

Mined Materials Geochemical Characterization Primer

Prepared for:
Hardrock Mining Geochemistry and
Hydrology Workshop 1: Evaluating Water Chemistry Predictions at
Hardrock Mine Sites

Sponsored by:
U.S. EPA Region 10, Office of Research and Development (ORD),
and the Office of Superfund Remediation and Technology Innovation

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February 13, 2013

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Linkages

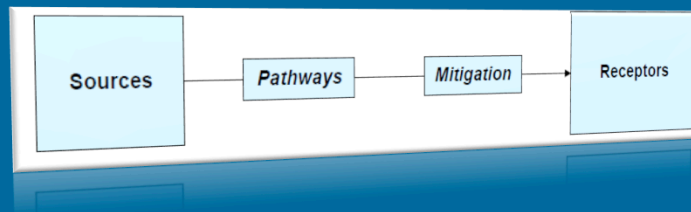


Grasberg Open Pit, New York Times, 12/27/05

- Geochemical characterization – modeling – mine management
- Purpose of characterization and modeling is to guide management decisions
 - Which rock goes where in the field? Will water treatment be needed? Will mitigation work?
- Results of some geochemical tests used for field decisions, others as inputs to block or geochemical models

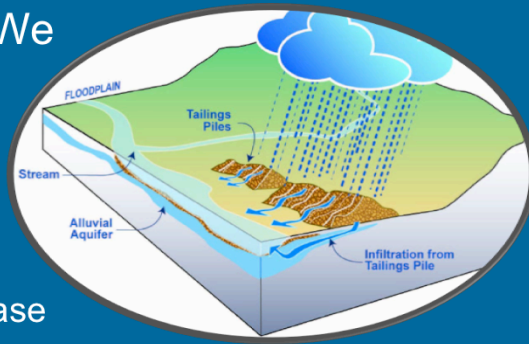
Geochemical Characterization of What?

- Mined materials (sources)
 - Tailings, waste rock, walls of open pits and underground workings, ore (why?), heap and dump leach materials, smelter slag, blended wastes, cemented backfill...



What Processes Are We Trying to Simulate?

- Earth processes
 - Dissolution, precipitation, acid/base
- Mining processes
 - Creation of tailings, waste rock, etc. – from crushed drill core
 - Blasting is rarely included – commonly missing contaminants of concern (NO_3/NO_2 , NH_4)
 - Heap leaching (CN)



The Real World: Waste Rock Yanacocha Mine, Peru

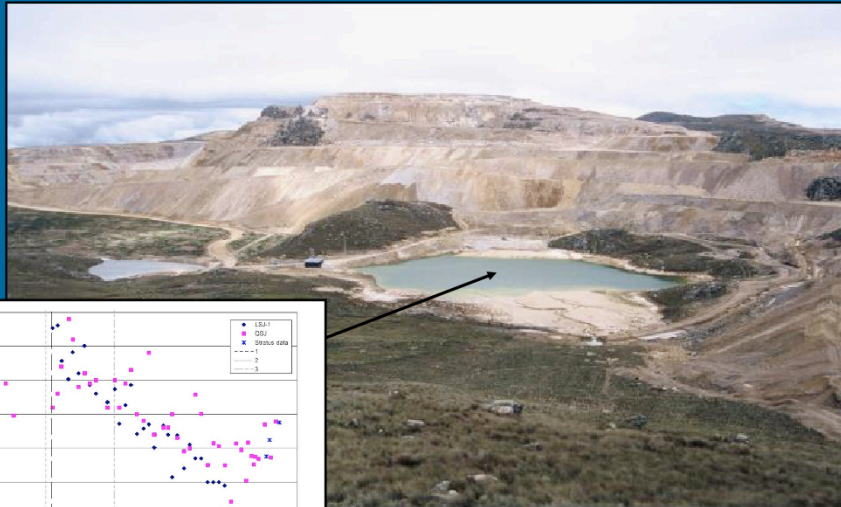
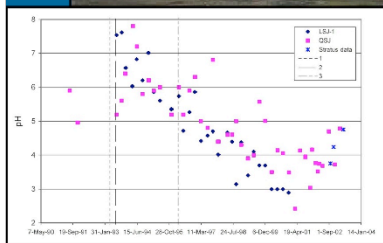
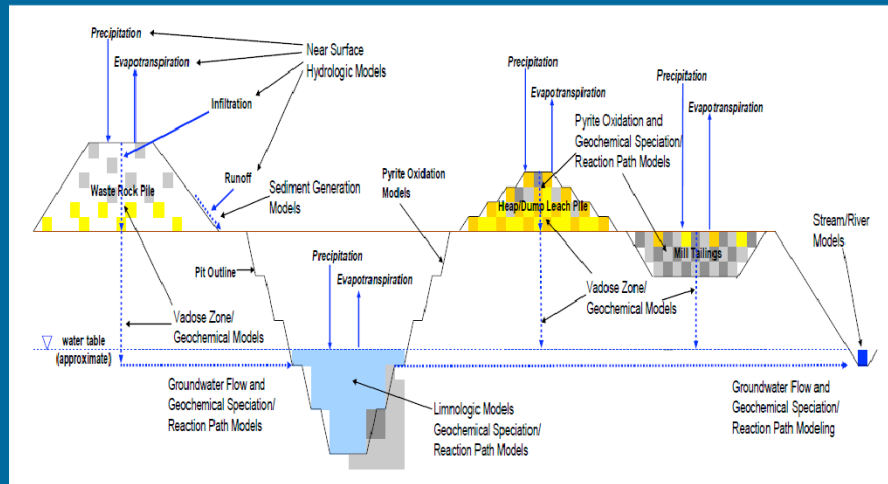


Photo by A. Maest



Sources, Pathways, Modeling



Kuipers and Maest, 2006

Characterization Overview



Tintaya Cu Mine, Peru; photo by A. Maest

- Focus on new and expanding mines
- Basics: test units, # samples
- What methods are used to characterize the geochemistry of mined materials?
- What are the advantages, limitations, and uses of each method?
- What kind of characterization should be done in each phase of mining?

Geochemical Test Unit



Yellowstone: <http://www.americansouthwest.net/wyoming/photographs700/purple-rock.jpg>

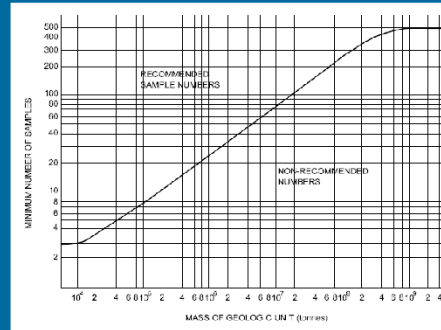
- Most important phase of predictions is sample selection – capture variability
- Rock types of distinctive lithology, mineralogy, and/or alteration, mineral availability (“liberation”)
- Should be as homogeneous as possible
- Could evolve during exploration/operation
- Examples: propylitically altered rhyolite, granodiorite with quartz-sericite-pyrite alteration
- Conduct full geochemical characterization on each unit

Sample Size

- *Suggested* samples/ton for each geochemical testing unit
- More homogeneous materials (tailings) require fewer samples
- Sample entire unit; put geochemical characterization information in block model

Mass of Each Separate Rock Type (tonnes)	Minimum Number of Samples
<10,000	3
<100,000	8
<1,000,000	26
10,000,000	80

Price and Errington, 1994.



US EPA, 2003 (BC AMD Task Force, 1989)

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How much is enough?

- No magic #
- Some statistical approaches
- Of course don't use this for # of HCTs, more geared toward ABAs and static testing

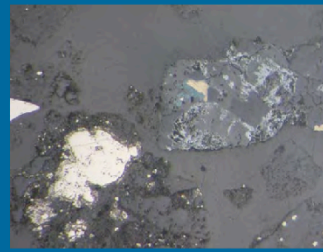
Geochemical Characterization Methods

- Static testing
 - Lithology and alteration zones
 - Whole rock analysis
 - Mineralogy
 - ABA, NAG tests
 - Short-term leach tests
- Kinetic testing
 - Humidity cell
 - Column tests
 - Field tests



Acid drainage at Eagle Mine, CO; photo by A. Maest

Lithology and Alteration Zones



Pebble deposit, Alaska; PLP, 2011, App. 11E; pyrite, chalcopyrite

- *What:* Rock types and alteration overprints
- *How:* Borehole logs, petrographic/mineralogic analysis, block model
- *Use:* ID geochemical test units
- *Limitations:* Sample representativeness

Whole Rock Analysis



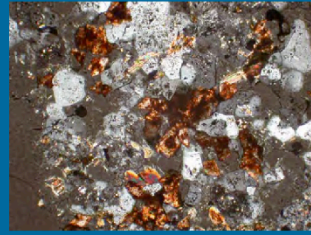
Pinson Mine, NV, heap leach monitoring: photo by A. Maest

- *What:* Total concentrations of metals, etc., in rock/waste
- *How:* Grind sample, acid digestion, analyze for metal, etc., content by XRF, ICP, -AES, -MS...
- *Use:* ID overall contaminant levels in rock types
- *Limitations:* Detection limits, interferences; does not provide information on mineralogy

Constituents of Interest/ Concern

- Start bigger, get smaller
- Solids, liquids (charge balance if liquids)
- Focus on potentially toxic constituents, AGP/ANP
- General: pH, SC, alkalinity, acidity, TDS
- Metals
 - Ag, Al, As, B, Ba, Be, Ca, Cd, Co, Cr, Cu, Fe, Hg, K, Mg, Mo, Mn, Na, Ni, Pb, Sb, Se, Ti, V, Zn...
- Non-metals
 - Cl, CN, F, NH₄, NO₃/NO₃, S, Si, SO₄ ...

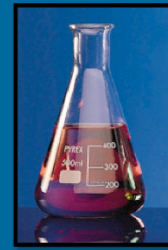
Mineralogy



*Pebble deposit, Alaska; PLP, 2011.
App. 11E; carbonate replaced by hematite*

- *What:* ID minerals and poorly crystalline substances present in rock/waste samples
- *How:* Optical microscopy, XRD, electron microscopy (SEM, TEM, HR-TEM), sulfide oxidation index/Rietveld analysis, AVIRIS (remote spectral imaging)
- *Use:* ID controls on solubility, identity source of AGP/ ANP, mineral availability (“liberation”)
- *Limitations:* Need specific expertise to interpret results, not great for secondary minerals, representativeness

Acid-Base Accounting (ABA)



- *What:* Total amount of acid-generating and acid-neutralizing material in a mined material
- *How:* Pulverize sample; add acid or H_2O_2 (AP), backtitrate with NaOH (NP)
- *Use:* Identify rock units with potential to generate acid; waste management
- *Advantages:* Well established, fast/cheap, operational definition for field management
- *Limitations:* Not for predicting long-term behavior

Kinds of sulfur: total, pyritic, sulfide, organic, sulfate

Part of acid-base accounting (ABA) testing; distinguishes between forms with more (pyritic, sulfide) and less (organic, sulfate) acid generation potential (AGP)

Issues: which form to use in AP (over/under-estimate AGP), does not confirm identity of minerals that contain the sulfur

Primary Sources of AP and NP

- Acidity

- Pyrite, pyrrhotite, marcasite, chalcopyrite, arsenopyrite...
- Certain Fe sulfate minerals
- Siderite

- Neutralization potential

- Calcite, dolomite
- Certain aluminosilicates (more likely at lower pH values)



Melanterite
<http://www.mindat.org/min-2633.html>



Pyrite in limestone
<http://www.mindat.org/min-3314.html>

Good summary: Plumlee, 1999.

~ ABA Testing Methods

- *Modified Sobek* (pH 7), Lapakko (pH 8.3), BCRI, BCRC, siderite correction
 - Most commonly used
- NCV (Newmont): no titration, infrared for C and S
 - Only includes carbonate minerals in NP
 - Can overestimate NP if siderite present
- NAG (Net Acid Generation): $\text{H}_2\text{O}_2 + \text{NaOH}$
 - Commonly used in Australia, screening only, fast
 - Does not distinguish between AP and NP



Equity Silver Mine, Canada;
photo by A. Maest

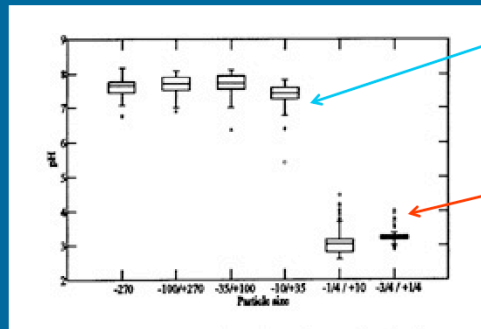
ABA/Static Testing: Main Sources of Uncertainty



Cananea Cu Mine, Mexico; photo by A. Maest

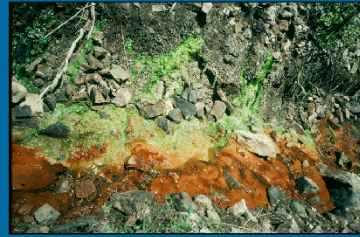
- Crushed sample – assumes all AP and NP available
 - Fracture surface vs. groundmass, encapsulation
- Final pH < 6: overestimate NP (silicates)
 - Modified Sobek and Lapakko pH 6 most reliable and conservative (Sobek > modified Sobek > BC Research > Lapakko)
- Mineralogy unknown – compare to “mineralogic” AP and NP
 - Especially important for low S, low NP wastes

Grain Size and Mineral Availability



Lapakko et al., 1998; <http://wvmdtaskforce.com/proceedings/98/98LAP/98LAP.HTM>

Interpretation of ABA Results



Acid drainage at Eagle Mine, CO; photo by A. Maest

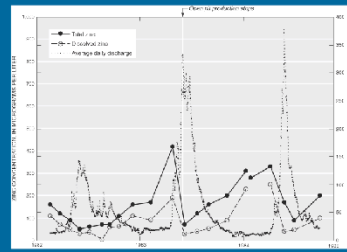
- Many options that rely on %S and/or NP, AP
- NP:AP, NNP (NP-AP), NCV ranges, etc.
- Ideally compare to kinetic testing results or actual mine drainage
- NP:AP
 - Likely not acid-generating: > 3 (or 2 or 4)
 - Uncertain: 1–3 (or 2 or 4)
 - Potentially acid-generating (PAG): < 1 (or 0)

Short-term Leach Testing

- *What:* Readily soluble components of mined materials; some states have regulatory levels (often 100x MCLs)
- *How:*
 - Synthetic precipitation leaching procedure (SPLP) (20:1 = water:rock ratio)
 - Nevada meteoric water mobility procedure (MWMP) (1:1)
 - California waste extraction test (WET) (10:1)
 - British Columbia special waste extraction procedure and modification (BC SWEP) (3:1)

Short-term Leach Testing (cont.)

- *Advantages/use:* Estimates leached concentration ranges from storm/hydrologic events
- *Limitations:*
 - Avoid use of unweathered materials
 - Not for predicting long-term behavior – only 18–48 hr tests
 - Water:rock ratio (Nevada MWMP has lowest w:r ratio, more conservative for arid climates)



Kinetic Testing

- *What*: Estimates long-term potential to generate acid and other contaminants
- *How*: Crush rock, apply water, measure
 - Laboratory kinetic tests
 - Humidity cell
 - Column (aerated, subaqueous)
 - Field kinetic test
 - Waste rock or tailings test piles
 - Wall washing
 - Minewall approach (Morin and Hutt, 2004)



2-yr kinetic tests, Montana Tunnels, MT;
Photo by A. Maest

Crush rock (<6.3 mm for waste rock, 150 mm for tails), place in column

HCT: 3-d alternating humid air/dry air cycles, flush every week, 20+ wks

Measure pH, sulfate, metals, etc. in leachate

Column tests – larger columns and particle size (<~25 mm), “trickle leach”

Kinetic Testing (cont.)

- *Advantages/uses*: Acid production rates, long-term weathering, **input to geochemical models**
- *Limitations*
 - Representativeness, focus on uncertain ABAs
 - Field/lab discrepancies: particle size
 - Length of tests: 20 weeks standard HCT length; too short for most materials, especially if higher NP
 - Lapakko: tailings with 1.3 wt% calcite and 6.6 wt% pyrite took 112 weeks to generate acid; mix of rotary kiln fines and rock with 2.1 wt% S from Duluth complex took 581 weeks to produce acid

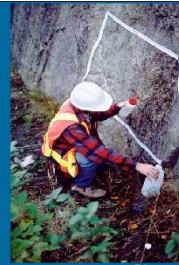
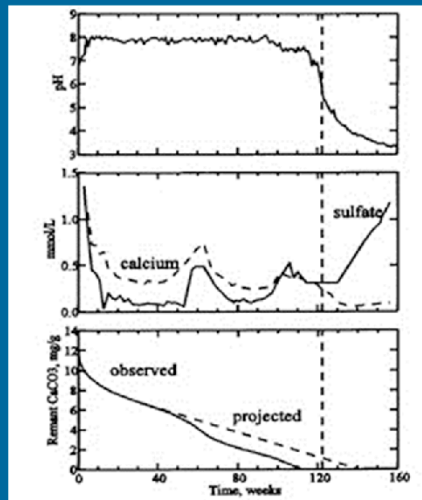


Photo:
<http://www.gardguide.com/index.php/>
Image: WallWashing.jpg

Should run kinetic tests on samples with full range of ABA results – need to know concentrations for input to geochemical models

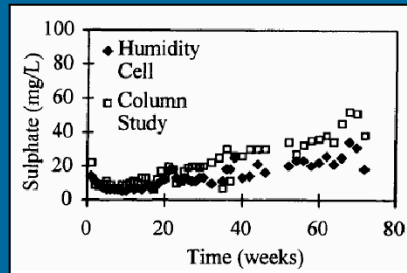
Kinetic Tests: Examples



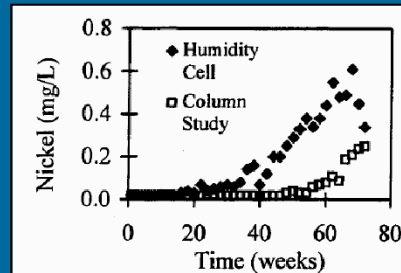
Lapakko et al., 1998; <http://wvmdtaskforce.com/proceedings/98/98LAP/98LAP.HTM>

- $\text{pH} < 6$ at week 122
- $[\text{Ca}] < [\text{SO}_4]$ shows NP rate $<$ AP rate
- All calcite depleted at week 112
- NP:AP = 0.09

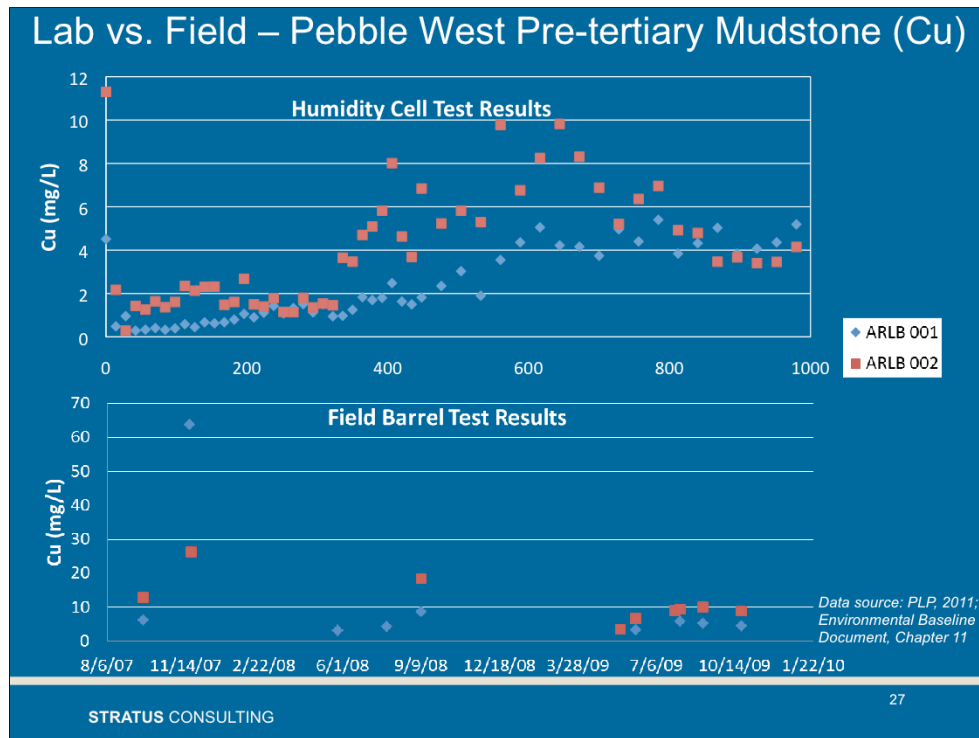
Kinetic Tests: Examples (cont.)



Nicholson and Rinker, 2000



- Metal leaching under neutral pH conditions
- Comparison of HCT and column test Ni and SO₄ concentrations



- Compare field and HCT splits
- First flush in weathered, then decreasing concentrations – “steady state” = last 5 week average
- Different trends if weathered or not
- Need to run even if ABA is PAG – to know concs for inputs to models

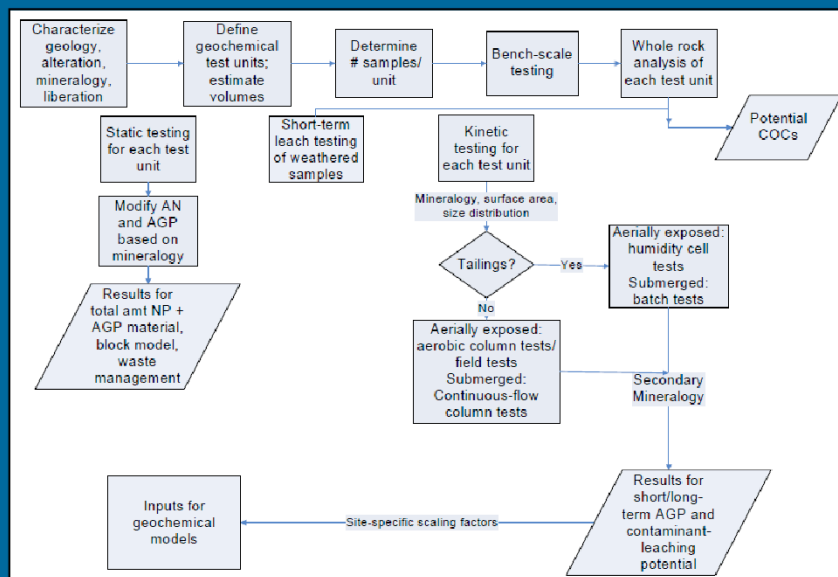
When to Characterize?



See Maest et al., 2005 for more detail; Tintaya Mine, Peru, ball mill; photo by A. Maest

- Exploration
 - Static testing (lithology, mineralogy, ABA...)
 - Geochemical testing units, block model
- Mine development
 - Continue static, start kinetic including field tests
- Operation
 - Continue lab/field testing; predicted/actual comparisons; waste leachate samples
- Closure
 - Continue lab/field comparisons

Geochemical Characterization Overview



Maest et al., 2005

Summary



*Rayrock Mine, NV, heap leach pad;
photo by A. Maest*

- Geochemical characterization aims to identify potential contaminants of concern and simulate range of concentrations under mining conditions
- Purpose is to inform mine management, including waste/ore placement, water quality monitoring, need for and type of water treatment and mitigation, effectiveness of mitigation measures
- Very few required tests or interpretation approaches
- Each method has advantages and limitations, and real crux is interpretation of results
- Need to compare predictions from tests to real conditions as mining proceeds