toxicity value
  - (Neff et al, Integr. Environ. Assess. Manage., 2005)
- Normalizes the concentrations by relative toxicity
- Provides a more realistic estimate of the relative contribution of each chemical to the observed biological responses





## INTEGRATED ENVIRONMENT AND HEALTH

o embryonic zebrafish exposed to environmental contaminant mixtures obtained from passive samplers deployed in a model river system show site-specific biological responses that can be associated with differences in the chemical profiles of the sites

BRIDGING: AN INNOVATIVE APPROACH TO QUANTIFYING RISK AT A MGP REMEDIATION PROJECT USING PASSIVE SAMPLING DEVICES

$$Exposure = \frac{C \times CF \times IR \times EF \times ED}{BW \times AT}$$

PSD mass-to-mass concentrations replace the shellfish/fish tissue concentrations normally used here



# CHEMISTRY + BIOLOGY INTEGRATED ENVIRONMENTAL AND HUMAN HEALTH



- Adding passive sampler concentrations to health risk models increases spatial and temporal precision, improves risk estimates, reduces animal collection, and reduces costs.
- Passive samplers can supplement fish/shellfish data in health assessments to provide specific spatial and temporal contaminant information and thereby help public health and remediation professionals more precisely evaluate and relate exposure and risk

# INTEREST IN PASSIVE SAMPLING DEVICES

## PSD PROVIDED TO SBR RESEARCHERS UPON REQUEST

#### Passive Sampling Devices Attributes:

- sequesters specific fraction (dissolved, bioavailable)
- comparative data rapidly developing
- iterative (captures episodic events)
- composite without mechanical equipment or on-site power requirements
- no seasonal issues with PSD compared w/ organisms
- <u>very low detection</u> limits possible with relative analytical ease
- minimizes decomposition or chemical change transport/storage
- · does not lump biological processes
- · Less expensive
- · Easier to replicate spatially
- Greener technology
- · adequately mimics biological
- not as many negatives as organisms with more positives than grabs...
- in-situ assessment of environmental contaminants that are most <u>biologically relevant</u>
- have ability to use in bio-assays
- easily able to do exposure concentration dependent bio-assays

#### Using PSDs for Risk Assessment

- Advantages
  - Inexpensive
  - Easy to analyze
  - Site specific
  - Bio-available fraction
  - No metabolic activity
  - Nondestructive





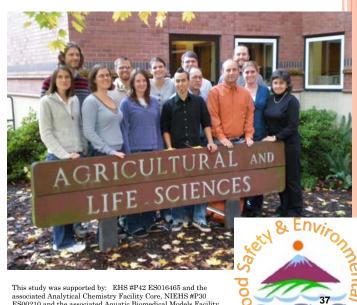




### **ACKNOWLEDGMENTS**

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Robert Tanguay **SARL** 



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# Chemical Reaction Kinetics Model uptake and release of contaminant

Air/Water

Passive Sampling Device

Air/water

$$C_{a/w} \xrightarrow{k_u} C_m \xrightarrow{k_e} C_{minate}$$

Rate to change of the concentration:

$$dC_m/dt = k_uC_w - k_eC_s$$



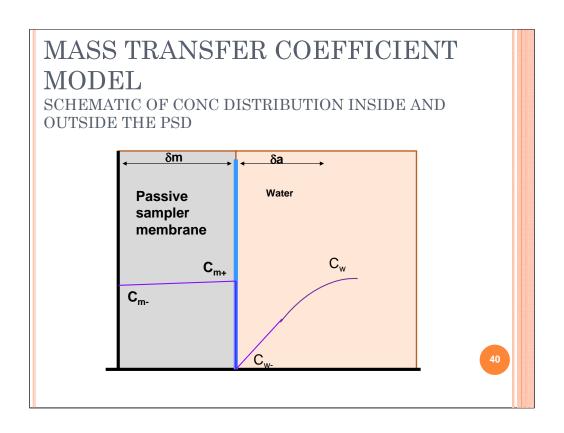
Conc at any t is determined by competing rates of uptake and release

### Performance Reference Compounds Uptake and Release of Contaminant

$$N = N_0 \exp(-k_e t)$$

Where  $N_0$  is the amount present at  $t=0,\,N$  is the amount present after

$$k_e = -ln(N/N_0)/t$$



## WATER (OR AIR) CONCENTRATION

$$R_s = V_m K_{mw} k_e$$

Rs = sampling rate

$$C_w = N/V_m K_{mw} (1 - exp((-R_s t/V_m K_{mw}))$$





# Where, What, and How



Illustration by W. Scavone

- Aquatic Ecosystems
  - Lakes
  - Estuaries
- Metals
  - Hg
- Approaches
  - Field studies
  - Mesocosm studies

## **Ecological Processes**

**Trophic Transfer** occurs when concentrations in predators are related to concentrations in their prey.

**Biomagnification** occurs when metal concentration increases with each level of a food chain.

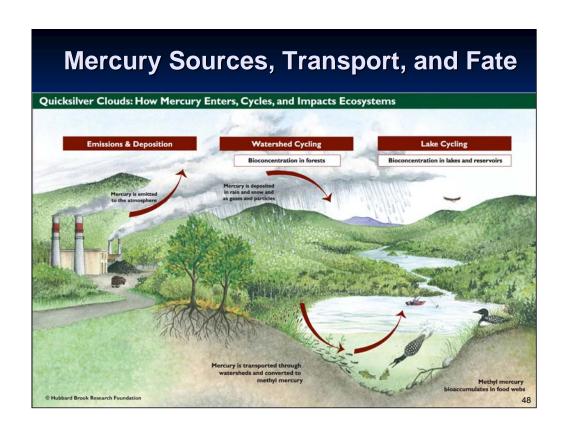
**Biomass dilution** occurs when metal concentration decreases with increasing plankton biomass

Feeding Pathway distinguishes the route through which metals are transferred in food webs (benthic vs. pelagic)

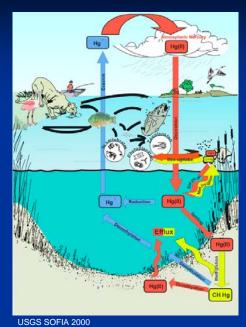


# **Chemical Forms of Hg**

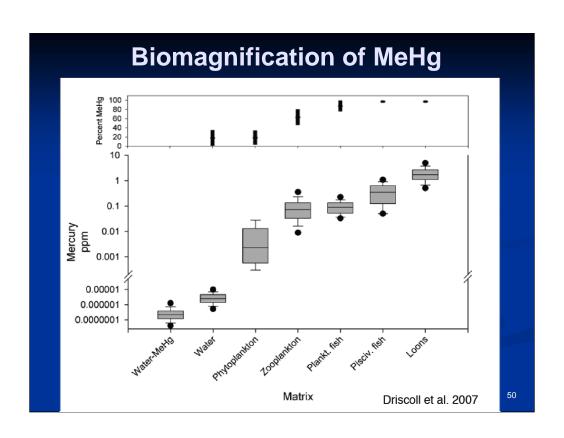
- Elemental Hg<sup>0</sup>
  - Mostly atmospheric
  - •Long residence time 0.5-2 yrs
- Oxidized Hg<sup>2+</sup>
  - •Dominant form in atmosphere and water
  - •Reactive gaseous Hg (RGM)
  - Particulate Hg (PHg)
- •CH<sub>3</sub>Hg<sup>+</sup> monomethylmercury (MeHg)
  - Toxic form
  - •Bioavailable form



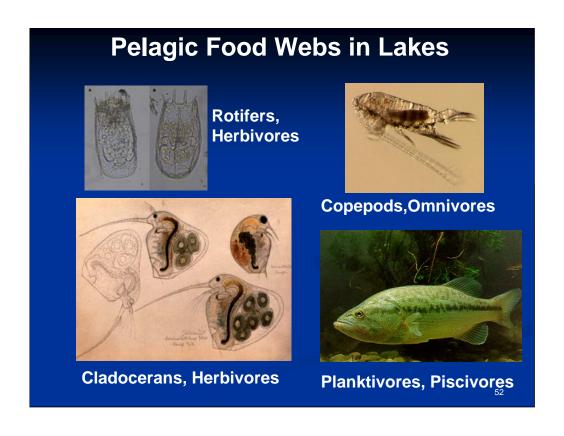
# **In-Lake Mercury Processes**



- Hg enters lake as Hg(II)
- ■Hg methylation occurs in sediment producing MeHg
- ■MeHg and Hg(II) are fluxed from sediments to water
- •MeHg and Hg(II) are taken up by particulates and phytoplankton
- ■MeHg is preferentially assimilated and transferred up the food web



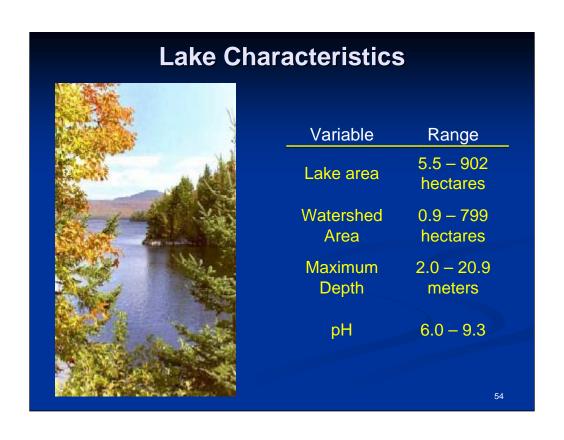
# 1. Field Approaches in Lakes to Investigate Metal Bioaccumulation and Trophic Transfer



# Field Survey of 20 Lakes in the Northeast US



- Environmental variables
- Phytoplankton and zooplankton metal concentrations
- Phytoplankton and zooplankton taxonomy
- Concentrations of metals in fish tissue



# Field Sampling and Sample Analysis

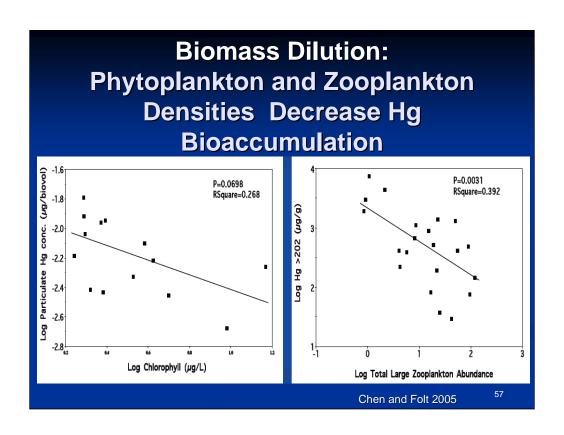


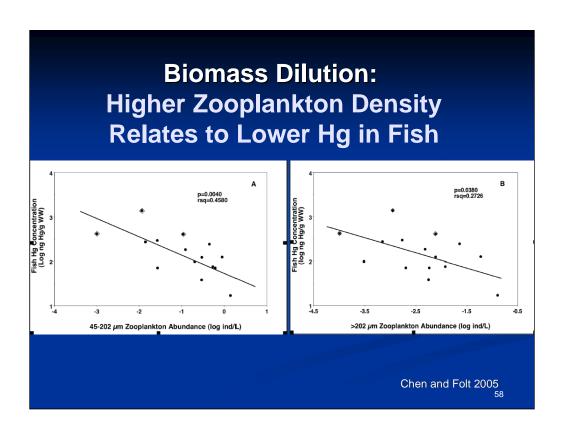
- Trace metal clean technique
- Plankton metal and taxonomy samples (45-202 and >202 μm)
- pH, TN, TP, DO, Cond., DOC, Chlorophyll
- Calculated variables (BMF, Food web structure)

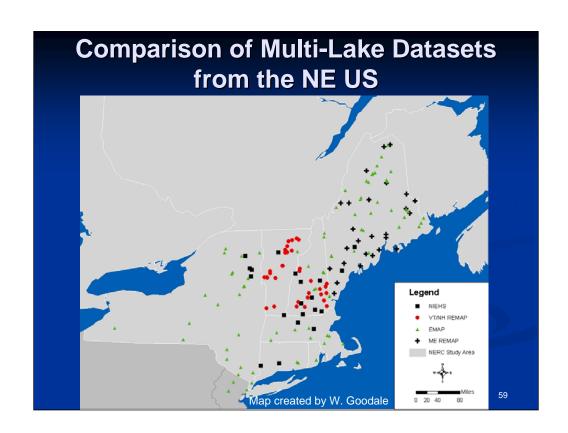
# Metal Analysis: Dartmouth Trace Element Analysis Core

- Acid digestions
- High resolution ICP-MS (Trace Metal Lab Facility)
- Hg analysis via cold/vapor ICP-MS





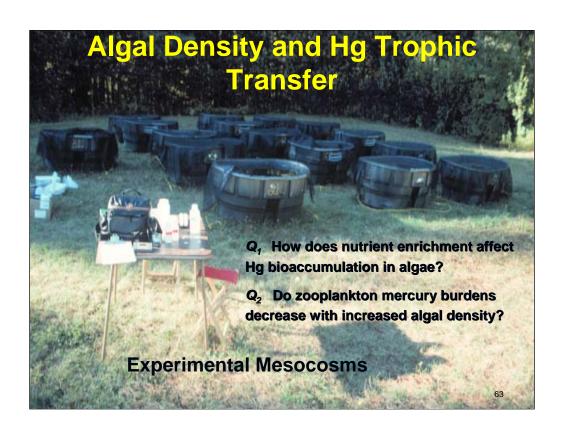


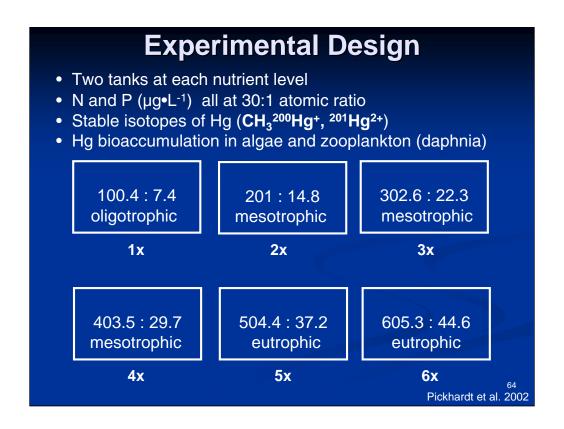


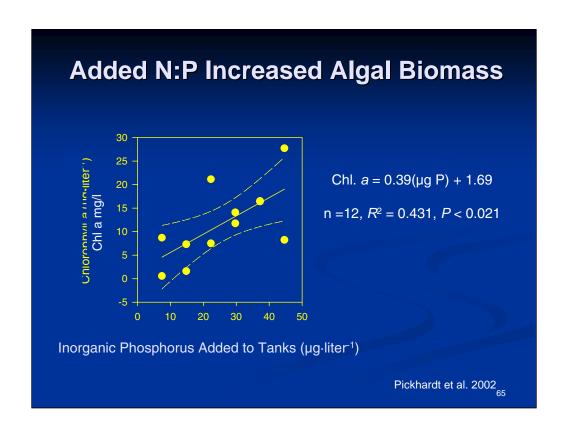
Variable Category	Variables	NIEHS	REMAP VT- NH	EMAP	ME REMAR
Physical	Lake area Watershed area	X X	X X	X X	Χ
	Depth (max.)	X	X	(WLR)	X
	Elevation			X	X
Chemical	pH	X	X	X	X
	Alkalinity	X	X (Epi ANC)	X (ANC)	X
	DOC SO4	X	X (Epi DOC) X (Epi SO4)	X	X
	Conductivity	X	X (Epi 304)	X	X
	Total P	X	X	X	
	Total N N:P	X		X	
Landuse	Population		X (911 count)		X
	Disturbed	X		X	
	Residential Hay/Perm. Past,		X	X (AG)	
	Comm/Ind/Transp.		X	X (URB)	
	Forest		X	X	
	Wetland	X	(Forest/Wetl.)	X	
	Road density	X	V	X	
	Linkages Max. chain length	X	X		
	Chorophyll	X		Х	X
	Algal biomass	X			
	Zoop. abundance	X	X	X	X
	Zoop. Biomass Richness	X	X		Χ
Hq conc.	Aqueous	X	X (Epi THg)		^
	Zooplankton	X	X		
	Fish	X (MX)	X (YP)	X (MX)	X (MX)

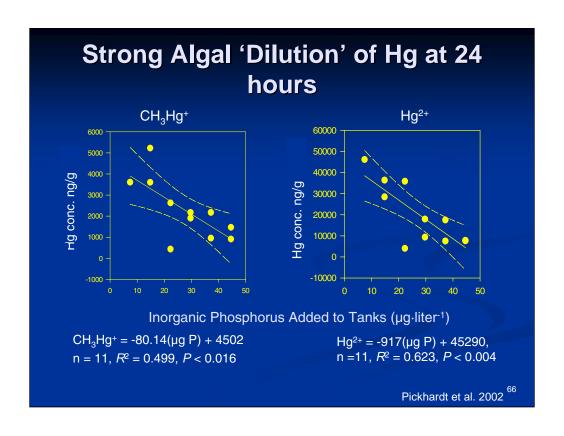
Common Variables Spearman Rank Correlation, P<0.05					
Variable Category	Common Variables	Relationship to Hg in Fish			
Physical	Lake area (3/4) Watershed area (2/3)	(+)			
Chemical	pH (3/4) ANC (alkalinity) (3/4) Conductivity (2/4) SO4 (2/2) Nutrients (2/3)	(-)			
Land Use	Residential, Agricultural, Commercial Industrial, Roads, Disturbed (3/3)	(-)			
Ecological	Zooplankton abundance (3/4)	(-)			
	Bonferroni Adjustment	61 Chen et al. 2005			

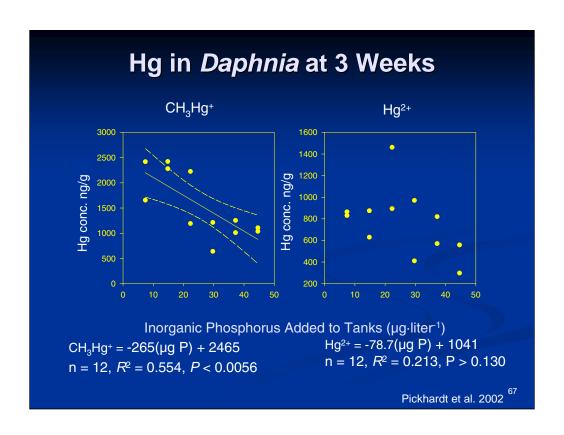
# 2. Experimental Approach in Mesocosms to Investigate Biomass Dilution and Trophic Transfer











# 3. A Field Approach in Coastal Marsh Ecosystems to Investigate Feeding Pathways of Trophic Transfer

Bioadvection of Contaminants via the Trophic Nekton Relay
Tropine Nekton Kelay
QuickTime™ and a TIFF (LZW) decompressor are needed to see this picture.
Illustration by W. Scavone



# Sampling in the Field





#### **Biotic Samples:**

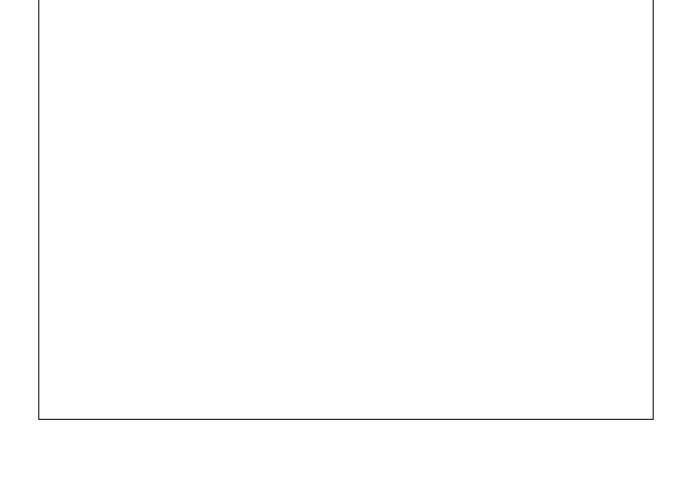
Primary consumers: Snails, Mussels Secondary consumers: Killifish, Green Crabs

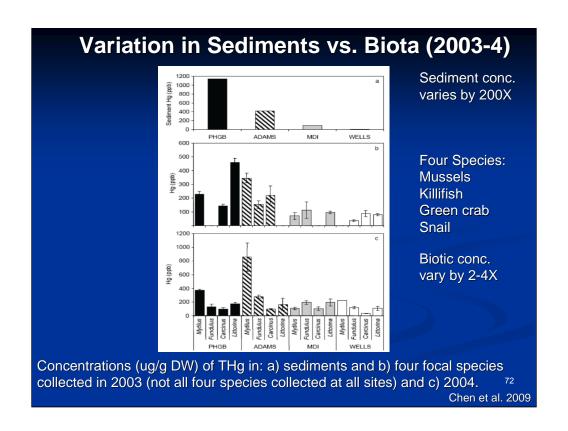
#### **Metal Analysis**

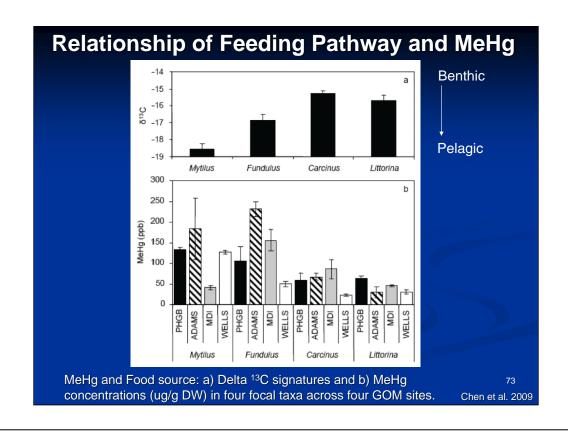
- -whole organisms freeze dried & homogenized
- -Hg speciation by isotope dilution purge and trap GC-ICPMS at Dartmouth

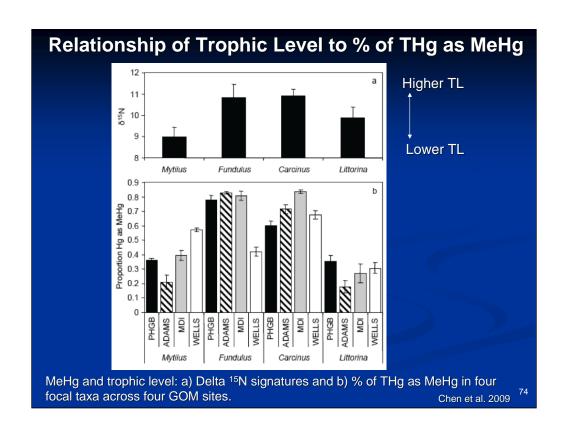
#### **Isotope Analysis**

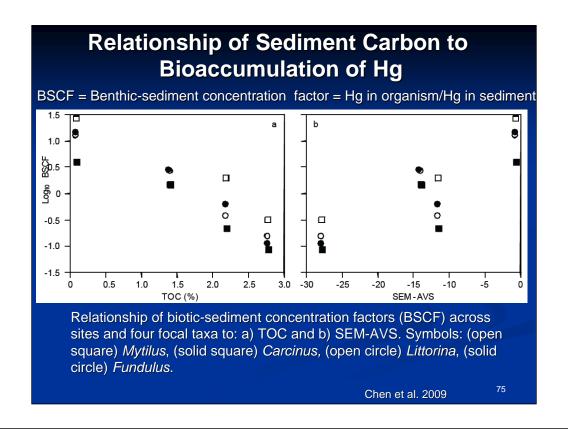
- -separate set of samples collected in field simultaneously
- -whole organisms freeze dried and homogenized  $\delta^{13}C,\,\delta^{15}N$











## **Overall Conclusions**

Metal fate in aquatic food webs is determined by a number of ecological processes:

**Trophic Transfer** 

Biomagnification.

**Biomass dilution** 

**Feeding Pathway** 

These processes are present across lake and estuarine systems

## **Acknowledgments**

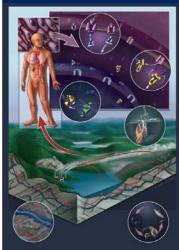


Illustration by W. Scavone

#### **Dartmouth Collaborators:**

Carol Folt, Brian Jackson, Vivien Taylor, Jason Williams, Deenie Bugge, Darren Ward

### Other Collaborators:

Paul Pickhardt, Lakeland College Neil Kamman, VT DEC Michele Dionne, Wells NERR Dave Evers, Biodiversity Research Institute Brandon Mayes, Long Trail Brewery

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