

Understanding and enhancing PFAS phytoremediation mechanisms using hemp plants

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Progress in Research
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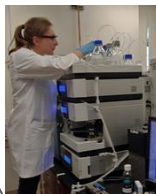
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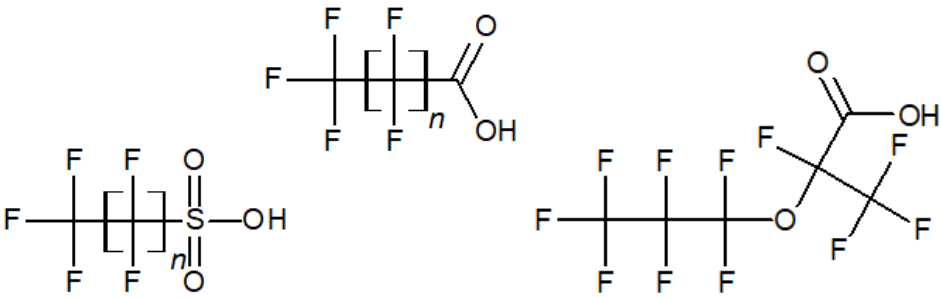
Cheng-Hsin Huang
Porous silica
synthesis and
characterization



Phil Zuccaro,
PFAS

PFAS: Per- and Polyfluoroalkyl Substances

- Molecules that contain carbon-fluorine chains
 - Over 10,000 listed on US EPA website
 - “Forever chemicals” that don’t degrade
- In use since the 1940s
- Used in many consumer products
- Widely spread environmental contaminants



Project Beginnings

- The burn house and parking lot located at the former Loring Air Force Base in northern Maine were used for firefighting drills and testing
- The land now belongs to the Aroostook Band of the Micmac Nation, an indigenous people
- Community members reached out to CAES for help with a hemp phytoremediation project focused on cleaning up PFAS





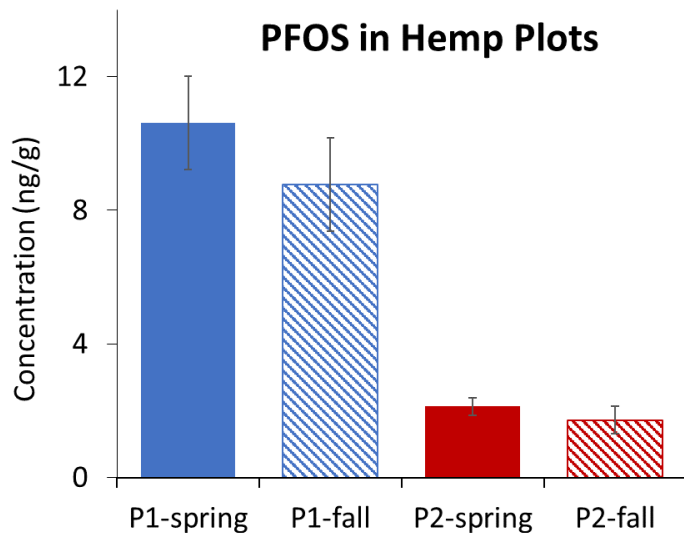
Field Team at Loring Air Force Base

Analytical Methods

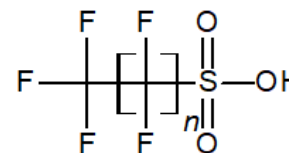
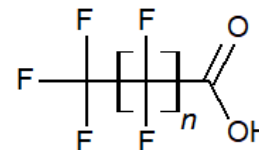


- LC-HRMS based analysis of soil and plant tissue
- Targeted list of ~30 PFAS
- Non-targeted analysis with FluoroMatch and Compound Discoverer
 - Found over 90 suspected PFAS in Loring soil
 - Aided in development of FluoroMatch software updates
- Methods also used in additional projects - like this one
- Always interested in additional collaborations!

Preliminary Data



| Compound | Bioconcentration Factor |
|-------------|-------------------------|
| PFBA (n=2) | 4 |
| PFPeA (n=3) | 5 |
| PFBS (n=3) | 10 |
| PFHxS (n=5) | 7 |
| PFOS (n=7) | 0.13 |



- Statistically significant decreases in soil PFOS in both hemp growth plots
- Lower bioconcentration for longer chain PFAS

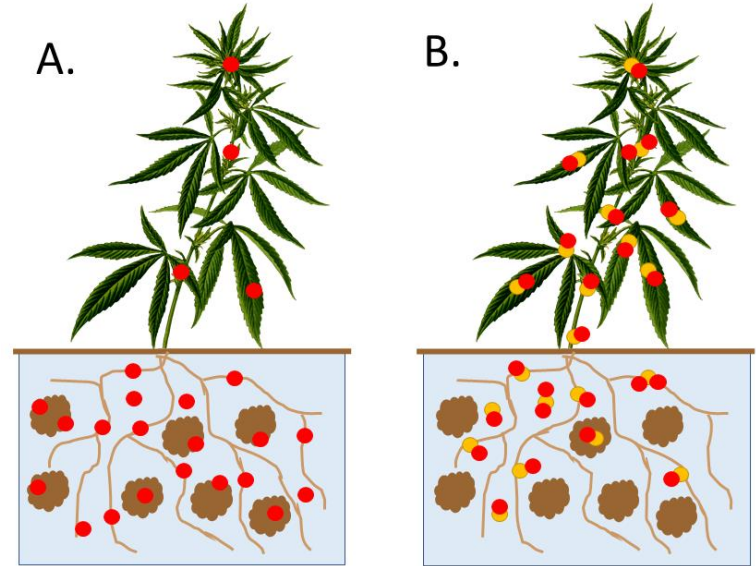
Overarching goal: use novel materials to make phytoremediation a viable solution for removing a wide range of PFAS from soil

Specific Aim 1: Synthesize novel nanomaterials (NNMs) designed to sorb PFAS and translocate through plants

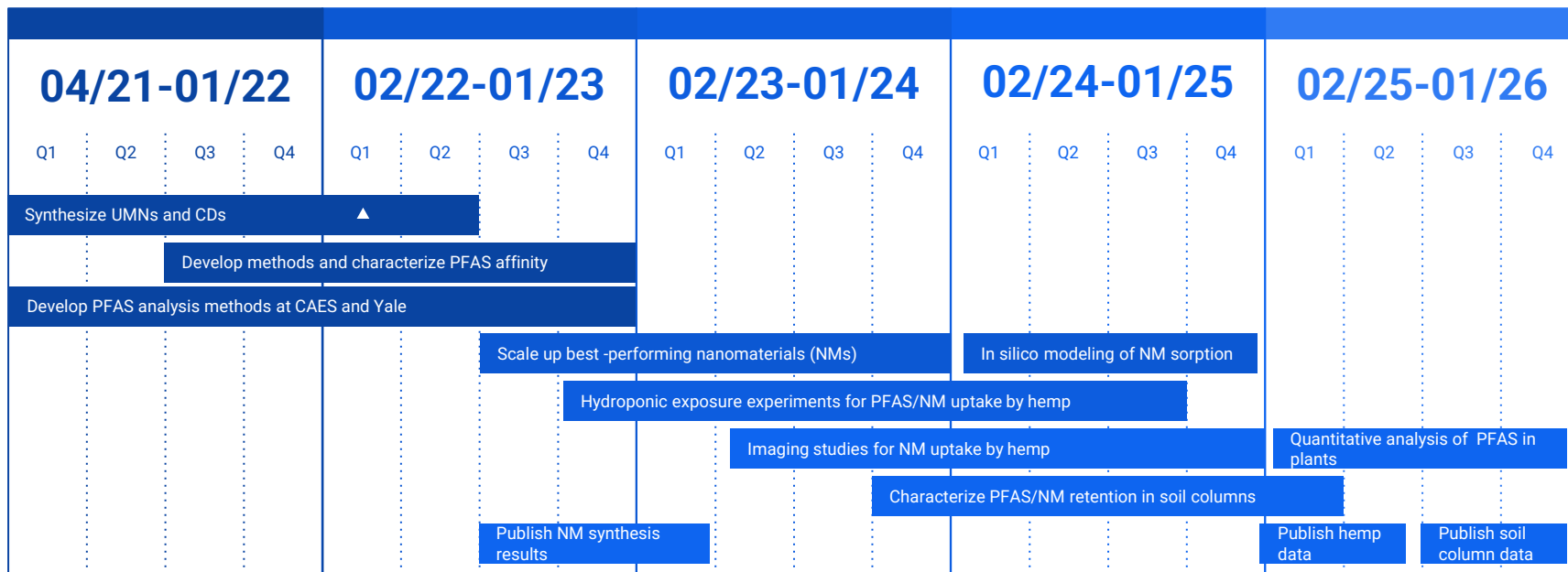
- **Haynes Group**, University of Minnesota - nanomaterial synthesis/characterization, analytical method development

Specific Aim 2: Determine capacity of NNMs to promote PFAS uptake into plants and retain PFAS in soils.

- **CAES team** - plant growth, PFAS measurement, soil column studies
- **Vasiliou Group**, Yale University - imaging studies and computational modeling

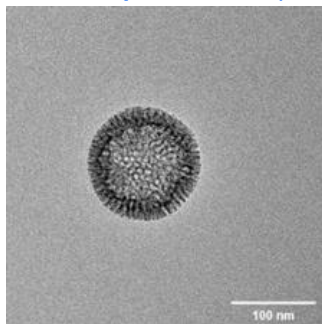


Overall Project Timeline

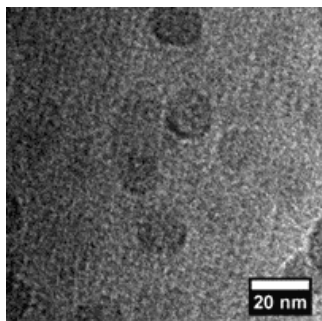


Overall Plan for Aim 1

Ultraporous Mesostructured
Silica Nanoparticles (UMNs):



Carbon Dots (CD):



PFOS



PFOA



PFBS



GenX



+



Characterize PFAS
affinity for UMNs with
varied:

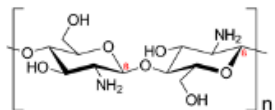
- Hydrophobicity
- Surface charge

Characterize PFAS
affinity for CDs with
varied:

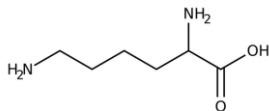
- Amine content
- Surface charge

Specific Aim 1: Carbon dots

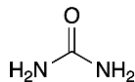
Carbon dot synthesis: microwave-assisted hydrothermal synthesis with amine rich precursors



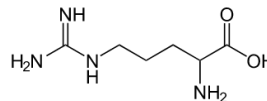
Chitosan



L-Lysine

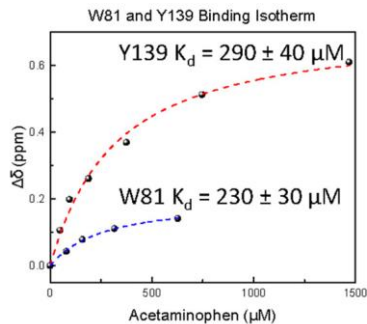


Urea

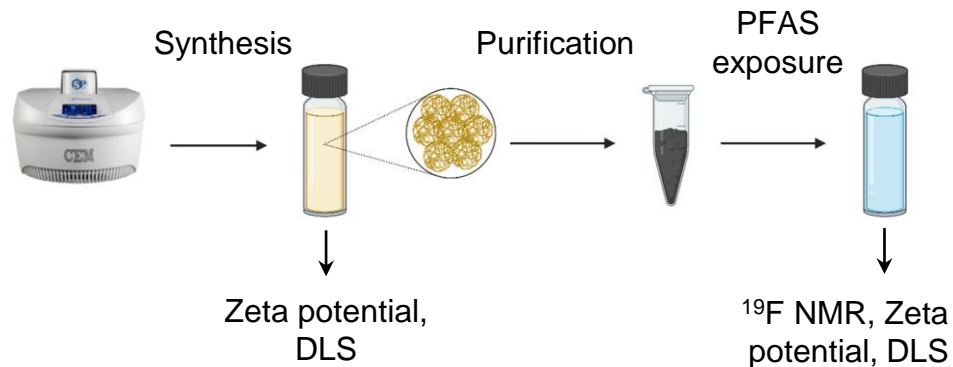


Arginine

Characterization and sorption assessment:
Zeta potential, dynamic light scattering (DLS), and ^{19}F NMR



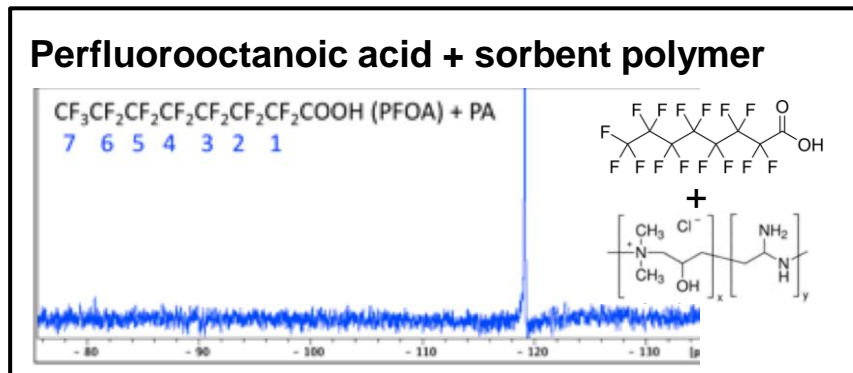
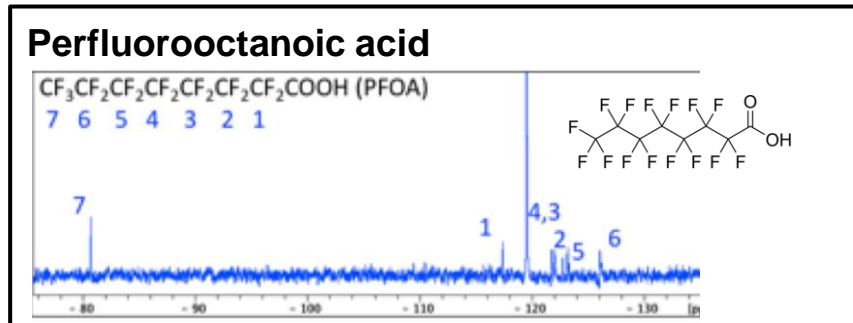
Experimental setup:



Why ^{19}F NMR?

Goal: determine most effective carbon dots at PFAS sorption

- Only detects fluorine
 - Assess CD/PFAS solution directly
- No sample processing (filter, column, etc)
- Gives insight into local environment of fluorine atoms
 - Changes in peak intensity and location
- Quantify interactions between PFAS and CDs to compare effectiveness



Using ^{19}F NMR to assess PFAS sorption

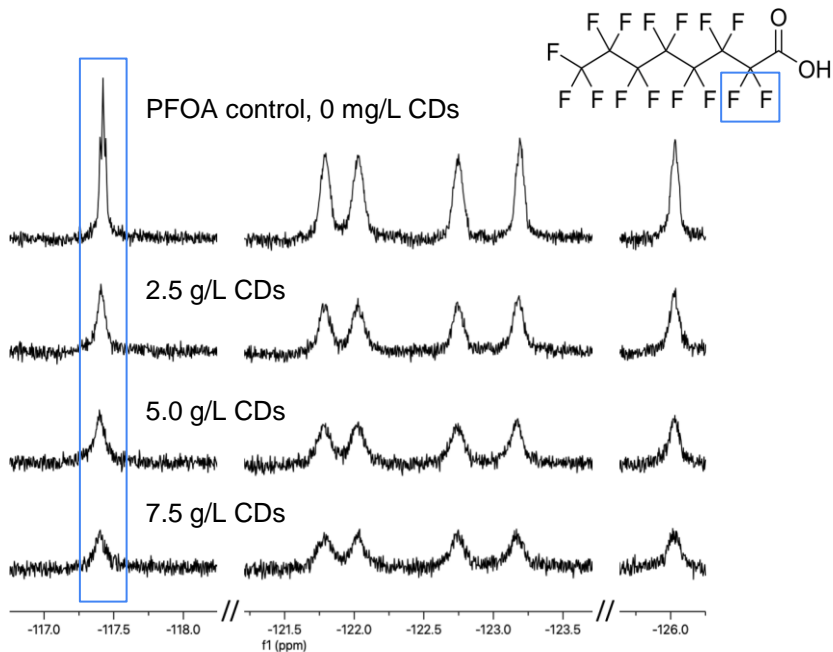


Fig 1. ^{19}F NMR of PFOA with increasing carbon dot concentrations

| | |
|------------------------------|--------|
| Chitosan, lysine, malic acid | |
| Zeta potential | +22 mV |
| Avg. size | 80 nm |

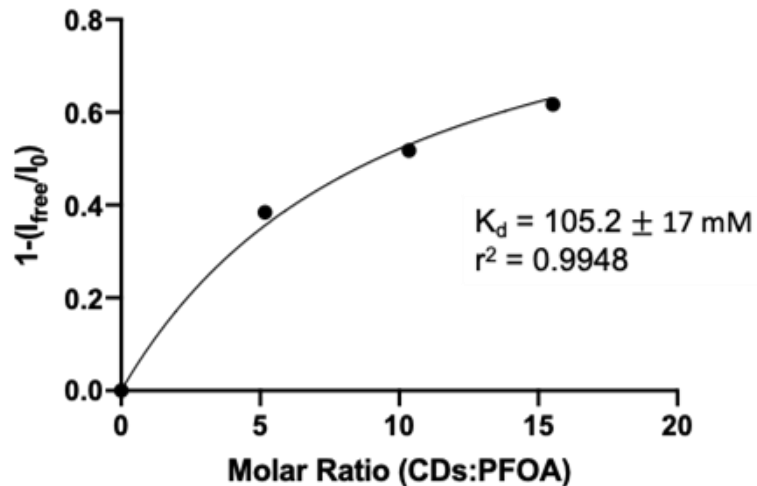
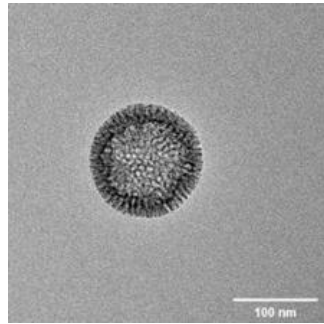
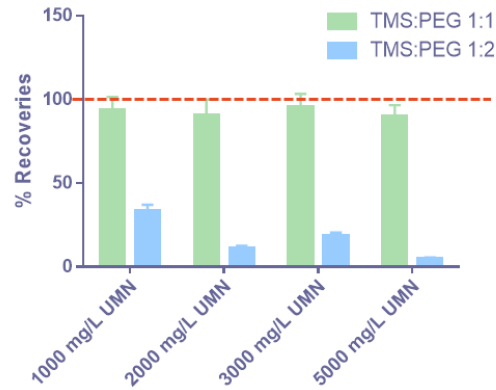


Fig 2. CD/PFOA binding curve from peak intensity changes ^{19}F NMR

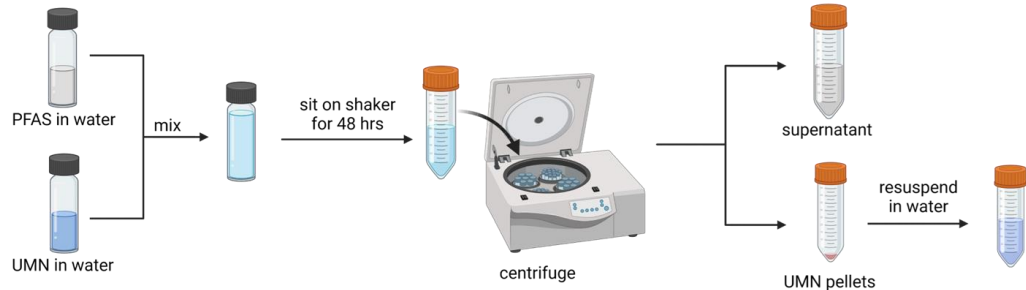
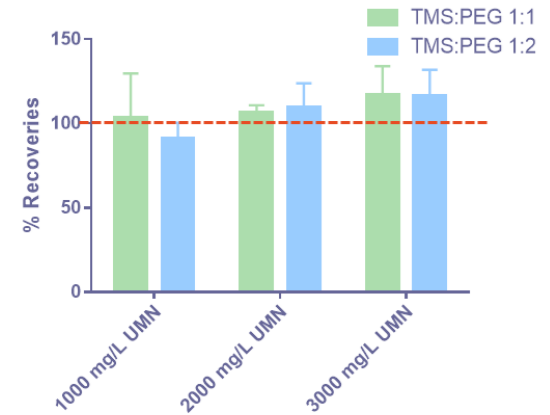
Specific Aim 1: Ultraporous Mesostructured Silica



Initial PFOA exposure data:

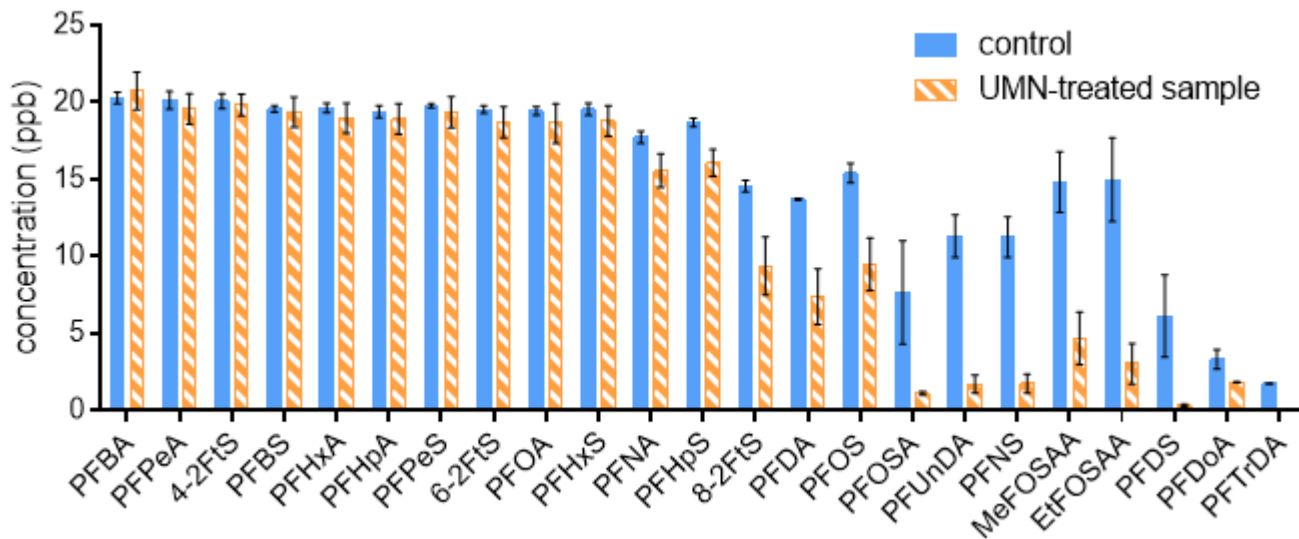


Initial GenX exposure data:



**% recovery < 100%: UMN
taking up PFAS**

CAES Preliminary Analysis of UMN/PFAS Sorption



The best early candidate silica nanoparticle has significant affinity for hydrophobic PFAS based on LC-HRMS analysis at CAES.

Acknowledgements

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