Treatment of Perchlorate in Water

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Perchlorate Treatment Systems are Operating

- Full-scale perchlorate treatment systems are commercially-available and operating, although treatment costs and operational issues may be significant
- Technologies in use include ion exchange and biological treatment

Progress Since 1997

- In 1997, little was known about perchlorate treatment
- Since 1997, many bench-scale and pilot-scale studies have been completed
- In 2002, full-scale systems are commercially-available and operating in California and elsewhere

Relevant Properties of Perchlorate The Bad News

- Not volatile
 - => Not removed by air stripping
- Highly soluble
 - => Not strongly adsorbed by GAC
- Stable in typical ground and surface waters
 - => Not readily chemically-reduced
 - => Does not readily biodegrade

Relevant Properties - The Good News

- Perchlorate is strongly retained by some synthetic resins
- Will biodegrade in an anoxic environment

Ion Exchange

- Removes undesirable ion(s) from water
- Undesirable ion replaced by chemicallysimilar but less toxic or nontoxic ion (e.g., Na⁺ replaces Ca⁺⁺)
- Ion exchange medium may be naturallyoccurring mineral or synthetic resin

Ion Exchange - Advantages

- Effective
- Commercially-available
- Familiar technology

Ion Exchange - Disadvantages

- Does not destroy the contaminant
- Produces a concentrated waste brine requiring further treatment or disposal
- Moderate to high cost

Full-Scale Ion Exchange Systems Operating in CA and NV

• 2,500 gpm system operated by La Puente

Valley County Water District (Los Angeles County)

• Full-scale system at Kerr-McGee facility (Henderson, NV)



La Puente Valley County Water District system

- Capacity: 2,500 gpm
- <u>Use of Treated Water</u>: drinking water supply
- Resin Regeneration: onsite
- Operational: 2000 to present
- <u>Vendor</u>: Calgon Carbon Corp
- Part of San Gabriel Valley/ Baldwin Park Superfund Site (1st phase of 26,000 gpm in treatment capacity)



La Puente Valley County Water District System - Cost and Performance

- Perchlorate Influent: approx. 100 ug/L
- Perchlorate in Treated water: < 4 ug/L
- Capital Cost: approx. \$2 million
- Operating Cost: approx. \$145/acre-foot



Brine Disposal/ Treatment

- At La Puente facility, waste brine discharged to dedicated brine disposal line
- Leading candidates for brine treatment are a high-temperature catalytic unit and anoxic biological treatment

Biological Treatment

- Microbes can chemically reduce undesirable chemical to less toxic or non-toxic chemicals
- Microbes of interest may require changes in physical/chemical environment to thrive (e.g., anoxic conditions, added nutrients)

Biological Treatment - Advantages

- Effective
- May be destructive technology (i.e., no waste brine produced)
- Potentially less expensive than ion exchange

Biological Treatment - Disadvantages

- Less familiar technology regulatory and community acceptance may be significant issues
- Requires post-reactor treatment to produce potable water

Full-Scale Biological Reactors Operating in California and Utah

- 3,600 gpm system operated by Aerojet-General Corporation (Northern CA)
- Continuously stirred tank reactors (CSTR) at Thiokol facility (Utah)

Aerojet-General Bioreactor System

- <u>Bioreactor Type</u>: fixed film on GAC operated as fluidized bed
- Total Capacity: 3,600 gpm
- End Products: primarily chloride and oxygen
- <u>Use of Treated Water</u>: aquifer recharge
- <u>Carbon Source/Electron Donor:</u> Denatured ethanol
- Operational: 1998 to present
- <u>Vendor</u>: Envirogen Inc.
- Part of Aerojet Superfund Site

Aerojet-General Bioreactor System - Cost and Performance

- Perchlorate Influent: approx. 2,500 ug/L
- Perchlorate in Treated water: < 4 ug/L
- Capital Cost: approx. \$5.5 million
- Reported Operating Cost: approx. \$65/acre-foot



Other Technologies

- Reverse Osmosis/Nanofiltration
- Biologically Active Carbon (BAC)
- Abiotic Reduction Using Ozone/GAC
- Chemical/Electrochemical Reduction

Sources of Perchlorate Treatment Technology Information

- May 2000 GWRTAC Technology Summary at: www.frtr.gov/perchlorate
- American Water Works Association Research Foundation at: http://www.awwarf.com/research/spperch.asp
- Vendors