

SRP Progress in Research Webinar Series

Utilizing Innovative Materials Science Approaches to Enhance Bioremediation: Session II - Chlorinated Compounds

Project

Enhancing bioremediation of groundwater co-contaminated by chlorinated volatile organic compounds (CVOCs) and 1,4-dioxane using novel macrocyclic materials

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Team

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Technical support



Problems and Solution



Problem #1: Low
1,4-dioxane concentration

Problem #2: CVOCs inhibit 1,4dioxane biodegradation

<u>Problem #3:</u> Opposite environmental conditions for biodegradation

Solution: <u>Component #1</u>: Cultures that efficiently degrade 1,4-dioxane at low concentrations

Component #2: Novel macrocyclic sorbents





Technical Approach



- Examined conformer stability and statistical distribution at ambient conditions. The most abundant (C7) conformer was used to model
 Pillar[6]arene (as shown in Table 1 and Figure 1).
- Probed for transport barriers relating to adsorption in different test macrocycles (as shown in Figure 2).
- Analyzed materials based on strength of adsorption CVOCs and 1,4dioxane on Pillararene-based macrocycles (as shown in Table 2).

Conformer of P6A	C1	C2	C3	C4	C5	C6	C7	C8
Relative Energy to ground state C7 (kJ/mol)	61.8	44	27.4	29.8	15.2	46.8	0	18.8
Population per billion at 298 Kelvin	0	113	41,554	35,463	6,340,558	19	990,663,288	2,919,006

Table 1 – Statistical (thermal/energetic) abundance of pristine Pillar[6]arene (P6A) conformers at 298K.



Figure 1 – C7 conformer ofFigure 2 –Pillar[6]arene (P6A).primary and

Figure 2 – Adsorption path energetics for 1,4-dioxane on a primary amine-substituted Pillar[6]arene (P6A) macrocycle.

	Pristine P6A	Dimethoxy P6A	P6A with	P6A with	P6A with
			primary amine	carboxylic acid	methylbromide
1,1-dichloroethylene	-44.77	-61.74	-66.03	-63.52	-66.97
Cis-dichloroethylene	-48.08	-62.44	-66.54	-68.51	-69.66
Trichloroethylene	-43.15	-69.85	-70.84	-73.52	-74.54
1,4-Dioxane	-59.69	-78.95	-82.30	-81.91	-85.20

Table 2 – Binding Energies (in kJ/mol) for 1:1 adsorption of adsorbate onPillar[6]arene (P6A)-based macrocyclic materials. (More negative value represents
stronger exothermic adsorption)5

Monomers:















Activated sludge





Pure culture identification (**5 cultures** identified so far)





1,4-dioxane degradation:
$\frac{dS_d}{dt} = -X \boldsymbol{q_d} \left[\frac{S_d}{K_s + S_d} \right]$
Biomass growth:
$\frac{dX}{dt} = \mathbf{Y} X \mathbf{q}_d \left[\frac{S_d}{K_s + S_d} \right] - \mathbf{b} X$

Pure culture	$\frac{q_d}{d} (mg \ dx/mg \ pr \ \cdot d)$	K_{s} (mg dx/L)	$\mathbf{Y}(mg \ pr/mg \ dx)$	b (d ⁻¹)
WC10G	0.47	8.24	0.36	0.02



Good fit for 1,4-dioxane degradation at low concentrations

Conclusions

- > Specific Aim 1
 - **Material modeling**: Macrocycles tend to exist in one conformer that dominates at relevant conditions. Adsorption shows no transport barrier, indicating energetics can drive selectivity. Pillar[6]arene and its variants are predicted to be weakly selective for 1,4-dioxane over CVOCs.
 - Material synthesis and characterization: β-Cyclodextrin adsorbents showed strong selectivity towards
 CVOCs in a mixture of 1,4-dioxane and CVOCs; they adsorbed CVOCs but not 1,4-dioxane.

- > Specific Aim 2
 - Microbial enrichment: We enriched six mixed cultures through feeding 1,4-dioxane at a low concentration.
 - Microbial isolation and characterization: We identified five pure cultures and characterized one pure culture (WC10G), which seems a good fit for degrading 1,4-dioxane at low concentrations: low K_s , b, q_d , and high Y.





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