

ENGINEERING SERVICE CENTER Port Hueneme, California 93043-4370

CONTRACT REPORT CR-04-002-ENV

ASSESSING THE FEASIBILITY OF DNAPL SOURCE ZONE REMEDIATION: REVIEW OF **CASE STUDIES**

by

GEOSYNTEC CONSULTANTS 130 Research Lane, Suite 2 Guelph, Ontario, Canada N1G 5G3

May 2004

Approved for public release; distribution is unlimited.



Prepared for:

NAVAL FACILITIES ENGINEERING SERVICES CENTER NFESC ESC 411 1100 23rd Ave Port Hueneme, CA 93043

ASSESSING THE FEASIBILITY OF DNAPL SOURCE ZONE REMEDIATION: REVIEW OF CASE STUDIES

Prepared by:



130 Research Lane, Suite 2 Guelph, Ontario, Canada N1G 5G3 GeoSyntec Project Number TR0132 April 2004

EXECUTIVE SUMMARY

The Naval Facilities Engineering Services Center (NFESC) hired GeoSyntec Consultants (GeoSyntec) to conduct a survey of the application of chlorinated solvent dense non-aqueous phase liquid (DNAPL) source remediation technologies. The overall goal of this project was to gather information on the relative successes of remediation applications under different site conditions in order to initiate the development of guidelines on DNAPL source treatment. Members of the environmental community that had attempted to remediate DNAPL source zone areas were invited to participate in a web-based survey, which remained open for several months. Survey respondents were asked a number of questions dealing with site-specific issues (*e.g.*, geology, hydrogeology, contaminant distribution, monitoring network) and technology specific issues (*e.g.*, type of remedial technology, remediation stage, cost, treatment effectiveness). The number of survey respondents totaled 213. Of the 213, complete sets of data were obtained for 118 locations. Data from 21 published case studies was also entered into the survey. This report also includes a brief overview of chlorinated solvent DNAPL contamination and remediation processes (Section 2).

The data compiled from the survey was analyzed to detect correlations between remedial application success and site conditions. The data is presented in detail in Section 4 of the report. The following observations and correlations were established upon data analysis:

- <u>Site Characteristics</u>: The majority of the sites had areal extents of 10,000 ft² to 100,000 ft² and the volume impacted was greater than 100,000 ft³. A large number (89%) of the sites consisted of unconsolidated material and the minority (11%) had consolidated material. Within the unconsolidated material sites, 39% of them comprised sandy soils, 19% silt, and 15% clay soils.
- <u>Source Zone Characteristics</u>: The majority of the locations had chlorinated ethenes in the source zone area but some had a combination of other DNAPLs (chloroethanes, chloromethanes and chlorobenzenes). The median concentrations reported at these locations were 56mg/L and 100 mg/L for PCE and TCE, respectively. Of the 75 sites with DNAPL distribution data, 83% had residual DNAPL, 61% had sorbed DNAPL, 44% had pooled DNAPL, 40% had DNAPL diffused into low K layers and 11% had DNAPL trapped in dead-end fractures. The average survey respondent reported having a maximum DNAPL depth between 10 to 100 ft bgs. However, close to 8% of the users reported having source zone areas between 100 to >1,000ft bgs. Approximately one third of the survey users (31%) reported having an irregular shape source zone areas.
- <u>DNAPL Source Zone Remedial Technologies Used</u>: A number of technologies were applied to remediate DNAPL source zones. Thermal technologies, *in situ*

chemical oxidation and bioremediation were each used in roughly 20% of the cases. Dual Phase extraction and excavation were used at 11% and 9% of the locations respectively. Surfactant flushing and zero valent iron (ZVI) were also used at a few locations.

- <u>Remediation Cost</u>: The average cost for full-scale applications was \$2.8M, with the largest cost being a \$15M water/DNAPL dual-phase treatment (on-going) at a site with a DNAPL impacted zone that was in the range of 100,000 to 1,000,000 ft². The lowest cost for full-scale applications was \$75K for a bioremediation application and a ZVI application, both at small sites. The majority of the pilot tests cost less than \$0.5M with no cost greater than \$2M. Thermal treatment, pump and treat, and dual-phase extraction applications appeared to be significantly more expensive than chemical oxidation cases on large sites.
- <u>Treatment Duration</u>: Estimated treatment durations were as follows: Dual Phase Extraction: 60 years; Pump & Treat: 158 years; Chemical Oxidation: ~4 years; Thermal Technologies: ~4 years; ZVI Technologies: ~4 years; and Bioremediation: ~4 years. Treatment duration was estimated based on 16 full-scale applications both on-going and completed.
- <u>Evaluating Success</u>: Evaluating success at the survey locations was difficult since a rigorous statistical analysis was not conducted due to budget limitations. Hence, success criteria were established based on estimated mass removal, decrease in mass flux and the occurrence of rebound. Only one remediation application, a chemical oxidation using permanganate, met the most stringent criteria for evaluating success (>80% mass removal, >81% reduction in mass flux, and no rebound). This site was a pilot scale demonstration in sand with a controlled DNAPL release. If the success criteria are relaxed slightly to include >61% reduction in mass flux then an additional remediation application (bioremediation) is added. This site was also a pilot scale application conducted in sand.
 - <u>Mass Removal</u>: Fourteen (14) locations had ≥80% source mass removal and of these, 4 sites claimed to have 100% source mass removal (one excavation pilot test, one thermal six-phase heating and one thermal conductive heating [both full-scale], and one chemical oxidation pilot test using permanganate). Of the 4 sites with 100% mass removal, three had unknown mass flux reduction and/or unknown rebound.
 - Mass Flux: Thirteen (13) locations had ≥81% reduction in mass flux. The 13 cases include 5 thermal six phase heating applications (1 full scale), 2 permanganate and 2 Fenton's applications, 2 bioremediation applications, 1 surfactant and 1 steam application.

- <u>Success v. Media</u>: None of the technology applications in consolidated material were deemed to be successful. This may be due to the fact that only 11% of the locations included in the survey had consolidated media, but it may also be due to the fact that remediating source zone areas in consolidated material remains a big challenge.
- <u>Meeting MCLs</u>: None of the remediation attempts presented in this survey/review achieved MCLs or regulatory site closure. Meeting MCLs was not always the reason source reduction was attempted, and there are other tangible and intangible criteria that are used to interpret success.

The data collected from this survey suggests that DNAPL remediation efficacy is extremely difficult to gauge, and although employing aggressive remedial technologies cannot achieve site closure, significant mass removal can be achieved.

TABLE OF CONTENTS

Page

1	INTRODUCTION 1		
	1.1	OBJECTIVES 1	
	1.2	SCOPE OF WORK AND REPORT ORGANIZATION	
2	OV	ERVIEW OF DNAPL REMEDIATION 4	
	2.1	BACKGROUND ON DNAPLS	
	2.2	DNAPL BEHAVIOR IN THE SUBSURFACE	
	2.3	DNAPL REMEDIATION PROCESSES	
	2.4	REMEDIAL SUCCESS	
	2.4.1	Definition of Remedial Success	
	2.4.2	Factors Impacting Remedial Effectiveness	
	2.4.3	Measurement of Remedial Success 12	
3	3 OVERVIEW OF DNAPL REMEDIAL TECHNOLOGIES		
	3.1	SURFACTANT AND CO-SOLVENT FLUSHING	
	3.2	BIOREMEDIATION	
	3.3	CHEMICAL OXIDATION	
	3.4	THERMAL TECHNOLOGIES	
	3.5	ZERO-VALENT IRON	
	3.6	DUAL PHASE/MULTIPHASE EXTRACTION	
4 SURVEY RESULTS		RVEY RESULTS	
	4.1	SURVEY AND LITERATURE DATA COLLECTION METHODS	
	4.1.1	Web-Based Survey	
	4.1.2	Review of the Literature	
	4.2	DATA QUALITY	
	4.2.1	Data Information Check	
	4.3	SUMMARY OF SURVEY RESULTS	
	4.3.1	Remedial Technologies Identified in Survey	

TABLE OF CONTENTS (CONT'D)

Page

	4.3.2 A	quifer Geology/Lithology	38		
	4.3.3 A	quifer Hydrogeology	40		
	4.3.4 D	NAPL Source Zone Characteristics	40		
	4.3.4.1	Source Area Chemistry	40		
	4.3.4.2	Source Area Size and Distribution Parameters	41		
	4.3.4.3	DNAPL Detection and Mass Estimation	43		
	4.3.5 Ti	reatment Cost	43		
	4.3.6 Ti	reatment Duration	44		
	4.3.7 Te	echnology Performance Evaluation	44		
	4.3.7.1	Perception of Technology Effectiveness	45		
	4.3.7.2	Factors Impacting Remedial Effectiveness	46		
	4.3.7.3	DNAPL Mass Removal	46		
	4.3.7.4	Mass Flux Decrease	47		
	4.3.7.5	Post-Treatment Monitoring and Rebound of Dissolved Phase			
	Concentrations				
	4.3.7.6	Achievement of Remedial Goals and Site Closure	48		
	4.3.7.7	Impacts to Secondary Groundwater Quality	49		
	4.3.7.8	Overall Technology Performance	49		
5	CONCL	USIONS	51		
6	RECOM	MENDATIONS	54		
7	REFERI	ENCES	55		

LIST OF TABLES

MPE Terms and Configurations

Table 3.2:	Optimal Conditions for Multiphase Extraction
Table 4.1:	Breakdown of Primary Technologies
Table 4.2:	Summary of the breakdown of the predominant geologic material in the source area
Table 4.3:	Number of Methods Used to determine DNAPL Presence
Table 4.4:	Summary of Remedial Cost Data
Table 4.5:	Summary of Perceived Success
Table 4.6:	Summary of Source Mass Removal

Table 3.1:

- Table 4.7:Summary of Mass Flux Decrease
- Table 4.8:
 Summary of Post Treatment Monitoring Stage
- Table 4.9:Summary of Rebound Data
- Table 4.10:
 Summary of impacts to secondary groundwater quality data

LIST OF FIGURES

- Figure 4.1: Survey Respondents by Role in Groundwater Remediation Community
- Figure 4.2: Geographic distribution of Survey Respondents
- Figure 4.3: Number of Sites with Technology Data Entered in the Survey
- Figure 4.4: Number of Sites with Lithology Data Entered Into the Survey
- Figure 4.5: Overall Distribution of DNAPL Composition
- Figure 4.6: Summary of DNAPL Composition Sorted by Technology
- Figure 4.7: Actual Cost of Technology Applications
- Figure 4.8: Summary of Perceived Success Sorted by Technology
- Figure 4.9: Summary of Source Mass Removal Sorted by Technology
- Figure 4.10: Summary of Mass Flux Decrease Sorted by Technology
- Figure 4.11: Summary of Rebound Sorted by Technology
- Figure 4.12: Summary of Secondary Groundwater Impacts Sorted by Technology

LIST OF APPENDICES

- Appendix A: Case Studies
- Appendix B: Screen Shots of the Web Based Survey
- Appendix C: Detailed Survey Data Summary
- Appendix D: Survey Data Comparison and Interpretation

1 INTRODUCTION

GeoSyntec Consultants (GeoSyntec) was retained by the Naval Facilities Engineering Services Center (NFESC) to conduct a survey of the application of chlorinated solvent dense non-aqueous phase liquid (DNAPL) source remediation technologies. The overall goal of this project was to assist NFESC to initiate the development of guidelines on DNAPL source treatment by gathering information on current technologies used to treat DNAPL sources and their relative successes under various site conditions.

1.1 Objectives

The development of promising source zone treatment technologies has changed the conventional belief that DNAPL removal is often "technically impracticable". Although obtaining maximum contaminant levels (MCLs) near and within source areas after the application of DNAPL treatment technologies has proven to be an elusive goal, ongoing development and testing of innovative remedial technologies suggest that source zone remediation can cause significant reduction of mass discharge and dissolved phase concentrations of volatile organic compounds (VOCs) from source areas. However, the selection of a source zone remedial technology that is cost effective and will produce the best results within the specified time frame for a specific site is not a simple task. The advantages and limitations associated with source remediation must be carefully considered in order to have reasonable expectations for the remediation. Limitations of remedial technologies due to such things as geologic heterogeneities, depth and lateral extent of contamination must be factored into the technology choice as well. Currently no guidelines for technology selection exist, and little information is available to provide direct comparisons of technology effectiveness for all site conditions. Guidelines for technology selection need to be developed to:

- *i)* simplify the technology selection process;
- *ii)* provide defensible arguments for: (a) technology selection; (b) development of official remedial objectives that can reasonably be met; and (c) defense for no action for source remediation where appropriate; and
- *iii)* increase the overall effectiveness and minimize the costs of DNAPL remediation attempts by optimizing the chosen remedial technology to the site conditions.

This report represents the first step towards developing such guidelines through conducting a review of DNAPL treatment technologies that have been field-deployed.

As such, the objectives of the review of DNAPL remedial technologies outlined in this report were as follows:

- *i)* a brief overview of chlorinated solvent DNAPL contamination and remediation processes (Section 2);
- *ii)* an overview of how each DNAPL remediation technology removes DNAPL from the subsurface, the advantages and limitations of the technology, the range of site conditions where it may/may not be applicable, and information on technology application, including a summary of select published case studies (Section 3 and Appendix A);
- *iii)* a review of data compiled from a web-based survey developed to collect information on field applications of DNAPL remediation, including some site characteristics, costs, and remedial effectiveness data (Section 4 and Appendices B, C and D);
- *iv)* conclusions (Section 5); and
- v) recommendations for further work based on the findings of our review and analysis (Section 6).

This report contains a compilation of published information, as well as information collected from survey respondents from all aspects of the groundwater community. The accuracy of the information and data collected from the survey and from the published case studies could not be independently verified for the majority of sites; however, the data was reviewed to identify obvious errors (as discussed in Section 4.2.1). The information is presented here as it was provided by the survey respondents.

1.2 Scope of Work and Report Organization

There are many technologies and approaches available to treat the vadose and dissolved phase contaminants derived from chlorinated organic solvent DNAPLs. However, the scope of work for the project was to address remedial technologies that are applied to treat DNAPL source areas only. The review of DNAPL remediation technologies was also limited to technologies that have been used to clean up contaminated sites with chlorinated solvent DNAPLs [*e.g.*, chlorinated ethenes such as trichloroethene (TCE), chlorinated ethanes such as 1,1,2-trichloroethane (1,1,2-TCA), chlorinated methanes such as dichloromethane (DCM), and chlorinated benzenes such as pentachlorobenzene (PeCB)] that are present beneath the watertable (*i.e.*, saturated zone). Technologies used to target contamination originally present in the unsaturated zone.

Information was gathered on sites where the presence of DNAPL was either suspected or known.

A review of peer reviewed/gray published literature (Sections 2 and 3) was conducted to compile information on the following:

- *i)* the mechanisms of DNAPL behavior in the subsurface;
- *ii)* the theory and application of DNAPL remedial technologies that have been attempted in a field setting; and
- *iii)* a review of published case studies.

A key effort of this project was the design of a web-based survey to collect information about DNAPL remedial applications at sites for which the data is not obtainable through the literature. Survey participants from the larger environmental community (*e.g.*, consultants, regulators, site owners, *etc.*) were solicited through a number of large email campaigns, and posting in widely distributed environmental newsletters and web pages. Data from published case studies was also entered into the survey, and the data compiled from the survey was analyzed to detect correlations between remedial application success and site conditions (Section 4). The information gathered from the literature review and the survey was used to evaluate whether sufficient information exists to develop technology selection criteria (Section 5) and to develop recommendations for further work in achieving this goal (Section 6).

2 OVERVIEW OF DNAPL REMEDIATION

An overview of the "DNAPL Paradigm", including background information on DNAPLs (Section 2.1) and their behaviour in the subsurface (Section 2.2) is presented below. Coupled with this overview is a discussion on the DNAPL remedial process, including removal mechanisms (Section 2.3) and remedial success parameters [*i.e.*, various definitions of "success" (Section 2.4.1), factors impacting remedial success (Section 2.4.2) and success metrics (Section 2.4.3)].

2.1 Background on DNAPLs

Federal and state regulations are one of the main drivers for undertaking remediation of contaminated groundwater sites. Remedial actions are also undertaken to eliminate long-term liability due to such things as human health impacts, diminution of property values, damage to natural resources, etc. Often the legal, regulatory and societal pressures to remediate contaminated sites within a relatively short time frame does not allow the responsible parties to fully understand the available technologies and set reasonably achievable remedial goals for a site. For groundwater that is considered to be a potential or actual source of drinking water, the standards for remediation are either maximum contaminant levels (MCLs) for regulated organic chemicals as specified under the Safe Drinking Water Act or risk-based concentrations based on site-specific human However, obtaining these goals has proven elusive for health risk assessments. chlorinated organic solvents, specifically chlorinated aliphatic hydrocarbons (CAHs) such as chlorinated ethenes / ethanes / methanes and chlorinated benzenes. CAHs are among the most common groundwater contaminants and, because of their physical characteristics (see Section 2.2 below), are also among the most difficult groundwater contaminants to remediate (Pankow and Cherry, 1996).

These CAHs are non-aqueous liquids that are denser than water, and are often used and released to the subsurface as "free-product" liquids (*i.e.*, DNAPLs). They are found at approximately 80% of all Superfund sites with groundwater contamination and are by far the most prevalent contaminants at Department of Defense (DoD) sites, occurring at more than 3,000 DoD sites in the United States (Environmental Protection Agency, 1997). Based on the NORM database (March 2003), the Navy has 867 chlorinated solvent impacted sites. Of the 399 sites impacted with TCE, it is estimated that 12% of the DoD sites have a DNAPL source remaining in the subsurface (based on groundwater TCE concentrations greater than 10 mg/L). Of the 213 sites impacted with PCE, it is estimated that 9% of the DoD sites have a DNAPL source remaining in the subsurface (based on groundwater PCE concentrations greater than 1.5 mg/L).

The recognition of the difficulties that cleanup of these CAHs pose resulted in the USEPA publishing many technical guidance documents on characterization and

remediation of DNAPL sites, and to recognize that DNAPL remediation presents significant technical challenges. Section 2.4.2 outlines some of the factors impacting remedial effectiveness. In 1993, the USEPA published a guidance document on technical impracticability (TI) to be used for assessing the feasibility of meeting established performance goals at Superfund sites where remediation was considered impracticable from "an engineering perspective, taking cost into consideration" (EPA TI Guidance Document, 1993). However, since 1990, only about 50 TI waivers have been granted, out of over 1500 Superfund sites (USEPA Report of an Expert Panel, In Press).

The USEPA, the Department of Energy (DoE), the DoD and various private sector organizations have invested substantial resources in DNAPL investigations and remediation attempts. However, to date the most common remediation techniques used for DNAPL impacted sites are containment technologies, such as pump and treat, slurry walls or permeable reactive walls. These remedial technologies, although often effective at limiting the spread of contamination at these sites, require at a minimum long term monitoring and institutional controls and may require very large operation and maintenance costs. Pump and treat and other containment processes have not fully remediated sites with DNAPL source areas. Several technologies have been proven to remove mass from the source zone; however, without near complete source zone removal, few if any sites have been remediated to drinking water standards or background levels. The USEPA has commissioned an expert panel to evaluate the benefits of partial source removal. Advocates of partial source removal suggest that this would ultimately speed up the remedial activities, reduce long-term risks, and reduce the mass transfer of contaminants into the aquifer, which would lower groundwater concentrations and reduce plume size.

2.2 DNAPL Behavior in the Subsurface

A DNAPL released to the subsurface will seek phase equilibrium, a condition in which all acting influences are canceled by others, resulting in a stable, balanced, or unchanging system. The DNAPL will remain as a NAPL, adsorb to soil, dissolve in groundwater, or volatilize into soil gas to the extent defined by the physical and chemical properties of the individual CAH and the subsurface environment.

DNAPL movement through the subsurface is primarily governed by the conflicting forces of gravity and capillary forces. The primary driving force for DNAPL migration is gravity. DNAPLs are denser than water and thus will tend to mobilize downwards as opposed to spreading on the water table like hydrocarbons. As a result, a DNAPL released to the subsurface has the capacity to sink below the water table where it can provide a long-term source of contamination. The primary arresting force is the capillary force (P_c), which is dependent on the interfacial tension at the interface between the DNAPL and the water or air. The capillary force will vary with changes in the radius of

curvature of the interface caused by variations in the soil pore size; small geological heterogeneities (often indistinguishable with site characterization methods) can have significant impact on the migration pathways taken by the DNAPL. This can lead to complicated DNAPL source zone architecture and significant DNAPL mass deeper and/or over greater lateral extent than expected.

As DNAPL travels through the subsurface, a small quantity is retained as residual saturation in every pore through which it flows; this portion of the DNAPL is referred to as residual DNAPL. As the DNAPL travels downward, it will accumulate on the top of soil layers or fractures with entry pressures (i.e., the capillary pressure that must be exceeded for DNAPL to pass through a constricting pore throat into a pore) greater than the capillary pressure, forming a "pool". This pool will spread laterally until it either: *i*) reaches the edge of the confining layer; or *ii*) until sufficient height of free product is accumulated on the layer to exceed the entry pressure of the layer, at which point the DNAPL will enter the confining layer and continue to migrate downward. The architecture of the source may vary depending on the volume of DNAPL released, the geologic heterogeneity encountered, and the length of time since the release occurred, DNAPL is typically found as multiple horizontal layers or lenses, connected by vertical pathways at residual saturation with one ore more pools above fine grained layers. Only DNAPL that occurs in pools is considered to be potentially mobile under natural conditions. The mobility of residual and pooled DNAPL can be enhanced through a variety of remedial techniques, including heat or chemicals, which modify the interfacial tension between the water and the DNAPL. DNAPL can also be trapped in dead end pores or fractures, or diffused into low permeability layers.

2.3 DNAPL Remediation Processes

A number of DNAPL source zone technologies have been developed to optimize the mass transfer, *in situ* destruction and advective/dispersive DNAPL removal mechanisms. These technologies can be subdivided into three overall categories: DNAPL removal technologies, *in situ* destruction technologies, and source zone immobilization technologies.

Remediation is carried out by one of more of the following mechanisms:

Dissolution: Technologies dependent on dissolution rely on increasing the solubility of the contaminant through the use of chemical additives (surfactants and/or co-solvents) or by increasing the temperature (thermal technologies). The dissolved contaminant is then extracted with the groundwater and treated *ex situ* as in conventional pump and treat systems. Technologies employing this mechanism include surfactant and co-solvent

flushing and the thermal technologies steam injection and three- and sixphase heating (also known as electrical resistive heating).

- *ii)* **Displacement:** Technologies that involve displacement of the DNAPL either reduce the capillary forces by reducing the interfacial tension between the DNAPL and groundwater (through the use of surfactants, cosolvents or heat), or increase the driving forces through enhancing the gradient across the DNAPL (*e.g.*, dual-phase extraction). The DNAPL is extracted along with the groundwater and treated *ex situ*. Technologies employing this mechanism include surfactant and co-solvent flushing, dual-phase extraction, and the thermal technologies steam injection and three- and six-phase heating.
- *iii)* **Volatilization:** Technologies designed to remove DNAPL *via* volatilization involve increasing the transfer of contaminants to the vapor phase and inducing the flow of vapor through the contaminated zone. The vapor is then extracted via wells for above ground treatment. It is possible to enhance the volatilization of the contaminants using heat. Technologies relying on this approach include the thermal technologies steam injection and three- and six-phase heating and dual-phase extraction.
- *iv)* **Excavation:** Contaminated soil containing DNAPL is extracted and treated *ex situ*. Contaminants remaining in the groundwater as dissolved phase are not removed.
- v) Destruction: In situ destruction technologies include technologies that destroy the contamination through chemical reaction or biodegradation. Most DNAPL components are relatively resistant to oxidation under naturally occurring conditions; however, through the addition of a strong oxidizer it is possible to degrade the contaminants. Most DNAPL components may be more readily broken down by reduction through the addition of a reducing compound such as zero valent iron (Fountain, 1998) or a chemical oxidant such as permanganate. Bioremediation is an *in situ* destruction process that involves the breakdown of contaminants through biologically-mediated reactions. This can occur naturally, be enhanced through the addition of an electron donor/acceptor, or be stimulated through the addition of both an active bacteria culture and an electron donor/acceptor. Technologies relying on this mechanism include the thermal technology electrical conductive heating, chemical oxidation, bioremediation and zero valent iron.

vi) **Immobilization:** Source zone immobilization does not actually remediate the DNAPL source zone but isolates the source from the surrounding environment, thereby mitigating risks. The source zone immobilization techniques include barrier wall, reactive barrier walls, pump and treat, and vitrification.

2.4 Remedial Success

2.4.1 Definition of Remedial Success

The definition of remedial success may change according to the site characteristics, regulatory demands, land use needs, *etc.* and may include the following factors:

- *Mitigate Risk to the Environment:* Risk management may include either partial or complete DNAPL mass removal, plus the reduction of dissolved phase mass flux downgradient of the source area. Containment of dissolved plumes and/or source areas may also be a remedial goal. Remedial success is often evaluated in terms of meeting established cleanup goals that may be based on MCLs or risk-based levels intended to reduce the risk to human health and the environment.
- *ii)* **Reasonable Treatment Duration/Application:** A remedial treatment method may be successful in terms of its ability to remove DNAPL mass from the subsurface; however, the time required to complete remediation may be unreasonably long. Restraints may be placed on the treatment duration by regulators or site owners who may wish to use the contaminated land for purposes that will be impacted by the presence of contamination. Also, ongoing invasive or disruptive remediation activities or systems may be undesirable for some site uses.
- *iii)* **Reasonable Cost of Implementation:** Similarly, effective remedial treatments may be unreasonably expensive to implement at some sites. Two of the ways that treatment costs are evaluated are net present value (NPV) or cost to complete (CTC). For net present costs, often used in industry, the costs on a per year basis may be more appealing (*i.e.*, \$250K per year for 100 yr) then a larger lump sum (*i.e.*, \$5 million in one year) up front for faster remediation. However, the government more often looks at the cost to complete the remediation and would more likely select the \$5 million over the \$25 million. It is difficult to compare the cost of remediation methods based on prior experience at other sites, since cost is influenced by factors such as the size of treatment area (area and volume) as well as the amount of DNAPL mass targeted within that area. Unit cost

measures (*e.g.*, cost per area, cost per volume, cost per unit mass) are a more effective means of comparing remedy costs.

2.4.2 Factors Impacting Remedial Effectiveness

A number of factors may have a significant impact on the effectiveness of the DNAPL remediation process. Many of these factors are listed below, along with brief discussions of their impact.

1) Inadequate delineation of the DNAPL location

One of the key factors in remediating DNAPL source zones is adequately defining the location and composition of the source, including both macro-scale and local distributions of DNAPL. However, as described in Section 2.2 above, the typical complex spatial distribution of DNAPLs in the subsurface makes determination of the actual DNAPL extent difficult. To collect sufficient data at the scale needed to identify the source zone, data must be collected on the order of meters to tens of meters and this is often costly and subject to error. Samples of the aqueous, gaseous, solid and immiscible phases need to be collected to determine the soluble, volatile, sorbing and immiscible components of the DNAPL.

Other limiting factors in the delineation of DNAPL source zones include the indirectness of the methods and the risk involved in direct source zone characterization. None of the remote sensing techniques developed for locating DNAPLS (*e.g.*, geophysical methods, partitioning interwell tracer tests, and natural radon abundance) have proven effective in defining the DNAPL extent at meaningful resolutions (Stroo *et al.*, 2003). The use of source zone delineation and characterization methods that disturb the subsurface environment (*e.g.*, soil borings, groundwater samples) within DNAPL source zones is risky due to the possibility of creating a pathway for migration of mobile DNAPL. This could significantly worsen the contamination problem, particularly at sites where the DNAPL was contained by a low permeability layer.

2) Proper characterization of important physical and chemical site attributes

Not only is the DNAPL source architecture highly dependent on geologic heterogeneities, the performance of most technologies is also highly affected by these heterogeneities and therefore the hydrogeology of the contaminated volume must also be carefully determined (Fountain, 1998). In addition to the physical attributes of the source zone such as heterogeneity and hydrogeology, the chemical composition of the DNAPL is also important for the selection, design and performance of remedial technologies. The chemical composition of the NAPL affects properties like interfacial tension, viscosity, density, wettability, solubility, vapor pressure, Henry's Law constants, biodegradability,

oxidation potential, and other properties that affect the performance and therefore the selection of different technologies.

3) Degree of reliance on dissolution or volatilization of DNAPL

For technologies that are dependent on the dissolution or volatilization of the DNAPL, the contaminant distribution is of additional importance. A number of factors will affect the degree to which DNAPL dissolution/volatilization can be accelerated in both porous and fractured geologic media, including the surface area of the DNAPL across which mass transfer can occur (i.e., DNAPL present in pools versus residual) and the fraction of DNAPL stored in soil layers and/or fractures disconnected from the principal groundwater flow pathways. The specific interfacial surface area (surface area to volume ratio) of DNAPL pools is much lower than that of residual DNAPL; this is exaggerated in fractures due to the two-dimensional nature of fractures. For rocks with a porous matrix, significant diffusion of DNAPL into the rock matrix can occur (Parker et al., 1994); removal of the DNAPL stored in the rock matrix is then limited by the rate of diffusion from the matrix. Similarly, low permeability soil layers or dead-end or hydraulically-disconnected fractures in bedrock can prevent advective removal of contaminants in the dissolved/vapor phase, which limits mass removal from DNAPL trapped in these locations. Destruction of the dissolved phase contaminant at the edge of the hydraulically-disconnected areas will accelerate dissolution and diffusion of the DNAPL from these locations.

When dissolution/volatilization of a pool of DNAPL by abiotic flushing is occurring, significant dissolution/volatilization will occur at only the leading edge of the DNAPL zone where clean water/air first contacts the DNAPL phase. Once the groundwater/air has bypassed the leading edge of the DNAPL, it contains dissolved phase/vapor contaminants. The driving force for dissolution/volatilization lowers beyond the leading edge, resulting in mass removal essentially only from the leading edge of the DNAPL zone.

4) Interaction at NAPL interfaces

Biological, physical and chemical interactions that occur at the interface between NAPLs and the aqueous phase can significantly impact source zone treatment effectiveness; however, these interactions are poorly understood (Stroo *et al.*, 2003). It is possible that degradation reactions that occur in the aqueous phase at the interface with DNAPL pools/residual can enhance interphase mass transfer by increasing concentration gradients and therefore increasing dissolution rates. Bioremediation has been shown to increase dissolution rates by 2 to 6.5 times the natural rates (*e.g.*, Yang and McCarty, 2001; Cope and Hughes, 2001; Carr *et al.*, 2000). Field and laboratory evidence indicates that *in situ* chemical oxidation (*i.e.*, permanganate) could increase dissolution rates by an order of magnitude more than the bioremediation rates (Schnarr *et al.*, 1998; MacKinnon

and Thomson, 2002; Lee *et al.*, 2003). Alternatively, treatment rates may be increased further through the use of reagents that preferentially partition to the NAPL-water interface (Stroo *et al.*, 2003).

Since DNAPL degradation may occur over the entire DNAPL:water interface, interphase mass transfer is enhanced across the entire interface. The surface area:volume ratio of the DNAPL can impact remedial effectiveness and/or duration. A smaller surface:volume ratio (*e.g.*, DNAPL pools) results in less enhancement of the DNAPL dissolution than for a larger ratio (*e.g.*, residual). The enhancement in the DNAPL dissolution rate may decrease over time due to the formation of reaction products as films at the interface (*e.g.*, permanganate reaction with chlorinated solvents forming an insoluble manganese dioxide layer over the DNAPL/ water interface, MacKinnon and Thomson, 2002, Lee *et al.* 2003; biofilms forming near the DNAPL interface, Chu *et al.*, 2002). The impact of film formation is more pronounced where DNAPLs are distributed as pools, rather than as residual, due to the smaller interfacial area.

5) Degradation of groundwater quality (secondary groundwater quality issues)

While the removal of DNAPL mass from the subsurface will directly mitigate the chlorinated solvent contamination problem, the removal process may result in the generation of secondary groundwater quality issues, which may prevent the use of the groundwater as a source of drinking water for health or taste and odor reasons. For example, the strongly reducing conditions created by anaerobic biodegradation of chlorinated solvents may mobilize large quantities of dissolved iron and manganese in some aquifers. Addition of high concentrations of electron donors can results in high BOD (biological oxygen demand), and methane generation that can cause safety-related problems. Similarly, chemical oxidation using permanganate results in the formation of manganese dioxide precipitate, which may be significant at sites where large quantities of permanganate are used to treat large DNAPL masses. The manganese dioxide may reduce over time to dissolved manganese if aquifer conditions ever become reducing (e.g., permanganate mass is consumed, and reducing groundwater influxes from upgradient). The use of zero-valent iron may result in high quantities of dissolved iron added to the groundwater. Impurities (e.g., regulated metals) in amendments injected into the subsurface during remediation may create their own groundwater contamination problem where sufficient mass of the amendments are employed (e.g., trace amounts of selenium contained in molasses, which is sometimes used as a nutrient to stimulate biodegradation). It is possible that these secondary groundwater impacts may mitigate over time and distance from the treatment zone; however, testing of any amendments added for regulated compounds as well as the treatment remedy over a smaller scale should be done to evaluate the potential for the formation of secondary groundwater issues.

6) Use of multiple technologies to complete remediation

Some remedial technologies can be effective at removing significant quantities of easily accessible DNAPL mass over a short time period (e.g., surfactant/co-solvent flushing, chemical oxidation, multi-phase extraction); however, they can be prohibitively costly to use for removal of the remaining mass that is less accessible. In these situations, it may be beneficial to follow up the initial remedy with a second, less expensive and less aggressive remedy as a polishing step (*e.g.*, bioremediation or natural attenuation). When designing for the possibility of using "treatment trains", attention needs to be given to the long-term impacts on the aquifer caused by the primary treatment remedy. For example, chemical oxidation may produce extremely low pH conditions (Fenton's reagent), a highly oxidizing environment, and/or significant masses of precipitated manganese dioxide (permanganate) that could potentially become a long-term source of dissolved manganese if the groundwater environment becomes reducing in the future. Chemical oxidation can reduce specific microbial populations and thermal technologies can sterilize the soil. These conditions may inhibit microbial activity, which would impact the effectiveness of bioremediation or natural attenuation as secondary polishing remedies.

7) Groundwater influx

Sites with large groundwater velocities or a large influx of groundwater into the treatment area can negatively impact the effectiveness of remedial activities. Thermal treatments rely on maintaining the temperature of the aquifer at a specific level in order to effectively volatilize, mobilize and/or destroy DNAPL *in situ*. High influx of lower temperature groundwater can impact the ability to maintain the aquifer at the optimal temperature. High groundwater flux also can transport in additional electron acceptors or other compounds, which increase cost for bioremediation by requiring the addition of extra electron donors. On a positive note, high groundwater flux can be beneficial for flushing or flooding a site with treatment amendments or in pump and treat scenarios.

8) Regulatory environment

Regulatory cleanup levels are highly variable depending on factors such as location, and land use. Thus, a remedy that is acceptable at one site may not be acceptable at another. The definition of remedial objectives with all-or nothing end-points for DNAPL remediation has resulted in a limited number of DNAPL sites where source remediation technologies are being applied

2.4.3 Measurement of Remedial Success

Understanding how well technologies perform with specific site and source characteristics is required to optimize the remedial technology application, as well as the

technology selection. To properly evaluate a source remediation technology, multiple metrics must be used to evaluate performance (EPA DNAPL panel, In Press), including the following:

- i) Estimates of DNAPL Mass Removed: Measured or defined as either the contaminant mass recovered, an estimate of the mass destroyed during treatment or an estimate of the mass remaining after treatment. Direct measurement of both the mass remaining and mass destroyed in situ is often not feasible. DNAPL mass initially present and remaining posttreatment may be the more important parameters, but are very difficult to quantify. Methods such as soil coring, partitioning inter-well tracer tests, geophysical methods, inference from dissolved phase concentrations, or push-in samplers and sensors may provide rough estimates of DNAPL mass present; however, costly detailed sampling is required at many sites due to the typical sporadic distribution of DNAPL in the subsurface. For in situ destruction technologies, the estimate of the mass destroyed must be inferred from other data (e.g., the formation of degradation products, changes in groundwater concentrations, soil boring samples, differences in pre- and post-treatment partitioning tracer tests) and again is dependent on the initial mass estimate. For treatment technologies that remove mass and treat ex situ, a measurement of the total mass recovered can be obtained. This can be used to then assume the mass remaining, but this value is dependent on the accuracy of the initial estimate of mass.
- *ii)* Changes in Dissolved Phase Concentrations and Mass Flux: The easiest metric to implement. The DNAPL mass remaining may be inferred from the dissolved phase concentration; however, this estimate is typically inaccurate since the relationship between DNAPL mass and dissolved phase concentrations is nonlinear. Dissolved phase data can also be sensitive to a number of outside factors that can lead to variabilities in space and time, including: (*i*) precipitation recharge; (*ii*) temporal changes in groundwater flow direction; (*iii*) external influences on the aquifer (*e.g.*, pumping); (*iv*) variability in the sampling method; and (*v*) rebound after treatment. The type of well installation used to obtain groundwater samples may also impact the sample data at sites with depth-variable contamination profiles due to dilution with less contaminated water.
- *iii)* Changes in the DNAPL Distribution, Composition, and Properties: Changes in the DNAPL other than mass reduction may occur throughout the course of remedial treatment; these changes can be monitored to evaluate remedial progress. Specifically, these changes include: (*i*)

DNAPL mobilization; (*ii*) changes in DNAPL composition; and (*iii*) changes in the DNAPL properties (*e.g.*, interfacial tension, solubility, volatility, *etc.*).

iv) Secondary Impacts on the Aquifer Geochemistry and Biochemistry: A more indirect measure of remedial success can be determined through the monitoring of secondary impacts on aquifer conditions. Many remedial methods affect the geochemistry of the aquifer. Examples of potential impacts include changes in the following parameters: (*i*) microbial populations; (*ii*) pH and oxidation-reduction potential (ORP); (*iii*) dissolved metal concentrations; (*iv*) groundwater temperature; (*v*) sulfate reduction; and (*vi*) production of degradation byproducts.

Not all metrics are appropriate for all technologies, and the optimal combination of metrics will depend on the chosen remedial technology(ies), aquifer hydrogeology characteristics, and contaminant properties (EPA DNAPL panel, In Press).

3 OVERVIEW OF DNAPL REMEDIAL TECHNOLOGIES

An important step in the process of developing technology selection criteria, is to understand the theory behind each remedial technology, the advantages and disadvantages of each technology, and the factors affecting implementation. This section contains a brief overview of this information for the following DNAPL remedial technologies:

- *i)* surfactant/co-solvent flushing (Section 3.1);
- *ii)* bioremediation (Section 3.2);
- *iii)* chemical oxidation (Section 3.3);
- *iv)* a number of thermal technologies (Section 3.4);
- *v)* multi-phase extraction (Section 3.5); and
- vi) zero-valent iron (Section 3.6).

Appendix A contains a limited number of case studies of each technology. These case studies are included to highlight some of the challenges and advantages associated with each technology, as well as to include examples of how each technology performed for various site characteristics. The information presented in each case study includes a summary of relevant site characteristics, technology application approaches, remedial goals, and lessons learned from each site.

3.1 Surfactant and Co-Solvent Flushing

Surfactant flushing (or surfactant-enhanced aquifer remediation [SEAR]) and cosolvent flushing are chemical enhancements to pump and treat and involve injection of a chemical solution, flushing through a source zone and recovery of the injected solution and targeted DNAPL. The application of a surfactant or co-solvent flush results in DNAPL remediation through one or more of the following mechanisms:

- *i)* Enhancement of the target chemicals' solubility by lowering the interfacial tension between water and the DNAPL;
- *ii)* Mobilization of the DNAPL by lowering the DNAPL:water interfacial tension;
- *iii)* Mobilization of the DNAPL by swelling the DNAPL volume and reducing its density as the added alcohol dissolves into the DNAPL (co-solvent only); and

iv) Stimulation of *in situ* bioremediation after the initial chemical flush, as many surfactants and co-solvents can also act as electron donors.

When surfactants are flushed through a source zone, the concentration and specific interactions of that surfactant with the target chemical help to determine whether the surfactant will enhance solubility or mobility. These surfactants achieve enhanced solubility and/or mobility by reducing the interfacial tension between the DNAPL and the water phase through a process referred to as micellar solubilization. The surfactants used for aquifer flushing are typically anionic surfactants, as the anionic surfactant is less likely to be sorbed to the soil matrix then the cationic surfactant, and include alcohol ether sulfates, alkane sulfonates and sulfosuccinates. These surfactants consist of a water soluble head and an oil soluble tail, and at concentrations typical for remedial application, will coalesce into aggregations referred to as micelles. Micelles can be envisioned as spheres having a polar exterior (hydrophilic portion of surfactant) and nonpolar interior (hydrophobic moiety). The type of surfactant aggregation changes as the concentration of the surfactant changes. Solubilization of DNAPL into the micelles occurs for oil-in-water micro emulsions, resulting in enhanced DNAPL dissolution. At somewhat higher surfactant concentrations, the creation of an ultra-low interfacial tension can result in mobilization of the DNAPL. A water-in-oil micro emulsion may also be created in which the surfactant will partition into the DNAPL phase, which is undesirable as the surfactant is essentially lost into the DNAPL with little impact on the DNAPL's solubility or mobility.

Co-solvents are similar to surfactants in that they can alter the properties of solution interfaces to affect both the solubility and interfacial tension at the DNAPL:water interface. Co-solvents are miscible in both water and DNAPL and may partition preferentially into one or the other depending on the type of alcohol and DNAPL and the alcohol concentration. As a result, alcohols increase the solubility of many DNAPLs, and can increase the mobility of the DNAPL if used at sufficiently high concentration to lower the DNAPL:water interfacial tension to essentially zero. The low density of many alcohols used as co-solvents can make it more difficult to target specific geologic horizons below the water table. However, the DNAPL density will decrease as the lower density alcohol partitions into the DNAPL, which has the advantage of decreasing the likelihood of downward DNAPL mobilization.

Enhanced mobility is commonly used as the primary approach for remediation in cases where the DNAPL zone is underlain by a competent capillary barrier, such as a clay aquitard. DNAPL mobilization is considered particularly useful in cases where larger accumulations (such as pools) of DNAPL are known to exist in the aquifer. In cases where no competent capillary barrier exists, enhanced solubility from the DNAPL is the preferred surfactant flushing approach. Enhanced solubility is also considered to be preferable in cases where the DNAPL is fairly distributed in the aquifer, likely existing

primarily as residual phase. In either case, hydraulic control during flushing is essential for application of this technology.

The advantages of surfactant and co-solvent flushing include the following:

- *i)* DNAPL removal can be rapid: Significant amount of DNAPL mass can be removed, typically within 4 to 15 pore volume flushes (CH2M Hill, 1997);
- *ii)* **Effective DNAPL removal:** Large amounts of DNAPL may be extracted from the subsurface during treatment; and
- *iii)* **Biodegradation may be stimulated:** Amendments remaining in the subsurface post-flushing may act as electron donors and stimulate biodegradation of any remaining contamination.

Disadvantages associated with surfactant and/or co-solvent flushing include:

- *i)* **There may be a risk of undesirable DNAPL mobilization:** The magnitude of the risk is dependent on site geology (*i.e.*, confining layers may mitigate the risk in some situations) and the type and concentration of surfactant/co-solvent used (*i.e.*, the use of a low density co-solvent may mitigate risk by decreasing the density of the DNAPL after partitioning into it).
- *ii)* **Handling of the extracted fluids may be problematic:** Surfactants are generally not recyclable, thus requiring disposal or *ex situ* treatment. Co-solvents can also be difficult to separate from extracted DNAPL due to its miscibility with the DNAPL.
- *iii)* Achieving the targeted flush may be difficult: Soil heterogeneity can impact the ability to deliver the amendments evenly throughout the targeted zone, resulting in uneven DNAPL removal. *In situ* dilution and dispersion can result in *in situ* amendment concentrations varying from the targeted concentrations, which may change the resulting DNAPL behavior from that desired (*e.g.*, dilution of alcohol may result in DNAPL solubilization rather than mobilization, or dilution of surfactants may result in the surfactant partitioning into the DNAPL rather than DNAPL dissolution into surfactant micelles).

3.2 Bioremediation

Biodegradation accelerates the remediation of DNAPL source areas through several major processes, including:

- *i)* Dechlorination of the parent chlorinated solvents (*e.g.*, PCE, 1,1,2,2-PCA) in the dissolved phase near the DNAPL/water interface steepens the dissolution gradient and increases the overall mass transfer of the solvent from the DNAPL to the dissolved phase (Carr *et al.*, 2000; Cope and Hughes, 2001), thus depleting the DNAPL at a faster rate; and
- *ii)* Dechlorination of the parent chlorinated solvents to species that have higher saturated solubilities and thus faster dissolution rates; therefore resulting in a directly proportional increase in the dissolution rate and decrease in the DNAPL longevity (Carr *et al.*, 2000; Cope and Hughes, 2001).

The maximum enhancement of DNAPL removal is primarily based upon the total effective surface area over which biodegradation can occur, biodegradation rate, and the partitioning behavior of the terminal chlorinated ethene. Other factors include the rate of groundwater flow, the nutrient availability, the concentration of alternate electron acceptors, if biofilms form near the water:DNAPL interface and the parent CAH and degradation product solubility.

Chlorinated solvents can be biodegraded through four mechanisms: (*i*) reductive dechlorination; (*ii*) aerobic cometabolism; (*iii*) anaerobic oxidation; and (*iv*) direct oxidation. Reductive dechlorination involves the sequential replacement of chlorine atoms on the organic molecule by hydrogen atoms. The reaction occurs primarily under anaerobic and reducing redox conditions that typically favor methanogenesis, although reductive dechlorination has been observed in bulk aerobic aquifers (the activity occurs within anaerobic micro-habitats). The chlorinated VOCs serve as electron acceptors for the halorespiring bacteria that carry out these degradation reactions; simple organic carbon compounds (*e.g.*, alcohols, fatty acids, sugars, petroleum hydrocarbons and natural organic carbon substances such as humic/fulvic acids) can serve as electron donors. Reductive dechlorination is the principal mechanism for biodegradation of most highly chlorinated compounds such as PCE, TCE, TCA and CTC.

A variation of sequential reductive dechlorination is dihaloelimination which involves the removal of two chloride atoms and the formation of ethene with a carbon– carbon double bond. Like reductive dechlorination, this reaction occurs under anaerobic and reducing redox conditions, although methanogenesis may not be required. Again, like reductive dechlorination, chlorinated VOCs will serve as electron acceptors for the bacteria that carry out these degradation reactions, while simple organic carbon compounds (*e.g.*, alcohols, fatty acids, sugars, petroleum hydrocarbons and natural organic carbon substances such as humic/fulvic acids) can serve as electron donors.

Cometabolic reactions can occur under aerobic and anaerobic conditions. Aerobic cometabolism of chlorinated organic compounds results in their oxidization to CO₂, water and chloride by non-specific microbial oxygenase enzymes produced by a variety of aerobic microorganisms. Typical cometabolites that induce enzymes that react with chlorinated VOCs include methane, ammonia, ethene, toluene and phenol. Aerobic cometabolism can occur naturally at the fringes of the plumes where redox conditions transition from anaerobic to aerobic, and a suitable cometabolite, oxygen, and the target VOC are mixed in relatively balanced proportions. Anaerobic cometabolism can occur as part of the halo-respiration process, or reactions with other highly reduced proteins used in methyl-transfer reactions. Halo-respiration will produce reduced forms of the parent VOC, and these reduced forms can react with the same enzymes that produced it. The microorganism may gain energy from the initial dechlorination steps, but not the cometabolic step.

Of the above processes, reductive dechlorination will be the most applicable for the treatment of most chlorinated solvent DNAPLs for the following reasons:

- *i)* **Energetically favorable:** microorganisms that cometabolize chlorinated solvents do not derive energy or carbon from the process. In contrast, halorespiring microorganisms derive significant energy during dechlorination. Anaerobic and direct oxidation reactions are also energetically favorable, but only occur for lower chlorinated solvents (*e.g.*, dichloroethenes, vinyl chloride, 1,2-dichloroethane) and will not be significant within a DNAPL source area where higher chlorinated solvents (*e.g.*, PCE, TCE) dominate.
- *ii)* Limited solubility constraints: nutrients used to stimulate or support reductive dechlorination (electron donors such as sugars, alcohols, fatty acids that are fermented to hydrogen and used for reductive dechlorination) are more soluble than the chlorinated solvents, so can be applied in amounts that equal or exceed reductive dechlorination demand, and that create concentration gradients that encourage microbial growth near or at the DNAPL:water interface (Chu *et al.*, 2003).
- *iii)* **Relative insensitivity to high concentrations:** whereas nondechlorinating microorganisms that compete for hydrogen are inhibited at concentrations approaching the aqueous solubility limit of PCE/TCE, dechlorinating microorganisms are not. Therefore, DNAPL source areas provide a unique environment within which halorespirers have a competitive advantage. A corollary to this advantage is an improved cost

effectiveness of nutrient addition because a greater percentage of the hydrogen produced during the fermentation of added electron donors is consumed by dechlorinating microorganisms and not wasted in support of other microbial processes such as methanogenesis (Yang and McCarty, 1998 and 2000).

The advantages of bioremediation include the following:

- *i)* **Cost-effectiveness:** Operating and maintenance costs can be relatively low, and equivalent to very simple groundwater extraction and treatment systems that only require air-stripping with no off-gas treatment.
- *ii)* **The ability to combine with other technologies:** Bioremediation can work synergistically with other DNAPL treatment technologies (*e.g.*, surfactant/co-solvent flushing, emulsified nano-scale iron) to speed up DNAPL treatment, or be used as a polishing step to cost effectively remove residual DNAPL left behind from more aggressive technologies.
- *iii)* **Reduction in treatment duration:** Studies have suggested that bioremediation could increase the mass removal rate by an order of magnitude, with a corresponding reduction in time to achieve remedial goals (*e.g.*, a 100 year pump and treat scenario is reduced to 10 years).

Disadvantages of bioremediation include the following:

- *i)* **Impacts to secondary groundwater quality:** Secondary water quality effects are related to the creation of lower reduction-oxidation conditions that may result in the solubilization of metals, such as iron and manganese above their secondary water quality limits, or the release of arsenic, a known carcinogen and toxic compound. Hydrogen sulfide can be created during the reduction of sulfate, and methane can degas from the groundwater and build up in the vadose zone forming an explosion hazard. In addition, there can be transient or steady-state production of toxic degradation byproducts (*e.g.*, VC) where required microbial communities do not exist.
- *ii)* There may be some safety and handling issues: Several types of electron donors are flammable (*e.g.*, alcohols) and require specialized containers for storage and procedures for handling.
- *iii)* **Biofouling of injection wells:** Well fouling from the formation of biofilms and precipitation of inorganic species inside and outside of

biofilms can be extremely problematic depending on the site geochemistry, and can result in significant O&M costs.

3.3 Chemical Oxidation

In situ chemical oxidation (ISCO) has been shown to destroy or degrade an extensive variety of hazardous wastes in groundwater and soil. ISCO refers to a group of specific technologies that each use specific combinations of oxidants and delivery techniques. Various oxidants have been used in laboratory and field applications to aggressively destroy chlorinated solvent DNAPLs, including permanganate (MnO_4), ozone, and Fenton's reagent (a combination of hydrogen peroxide [H_2O_2] and a ferrous iron catalyst). The oxidants react with the contaminants and convert them to innocuous compounds commonly found in nature such as carbon dioxide (CO_2), water and inorganic chloride. Some of the organic contaminants that can be treated using ISCO include BTEX (benzene, toluene, ethylbenzene, and xylenes), chlorinated solvents (*e.g.*, PCE, TCE), MTBE (methyl-ter-butyl-ether), and PAH (polyaromatic hydrocarbons) compounds. Permanganate has really only been shown to be effective at degrading chlorinated ethenes. Fenton's reagent has been used to degrade most chlorinated compounds with the exception of chlorinated methanes.

Similar to biodegradation, chemical oxidation accelerates the remediation of DNAPL source areas through dechlorination of the chlorinated solvents in the dissolved phase near the DNAPL/water interface. The destruction of the dissolved DNAPL at the interface steepens the dissolution gradient, increasing the overall mass transfer of the solvent from the DNAPL to the dissolved phase, and thus depleting the DNAPL at a faster rate. The maximum enhancement of DNAPL removal is primarily based upon the total effective surface area over which oxidation can occur and the reaction rate. Other factors include effective delivery of the oxidant to the contaminated media, consumption of the oxidant by other organic material in the aquifer, and the CAH solubility.

Fenton's Reagent

The basis of the Fenton's technology involves free radical generation and direct oxidation with hydrogen peroxide. Hydrogen peroxide is an effective oxidizing agent; however, to achieve contaminant reduction in a reasonable time, iron or iron salts are used as a catalyst (the combination is referred to as Fenton's reagent). Fenton's reagent chemistry (equation 3.1) is well documented as a method for producing hydroxyl radicals by reaction of peroxide (H₂O₂) and ferrous iron (Fe²⁺) (ITRC, 2001). The basic reaction is as follows:

$$H_2O_2 + Fe^{2^+} => Fe^{2^+} + OH + OH \bullet$$
 (3.1)

The hydroxyl radicals $(OH \cdot)$ serve as powerful, effective and nonspecific oxidizing agents. There are many reactions that occur during the oxidation of a contaminant and either ferrous or ferric iron can react with the peroxide to produce oxidizing radicals:

$$RHX + H_2O_2 <==>H_2O + CO_2 + H^+ + X$$
(3.2)

where RHX represents an organic compound and X represents a halide (such as chloride). Compounds such as TCE and PCE are converted to CO₂, water and hydrogen and chloride ions.

An effective treatment stoichiometry for *in situ* Fenton's is influenced by variables such as pH, contaminant concentrations, and total organic carbon (TOC) (Kakarla *et al.*, 2002). A pH in the range of 3 to 4 is ideal for free radical generation (Watts *et al.*, 1990), which is impractical to maintain under field conditions due to the enormous buffering capacity associated with most native soils. It has been demonstrated that free radical generation and contaminant oxidation can be promoted without acid addition using a modified Fenton's process that uses iron catalysts, which are effectively chelated in the pH range of 5 to 7, and a stabilized hydrogen peroxide (Kakarla *et al.*, 2002). Fenton's reagent produces a strong reaction, which is not persistent in the subsurface and thus can be difficult to distribute to the source area.

Ozone

Ozone (O_3) is one of the strongest oxidants available for ISCO (ITRC, 2001). Ozone can oxidize organic contaminants by either direct oxidization by ozone or by generation of free radical intermediates. The hydroxyl radicals are nonselective oxidizers that rapidly attack organic contaminants and break down their carbon-to-carbon bonds. Ozone can oxidize compound such as aromatics and chlorinated alkenes, although oxidation by hydroxyl radicals is faster than oxidation by ozone itself (ITRC, 2001).

Permanganate

Permanganate is an oxidizing agent that has an affinity for oxidizing organic compounds containing carbon-carbon (C=C) double bonds, aldehyde groups or hydroxyl groups (ITRC, 2001). The reaction between permanganate and chlorinated ethenes involves an electrophilic attack on the ethene's C=C double bonds and the formation of a cyclic hypomanganate ester. Rapid hydrolysis of the cyclic ester results in the production of carbon dioxide (CO₂). There are two forms of permanganate, potassium permanganate (KMnO₄) and sodium permanganate (NaMnO₄). The stoichiometric reactions describing the oxidation of TCE and PCE by MnO₄⁻ are given by Yan and Schwartz (1999):

$$C_2Cl_3H + 2MnO_4^- \rightarrow 2CO_2(g) + 2MnO_2(s) + 3Cl^- + H^+$$
 (3.3)

$$3C_2Cl_4 + 4MnO_4^- + 4H_2O \rightarrow 6CO_2(g) + 4MnO_2(s) + 12Cl^- + 8H^+$$
 (3.4)

where PCE and TCE are presented by their chemical formulae C_2Cl_4 and C_2Cl_3H , respectively. This reaction indicates that oxidation of these compounds by MnO_4^- is accompanied by the production of manganese dioxide (MnO_2) solid, CO_2 gas, hydrogen (H^+) and chloride (CI^-). In comparison to Fenton's reagent, permanganate is a weaker oxidant; however, it is more persistent in the subsurface and therefore may be better able to target the source area.

The effectiveness of ISCO treatment is influenced by the following factors:

- *i)* Adequate contact between oxidants and contaminants must be achieved: Subsurface heterogeneities, preferential flowpaths, or low soil permeability can result in uneven flushing of the oxidant through the subsurface, resulting in untreated contaminants.
- *ii)* **Groundwater geochemistry can impact oxidant consumption:** An understanding of the geochemical conditions at the site is essential since the applied reagents could be consumed by natural organic matter or dissolved iron rather than the contaminants, resulting in poorer than expected treatment. Groundwater geochemistry may also need to be adjusted to more optimal conditions prior to treatment (*e.g.*, lowering of pH during application of Fenton's reagent).
- *iii)* **Design of the oxidant delivery approach:** In order to achieve adequate contact between the oxidant and the contamination, an adequate fraction of the pore-volume of the target area must be filled or flushed with the oxidant. However, care must be taken not to displace the contamination with excessive amounts of oxidant injection.
- *iv)* **Oxidant concentration:** Oxidant concentrations need to be high enough to meet the natural oxidant demand of the aquifer, as well as the demand of any contaminant encountered during flushing. However, excessive concentrations of oxidant are not desirable due to potential impacts on secondary groundwater quality (*e.g.*, color, pH, dissolved metals) and higher costs.
- *v*) Initial DNAPL distribution: Nearly all residual DNAPL mass can be destroyed *in situ* with an expectation of a comparable level of mass flux reduction (Thomson *et al.*, 2000). In comparison, it is likely that a lower level of mass removal and mass flux reduction may be achieved for pooled DNAPL accumulations (Thomson *et al.*, 2000). The differences in mass flux removal are primarily due to the difference in the DNAPL-water interfacial surface area.

The advantages and disadvantages of the various chemical oxidation methods are as follows:

Fenton's Reagent

Advantages:

- *i)* The oxidant materials are inexpensive and readily available.
- *ii)* A wide range of chemicals including chlorinated solvents, polyaromatic hydrocarbons and petroleum products can be treated.

Disadvantages:

- *i)* The technology is limited by the interference of subsurface impurities and carbonate since bicarbonate and organic matter will create competing reactions that hinder performance.
- *ii)* An extremely exothermic reaction occurs, which can create safety and handling issues.
- *iii)* Groundwater pH post-treatment can be quite low (<5), which can effectively sterilize the soil and limit secondary treatment choices.

Ozone

Advantages:

i) The gaseous nature of ozone allows for ease of delivery through the vadose zone compared with the liquid oxidants (Looney and Falta, 2000).

Disadvantages:

- *i)* The half-life of ozone substantially limits its ability to migrate through the soil, thus this oxidant is generally considered useful only for small scale or vadose zone applications.
- *ii)* Ozone generation system requires a large capital investment.
- *iii)* Ozone can be an indoor air quality issue.
- *iv)* Ozone is highly reactive with aquifer solids and groundwater constituents.

Permanganate

Advantages:

- *i)* Permanganate is less reactive with aquifer solids than other oxidants, resulting in improved oxidant delivery to the target contaminants due to its persistence.
- *ii)* It is typically more stable and safer to handle than Fenton's reagent, does not require pH adjustment with concentrated acid and produces less heat and insoluble gas in the treatment zone.

Disadvantages:

- *i)* Permanganate treats a narrower range of contaminants than the other oxidants. Although it can treat chlorinated ethenes, permanganate is not effective at treating chlorinated ethanes and may have limited effectiveness against BTEX.
- *ii)* Permanganate can be expensive.
- *iii)* Permeability reductions can occur near DNAPL source zones due to the formation of MnO₂ precipitates (*e.g.*, MacKinnon and Thomson, 2000; Dai and Reitsma, 2002; Lee *et al.*, 2003) and/or rapid production of CO₂(g) (Dai and Reitsma, 2002), resulting in less effective treatment over time.
- *iv)* Recirculation systems are prone to fouling with MnO₂ precipitates.
- *v)* Strongly oxidizing conditions are created that can persist post-treatment, which may impact the effectiveness or choice of polishing technology (if required).
- *vi)* Dissolved metals mobilization may occur in some aquifers, depending on the mineral content of the geological material present.
- *vii)* Manganese precipitated as MnO₂ may mobilize as dissolved manganese if the groundwater geochemistry becomes reducing upon termination of the treatment.

3.4 Thermal Technologies

The thermal technologies that are most commonly applied for remediation of DNAPLs include steam flushing, electrical resistance heating (ERH; both three-phase and six-phase heating) and electrical conductive heating (ECH; also referred to as *in situ* thermal desorption and thermal conductive heating). While there are a number of other thermal technologies including *in situ* vitrification, radio frequency heating and hot-air injection, these are not commonly applied and were not cited in any of the survey's responses; thus we have not included specific details for these technologies in this document.

All thermal technologies involve increasing the soil and groundwater temperature in the target area. Heating can result in DNAPL remediation through enhancement of both extraction and *in situ* destruction processes through the following mechanisms (Udell & Stewart, 1998; Battelle, 2002; Stegemeier and Vinegar, 2001; Roote, 2003; USEPA, 1999):

- *i*) increasing vapor pressure and volatilization rates of low boiling point chemicals;
- *ii)* conversion of groundwater to steam and subsequent steam distillation of target chemicals;
- *iii)* desorption of target chemicals from sorption sites;
- *iv)* decreases in viscosity of separate phase chemicals which can increase mobility;
- *v)* increases in soil permeability through partial (steam, ERH) or complete drying (ECH) of the soil matrix;
- *vi)* increases in both aqueous solubility and aqueous and gaseous molecular diffusion coefficients to increase dissolution and diffusion rates;
- *vii)* enhanced *in situ* biodegradation of target chemicals in the case of moderate (typically <100°C) temperature increases;
- *viii) in situ* thermal destruction of target chemicals through hydrous pyrolysis/oxidation, particularly when higher temperatures are applied (i.e. greater than 100°C); and

ix) physical displacement of DNAPL mobilized by active flushing of the target zone due to the induced gradient from SVE and/or steam injection.

ERH and steam flushing rely on the water to transport heat, and therefore are only effective while soil moisture remains. In contrast, ECH is achieved through heating of the soil, and therefore can be applied at much higher temperatures to achieve DNAPL remediation (Roote, 2003; Stegemeier and Vinegar, 2001). ECH and ERH are suitable for application in both high and low permeability media (*i.e.* clays), as the thermal conductivity of soils tends to be fairly uniform as compared to hydraulic permeabilities. As steam flushing is reliant on hydraulic transport, it is less applicable to low permeability media. For all thermal technologies, groundwater influx into the treatment zone is a key factor in the successful application of the technology as this material needs to be heated to continue the remedial process. In cases where groundwater velocities or surface recharge is high, resulting in a high influx of unheated groundwater into the treatment zone, special controls (extraction wells) may be required so that the thermal technology can be effective in maintaining adequate heat in the treatment zone. If controls are likely to be insufficient for limiting groundwater influx then thermal technologies may not be applicable to the site.

Advantages of the various thermal technologies include the following:

- *i)* **Potential for rapid remediation:** DNAPLs may be removed within months to years, in comparison to the years to decades for less aggressive (*e.g.*, bioremediation, chemical oxidation) technologies.
- *ii)* **Effective in low permeability media:** ERH and ECH technologies rely primarily on electrical and heat conduction, which is less sensitive to heterogeneities in soil permeability as the thermal conductivity of soils tends to be fairly uniform as compared to hydraulic permeabilities. Steam flushing is primarily reliant on hydraulic transport to remove the DNAPLs, this method is less effective for low permeability media (Roote, 2003).

Disadvantages of the thermal technologies include:

i) **Relatively high cost:** Both capital expenditures and operations and maintenance can be costly. Capital expenditures typically include above ground infrastructure to generate steam/heat/electricity for subsurface application, soil vapor treatment, extracted groundwater treatment, piping and off-gas control; below ground infrastructure includes a high density of application points and temperature monitoring points. Operation and

maintenance costs are dominated by the power requirements to generate steam/heat/electricity to the subsurface.

- *ii)* **Health and safety is a concern:** The equipment used to generate the thermal effect requires careful handling and, for ERH, the application area must be treated as an exclusion zone (no entry) to avoid electrocution risks (Battelle, 1999). In addition, the vapors generated during the remedial program can transport to ground surface resulting in potential exposure to operators as well as the surrounding area. Thermal applications typically have ground surface emission controls as well as air monitoring to limit the potential for exposure.
- *iii)* Undesired mobilization of DNAPL can occur: An increase in the temperature during the thermal technology application can lead to changes in contaminant transport properties (viscosity, solubility, diffusion coefficients) that can result in spreading of the DNAPL outside of the original defined source area (Kaslusky and Udell, 2002). Careful treatment design can minimize the degree of DNAPL remobilization.
- *iv)* Sterilization of the soil can occur: Typically, the application of heatbased remediation technologies will increase the subsurface soil and groundwater temperatures to above 65° C for extended time periods (several months to a year), which will effectively sterilize the soil in the treatment area. This will impact the effectiveness of microbiallydependent post-treatment polishing remedies (*e.g.*, natural attenuation, bioremediation; Dettmer, 2002).
- *v)* **Formation of undesired intermediates:** Chlorinated ethenes can be reactive at relatively moderate temperatures and form chlorinated intermediates that can persist. For example, the formation of hexchlorobutadiene or other toxic compounds (Constanza et al., 2003a, b).

3.5 Zero-Valent Iron

Zero-valent iron (ZVI) has traditionally been used in the treatment of groundwater plumes as part of permeable reactive barriers. ZVI technologies that are now being tested and applied for remediation of DNAPLs include direct injection of particulate iron, mixing of iron with clay slurries or incorporating nano-scale ZVI into an oil emulsion prior to injection.

ZVI in close proximity to DNAPL source areas accelerates the remediation of DNAPL source areas through several major processes, including:

- *i)* DNAPL partitions into oil droplets containing nano-scale ZVI within them. The DNAPL reacts with the iron causing dechlorination of the parent chlorinated solvents (*e.g.*, PCE, 1,1,2,2-PCA). Encapsulation of ZVI inside of a hydrophobic fluid allows the iron to be in closer contact with the DNAPL, thus reducing the influence of the size and accessibility of the DNAPL/water interface on the DNAPL removal rate.
- *ii)* Dechlorination of the parent chlorinated solvents in the dissolved phase near the DNAPL/water interface steepens the dissolution gradient and increases the overall mass transfer of the solvent from the DNAPL to the dissolved phase, depleting the DNAPL at a faster rate; and
- *iii)* Dechlorination of the parent chlorinated solvents to species that have higher saturated solubilities and thus faster dissolution rates; therefore resulting in a directly proportional increase in the dissolution rate and decrease in the DNAPL longevity.

Laboratory and field research has demonstrated that zero-valent metals will reductively dehalogenate dissolved chlorinated solvents such as PCE and TCE to ethene (Gillham and O'Hannesin, 1994; Gillham 1995; Roberts *et al.*, 1996). The main dehalogenation reaction pathways occurring at the iron surface require excess electrons produced from the corrosion of the zero-valent iron in water as follows:

$$Fe^{0} \rightarrow Fe^{2+} + 2e^{-}$$
(3.5)
$$Fe^{2+}_{(surface)} \rightarrow Fe^{3+}_{(aqueous)} + e^{-}$$
(3.6)

Hydrogen gas is produced as well as OH⁻, which results in an increase in the pH of the surrounding water according to the following reaction:

$$2H_2O + 2 e^- \rightarrow H_{2(gas)} + 2OH^-$$
(3.7)

Some portion of the chlorinated ethenes are degraded by a step wise dehalogenation process according to:

$$\mathrm{RCl} + \mathrm{H}^{+} + 2\mathrm{e}^{-} \rightarrow \mathrm{RH} + \mathrm{Cl}^{-}$$
(3.8)

In the dehalogenation step (equation 3.8), the "R" represents the molecular group to which the chlorine atom is attached (*e.g.*, for TCE, "R" corresponds to CCl_2CH). For complete dehalogenation of TCE, reaction (3.8) must occur three times, with the end product being ethene. Through this process, the target chemicals undergo sequential dechlorination steps, resulting in the formation of non-chlorinated hydrocarbon products (*e.g.*, ethene, ethane, and straight chain hydrocarbons). The degradation of TCE can also

occur via β -elimination where TCE is converted to chloroacetylene, which is dehalogenated to acetylene. Acetylene is subsequently degraded to ethene and ethane.

For all direct applications of ZVI, destruction of the DNAPL can be fairly rapid, especially with the use of nano-scale iron (Choe *et al.*, 2001). When ZVI is used with a clay slurry or as EZVI, there is a reduction in contaminant migration during treatment through reduced groundwater flow and partitioning into the oil phase, respectively (Liberati 2003; O'Hara *et al.*, 2003; Wadley and Gilham, 2003). ZVI can be applied in low permeability and consolidated media through the use of an injection technology that will also cause either fracturing or enhanced mixing such as pneumatic or hydraulic fracturing (Chen and Markesic, 2001). Due to the redox conditions that are developed with the application of ZVI, favorable conditions for follow on polishing treatments can exist.

Advantages of ZVI technologies include the following:

- *i)* **Destruction of the DNAPL can potentially be rapid:** This is particularly true where nano-scale ZVI is used due to the large surface area of the iron particles that are available for reaction with the DNAPL (Choe *et al.*, 2001).
- *ii)* **Reduction of contaminant migration during treatment:** For the EZVI method, the DNAPL will tend to partition into the injected oil, resulting in an immediate reduction in mass flux downstream of the source area. The oil slowly degrades over time, creating electron donors and stimulating biodegradation of the DNAPL, further mitigating contaminant migration (O'Hara *et al.*, 2003). For ZVI mixed with clay, the clay reduces the permeability of the soil and therefore reduces the groundwater flow (Liberati 2003; Wadley and Gilham, 2003).
- *iii)* Effective in low permeability media: Depending on the methods used to distribute the ZVI in the subsurface, ZVI and EZVI can be applied at low permeability sites within created fractures (*i.e.*, pneumatic or hydraulic fracturing and injection).
- *iv)* Creates conditions that are amenable to the use of polishing technologies: ZVI can create favorable conditions (*e.g.*, strongly reducing environment) for organisms that would mediate biological degradation for a bioremediation or natural attenuation remedy. The EZVI method has the added benefit of stimulating bioremediation.

Disadvantages of the ZVI remedial technologies include:

- *i)* **Spreading of DNAPL may occur:** DNAPL mobilization may occur during the direct injection of ZVI when large injection fluid volumes are used. This risk can be limited through the use of engineered controls such as injection pattern design.
- *ii)* **Cost of consumables can be high:** Only a limited number of manufacturers of nano-scale iron exist worldwide, and nano-scale iron can therefore be expensive.

3.6 Dual Phase/Multiphase Extraction

Multiphase extraction (MPE) was developed as an extension of the soil vapor extraction (SVE) technology, which is commonly used to remediate source zones in the vadose zone. The terms dual phase and MPE are used interchangeably in this document.

MPE technology involves the application of a vacuum at one or more wells to extract groundwater, DNAPL and/or vapor from a source area. During the operation of MPE systems, DNAPL remediation is achieved through one or more of the following mechanisms:

- *i)* Enhanced groundwater flushing through the source area results in increased dissolution rates from DNAPL and total mass extracted from the aquifer;
- *ii)* Mobilization and recovery of DNAPL phase due to the vacuum extraction; and,
- *iii)* Volatilization and extraction of exposed DNAPL in areas where the water table has been depressed.

MPE is a technology of many synonyms, these various synonyms have arrived from applications in different industries (i.e. oil industry) and differences in the application approach for MPE, as summarized in Table 3.1. In some cases the names are associated with a specific trademark or patent on some component of the MPE application approach. Table 3.2 outlines the optimal conditions for application of an MPE approach.

The advantages of an MPE approach include the following (EPA, 1999):

i) **Effective on moderate to low permeability soils:** It is easier to achieve and maintain a vacuum for lower permeability soils.

ii) **Total fluids recovery is optimized:** Minimal drawdown and thus free product smearing, and aquifer transmissivity is maximized at the wellhead.

The disadvantages of an MPE approach include:

- *i)* **Energy demands are high:** Vacuum pumps and/or blowers are required, which require a lot of energy to operate.
- *ii) Ex situ* treatment of extracted fluids may be difficult: NAPL emulsions and VOC-laden vapors require treatment.
- *iii)* Initial start-up and adjustment periods are required: These may be prolonged.
- *iv)* Capital costs can be high: *Ex situ* treatment infrastructure is required along with vacuum extraction equipment.
- v) **Treatment may be limited to shallower depths:** Depth limitations apply.

4 SURVEY RESULTS

The successful translation of technology theory to the field is dependent on a number of factors. The relative impact of these factors on the remedial success of a technology can only be assessed through a performance evaluation under a wide range of site conditions. This information was compiled through a review of DNAPL remediation case studies, which is presented in this section. Section 4.1 presents an overview of the methods used and the data collected during the review process. Section 4.2 includes a summary of the information compiled, and an analysis of the data to determine whether trends exist with respect to site characteristics, cost, and technology success.

4.1 Survey and Literature Data Collection Methods

Information on specific DNAPL remediation case studies was collected using two approaches:

- *i)* A web-based survey was developed to compile a database of information on various aspects of DNAPL remediation that could impact the technology selection process (Section 4.1.1); and
- *ii)* A review of the literature and existing publicly available case study databases was performed to collect information from published case studies (Section 4.1.2). The details of these case studies were then entered into the web-based survey to add this information to the database.

4.1.1 Web-Based Survey

The type of information collected using the survey included the following information:

- *i)* site location, survey respondent role in the remediation community, the remedial technology applied, and stage of remediation;
- *ii)* the site characteristics specifically outlining the geology, hydrogeology, and groundwater geochemistry of a site;
- *iii)* the DNAPL source zone characteristics including its composition, location, size of aquifer impacted by DNAPL, and DNAPL distribution;
- *iv)* the cost and duration of treatment; and
- *v)* various remedial success parameters.

Screen shots of the actual survey questions are contained in Appendix B. Details of the information collected are included in Section 4.2

The objectives of the survey were as follows:

- *i)* Collect information on innovative and demonstrated source remediation technologies (*e.g.*, thermal, chemical oxidation, surfactant flushing, bioremediation, excavation, *etc.*) that have been tested or applied at chlorinated solvents DNAPL contaminated sites;
- *ii)* Obtain multiple perspective views of each technology application by encouraging personnel from all levels of the environmental field to participate in the survey; and
- *iii)* Obtain enough information to allow for evaluation of technology performance and impact of site characteristics, while at the same time limiting the amount of time required to enter data into the survey to a reasonable effort.

The survey of case studies was posted on the world wide web for a total of six months and the information package with the web link was emailed to approximately 700 people, mailed to over 3,000 people and posted on a number of web pages and news letters including:

- *i)* the Environmental Protection Agency (EPA) clu-in.org Tech Direct newsletter, which was sent to over 15,000 people;
- *ii)* the Battelle conference web page;
- *iii)* the Interstate Technology Regulatory Council (ITRC) web page;
- *iv)* and the Environmental Security Technology Certification Program (ESTCP) web page.

Targeted survey respondents included: Department of Defense (DoD) remedial project managers (RPMs), non-DoD RPMs, site owners, technology vendors, environmental consultants, academics and regulators. A total of 192 representatives from all areas of the remediation community accessed the survey and provided some site information (*i.e.*, a minimum of a site name); however, only a portion of these (61 out of 192 survey respondents) specified either a technology or technology specific information. In addition to individuals logging into the survey and entering information, GeoSyntec entered data into the survey from 21 published case studies (see Section 4.1.2 below). Including the published case studies as individual survey respondents increases the total

number of survey respondents to 213, and the number of respondents who specified a technology to 82.

Figure 4.1 presents a summary of the survey respondents by their role in the groundwater remediation community. Consultants represented the largest percentage of the respondents (35%), followed by the published case studies (26%), DoD RPMs (16%), site owners/RPMs (12%), vendors (9%), and regulators (2%). Notably absent from the respondents who specified a technology (see Figure 4.1) was members of the academic community; however, there are still representatives of all other areas of the remediation community.

Figure 4.2 shows the geographical distribution of the sites entered into the survey. The majority of sites were located in the United States, with a small number in Canada and one from Australia.

4.1.2 Review of the Literature

Case study information was also obtained from a review of the published literature, which included peer-reviewed and grey literature from a number of journals, conference proceedings, regulatory documents, vendor web pages and the following publicly available databases:

- *i)* Remediation and Characterization Technology Database (EPA REACH IT);
- *ii)* Federal Remediation Technologies Roundtable (FRTR) Case Studies;
- *iii)* Field Applications of *In Situ* Remediation Technologies: Permeable Reactive Barriers;
- *iv)* Fractured Bedrock Focus Area;
- *v)* In Situ Thermal Treatment Site Profile Database;
- *vi)* Innovative Remediation Technologies: Field-Scale Demonstration Projects in North America;
- *vii)* State Coalition for Remediation of Drycleaners;
- *viii)* Superfund Cleanup Technologies; and
- *ix)* Technology Focus: The Remediation Technology Information Center.

Information from case studies with more complete data available, and where it was obvious that the treatment of DNAPL was attempted, were entered into the survey database. A total of 21 case studies were inputted by GeoSyntec personnel. Table C.35 identifies the 21 sites and the technologies that were applied at each site.

4.2 Data Quality

4.2.1 Data Information Check

The data from the survey and case studies included in this report were not screened to verify their validity except for the following:

- *i)* **Outliers:** Obvious outliers (*e.g.*, abnormally small treatment volumes, or groundwater chemistry orders of magnitude above solubility limits) were either corrected where the correction was obvious [*e.g.*, the user selected the wrong unit of measure (*e.g.*, concentrations above the solubility limits), modify the chosen units of measure], the survey respondent was contacted to verify the data, or were omitted in the analysis where the correct data was not obvious/verifiable. In total, data was omitted once for a volume of <1 ft³, and 12 sites had the units changed for the dissolved groundwater chemistry values (*e.g.*, groundwater data over solubility limits).
- *ii)* **Duplicate information:** Where duplication of site and technology information occurred, an attempt was made to determine which of the datasets provided was the most complete. Only the more complete dataset was included in the analysis, except where duplicate information may be of interest (*e.g.*, information on the perception of success of a remedial technology at a particular site entered by two survey respondents representing different roles in the remedial process, such as consultant and regulator). There were a total of three sites with multiple entries, one site had three individuals entering data and the other two sites each had two individuals entering data.

4.3 Summary of Survey Results

There was considerable data collected on the site geology/lithology, hydrogeology, geochemistry, monitoring methods, DNAPL composition, distribution and other source zone characteristics and characterization methods. The information summarized in this section highlights the major trends and conclusions drawn from the data collected and the overall performance of the applied technologies. Appendix C and D respectively present detailed summaries of all of the data collected from the survey on a question by question basis, and an interpretation of the data in terms of trends observed in impacts of site

characteristics on technology performance/selection, costs, and treatment duration. Tables and figures with detailed breakdowns of the data entered into the survey, sorted by technology, are included in Appendices C and D respectively.

The following caveats are implicit in the data summary provided below:

- *i)* Variations in the number of total data available for each survey question arise due to the following:
 - one survey respondent can enter data for multiple sites;
 - one individual site location can have data for multiple technology applications; and
 - few survey questions required answers before allowing the respondent to proceed forward through the survey, resulting in variable totals per question.
- *ii)* For ease of discussion, the term "site" is used here to refer to a unique combination of site location and technology application. One site location may have multiple technology applications, either used sequentially or targeting different portions of the DNAPL source area(s); however, each technology application is treated as a separate site.
- *iii)* Given the limited amount of available data on field applications of DNAPL remediation, the data collected may not be a statistically significant representation of all field applications. In place of the rigorous statistical analysis, we have opted instead to present the data as collected, and note prominent trends and correlations between factors where the data exists. It should be noted, however, that our conclusions may be impacted by the small data sample size, and are therefore not necessarily representative of all DNAPL sites.

4.3.1 Remedial Technologies Identified in Survey

Survey respondents were asked to specify the technology that was applied at their site and the stage of remediation. Remedial technologies were specified for 118 of the sites entered into the survey. For purposes of analysis, remedial technologies that employed similar principles for DNAPL removal were combined under one "primary technology" (*e.g.*, chemical oxidation using permanganate or Fenton's reagent were defined under the same primary technology of chemical oxidation). Figure 4.3 illustrates the distribution of technologies.

Bioremediation, thermal and chemical oxidation were the most often applied technologies (21%, 23%, and 21%, respectively). Surfactant flushing and ZVI technologies were the least applied technologies (3.4% and 5%, respectively). No information was collected in the survey on sites where co-solvent flushing was applied. Information was also provided for dual-phase extraction, excavation, pump and treat, and 6 undefined other technologies.

Table 4.1 and Figure 4.3 provides a summary of the breakdown of the primary technologies: thermal, chemical oxidation, dual-phase extraction, other, and zero-valent iron into "secondary" technology descriptions that represent variations of the primary technology. Of the thermal technologies, resistive heating (six- and three-phase) and steam flushing were the most frequently applied, whereas conductive heating and low-temperature six-phase heating were each applied at one site. Of the chemical oxidation technologies, permanganate appears to be the most frequently chosen oxidant, while data for only one ozone treatment site was entered into the survey. There does not appear to be a preferred method for dual-phase extraction approaches. Of the zero-valent iron variations, non-specified use of ZVI was the most frequently applied alternative. EZVI using nano-scale iron is still in the development stage with only one pilot test completed to date.

Survey respondents were asked to specify, from a range of options, their reasons for selecting a particular technology. Appendices C and D contain detailed breakdowns of the responses.

Survey respondents were asked to specify the stage of remediation for each site (*i.e.*, pilot *versus* full-scale and completed *versus* ongoing). Of 80 total DNAPL treatment attempts, 31 were full-scale applications and 49 were pilot tests. Data was collected for only 2 full-scale systems and 33 pilot tests where the remediation was considered to be complete and post-treatment monitoring was not on-going.

4.3.2 Aquifer Geology/Lithology

The aquifer geology data collected with the survey consisted of the type of media (consolidated versus unconsolidated), degree of fracturing, heterogeneity and matrix permeability. These data are summarized in detail in Section 1 of Appendix C. DNAPL source areas were predominantly located in unconsolidated media (89%), with sand being most prevalent (45%) (Table 4.2, Figure 4.4). Only 11% had predominantly consolidated media in the source area.

Section 2.1 of Appendix D presents the interpretation of geology and hydrogeology on technology choice and performance. The technologies chosen for the following predominant source area lithologies were:

- *i)* **Only unconsolidated media:** dual phase extraction, pump and treat, and surfactant flushing.
- *ii)* One sedimentary rock site, remainder only unconsolidated media: excavation, ZVI and thermal.
- *iii)* Large range of geologic media types, including both unconsolidated and consolidated: bioremediation and chemical oxidation.

A number of interesting trends arise from the analysis of technology performance with respect to site geology/lithology (Section 2.1 of Appendix D). For example, sites with consolidated media were generally described as follows:

- *i)* no successful fractured bedrock sites are reported;
- *ii)* no one entered data of estimates of the DNAPL source mass removal at a fractured bedrock site;
- *iii)* neither of the fractured bedrock sites with estimates of mass flux observed any reduction in mass flux; and
- *iv)* the occurrence of post-treatment rebound was unknown.

In comparison, sites with predominantly unconsolidated media could be described as follows:

- *i)* 28 of 43 sites with unconsolidated media were perceived to be successfully remediated, with the remaining almost one-third perceived to be fairly successful (one poor success as well);
- 59% of 20 sites had a >90% reduction in the DNAPL source mass, with the majority of the remainder (90% in total) experiencing at least a 50% reduction in DNAPL mass;
- *iii)* the majority (68% of 19 sites) of the sites had a mass flux reduction of 80 to 100%, with only one site experiencing less than a 40% decrease; and
- *iv)* 14 of 20 sites did not experience post-treatment rebound.

4.3.3 Aquifer Hydrogeology

The aquifer hydrogeology data collected with the survey are summarized in Section 2 of Appendix C. Section 2.2 of Appendix D presents the interpretation of geology and hydrogeology on technology choice and performance.

Section 2.2 of Appendix D contains an analysis of the impact of the aquifer hydrogeology on remedial success. No trends in the data are discernible, likely due more to a lack of data than a lack of impact of the hydrogeology. The breakdown of the remainder of the success parameters is not shown, since the data available for these factors are even fewer in number.

4.3.4 DNAPL Source Zone Characteristics

A series of questions were asked in the survey pertaining directly to the DNAPL source zone. These included the following:

- *i)* Source area chemistry (*i.e.*, DNAPL contaminant profiles, groundwater geochemistry);
- *ii)* Size and distribution parameters of the DNAPL impacted zone (*i.e.*, areal extent, volume, maximum depth of the DNAPL, DNAPL distribution and shape); and
- *iii)* DNAPL mass estimation (*i.e.*, methods of determining presence and mass, and an estimation of pre-remediation DNAPL mass).

4.3.4.1 Source Area Chemistry

Section 3 of Appendix C contains a summary of the composition of the DNAPL and dissolved phase data. A total of 75 sites had both technology and DNAPL composition data specified. Figures 4.5 and 4.6 illustrate the breakdown of the sites with a specified technology by the DNAPL composition. A total of 80 sites had groundwater chemistry data specified. The majority (68%) of the sites were impacted with only chlorinated ethenes; a smaller percentage had mixed DNAPL compositions (29%). Four sites did not have chlorinated ethenes, of which three of the sites had DNAPL consisting solely of chloroethanes and one site had both chloromethanes and chlorobenzenes. As expected, the groundwater contaminant data was similar to the breakdown of the DNAPL composition data.

Section 2.3.1 of Appendix D contains an analysis of the distribution of technologies with DNAPL and dissolved phase composition. The conclusions may be summarized as follows:

- *i)* Bioremediation, thermal, excavation and dual phase extraction were applied at sites contaminated with all four groups of DNAPLs;
- *ii)* ZVI and surfactant flushing were applied at sites with only chloroethene contamination; and
- *iii*) Chemical oxidation was applied at sites with chloroethenes, chloroethanes, and chloromethanes as components of the DNAPL, and also at sites with these components plus chlorobenzenes as dissolved phase contamination. The two sites without chloroethenes were sites impacted by chloroethanes; at each of these sites, remediation was undertaken by chemical oxidation using Fenton's Reagent.

Section 3 of Appendix C also contains a breakdown of the dissolved phase groundwater contaminant data by the maximum, minimum, average and median concentrations, along with a comparison to the single-component solubility of each compound (*i.e.*, the theoretical maximum concentration that may be achieved from dissolution of DNAPL).

4.3.4.2 Source Area Size and Distribution Parameters

Section 3 of Appendix C contains the details on the data collected on the areal extent and volume of the DNAPL-impacted zone, the maximum depth of the DNAPL, and the DNAPL distribution. Section 2.3 of Appendix D provides an analysis of these factors with respect to technology choice and effectiveness. A summary of the findings is included below:

Areal extent of the DNAPL Impacted Zone

Areal extent ranged from 0.001 ft^2 to over 1,000,000 ft^2 , but the majority (31 % of 75 sites) had a DNAPL areal extent of 10,000 to 100,000 ft^2 . Sites with areal extents less than 100 ft^2 were generally technology demonstrations; the accuracy of the one site with an area less than 1 ft^2 could not be independently verified. Areal extent was unknown for 48 (39%) of 123 sites.

On a per technology basis, bioremediation and chemical oxidation have been applied at the largest range in areal extent. Of the 7 sites with areal extents of source zones greater than 1,000,000 ft², three of the sites were remediated using excavation, two with dual phase extraction and one each with bioremediation and thermal. There are no obvious trends in the success data in terms of the areal extent.

Volume of the DNAPL Impacted Zone

The volume of the DNAPL impacted zone was unknown for 55 (47%) of the 123 sites. The responses ranged from 1 ft³ to over 1,000,000 ft³, but the majority (67% of 63 sites) of the sites had a DNAPL impacted volume of greater than 100,000 ft³.

Chemical oxidation and excavation have been applied at sites with the largest range in volumes of DNAPL impacted soil, followed by bioremediation and dual phase extraction. Thermal treatments have been mainly applied at sites with greater than 100,000 ft³ volume of DNAPL impacted soils. There were no trends in the success data with respect to DNAPL volume.

Maximum DNAPL Depth

The maximum depth of the DNAPL zone was unknown for 43 (36%) of 118 sites that information was provided for .The responses ranged from 1 ft to 10,000 ft bgs, however, the majority (77% of 77 sites) of the sites had a DNAPL depth between 10 and 100 ft bgs.

Bioremediation has been applied at sites with the greatest depths of DNAPL distribution. Chemical oxidation, thermal, and excavation have also been applied at sites with deep DNAPL sources. Correlation with success data are presented in Appendix D.

DNAPL Distribution

One of the questions in the survey asked how the DNAPL was distributed in the subsurface, in pools, as residual, sorbed to the soil particles, diffused into low K layers (rock matrix or clay) or in dead-end fractures. The respondents were asked to choose all of the parameters that applied to their site, resulting in greater than 100% totals. Of the 179 responses to this question representing 75 sites, the majority of the sites had residual (83%) or sorbed (61%) DNAPL. Of the remainder of the responses, 44% of the sites had pooled DNAPL, 40% had DNAPL diffused into low K layers and only 11% had DNAPL trapped in dead-end fractures.

Surfactant flushing was applied at sites with DNAPL present as either pools or at residual saturation, whereas bioremediation, chemical oxidation, dual phase extraction and excavation were attempted at sites with DNAPL distributed in all of the classifications. No trends were observed with respect to DNAPL distribution and success parameters.

4.3.4.3 DNAPL Detection and Mass Estimation

Table 4.3 provides a summary of the methods used to identify DNAPL presence and shows that 2 of a total of 122 sites did not estimate the presence of DNAPL. DNAPL presence at these 122 sites was most frequently inferred from groundwater chemistry and site history (at 69 and 63 sites, respectively). However, known spills and direct observation of DNAPL in wells and soil samples were used at a large number of sites as well (32, 30 and 37 sites respectively).

Section 2.3.7 of Appendix D contains a summary of the relative frequency that each DNAPL sampling method is used at a site.

Section 2.3.7 of Appendix D also contains a discussion of the correlation between the various remedial success parameters and the sampling method used to detect and locate the DNAPL source area. While it should be noted that there was minimal data available to analyze, it is interesting to note that at the sites where the treatment was perceived to be successful, the average method for locating the DNAPL source area consisted of an approximately even mixture of fully-screened monitoring wells, nested monitoring wells, depth-discrete samples and soil samples. Conversely, the less successful sites predominantly used fully-screened monitoring wells on average.

4.3.5 Treatment Cost

Survey respondents were asked to specify the costs (in US dollars) for design and implementation of the remediation, omitting the costs required for the initial site characterization. The DNAPL treatment cost data from all sites is shown in a histogram in Figure 4.7, and summarized in Table 4.4. The costs in Table 4.4 and Figure 4.7 are broken down into the scale of treatment (*i.e.*, full-scale *versus* pilot test) as well as the size of the DNAPL impacted zone (small applications had areas < 10,000 ft² and volumes < 100,000 ft³; large applications had areas > 10,000 ft² and volumes > 100,000 ft³).

Cost data was provided for 16 sites at which full-scale treatment was either completed (1 site), treatment was complete but post-treatment monitoring was on-going (6 sites), or treatment was on-going (9 sites). The average cost for all of the 16 full-scale applications was \$2.8M. Full-scale applications on smaller sites had costs that were generally an order of magnitude smaller than the larger sites. Of the large, full-scale applications, remedial treatment costs were generally in the millions of dollars, with the largest cost being a \$15M water/DNAPL dual-phase treatment (on-going) at a site with a DNAPL impacted zone that was in the range of 100,000 to 1,000,000 ft². The smallest cost was reported as being \$75K at two sites where full-scale treatment was completed and post-treatment monitoring was ongoing. The sites were a ZVI treatment for a site with a DNAPL impacted zone that was in the range of 100 to 1,000 ft² and a bioremediation treatment with an unknown size. Thermal, pump and treat and dual-phase

extraction applications appeared to be significantly more expensive than the chemical oxidation applications on large sites; however, these costs will be strongly impacted by the actual size of the targeted treatment zone, and therefore do not necessarily reflect all sites.

Data was also collected for a total of 31 pilot tests, of which the majority of the tests were completed (all but 3 sites). None of the pilot tests had costs greater than \$2M, with the majority being less than \$0.5M. It should be noted that a significant portion of the pilot tests under \$0.5M were small-scale (<1,000 ft²) technology demonstrations/field research projects. If the technology demonstrations are removed from the count, the pilot test costs were generally within the range of \$150,000 to\$2M. The breakdown of costs data is presented in Appendix D.

Unit costs based on the area and volume of the DNAPL impacted zone of the aquifer and DNAPL mass removed were calculated for the full-scale applications to provide a fairer basis for cost comparison between technologies (see Section 1.2.1 of Appendix D). However, the lack of data prevented any meaningful comparison between technologies. The unit costs appeared to be affected by the size of the DNAPL impacted zone, with smaller sites having much larger unit costs.

4.3.6 Treatment Duration

Section 1.2.2 in Appendix D outlines the treatment duration data. No trend in the pilot test data is discernible, except perhaps that bioremediation pilot tests were generally longer than the remaining technologies. The one conclusion that can be drawn from the comparison of full-scale applications is the significant difference between the duration of the remedial technologies that employ *in situ* destruction or enhancement of the DNAPL mobilization/flushing mechanisms (*e.g.*, bioremediation, chemical oxidation, excavation, thermal and ZVI technologies) versus that of technologies that rely on flushing as the principal DNAPL removal mechanism (*i.e.*, pump and treat and dual-phase extraction). The expected treatment durations of the three sites using water/DNAPL dual-phase extraction and pump and treat were 45 and 75 years (dual-phase) to 158 years (pump and treat). The remainder of the technologies had expected durations of less than 4 years. It should be noted that pump and treat has been used on that site for five years to date, while the two dual-phase extraction sites have been in operation for 10 and 15 years, and are still operating.

4.3.7 Technology Performance Evaluation

Several remedial effectiveness criteria were evaluated to determine what impact technology choice and site characteristics may have on the degree of success of remediation. General technology performance criteria included the following:

- *i)* the perception of technology effectiveness and performance (Section 4.3.7.1); and
- *ii)* factors impacting remedial performance (Section 4.3.7.2).

More specific technology performance criteria were also evaluated, including the following:

- *i)* extent of DNAPL mass removal (Section 4.3.7.3);
- *ii)* decrease of mass flux (Section 4.3.7.4);
- *iii)* occurrence of rebound of dissolved phase concentrations (Section 4.3.7.5);
- *iv)* achievement of remedial goals and/or site closure (Section 4.3.7.6); and
- *v)* impacts to secondary groundwater quality (section 4.3.7.7).

DNAPL remediation technologies may be effective in terms of only one or more of the above criteria; however, the most effective technology would meet all criteria. Therefore, each technology was also evaluated in how well it met all criteria (Section 4.3.7.8).

4.3.7.1 Perception of Technology Effectiveness

Section 1.3.1 of Appendix D contains a detailed analysis of the rating of technology performance at each site. Table 4.5 and Figure 4.8 summarize the perceived success of the technology applications. The conclusions are summarized below:

- i) Of the sites where success was evaluated, more than 50% of the sites were rated successful for bioremediation (78% of 9 sites), chemical oxidation using Fenton's reagent (63% of 8 sites), excavation (75% of 4 sites), ZVI (67% of 3 sites), and thermal (100% of 4 sites of six-phase heating, 50% of 2 sites of steam flushing, and 100% of 1 site each of conductive and low-temperature six-phase heating) technologies.
- *ii)* Only bioremediation, excavation and thermal technologies (six-phase, conductive and low-temperature six-phase heating) were rated successful for more than 75% of the sites.
- *iii)* Technologies that were predominantly rated as having a fair success were chemical oxidation using permanganate (57% of 7 sites), surfactant flushing (100% of 2 sites), the thermal technology resistive heating (1 site), and EZVI (1 site).

iv) The three sites using dual-phase water/air extraction had ratings of good, fair and poor successes.

There was not enough data to discern any correlation between role in the environmental community and perception of success. Survey respondents were also asked to rate their technology according to a list of effectiveness criteria, which are summarized in Section 1.3.1 of Appendix D.

4.3.7.2 Factors Impacting Remedial Effectiveness

Section 1.3.2 of Appendix D provides a summary of specific limitations of each technology. Also included in Section 1.3.2 of Appendix D is a summary of the impact of external factors (e.g., presence of site infrastructure, proximity to surface water, etc.) on remedial effectiveness. No trends were apparent, other than that most technologies were impacted by budget/remediation costs.

4.3.7.3 DNAPL Mass Removal

Table 4.6 and Figure 4.9 summarize the source mass removal data that were specified by respondents to the survey. Section 1.3.3 of Appendix D contains a detailed summary of the data and analysis. There were a total of 20 sites with source mass removal data. The majority (70%) of sites had >80% mass removal. The technologies used at these sites were chemical oxidation (2 permanganate, 4 Fenton's reagent), thermal (one each of six phase heating, steam and conductive heating), excavation (3 total), surfactant flushing (1 site) and bioremediation (1 site).

Only four sites achieved 100% mass removal (one excavation pilot test, one sixphase and one conductive heating [both full-scale], and one chemical oxidation pilot test using permanganate). Of these sites, rebound was not evaluated at two (50%) of the sites; the DNAPL was present only as residual in three (75%) of the sites and had an unknown distribution at the fourth. Two (50%) of the sites were full-scale treatments (one with post-treatment monitoring ongoing) and two (50%) of the sites were pilot-scale treatments.

DNAPL/air dual-phase extraction performed poorly with less than 10% mass removal. The low-temperature six-phase heating pilot test also performed poorly in terms of mass removal; however, it should be noted that this pilot test was terminated prematurely in order to focus on another remedial technology. Therefore, the extent of DNAPL mass removal does not necessarily reflect what may have been achieved if the pilot test had continued.

4.3.7.4 Mass Flux Decrease

Table 4.7 and Figure 4.10 summarize the mass flux data that were specified by respondents to the survey. Section 1.3.4 of Appendix D contains a detailed overview of the data and analysis. There were a total of 21 sites with mass flux decrease data. The majority (62%) of sites had a mass flux decrease greater than 80%. The technologies applied at these sites included chemical oxidation (2 each permanganate and Fenton's reagent), thermal applications (5 six phase heating and 1 steam), bioremediation (2), and surfactant flushing (1). However, rebound was evaluated at only three of these 13 sites (two with no rebound, one with rebound). The DNAPL distribution was unknown for eight sites and present as residual at four sites and residual and sorbed at one site. All but one site were pilot scale treatments.

In comparison, 86% of sites had a mass flux decrease greater than 50%. The only technologies that did not achieve >80% mass flux decrease in at least half of the sites were excavation and EZVI. Two sites had no mass flux decreases (chemical oxidation using permanganate was applied at one, excavation the other). Both of these technologies were targeting DNAPL in a predominantly fractured bedrock environment. The excavation was on-going. Success had not yet been evaluated for the chemical oxidation application and the excavation site was considered a fair success.

Mass flux as a performance metric for DNAPL remediation is still relatively new and its use as a performance metric to evaluate source depletion technologies is not generally accepted practice (EPA DNAPL Panel, In Press). Contaminant mass flux is the locally defined contaminant discharge per unit area (mass per unit area per time). It is possible that a number of the survey responses about measured decreases in mass flux were actually more applicable to either a decrease in mass discharge or in groundwater concentrations.

4.3.7.5 Post-Treatment Monitoring and Rebound of Dissolved Phase Concentrations

For many of the parameters used to evaluate success, a certain level of posttreatment monitoring is necessary to determine whether the level of mass decrease or reduction in mass flux was a permanent change or a temporary change due to things such as dilution. Table 4.8 summarizes the post-treatment monitoring stage data that were specified by respondents to the survey. Post-treatment monitoring was generally terminated within 4 years (29 of 31 sites). One site had monitoring on-going for 10 years. Section 5 in Appendix C provides more details on the data.

Table 4.9 and Figure 4.11 summarize the rebound data (see Section 1.3.5 in Appendix D for more details). There were a total of 21 sites with rebound data. The majority (67%) of sites did not have rebound occur. The technologies where rebound did occur were permanganate (83% of 6 sites), excavation (50% of 2 sites), and low-

temperature six-phase heating (100% of 1 sites). Again, it should be emphasized that the low-temperature six-phase heating site was a pilot test that was terminated prematurely; therefore, the occurrence of rebound does not necessarily reflect the result of a completed application. Of the sites with rebound, four (57%) had DNAPL present in pools and two (29%) had DNAPL diffused into low K layers. However, of the 14 sites with no observed rebound, three sites (21%) had DNAPL in pools and four (29%) had DNAPL diffused into low K layers so DNAPL distribution may not be a good indicator on its own for evaluating the difficulty of achieving success.

4.3.7.6 Achievement of Remedial Goals and Site Closure

Survey respondents were only asked to evaluate success based on source mass removal, mass flux decrease, rebound and perceived success. Subsequent to the closure of the survey, an email request was sent out to the 53 participants who provided success data asking them the following questions:

- *i)* State whether they achieved dissolved phase concentrations reduced to less than maximum concentration levels (MCLs) upon completion of the remediation scheme.
- *ii)* If MCLs were achieved, state the stage of treatment (*i.e.*, pilot test versus full-scale) and the length of post treatment monitoring.
- *iii)* If MCLs were not achieved, were remedial goals met?
- *iv)* If MCLs were not achieved, but regulatory closure was, state how regulatory closure of the site was achieved.

In total, 8 sites had responses to these questions. Although out of the 53 sites, 4 sites were believed to have had 100% source mass removal and 13 sites had greater than 81% mass flux reduction, only one site claimed to have achieved MCLs after remediation. The site where MCLs were achieved was an excavation application with silt as the dominant lithology; however, the site did not achieve regulatory goals with respect to cis 1,2-DCE which has an MCL of 70 ppb and a site clean up goal of 5 ppb. The post remediation monitoring is ongoing, several years after the completion of remediation.

As a number of the sites were pilot scale demonstrations, achieving MCLs and site closure would not be expected at that stage of remediation. Of the 8 responses to the questions about MCLs, five stated that the pilot scale remediation attempts did meet the project goals, which were set above MCLs. Of these, two were bioremediation applications, one was a dual phase application, one was a permanganate application that is now going to full scale and one is an application of six phase heating that is now going to full scale. Of the remaining two sites, one was a Fenton's application which did not

achieve remedial goals or MCLs and one was a full scale application of bioremediation were MCLs and regulatory closure are expected to be achieved within three years.

4.3.7.7 Impacts to Secondary Groundwater Quality

Table 4.10 provides a list of the secondary groundwater quality parameters that the survey respondents had to choose from, of which they were to check all that apply. Figure 4.12 summarizes the secondary groundwater impacts sorted by technology. Section 1.3.7 in Appendix D contains a more detailed discussion of the data, broken down by technology.

In general the most prevalent secondary groundwater impacts included: 1) generation/mobilization of undesirable compounds; 2) changes in groundwater aesthetic parameters 3) changes in soil hydraulic properties; and, 4) changes in DNAPL distribution.

In general, bioremediation, chemical oxidation (particularly permanganate) and dual phase extraction had the most secondary groundwater issues identified, although some of them were noted to be temporary (*i.e.*, groundwater color changes with the application of permanganate).

4.3.7.8 Overall Technology Performance

It is possible for remediation at a site to meet one or more of the success criteria, yet fail to meet other criteria. For example, a large amount of DNAPL mass may be removed from the subsurface, but rebound may occur coupled with insignificant mass flux decrease. The optimal result from remedial activities would be a large amount of DNAPL mass removed, significant mass flux reduction, and no post-treatment rebound of dissolved phase concentrations. To evaluate the level of success that was achieved for various combinations of site conditions and technologies, the sites were sorted according to varying degrees of each success criteria. Sites with "unknown" responses to any of the success criteria eliminated the site from consideration. The following results were found:

- *i)* Greater than 80% mass removal AND a greater than 61% reduction in mass flux AND no observed rebound AND a perceived success: Only two sites met the most stringent of the success criteria. Bioremediation and chemical oxidation (permanganate) were applied at these sites. Both sites had sandy aquifers and were pilot-scale applications.
- *ii)* 80% mass removal AND a greater than 61% reduction in flux AND perceived a success: Four sites met these criteria, with the technologies being chemical oxidation (permanganate two sites), bioremediation

(one site) and excavation (one site). All four sites had sandy aquifers. All but the site with excavation were pilot-scale applications.

- *iii)* 80% mass removal AND a greater than 61% reduction: Six sites in total met these criteria, with the technologies applied being chemical oxidation (permanganate two sites), bioremediation (one site), excavation (one site), surfactant flushing (one site) and thermal six-phase heating (one site). All sites except for the surfactant flush had sandy aquifers; the surfactant flush was conducted in a clay aquifer. Only the excavation and the thermal six phase heating were full-scale applications, the remaining were pilot-scale.
- 80% mass removal AND no observed rebound AND a perceived success: Seven sites met these criteria. These sites include the three chemical oxidation sites (one permanganate, two Fenton's reagent), one bioremediation, two thermal applications (steam and conductive heating) and one excavation. If the perceived success criterion is removed from this evaluation, another chemical oxidation application (Fenton's reagent) is included. The sites were all unconsolidated, and the lithologies comprised three sand, three clay and one silt sites. Five of the sites were completed full scale applications and two were pilot scale applications.
- W) Greater than 61% reduction in flux AND no observed rebound AND a perceived success: Two bioremediation, one chemical oxidation (permanganate), and one thermal (six phase heating) sites met these criteria (four in total). Two of the sites had sandy aquifers, the other two had clay. Two of the applications were full scale and two were pilot scale.

The two predominant trends that appear to be consistent throughout all sites that met all degrees of success criteria were:

- *i)* Not one of the "successful" sites had remediation applied in unconsolidated media. The prevalent lithology was sand, with some sites with predominantly clay.
- *ii)* The majority of the "successful" sites were applied at the pilot-scale.Only excavation and thermal technologies met even the least stringent of the criteria for full-scale applications.

5 CONCLUSIONS

The principal goal for this review and survey of chlorinated solvent DNAPL remediation case studies was to gather information on the current technologies used to treat DNAPL sources and their relative success to provide the first step towards developing technology selection guidelines. The following general conclusions can be made from the gathered data.

- The web-based survey proved to be a useful tool for collecting and compiling a cross section of case study data, in a consistent manner that reflected the impressions of users and reviewers of performance data.
- DNAPL source zone remediation has been attempted at a wide range of sites using a number of different technologies with varying success. Although remediation of source areas to below MCLs is an elusive goal, ongoing development and testing of innovative remedial technologies suggests that source zone remediation can cause significant reduction of mass discharge and dissolved phase concentrations of volatile organic compounds (VOCs) from source areas.
- Thermal remediation technologies are among the most common DNAPL treatment technologies; however, the use of innovative technologies is becoming more common. Based on the results of the survey, thermal technologies were the most often applied, chemical oxidation was a close second along with a surprising number of bioremediation applications (23%, 21%, and 21%, respectively). Surfactant flushing and ZVI technologies were the least applied technologies (3.4% and 5%, respectively). Information was also provided for dual-phase extraction, excavation, pump and treat, and 6 undefined other technologies. No information was collected in the survey on sites where co-solvent flushing was applied.
- DNAPL-impacted sites range greatly in size, with areal extents from <100 ft² to over 100,000 ft² and volumes from <10 ft³ to >1,000,000 ft³. The majority of the sites had a DNAPL areal extent of 10,000 to 100,000 ft² and a DNAPL impacted aquifer volume of greater than 100,000 ft³. Many of the sites for which data was collected were pilot scale demonstrations. Sites with areal extents less than 100 ft² were generally technology demonstrations and the sizes of the site reported in the survey are not necessarily indicative of the extent of DNAPL contamination at the site. Of the 123 sites with data on the DNAPL areal extent, 48 (39%) said that the areal extent was unknown and 55 (46.6%) said that the volume was unknown.

- A key parameter when designing a remediation strategy is the DNAPL distribution in the subsurface. Although the most common distribution form for DNAPL is as residual or sorbed mass (83% and 61%, respectively based on survey responses), a large percentage (44%) of the sites estimated that DNAPL was present in the subsurface as pools. In addition, 40% had DNAPL diffused into low K layers and 11% had DNAPL trapped in deadend fractures.
- The use of mass flux (*i.e.*, the locally defined contaminant discharge per unit area) as a performance metric to evaluate source depletion technologies is still relatively new and is not a generally accepted practice. However there were a total of 21 responses with a measured decrease in mass flux observed after remediation. Although 2 sites (10%) had 0% decrease in mass flux, 13 sites (62%) had 81% to 100% decreases in mass flux. It is possible that a number of the responses about a measured decrease in mass flux were actually more applicable to either a decrease in mass discharge or a decrease in groundwater concentrations.
- Lithology/geology, hydrogeology and DNAPL source zone architecture (size, depth, distribution, *etc.*) are all parameters that impacted successful DNAPL remediation; however, information on these factors was either unknown or not measured for a large percentage of the case example responses. The impact of these parameters on the success of DNAPL remediation appears to be unknown or ignored by vendors, regulators, or users of these technologies.
- None of the remediation attempts presented in this survey/review achieved MCLs or regulatory site closure.
- Although meeting MCLs is not always the reason source reduction is attempted, there are other tangible and intangible benefits, such as mass removal, regulatory favor, risk reduction, which can be derived from such an attempt.
- Although remediation was perceived to be successful at 28 sites:
 - Only one remediation (4%) application, a chemical oxidation case using permanganate), met the most stringent criteria for evaluating success). This site was a pilot scale demonstration in a sand with an emplaced DNAPL source.

- Two (2) cases (9.5%) met criteria of >61% reduction in mass flux, the prior chemical oxidation case and an additional bioremediation application. Both sites are pilot scale applications conducted in sand.
- o Fourteen (14) sites (67%) had ≥80% source mass removal and only 4 sites (19%) claimed to have 100% source mass removal. Of the 4 sites with 100% mass removal, three had unknown mass flux reduction and/or unknown rebound.
- Thirteen (13) sites (62%) had \geq 81% reduction in mass flux.
- Full-scale applications typically cost millions of dollars, with the largest cost being a \$15M water/DNAPL dual-phase treatment (on-going) at a site with a DNAPL impacted zone that was in the range of 100,000 to 1,000,000 ft². None of the pilot tests had costs greater than \$2M, with the majority being less than \$0.5M.
- In terms of treatment duration, from the comparison of full-scale applications there is a significant difference between the duration of the remedial technologies that employ in situ destruction or enhancement of the DNAPL mobilization/flushing mechanisms (e.g., bioremediation, chemical oxidation, excavation, thermal and ZVI technologies) versus that of technologies that rely on flushing as the principal DNAPL removal mechanism (i.e., pump and treat and dual-phase extraction). The expected treatment durations for technologies relying on flushing ranged from 45 to 158 years in comparison with the remainder of the technologies which had expected durations of less than 4 years.

The data collected from this survey suggests that DNAPL remediation efficacy is extremely difficult to gauge, and although employing aggressive remedial technologies has not achieved site closure, significant mass removal can be achieved.

6 RECOMMENDATIONS

To further refine the benefit of source treatment requires that:

- *i)* **Update the status of on-going field applications to expand upon the database.** The following list contains suggestions as to how to do this:
 - Follow up on sites listed as ongoing in the survey to track progress, and on sites with incomplete data.
 - Expand on the case study data collection, either through re-releasing the web-based survey for public input, continuing to review the literature for case study information, and/or directly targeting particular sites and obtaining case study information through site personnel interviews. Insufficient amounts of data were collected for much of the information requested, which can result in the results not being statistically significant.
- *ii)* **Collecting data on a finer scale to allow statistical data analysis.** Data was collected using "ranges" of input values that where selected by the user. The selection of ranges reduces the ability to quantify factors that affect remedial performance and predict success, such as correlation between technology performance metrics (*e.g.*, remedial success and lithology).
- *iii)* Collaborate with others collecting this type of information to pool resources and share data. Partnering with EPA, the ITRC, and others to aid in filling in data gaps or collecting data on a finer scale. The combined efforts will help to improve the state of the knowledge and help to streamline the remediation process.

7 REFERENCES

Abriola, L.M., C.D. Drummond, K.M. Rathfelder, L.D. Lemke, E.J. Hahn, and K.D. Pennell (2000). Surfactant enhanced remediation: application of numerical models in pilot test design. *In* Groundwater 2000, Proc. Of the International Conference on Groundwater Research, Copenhagen, Denmark, 6-8 June 2000, *P.L. Bjerg, et al., eds., 33-36, A.A. Balkema, Rotterdam, The Netherlands.*

Abriola, L.M., C.D. Drummond, L.D. Lemke, K.M. Rathfelder, K.D. Pennell, E. Petrovskis, and G. Daniels (2001). Surfactant enhanced remediation: application of mathematical models in the design and evaluation of a pilot test. *In* Groundwater Quality 2001, Pre-prints of the Third International Conference on Groundwater Quality, 18-21 June, University of Sheffield, United Kingdom, *6-8*.

Baker, Ralph S., and Myron Kuhlman 2002. A Description Of The Mechanisms Of In-Situ Thermal Destruction (ISTD) Reactions. in H. Al-Ekabi (Ed.), Current Practices in Oxidation and Reduction Technologies for Soil and Groundwater, and presented at the 2nd International Conf. on Oxidation and Reduction Technologies for Soil and Groundwater, ORTs-2, Toronto, Ontario, Canada, Nov. 17-21, 2002.

Battelle, 2002. Steam Injection Treatment of a DNAPL Source Zone at Launch Complex 34, Cape Canaveral Air Station. Draft Final Technology Evaluation Report. August 20 2002

Battelle, 2003. Demonstration of Resistive Heating Treatment of DNAPL Source Zone at Launch Complex 34 in Cape Canaveral Air Force Station, Florida. Final Innovative Technology Evaluation Report. February 19, 2003.

Battelle and Duke Engineering Services, 2002. NFESC Technical Report TR-2206-ENV: Surfactant-Enhanced Aquifer Remediation (SEAR) Design Manual. April, 2002.

Carr, C.S., S.Garg, and J.B. Hughes, 2000. Effect of dechlorinating bacteria on the longevity and composition of PCE-containing nonaqueous phase liquids under equilibrium dissolution conditions, Environ. Sci. and Tech., 34(6), 1088-1094.

CH2M HILL, 1997. Technology Practices Manual for Surfactants and Cosolvents. February, 1997.

Chen, S., Markesic S. (2001) An innovative Approach to the Emplacement of Zero-Valent Iron for the Treatment of Chlorinated Volatile Organic Compounds Presented at the 1st International Conference on Oxidation Reduction Technologies for Insitu Treatment of Soil & Groundwater, Niagra Falls. Choe S., Lee S-H., Chang Y.-Y., Hwang K-Y., and Khim J. 2001. Rapid Reductive Destruction of Hazardous organic compounds by nanoscale Fe^o. Chemosphere 42 (2001) 267-372.

Chu, M., P.K. Kitanidis, and P.L. McCarty, 2003. Effects of biomass accumulation on biologically enhanced dissolution of a PCE pool: a numerical simulation. *Journal of Contaminant Hydrology*, 65:75-100.

Costanza, J, K.D. Pennell, J.A. Mulholland 2003a. Laboratory Study on the Transformation of Organic Contaminants Under Thermal Source Zone Removal Conditions. In Proceedings of the 2003 Georgia Water Resources Conference, April 23-24, Athens, GA.

Costanza, J, K.D. Pennell, J.A. Mulholland, and E.L. Davis. 2003b. Systematic laboratory study on the transformation of organic contaminants under thermal source zone removal conditions. Platform presentation 225th American Chemical Society Meeting, March 23-27, New Orleans, LA.

Cope, N., and J.B. Hughes, 2001. Biologically-enhanced removal of PCE from NAPL source zones, Environ. Sci. and Tech., 35(10), 2014-2020.

Dai, Q., and S. Reitsma, 2002. Two-dimensional experimental studies of permanganate flushing of pooled DNAPL, in the proceedings of the Third International Conference on Remediation of Chlorinated and Recalcitrant Compounds, Monterey, CA, May, 2002.

Dettmer, Karen, 2002. A Discussion of the Effects of Thermal Remediation Treatments on Microbial Degradation Processes. August 2002 for U.S. Environmental Protection Agency Office of Solid Waste and Emergency Response Technology Innovation Office Washington, DC <u>www.clu-in.org</u>

Drummond, C.D., L.D. Lemke, K.M. Rathfelder, E.J. Hahn and L.M. Abriola (2000). Simulation of surfactant-enhanced PCE recovery at a pilot test field site. *In* Treating Dense Nonaqueous-Phase Liquids (DNAPLs): Remediation of Chlorinated and Recalcitrant Compounds (ed. By G.B. Wickramanayake, A.R. Gavaskar, and N. Gupta). 77-84, Battelle Press, Columbus.

Edwards, E.A. and E.E. Cox, 1997. Field and laboratory studies of sequential anaerobic-aerobic chlorinated solvent biodegradation, Accepted for publication in the proceedings of the Fourth International *In situ* and On-Site Bioremediation Symposium, New Orleans, LA. April 1997.

Ellis, D.E., E.J. Lutz, J.M. Odom, R.J. Buchanan Jr., and C.L. Bartlett, 2001. Bioaugmentation for accelerated *in situ* anaerobic bioremediation, Environ. Sci. and Tech., accepted for publication. Falta, R.W., C.M. Lee, S.E. Brame, E. Roeder, J.T. Coates, and C. Wright. Field Test of High Molecular Weight Alcohol Flushing for Subsurface NAPL Remediation, Water Resources Research, Vol. 35, No. 7, July, 1999.

Fountain, J.C., 1998. Technologies for Dense Nonaqueous-Phase Liquid Source Zone Remediation. Groundwater Remediation Technology Analysis Center: Pittsburgh, PA.

Gilham, R.W. 1995. Resurgence of research concerning organic transformations enhanced by zero-valent metals and potential application in remediation of contaminated groundwater. Natl. Meet. – Am. Chem. Soc. Div Environ Chem. 1995., 35, 691-694.

Gillham R.W., S.F. O'Hannesin. 1994. Enhanced degradation of halogenated aliphatics by zero-valent iron. Ground Water 32(6), 958-967. 958.

Hood, E.D., 2000. *Permanganate Flushing of DNAPL Source Zones: Experimental and Numerical Investigation*, Ph.D. Thesis, Department of Civil Engineering, University of Waterloo, Waterloo, Ontario.

Hood, E.D., N.R. Thomson, and G.J. Farquhar, 1997. *In situ* oxidation: an innovative treatment strategy to remediate trichloroethylene and perchloroethlyene DNAPLs in porous media, in the proceedings of The 6th Symposium on Groundwater and Soil Contamination, Montreal, Quebec.

Hood, E.D., N.R. Thomson, D. Grossi and G.J. Farquhar, Experimental determination of the kinetic rate law for oxidation of perchloroethylene by potassium permanganate, Chemosphere, 40(12):1383-1388, 1999.

Interstate Technology and Regulatory Cooperation (ITRC) 2001. Technical and Regulatory Guidance for *In situ* Chemical Oxidation of Contaminated Soil and Groundwater. Prepared by the *In situ* Chemical Oxidation Work Team. June 2001.

IT Corporation, 2000. *In situ* Oxidation Demonstration Test Final Report Treatment Cell C Launch Complex 34 DNAPL Source Zone Oxidation Project.

Jawitz, J.W., R.K. Sillan, M.D. Annable, P.S.C. Rao, K. Warner. 2000. *In situ* Alcohol Flushing of a DNAPL Source Zone at a Dry Cleaner Site. Environmental Science & Technology, 2000, 34, pp. 3733-3739.

Kakarla, Prada K., Andrews T., Greenberg R.S., Zervas D.S., 2002. Modified Fenton's Processes for In-Situ Chemical Oxidation – Laboratory and Field Evaluation. Remediation Autumn 2002. 23-36.

Kaslusky, S.F., Udell, K.S.. 2002. A theoretical model of air and steam co-injection to prevent the downward migration of DNAPLs during steam-enhanced extraction. Journal of Contaminant Hydrology 55(2002) 213-232.

Lee, E.S., Y. Seol, Y.C. Fang, and F.W. Schwartz, 2003. Destruction efficiency and dynamics reaction fronts associated with the permanganate oxidation of trichloroethylene, Environmental Science and Technology, 37(11):2540-2546.

Liberati, Michael R. 2003. Use of ZVI and Clay for Source Zone Remediation. Presented at RTDF Permeable Reactive Barriers (PRBs) Action Team Meeting, Niagara Falls, NY. October 15-16, 2003.

Looney, B.B., R.W. Falta, 2000. <u>Vadose Zone Science and Technology Solutions</u>. Editors. Battelle Press.

MacKinnon, L.K., and Thomson, N.R., 2002. Laboratory-scale *in situ* chemical oxidation of a perchloroethylene pool using permanganate, Journal of Contaminant Hydrology, 56:49-74.

Major, D.W., E.E. Cox, E. Edwards, and P.W. Hare, 1995. Intrinsic dechlorination of trichloroethene to ethene in a bedrock aquifer, Intrinsic Bioremediation, R.E. Hinchee, J.T. Wilson, and D.C. Downey (eds), Battelle Press, Columbus, OH, 197-203.

Major, D.W., E.H. Hodgins, and B.J. Butler, 1991. Field and laboratory evidence of *in situ* biotransformation of tetrachloroethene to ethene and ethane at a chemical transfer facility in North Toronto, *In situ* and on Site Bioreclamation, R. Hinchee and R. Olfenbuttel (Eds.), Buttersworth-Heineman, Stoneham, MA.

Nelson, M.D., B.L. Parker, T.A. Al, J.A. Cherry, and D. Loomer, 2001. Geochemical reactions resulting from *in situ* oxidation of PCE-DNAPL by KMnO4 in a sandy aquifer, Environ. Sci. and Technol., 35, 1266-1275.

O'Hara, Suzanne, Thomas Krug, David Major, Jackie Quinn, Cherie Geiger, and Chris Clausen. 2003. Performance Evaluation of Dehalogenation of DNAPLs Using Emulsified ZVI. Presented at RTDF Permeable Reactive Barriers (PRBs) Action Team Meeting, Niagara Falls, NY. October 15-16, 2003.

Pankow, J.F. and J.A. Cherry. 1996. <u>Dense Chlorinated Solvents and Other DNAPLs</u> in Groundwater, Editors. Waterloo Press.

Roberts, L.A., L.A. Totten, W.A. Arnold, D. R. Burris, and T.J. Campbell. 1996. Reductive Elimination of Chlorinated Ethylenes by Zero Valent Metals. Env. Sci. and Tech. Vol 30 No. 8. Roeder, E., R.W. Falta, C.M.Lee, and J.T. Coates, DNAPL to LNAPL Transitions During Horizontal Cosolvent Flooding, Groundwater Monitoring and Remediation, Winter, 2001.

Roote, Diane. 2003. *In situ* Thermal-Enhanced Remediation Technologies, Prepared for U.S. D.O.E., September 2003

Schroth, M.H., M. Oostrom, T.W. Wietsma, and J.D. Istok, *In situ* oxidation of trichloroethene by permanganate: effects on porous medium hydraulic properties, Journal of Contaminant Hydrology, 50:79-98.

Shoemaker, Steve, Richard Landis. 2002. Results of the Kinston Jetted PRB and Source Treatment. Presented at RTDF Permeable Reactive Barriers Workshop, November 6, 2002.

Stegemeier, George L. and Harold J. Vinegar. 2001. Thermal Conduction Heating For In-Situ Thermal Desorption Of Soils Hazardous & Radioactive Waste Treatment Technologies Handbook, Ch. 4.6-1, Oh, Chang H. (Ed.) Copyright 2001, CRC Press, Boca Raton, Florida.

Stroo, H.F., Unger M., Ward C.H., Kavanaugh M.C., Vogel C., Leeson A., Marqusee J., Smith B.P., 2003. Remediation Chlorinated Solvent Source Zones. Env. Sci. and Tech., June 1, 2003. 225A - 230A.

Thermal Remediation Services, Inc.. White Paper: Three-Phase Heating? Six-Phase Heating? Which is Better? [rev 2001; cited 2003 Nov 20]. Available from: <u>http://www.thermalrs.com/TRSPages/WhitePapers.html</u>.

Udell, K.S. & L.D. Stewart. 1998.Combined steam injection and vacuum extraction for aquifer cleanup. Subsurface Contamination by Immiscible Fluids pp. 327-335

U.S. Army Corps of Engineers, 1999. Engineering and Design: Multi-Phase Extraction. EM 1110-1-4010, 1 June 1999.

U.S. Department of Energy, 2003. Pinellas Environmental Restoration Project: Northeast Site Area A NAPL Remediation Final Report. GJO-2003-482-TAC, GJO PIN 13.12.10. September 2003.

U.S. Department of Energy, 1999a. Innovative Technology Summary Report; *In situ* Chemical Oxidation Using Potassium Permanganate. DOE/EM-0496. September 1999.

U.S. Department of Energy, 1999b. Innovative Technology Summary Report; Fenton's Reagent. DOE/EM-0484. October 1999.

U.S. Environmental Protection Agency. (U.S. EPA). 1993. Guidance for Evaluating the Technical Impracticability of Groundwater Restoration. OSWER Directive 9234.2-25.

U.S. EPA 1997. Cleanup of the Nation's Waste Sites: Markets and Technology Trends. EPA 542-R-96-005; U.S. Government Printing Office: Washington DC, 1997.

U.S. EPA, 1999. Multi-Phase Extraction: State-of-the-Practice. EPA-542-R-99-004, June, 1999.

U.S. EPA. In Press. The DNAPL Remediation Challenge: Is there a Case for Source Depletion? Edited by Michael C. Kavanaugh, and P. Suresh C. Rao, Expert Panel In Press.

Vinegar, H.J., Stegemeier, G.L., Stevenson, J.D., Carl, F.G., and Dudley, R.J. (1999), "In-Situ Thermal Desorption of Soils Impacted with Chlorinated Solvents," Paper #99-450, presented at the Air and Waste Management Association Annual Meeting, St. Louis, June 22.

Wadley, Sharon L.S., Robert W. Gillham. 2003. Laboratory and Field Investigations of DNAPL Source Zone Remediation Using Granular Iron. Presented at RTDF Permeable Reactive Barriers (PRBs) Action Team Meeting, Niagara Falls, NY. October 15-16, 2003.

Watts, R.J., Udell, M.D., Rauch, P.A., and Leung, S.W. 1990. Treatment of pentachlorophenol-contaminated soils using Fenton's reagent. *Hazardous Waste Hazardous Materials*, 7(4), 335-345.

Yan, Y.E., and F.W. Schwartz, 1999. Oxidative degradation and kinetics of chlorinated ethylenes by potassium permanganate, J. Contam. Hydrol., 37, 343-365.

Yang, Y., and P. L. McCarty, 1998. Competition for hydrogen within a chlorinated solvent dehalogenating mixed culture, Environmental Science and Technology, 32 (22): 3591-3597.

Yang, Y., and P.L. McCarty, 2000. Biologically enhanced dissolution of tetrachloroethene DNAPL, Environ. Sci. and Technol. 34(14)2979-2984.

Term	Configuration		
Dual Phase Extraction	 Non-specific MPE term 		
Drop-Tube Entrainment Extraction	 Single pump configuration 		
Well-Screen Entrainment Extraction	 Extraction of vapor and groundwater from a sealed well with induced vacuum. Groundwater is aspirated into the 		
	vapor stream at the well screen.		
High-Vacuum Dual Phase Extraction	 Two pump configuration with a submersible pump for 		
(HVDPE)	groundwater recovery. High vacuum application (18 to		
	26 in Hg)		
Low-Vacuum Dual Phase Extraction	 Low permeability, fractured systems 		
(LVDPE)	Two pump configuration with a submersible pump for		
	groundwater recovery. Low vacuum application (2 to 12 in Hg)		
Two-Phase Extraction (TPE)	 Single pump configuration with high vacuum application (18 to 26 in Hg) 		
Bioslurping	 Single pump configuration with high vacuum application 		
	(18 to 26 in Hg), generally applied to LNAPLs		
VE/GE ("Veggie"); Downhole-Pump	Two pump configuration with a submersible pump for		
Extraction	groundwater recovery		
Vacuum Enhanced Pumping (VEP)	 Non-specific MPE term 		
Vacuum Enhanced Recovery (VER)	Non-specific MPE term		

Table 3.1: MPE Terms and Configurations (EPA, 1999; USACE, 1999)

Table 3.2: Optimal Conditions	for Multiphase Extraction	(from p. 13, EPA, 1999)

Parameter	Optimal Conditions for MPE		
Hydraulic Conductivity	• Moderate to low ($K=10^{-3}$ to 10^{-5} cm/s)		
Transmissivity	 Low (≤ 500 gpd/ft) 		
Geologic Setting	 Sands to clays 		
Vadose Zone Soil Permeability to Air	 Moderate to low (k<1 darcy) 		
Formation Characteristics	 Low permeability, fractured systems 		
	 Interbedded sand and clay stringers 		
	 Limited saturated thickness 		
	 Shallow water table 		
	 Thick capillary zone (up to several feet) 		
	 Perched NAPL or groundwater layers 		
Drawdown/Recovery Rate	 Conditions producing steep or high drawdown in wells 		
	 Low groundwater recovery rates achieved with 		
	conventional pumping		
Contaminant Vapor Pressure	 > 1 mm Hg at 20°C 		
Contaminant Volatility	• >0.01 at 20°C		
	• >2x10-4 atm m^3 /mol at 20°C		

Primary Technology	Secondary Technology	Number of Sites Where Technology was Applied
	Six-Phase Heating	10
	Steam	8
Thermal	Three-Phase Heating	5
Therman	Resistive Heating	2
	Conductive Heating	1
	Low-Temp Six-Phase	1
	Permanganate	15
Chemical Oxidation	Fenton's Reagent	9
	Ozone	1
	Water/DNAPL	5
Dual-Phase Extraction	Water/air	5
Extraction	DNAPL/air	3
Other	Undefined	6
Other	Pump and Treat	1
	Zero-Valent Iron	4
Zero-Valent Iron	EZVI/Nano-Scale Iron	1
non	ZVI/Clay Source Treatment	1

 Table 4.1:
 Breakdown of Primary Technologies

Aquifer Media Type	Lithology		Sites with Unspecified Technology	Sites with Technology Specified	Total Sites
		Metabasalt	0	1	1 (50.0%)
	Metamorphic	Metashale	0	0	0 (0.0%)
	Wietamorphie	Other	0	1	1 (50.0%)
		TOTAL	0	2	2 (12.5%)
		Shale	1	5	6 (54.5%)
		Sandstone	0	0	0 (0.0%)
73	Sedimentary	Limestone (non-karstic)	1	0	1 (9.1%)
Consolidated	Sedimentary	Limestone (karstic)	0	3	3 (27.3%)
uso		Other	0	1	1 (9.1%)
Co		TOTAL	2	9	11 (68.8%)
		Granitic	0	1	1 (100%)
	Igneous	Basaltic	0	0	0 (0.0%)
		Intermediate	0	0	0 (0.0%)
		Other	0	0	0 (0.0%)
		TOTAL	0	1	1 (6.3%)
	Unknown	TOTAL	1	1	2 (12.5%)
	TOTAL		3	13	16 (11.1%)
	Gravel		0	8	8 (6.3%)
q	Sand		12	45	57 (44.5%)
date	Till Silt Clay Unknown TOTAL		1	7	8 (6.3%)
solia			7	22	29 (22.7%)
con			4	18	22 (17.2%)
Un			0	4	4 (3.1%)
			24	104	128 (88.9%)

Table 4.2: Summary of the breakdown of the predominant geologic material in the source area.

Method	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
Site history	1	62	63 (51.6%)
Known spill(s)	0	32	32 (26.2%)
Inferred from dissolved contaminant concentrations	1	68	69 (56.6%)
Direct observation of DNAPL in soil sample	0	37	37 (30.3%)
Extracted from monitoring wells	1	30	31 (25.4%)
Membrane Interface Probe	0	10	10 (8.2%)
Ribbon NAPL Samplers/FLuTEs	0	2	2 (1.6%)
Partitioning tracers	0	5	5 (4.1%)
Presence not estimated	0	2	2 (1.6%)
Total Number of Sites	5	117	122

 Table 4.3:
 Number of Methods Used to determine DNAPL Presence

Treatment	Total Cast (USD)	Size	of DNAPL Z	Zone ^a	TOTAL
Scale	Total Cost (USD)	Small	Large	Unknown	SITES
	<\$0.5M	3	2	2	7
	\$0.5M to \$1M	0	1	0	1
ale	\$1M to \$2M	0	2	0	2
Full-Scale	\$2M to \$4M	0	2	0	2
Ful	\$4M to \$10M	0	2	1	3
	>\$10M	0	1	0	1
	Total	3	10	3	16
	<\$0.5M	10	3	3	16
	\$0.5M to \$1M	1	8	2	11
est	\$1M to \$2M	1	2	1	4
Pilot Test	\$2M to \$4M	0	0	0	0
Pil	\$4M to \$10M	0	0	0	0
	>\$10M	0	0	0	0
	Total	12	13	6	31

 Table 4.4:
 Summary of Remedial Cost Data

^aDNAPL zone size classifications were: small = area < 10,000 ft² and volume < 100,000 ft³ large = area > 10,000 ft² and volume > 100,000 ft³.

GeoSytnec Consultants

TABLE 4.5 SUCCESS EVALUATION WITH ONLY PERCEIVED SUCCESS Navy Review of Case Studies

				DNA	DNAPL Distribution	tion	Γ		
				Diffused					Post
			Dead-end	Dead-end into Low					Treatment
Perceived Success	Technology	Lithology	Lithology fractures K Layers In pools Residual	K Layers	In pools	Residual	Sorbed	Stage	Monitoring
Success	Dual Phase-Water/air extraction	Till		Х		×	Х	Pilot test completed	
Success	Bioremediation	Sand						Pilot test completed	
Success	Bioremediation	Clay		×		×	×	Pilot test completed	
Success	Bioremediation	Sand		×		×	×	Pilot test completed	On going
Success	Chemical Oxidation-Fenton's Reagent	Sand						Full-scale completed	On going
Success	ZVI -Zero-valent iron	Sand				×	×	Full-scale completed	On going
Success	Thermal-Six phase heating	Silt			×	×		Pilot test completed	On going
Success	Bioremediation	Sand				×		Pilot test completed	On going
Fair Success	Dual Phase-Water/air extraction	Gravel						Pilot test completed	
Fair Success	Chemical Oxidation-Permanganate	Sand						Pilot test completed	
Fair Success	Thermal-Resistive heating	Clay					Х	Pilot test completed	

GeoSyntec Consultants

TABLE 4.6 SUCCESS EVALUATION IN TERMS OF SOURCE MASS REMOVAL Navy Review of Case Studies

							DNAPL Distribution	oution				
Source mass		Mass Flux		Perceived		Dead-end	Diffused Into					Post Treatment
removal	Technology	Decrease	Rebound	Success	Lithology	fractures	Low K Layers In pools Residual	In pools	Residual	Sorbed	Stage	Monitoring
	Chemical Oxidation-Permanganate	81-100%	No	Success	Sand				×		Pilot test completed	
/000	Thermal-Six phase heating	81-100%	Unknown	Not yet evaluated Sand	Sand						Full-scale completed	On going
% 00 I	Thermal- Conductive Heating	Unknown	No	Success	Clay				×		Full-scale completed	
	Excavation	Unknown	Unknown	Success	Sand				×		Pilot test completed	
	Surfactant Flushing	81-100%	Unknown	Fair Success	Clay				×		Pilot test completed	
	Chemical Oxidation-Permanganate	81-100%	Yes	Success	Sand				×	×	Pilot test completed	
	Bioremediation	61-80%	No	Success	Sand						Pilot test completed	On-going
%06<	Excavation	Unknown	No	Success	Silt		×	×	×	×	Full-scale completed	On-going
	Chemical Oxidation-Fenton's Reagent	Unknown	No	Success	Clay		×		×	×	Full-scale completed	On-going
	Thermal-Steam	Unknown	No	Success	Clay		×				Full-scale completed	On-going
	Chemical Oxidation-Fenton's Reagent	Unknown	No	Success	Sand						Full-scale completed	On-going
	Excavation	61-80%	Yes	Success	Sand		×		×	Х	Full-scale completed	On-going
>80 <90%	Chemical Oxidation-Fenton's Reagent	Unknown	No	Fair Success	Silt				×		Pilot test completed	
	Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	Success	Silt						Full-scale completed	
	Thermal-Steam	81-100%	Unknown	Fair Success	Sand				×		Pilot test completed	
>50 <80%	Chemical Oxidation-Permanganate	Unknown	Yes	Fair Success	Sand			×	×	×	Pilot test completed	
	Chemical Oxidation-Permanganate	61-80%	Yes	Fair Success	Sand			×	×	×	Pilot test completed	
>25<50	ZVI - EZVI/nano-scale iron	41-60%	No	Fair Success	Sand			Х	×		Pilot test completed	
>10<25	Thermal - Low temp-six phase heating	Unknown	Yes	Success	Silt			×	×	×	Pilot test completed	
<10	Dual Phase-DNAPL/air extraction	Unknown	Unknown	Poor Success	Silt		×	×	×	Х	Pilot test completed	On going

TABLE 4.7 SUCCESSEVALUATED IN TERMS OF MASS FLUX DECREASE Navy Review of Case Studies

Post Treatment On going Monitoring On going On going On going On going Pilot test completed Pilot test completed Full-scale completed Pilot test completed Pilot test completed Pilot test completed Full-scale completed Full-scale completed Full-scale completed Pilot test completed Stage Sorbed × ×× × Residual × × × × \times × × × **DNAPL** Distribution In pools × × × Diffused Into Llayers Low K × × Dead-end Fractures × Fair SuccessSandNot yet evaluatedMetamorphicFair SuccessLimestone Lithology Unknown Unknown Unknown Sand Sand Gravel Clay Sand Sand Sand Sand Clay Sand Clay ₽ Clay Silt Not yet evaluated Success Success Not yet evaluated Success Not yet evaluated Fair Success Perceived Success Success Success Success Success Success No Unknown Unknown Unknown No Unknown Unknown Rebound Unknown Unknown Unknown Unknown Unknown Unknown Unknown Unknown Yes Yes No ٩ å Unknown >90% >25 <50% Unknown Unknown >80 <90% >25 <50% Unknown Unknown Unknown >50 <80% Unknown Unknown Unknown Unknown Unknown Unknown Removal %06< Source Mass 100% 100% Thermal-Six phase heating Chemical Oxidation-Fenton's Reagent Chemical Oxidation-Fenton's Reagent ZVI - EZVI/nano-scale iron Chemical Oxidation-Permanganate Excavation Chemical Oxidation-Permanganate Bioremediation Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Thermal-Six phase heating Bioremediation Excavation Thermal-Six phase heating Thermal-Six phase heating Thermal-Six phase heating Technology Surfactant Flushing Surfactant Flushing **Themal-Steam** Bioremediation Bioremediation Decrease 81-100% 41-60% 61-80% Mass Flux %0

Post –Treatment Monitoring Stage	Number of Sites
Completed	25 (47.2%)
Ongoing	13 (24.5%)
Not Conducted	15 (28.3%)
Total	53

 Table 4.8:
 Summary of Post – Treatment Monitoring Stage

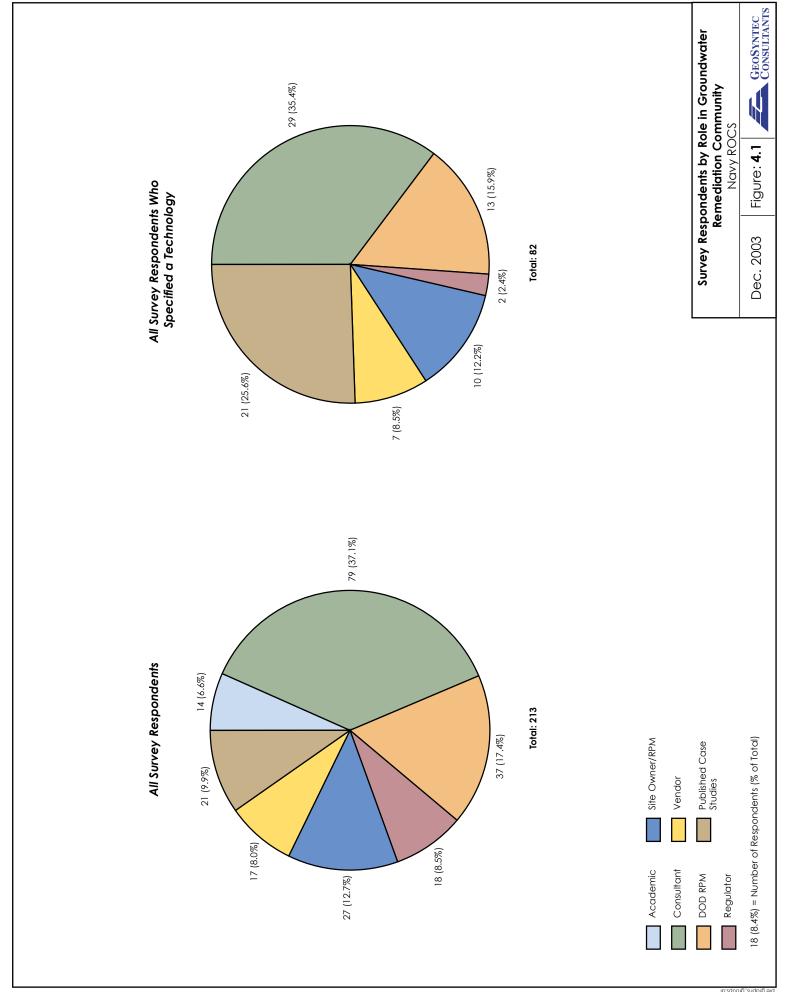
GeoSytnec Consultants

TABLE 4.9 SUCCESS EVALUATED IN TERMS OF REBOUND Navy Review of Case Studies

Source Mass folgyMass Flux RemovalPerceived SuccessLithology fracturesDead-end Low KLayersDiffused into ImpoolsResidual ResidualSocrease SuccessSuccess FacturesLithology madramDead-end fracturesDiffused into moolsResidualSocread SuccessSuccess SandSand SuccessDead-end successDiffused into successResidualSocread SuccessSuccess SandSand SuccessDead-end SuccessDiffused into SuccessResidualSocread SuccessSocread SandSocread SuccessSocread Sand<								DNAPL	DNAPL Distribution	L			
Forceived Technology Source Mass Fractures Source Mass Removal Dead-end Decrease Diffused into Evential Diffused into Fractures Diffused into Fractures <t< th=""><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th></th><th>Post</th></t<>													Post
TechnologyTechnologyRemovalDecreaseSuccessLithologyIthologyIthologyResidualSorbedChemical Oxidation-Permanganate>90%81-100%SuccessSandXXXXXChemical Oxidation-Permanganate>00%61-80%SuccessSandXXXXXXExcavation>80 r50%61-80%SuccessSandXXX<			Source Mass	Mass Flux	Perceived		Dead-end	Diffused into					Treatment
Chemical Oxidation-Permanganate>90%81-100%SuccessSandNNNNExcavationNuknownUhknownNuccessSandXXXXXExcavation>80 < 90%UhknownNuknownFair SuccessSandXXXXXEnemical Oxidation-Permanganate>56 < 60%UhknownFair SuccessSandXXXXXXThemial Oxidation-Permanganate>25 < 50%61-80%Fair SuccessSandXXXXXXXThemial -Low tempsix phase heating>10 < 25%UhknownBir SuccessSandXXXXXXXXChemical Oxidation-Permanganate100%UhknownBir SuccessSandXXXXXXXXXChemical Oxidation-Permanganate100%UhknownSuccessSandXXX<	Rebound		Removal	Decrease	Success	Lithology		Low K Layers	In pools	Residual	Sorbed	Stage	Monitoring
Chemical Oxidation-PermanganateUnknownUnknownSuccessShaleXXXXXExcavation>80 < 90%		Chemical Oxidation-Permanganate	%06<	81-100%	Success	Sand				×	Х	Pilot test completed	
Excavation>80 < 50%61-80%SuccessSandXXXXChemical Oxidation-Permanganate>50 < 60%		Chemical Oxidation-Permanganate	Unknown	Unknown	Success	Shale	×	×	×	×	×	Pilot test completed	
Chemical Oxidation-Permanganate>50 <80%UnknownFair SuccessSandXXXXChemical Oxidation-Permanganate>25 <50%		Excavation	>80 <90%	61-80%	Success	Sand		×		×	×	Full-scale completed	On going
Chemical Oxidation-Permanganate>25 < 50%61-80%Fair SuccessSandXXXXThermal - Low temp-six phase heating>10 < 25%	Yes	Chemical Oxidation-Permanganate	>50 <80%	Unknown	Fair Success	Sand			×	×	×	Pilot test completed	
Themal - Low temp-six phase heating>10 <25%UnknownSuccessSiltNXXXChemical Oxidation-PermanganateUnknownUnknownFair SuccessGravelNNNNNChemical Oxidation-PermanganateUnknownB1-100%B1-100%SuccessGravelNNNNOther - Conductive Heating100%UnknownSuccessGravelNNNNNOther - Conductive Heating100%UnknownSuccessGravelNNNNNDisremediation>90%61-80%SuccessSiltXXXXXExcavation>90%UnknownSuccessSiltXXXXXChemical Oxidation-Fentori's Reagent>90%UnknownSuccessSiltXXXXXChemical Oxidation-Fentori's Reagent>80%UnknownSuccessSiltXXXXXXChemical Oxidation-Fentori's Reagent>80%UnknownSuccessSiltXXXXXXZVI - EZVI/Inano-scale iron225 <50%		Chemical Oxidation-Permanganate	>25 <50%	61-80%	Fair Success	Sand			×	×	×	Pilot test completed	
Chemical Oxidation-PermanganateUnknownUnknownFair SuccessGravelImageImageImageChemical Oxidation-Permanganate100%81-100%SuccessSandImageImageImageChemical Oxidation-Permanganate100%UnknownSuccessSandImageImageImageOther - Conductive Heating100%UnknownSuccessClayImageImageImageBioremediation>90%61-80%SuccessSandImageImageImageChemical Oxidation-Fentor's Reagent>90%UnknownSuccessSandImageImageImageChemical Oxidation-Fentor's Reagent>90%UnknownSuccessSandImageImageImageImageChemical Oxidation-Fentor's Reagent>80%UnknownSuccessSandImageImageImageImageChemical Oxidation-Fentor's Reagent>80%UnknownSuccessSandImageImageImageImageZVI - EZVI/Inano-scale iron>25<650%		Thermal - Low temp-six phase heating	>10 <25%	Unknown	Success	Silt			×	×	×	Pilot test completed	
		Chemical Oxidation-Permanganate	Unknown	Unknown	Fair Success	Gravel						Pilot test completed	
Other - Conductive Heating100%UnknownSuccessClayMXXBioremediation>90%61-80%SuccessSandMXXXExcavation>90%UnknownSuccessSiltXXXXExcavation>90%UnknownSuccessSiltXXXXChemical Oxidation-Fenton's Reagent>90%UnknownSuccessClayXXXXChemical Oxidation-Fenton's Reagent>90%UnknownSuccessClayXXXXChemical Oxidation-Fenton's Reagent>80<<90%		Chemical Oxidation-Permanganate	100%	81-100%	Success	Sand				×		Pilot test completed	
Bioremediation>90%61-80%SuccessSand </td <td></td> <td>Other - Conductive Heating</td> <td>100%</td> <td>Unknown</td> <td>Success</td> <td>Clay</td> <td></td> <td></td> <td></td> <td>×</td> <td></td> <td>Full-scale completed</td> <td></td>		Other - Conductive Heating	100%	Unknown	Success	Clay				×		Full-scale completed	
Excavation>90%UnknownSuccessSiltXXXXChemical Oxidation-Fenton's Reagent>90%UnknownSuccessClayXXXXThermal-Steam>90%UnknownSuccessClayXXXXXChemical Oxidation-Fenton's Reagent>90%UnknownSuccessClayXXXXChemical Oxidation-Fenton's Reagent>90%UnknownSuccessSandXXXXChemical Oxidation-Fenton's Reagent>80<690%		Bioremediation	%06<	61-80%	Success	Sand						Pilot test completed	On going
Chemical Oxidation-Fenton's Reagent >90% Unknown Success Clay X X X X Thermal-Steam >90% Unknown Success Clay X X X X Chemical Oxidation-Fenton's Reagent >90% Unknown Success Clay X X X X Chemical Oxidation-Fenton's Reagent >90% Unknown Success Sand X X X X ZVI-Scale iron >80<<90%		Excavation	%06<	Unknown	Success	Silt		×	×	×	×	Full-scale completed	On going
Thermal-Steam >90% Unknown Success Clay X M M Chemical Oxidation-Fenton's Reagent >90% Unknown Success Sand X X X X X Chemical Oxidation-Fenton's Reagent >90% Unknown Success Sand X X X X ZVI - EZVI/nano-scale iron >25 <50%		Chemical Oxidation-Fenton's Reagent	%06<	Unknown	Success	Clay		×		×	×	Full-scale completed	On going
Chemical Oxidation-Fenton's Reagent >90% Unknown Success Sand Image: Constraint of the state o		Thermal-Steam	%06<	Unknown	Success	Clay		×				Full-scale completed	On going
Chemical Oxidation-Fenton's Reagent >80 <90% Unknown Fair Success Silt X X ZVI - EZVI/nano-scale iron >25 <50%	QN	Chemical Oxidation-Fenton's Reagent	%06<	Unknown	Success	Sand						Full-scale completed	On going
D-scale iron >25 <50% 41-60% Fair Success Sand X X ase heating Unknown 81-100% Success Clay Y Y ase heating Unknown 61-80% Success Clay Y Y Y Unknown 61-80% Success Clay Y Y Y Y Unknown Unknown Success Sand Y Y Y Y Unknown Unknown Success Sand Y Y Y Y	2	Chemical Oxidation-Fenton's Reagent	>80 <90%	Unknown	Fair Success	Silt				×		Pilot test completed	
ase heating Unknown 81-100% Success Clay (10 10 10 10 10 10 10 10 10 10 10 10 10 1		ZVI - EZVI/nano-scale iron	>25 <50%	41-60%	Fair Success	Sand			×	×		Pilot test completed	
Unknown 61-80% Success Clay Clay Unknown Unknown Success Sand X X X Unknown Unknown Success Sand X X X X Unknown Unknown Success Gravel Y Y X X		Thermal-Six phase heating	Unknown	81-100%	Success	Clay						Pilot test completed	
Unknown Unknown Success Sand X X X X X X X X X X X X X X X X X X X		Bioremediation	Unknown	61-80%	Success	Clay						Full-scale completed	On going
Unknown Unknown Success Gravel		Bioremediation	Unknown	Unknown	Success	Sand		×	×	×	×	Pilot test completed	On going
		Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	Success	Gravel						Pilot test completed	
		ZVI -Zero-valent iron	Unknown	Unknown	Success	Sand				×	×	Pilot test completed	

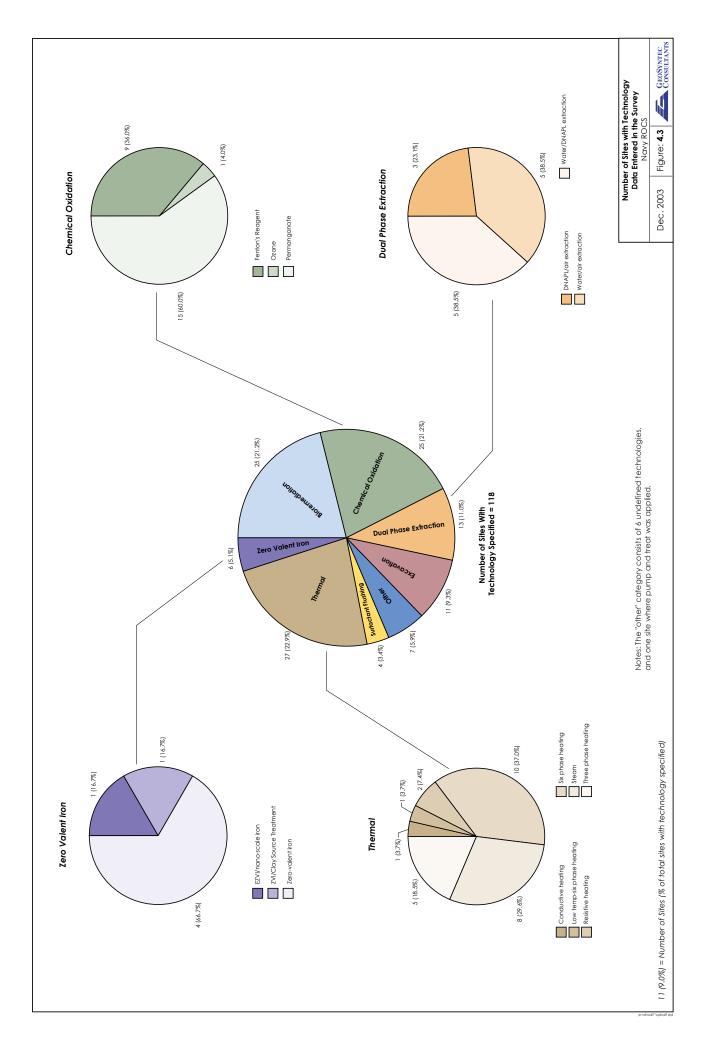
Secondary Groundwater Quality Indicator	Number of Sites With The Indicator Selected
Methane generation	8 (22.8%)
Hydrogen sulfide generation	1 (2.9%)
Dissolved iron generation	7 (20.0%)
Dissolved manganese generation	7 (20.0%)
Elevated concentration of other metals	8 (22.9%)
Increased Biochemical Oxygen Demand	4 (11.4%)
Change in pH	7 (20.0%)
Reduced soil porosity	6 (17.1%)
Change in hydraulic conductivity	5 (14.3%)
Change in groundwater color	8 (22.9%)
Increased dissolved solids	3 (8.6%)
Redistribution of DNAPL	4 (11.4%)
Increase in source area volume	0 (0.0%)
Decrease in source area volume	12 (34.3%)
Other	5 (14.3%)
TOTAL SITES WITH SECONDARY GROUNDWATER IMPACTS	35

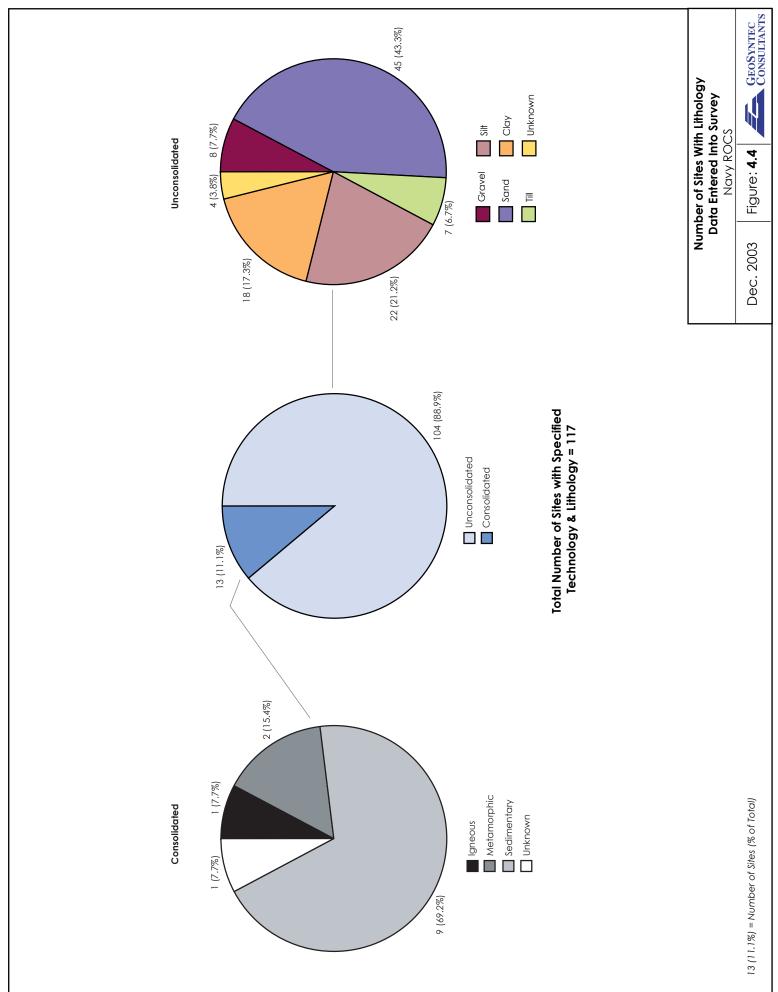
Table 4.10: Summary of impacts to secondary groundwater quality data.

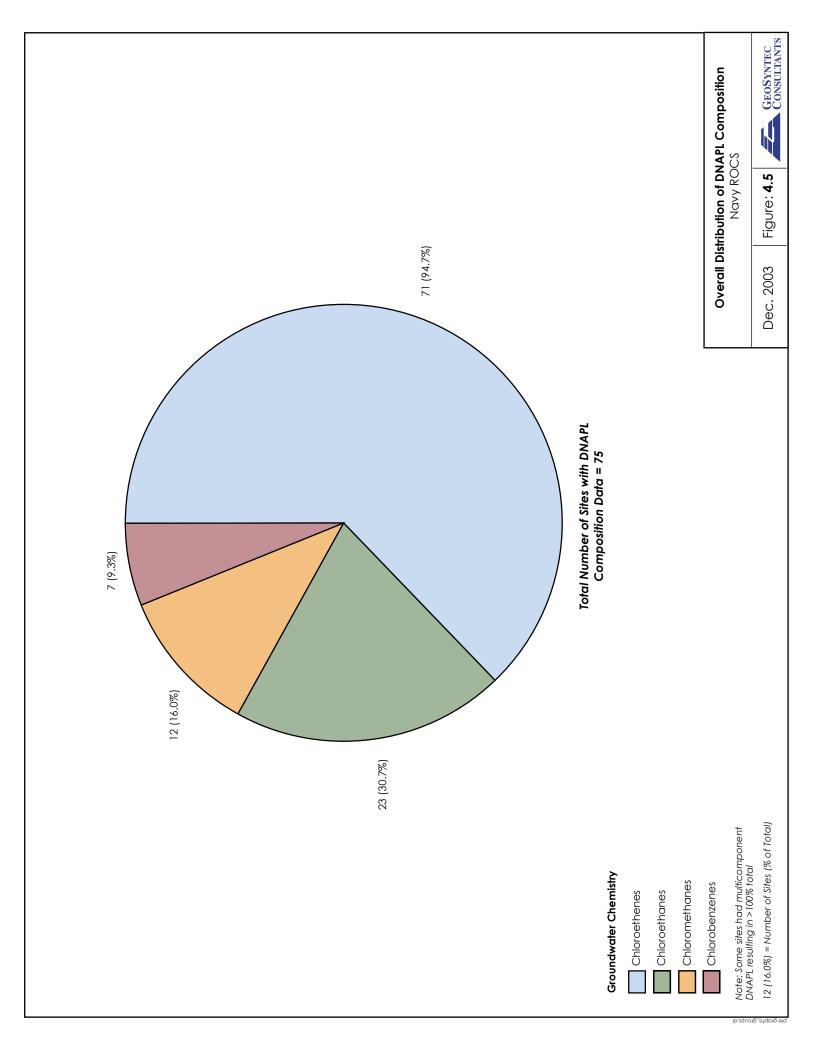


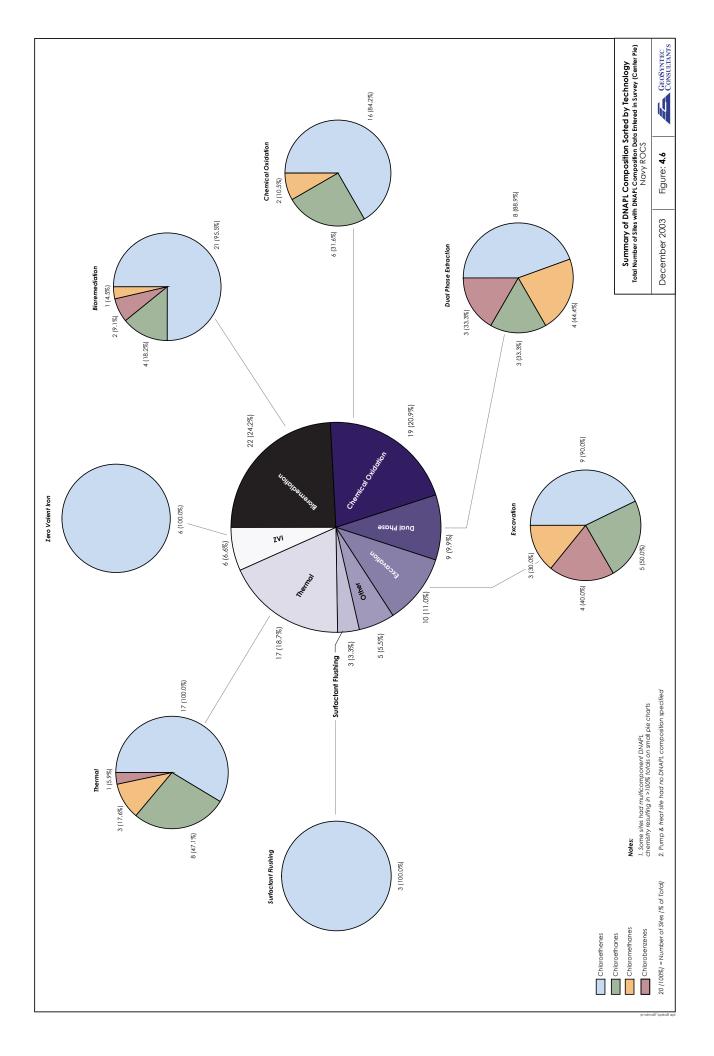
ibhs_groups.ai

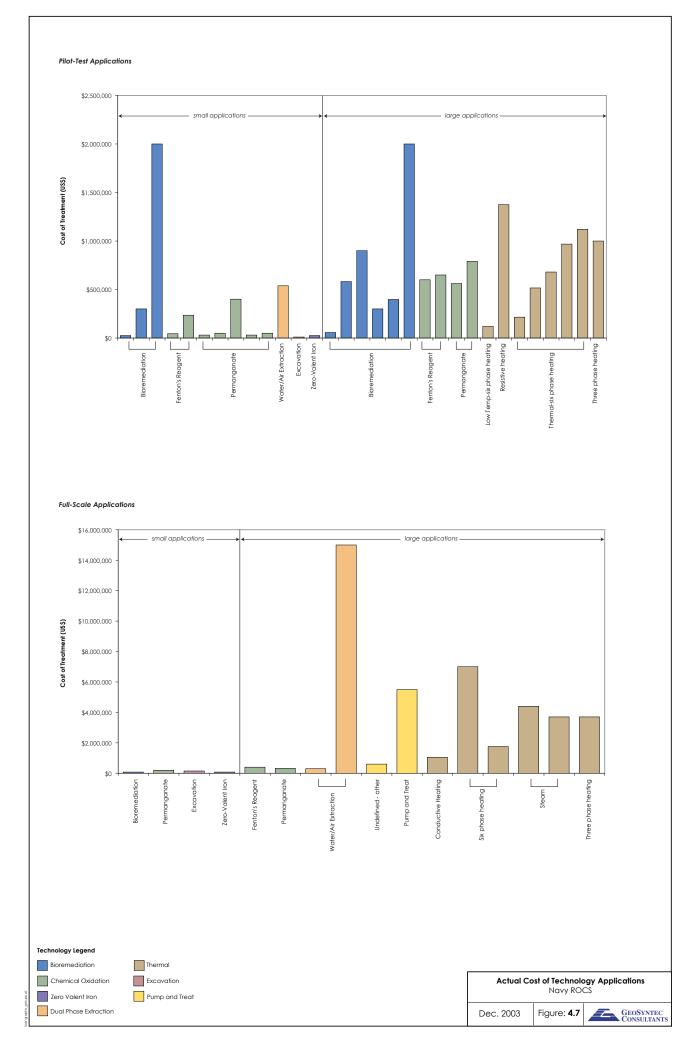


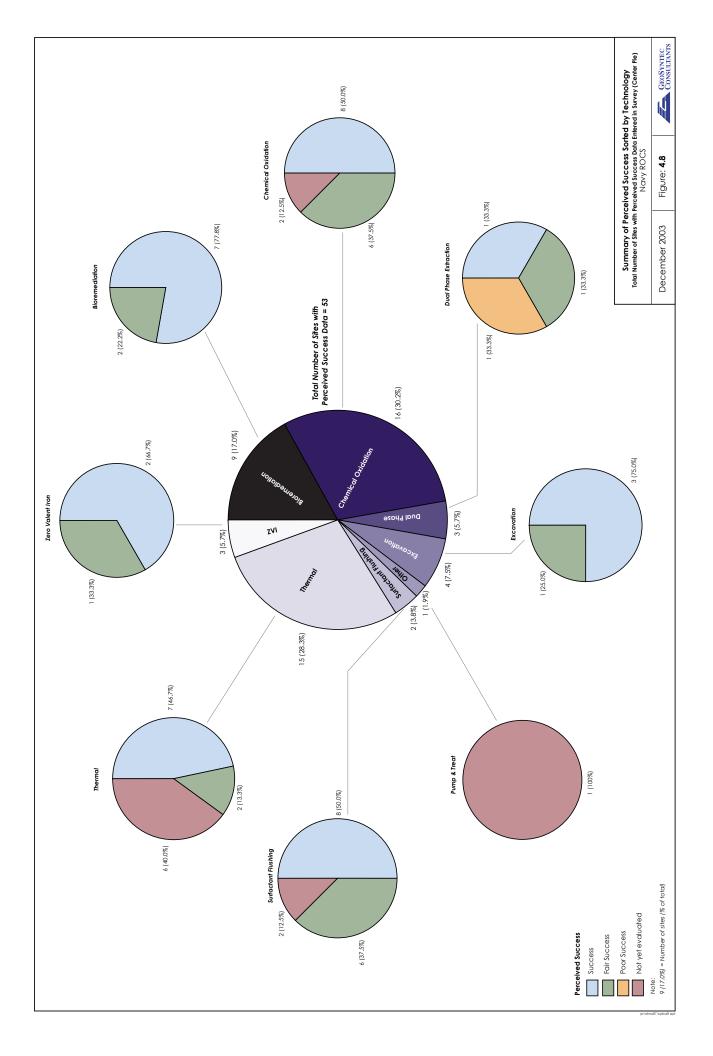


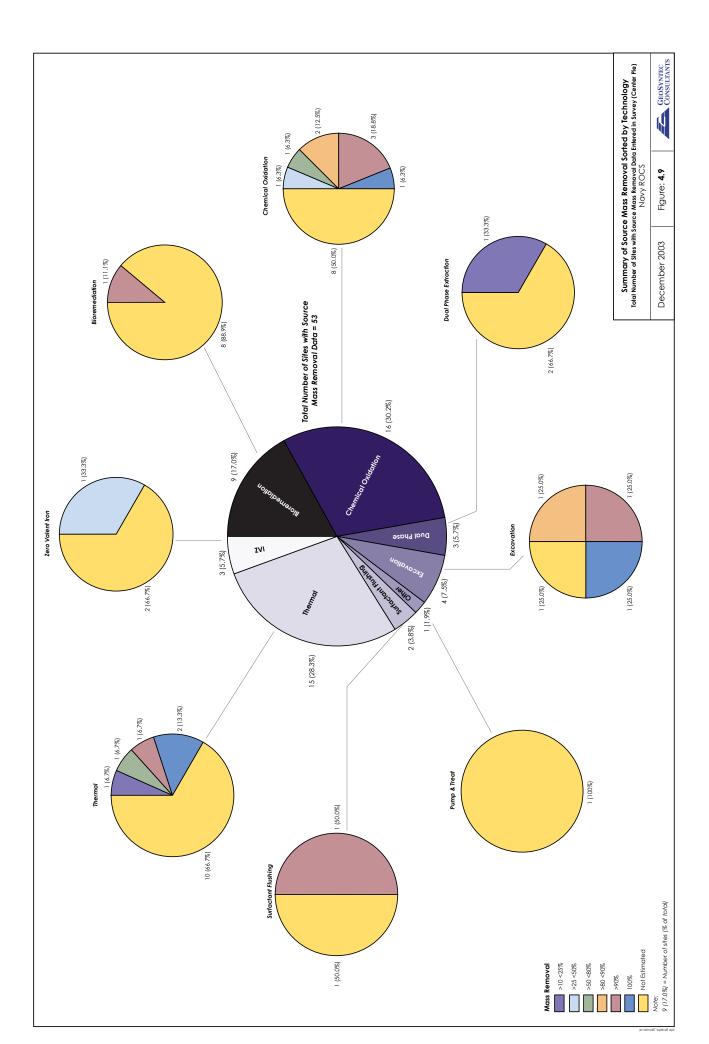


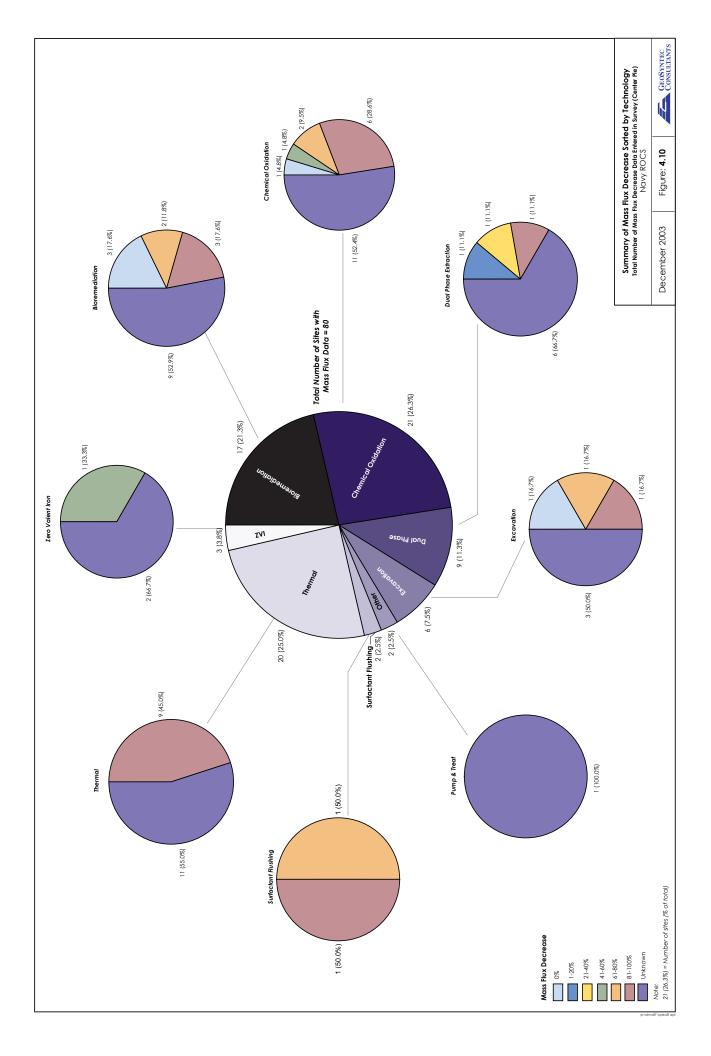


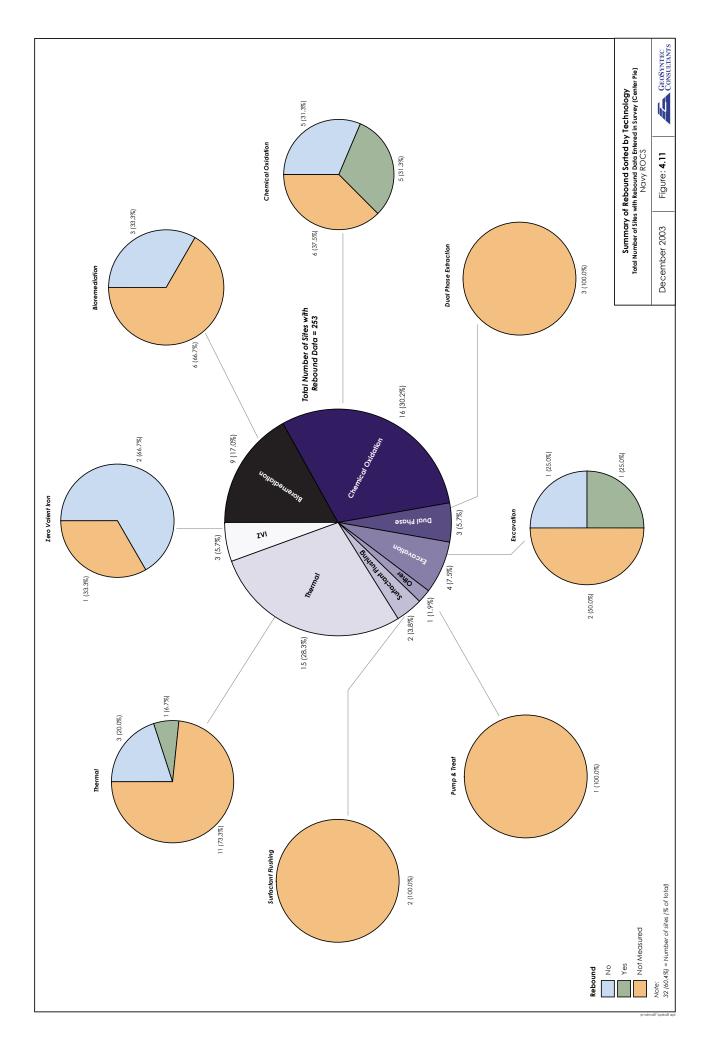


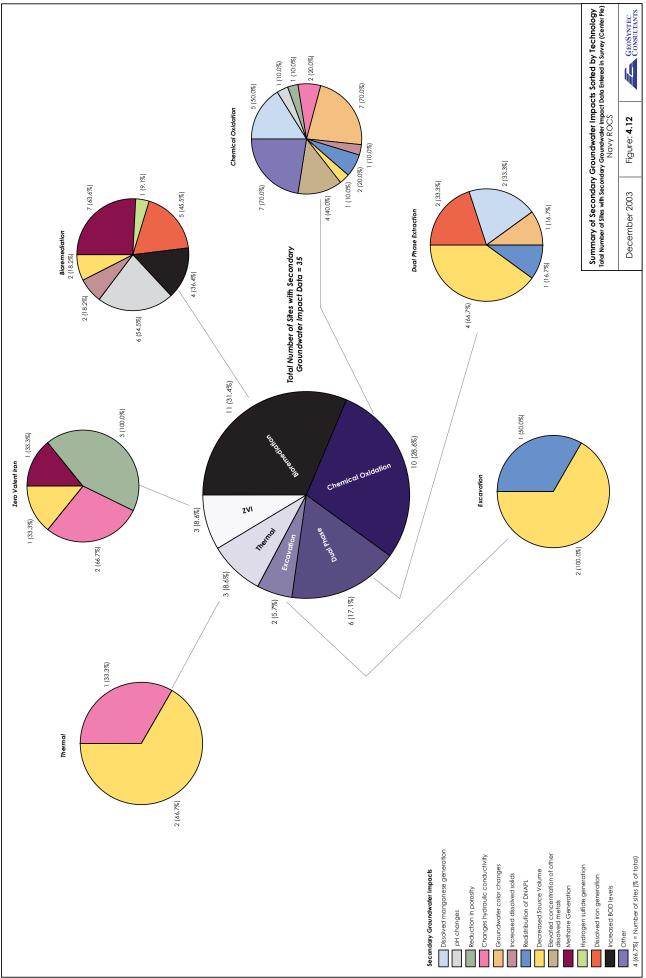












APPENDIX A

Case Studies

APPENDIX A: Case Studies

Appendix A contains a limited number of case studies per technology discussed in Section 3. These case studies are included to highlight some of the challenges and advantages associated with each technology, as well as to include examples of how each technology performed for various site characteristics. The information presented in each case study includes a summary of relevant site characteristics, technology application approaches, remedial goals, and lessons learned from each site.

1 Surfactant/Co-Solvent Flushing

One case study each of surfactant and co-solvent flushing are summarized below. The first case study is an outline of the application of a SEAR (surfactant enhanced aquifer remediation) at the Bachman Road Site in Oscada, MI, as described in the U.S EPA report *The DNAPL Remediation Challenge: Is there a Case for Source Depletion?* (In Press) The second case study is an outline of the application of co-solvent (alcohol) flushing at the Sages dry cleaner Site in Jacksonville, FL, as described by Jawiz *et al.* (2000). Table A.1 summarizes the site conditions and remedial activities at the site. Sections 1.1 and 1.2 below include summaries of the treatment operation and lessons learned from each application.

Parameter	Bachman Road Site	Sages Dry Cleaner Site
Technology description	SEAR for solubilization of source zone	Alcohol flushing of source zone
Soil type	Fine to medium grained glacial outwash sands	Fine to very fine sand to 9 m bgs; very fine to silty sand to 10.7 m bgs; thin clay layer at 10.7 m bgs; very fine to silty sand below
Hydraulic conductivity	15-150 ft per day	6 m/day in fine to very fine sand 3 m/day in very fine to silty sand
Hydraulic gradient	not provided	not provided
Depth to groundwater	11 ft bgs	not provided; source zone saturated
Well yield	not provided	Total system flow 15.1 L/min (3 injection wells; 6 recovery wells)
Contaminants of concern	PCE	PCE

Table A.1: Summary of Surfactant and Co-solvent Case Studies

TR0132

Parameter	Bachman Road Site	Sages Dry Cleaner Site
NAPL evidence	PCE concentration in groundwater to 88ppm;	Free-phase in one supply well; groundwater concentration from 70 to
	Free phase in one well and two soil sub- samples from coring	150 mg/L; 196 pre-treatment soil samples with average concentration of 2.8 mg/g
Volume treated	7,500 ft ³	2150 ft ³
Area treated	20 x 25 ft	9 x 24 ft
Depth treated	11-26 ft bgs	7.6 to 9.9 m bgs
DNAPL mass targeted	not provided	Estimated from pre-treatment soil samples as 42 L (67 kg)
Remediation Infrastructure	One extraction well, three injection wells, with three surfactant injection wells between the extraction and injection wells	Three central injection wells; 6 perimeter extraction wells; 7 multilevels (5 each); 2 pore volumes of alcohol added; MPPE above ground treatment system
Remediation Duration	November 1996 to October 1998	8 days active flushing
	Design and installation \$300,000	
Remedial Costs	O&M, reporting, analytical \$450,000	not provided
	Carbon disposal \$100,000	
Performance results: 1) met remedial goals	✓	✓ (removal of substantial fraction of DNAPL)
2) exceeded remedial goals		
3) met MCLs in at least some areas		
4) treatment efficiencies: TCE, etc.	Aqueous VOC concentrations decreased from 4,000 μ g/L to 650 μ g/L; 1164 lbs TCE mass removed	~60% by soil core and partitioning tracer test
5) DNAPL mass remaining	not quantified	~40%

TR0132

1.1 Case Study 1: Bachman Road Site, Oscada MI

General operation of the application of SEAR proceeded as follows:

- *i)* Groundwater extraction well operation was initiated at 19.7 L/min followed shortly by startup of three groundwater injection wells (1.9 L/min each). Three surfactant injection wells, placed between the injection and extraction wells were not initially used.
- *ii)* Surfactant injection was preceded by a short-term partitioning tracer test to estimate PCE mass.
- *iii)* Injection of a 6% (by volume) Tween 80 solution into all three surfactant injection wells over 5 days.
- *iv)* Operation of one surfactant injection well was then discontinued, and surfactant addition continued into the other two injection wells for an additional 5 days.
- *v)* In total, 1.5 pore volumes of surfactant solution were added.
- *vi)* Water injection was continued for an additional 2 days after surfactant addition.
- *vii)* The extraction well was operated for an additional month to ensure surfactant and solubilized DNAPL capture.

Hydraulic capture of the solubilized DNAPL was demonstrated with >95% capture of the injected surfactant. Multi-level sampler measurements confirmed the surfactant sweep. A two order of magnitude reduction in groundwater concentrations from multi-level samplers was observed post-demonstration. Post-demonstration degradation products have been observed, suggesting that the surfactant flood stimulated microbial activity. Based upon the results of the pilot test, a full scale application of the technology was being designed at the time of the report.

The predominant lesson learned during the technology application was the importance of complete source zone characterization, since the presence of an untreated upgradient source zone resulted in relatively high PCE concentrations at the extraction well post-demonstration.

TR0132

1.2 Case Study 2: Sages Dry Cleaner Site, Jacksonville, Fl.

General operation of the application of the alcohol flush proceeded as follows:

- *i)* Pre- and post-demonstration coring and partitioning tracer tests were conducted to evaluate the presence and distribution of PCE in the source zone.
- *ii)* Three injection wells and six extraction wells were used to create hydraulic capture of the source area, with a 2:1 extraction to injection ratio. Injection wells were screened 0.3 m deeper to enhance extraction of the DNAPL. Injection wells were fitted with packers over this period to focus alcohol injection into areas of highest DNAPL saturation, with water injected above the packer.
- *iii)* The ethanol concentration was gradually increased to 95% over the first 10 hours of injection to minimize fluid density differences between injected and resident fluids.
- *iv)* Packers were set at 0.3 m from the bottom of the well for the first 6 hours to flush alcohol immediately below the deepest DNAPL to create a barrier to downward mobilization.
- *v)* Packers were then raised at a rate of approximately 15 cm/hr to maximum heights of 1.7 m at two injection wells and 0.9 m at the third injection well.
- *vi)* After 70 hours of flushing, the packers were lowered back down to 0.3 m at the same rate.
- *vii)* A total of 34 kL (2 pore volumes) of 95% ethanol was delivered into the three injection wells during a period of about 3 days.
- *viii)* Water flooding was initiated at 3.5 days and continued for 4.5 days (4.48 pore volumes).
- *ix)* The total volume flushed was 76 kL of water and ethanol, combined.
- *x)* Monitoring of the demonstration area for biological enhancement was continuing at the time the report was issued.

The remedial goals of the demonstration were as follows:

TR0132

A.4

- *i)* Removal of a significant amount of PCE mass from the source area;
- *ii)* Maximize the efficiency of the DNAPL extraction;
- *iii)* Minimize the potential for DNAPL mobilization; and
- *iv)* Minimize waste disposal costs.

These goals were generally met by the demonstration, although waste costs were increased by the volume of groundwater extracted to ensure hydraulic capture.

Lessons learned during the technology application included the following:

- *i)* High extraction to injection fluid ratio dilutes the extracted fluid, requiring additional treatment capacity and cost.
- *ii)* Results of the pre-demonstration monitoring indicated that the architecture of the DNAPL was discretely distributed, which made it likely that a flushing technology would be inefficient, emphasizing the importance of source zone characterization prior to technology selection.

2 **Bioremediation**

Two case studies of field applications of enhanced *in situ* bioremediation (EISB) are summarized below. The first case study is the application of EISB to treat a DNAPL source area within overburden and bedrock at the Caldwell Trucking Superfund Site in New Jersey as described by Finn *et al*,(2003). The second case study was conducted at the NASA LC 34 complex in Florida (Battelle, in press). Several applied field demonstrations have recently been completed or are currently being conducted to assess EISB of DNAPL source areas in both porous and fractured bedrock media. Table A.2 summarizes the site conditions and remedial activities at these sites.

TR0132

Parameter	Caldwell Trucking	NASA LC34
Technology description	EISB + bioaugmentation with KB-1 [™] . Donors used: methanol, ethanol lactate	EISB + bioaugmentation with KB-1 [™] . Donor used: ethanol
	Design: batch addition	Design: recirculation
Soil type	Overburden: basal sand and gravel	Sand
	Bedrock: fractured basalt	
Hydraulic conductivity	Overburden: geomean 1x10-3 cm/sec	5.3x10-3 cm/sec
	Bedrock: geomean 2.4x10-5 cm/sec	
II. In the section	Overbuden: 0.011 ft/ft	<0.0001
Hydraulic gradient	Bedrock: 0.017 to 0.56 ft/ft	<0.0001
Depth to groundwater	15 to 20 feet bgs	4 ft
Contaminants of concern	PCE, TCE, 1,1,1-TCA	TCE, cis-DCE, VC
NAPL evidence	60% of aqueous saturation for TCE	Direct measurement of soil concentrations indicative of DNAPL; groundwater concentration at solubility in some locations; known history of TCE DNAPL release into the aquifer
Volume treated	Ca. 16,000 cubic yards	22 ft x 20 ft x 10 ft =4,400 cu. ft (163 cu yd)
Area treated	120x60 ft ²	440 ft ²
Depth treated	120 feet bgs	16 to 26 ft bgs
DNAPL mass targeted	Yes	Yes
Remediation infrastructure	7 injection wells (multi-level) 7 monitoring wells	3 injection and 3 extraction wells, 30 monitoring points
Remediation duration	Pre-full scale evaluation 30 months- currently on-going	Pilot August 2002 to October 2003 (completed)

 Table A.2 Summary of Bioremediation Case Studies

TR0132

Parameter	Caldwell Trucking	NASA LC34
Performance results	PCE and TCE: 99.8% reduction from 27,000 ug/L to 50 ug/L; and 680,000 ug/L to 1,400 ug/L, respectively. Corresponding increase in cDCE. VC, and ethene. cDCE and VC are declining, ethene increasing; Result is the expected ROD amendment replacing P&T with bioremediation to treat source area.	Independent determination that VOCs were non-detect in soil within the treatment zone; No evidence that total VOC concentrations as high as 291 mg/L inhibited biodegradation; Ethene concentrations in excess of the stoichiometric chloroethene concentrations measured during the Baseline phase were observed in some performance monitoring wells, suggesting that biodegradation resulted in a mass transfer enhancement; Limited methanogenesis was observed during electron donor addition; Molecular techniques indicated that methanogenic Archaebacteria were not present. Some methanogenesis was observed following bioaugmentation.

3 Chemical Oxidation

One case study each of field applications of permanganate and Fenton's reagent technology are summarized below. No case study was found where ozone was used to treat a chlorinated solvent DNAPL source zone below the water table.

The first case study is an outline of the pilot-scale application of Fenton's reagent at the Savannah River Site in Aiken, South Carolina (US DOE, 1999a). The second case study is a summary of a potassium permanganate pilot test conducted at the Portsmouth Gaseous Diffusion Plant in Piketon, Ohio (US DOE, 1999b). Table A.3 summarizes the site conditions and remedial activities at both of these sites. Sections 3.1 to 3.2 below include summaries of the treatment operation and lessons learned from each application.

Table A.3: Summary of Chemical Oxidation Case Studies

Parameter	Savannah River Site	Portsmouth Gaseous Diffusion Plant
Technology description	Fenton's Reagent	Potassium Permanganate
Soil type	permeable sands with low fines alternating with clayey sand and clay units	silt and clay layer (25 to 30 ft thick), overlying sand and gravel (2 to 10 ft thick),

Parameter	Savannah River Site	Portsmouth Gaseous Diffusion Plant
Hydraulic conductivity	not provided	20 ft/day for horizontal well tests and between 24 and 411 ft/day for vertical well tests
Depth to groundwater	130 ft bgs	not provided
Contaminants of concern	TCE and PCE - DNAPL composition 95% TCE and 5 % PCE	TCE
NAPL evidence	Observed DNAPL in bottom of wells; groundwater concentrations of 120 mg/l PCE and 21 mg/l TCE	Observed DNAPL in wells; 700 mg/l TCE in groundwater.
Volume treated	68,702 ft ³	119,000 ft ³
Area treated	50 X 50 ft ²	90 x 220 ft ²
Depth treated	124 to 152 ft bgs	~30 to 35 ft bgs
DNAPL mass targeted	600 lbs	273 lbs of TCE
Remediation infrastructure	 4 injection wells; 3 groundwater monitoring wells; 3 vadose zone monitors; proprietary injection process 	 2 horizontal recirculation wells – 1 injection one extraction well each 220 ft long; 22 boreholes for pre-demonstration concentrations; 14 monitoring wells above ground mixing system including solids feeder, mixing tank and jet pump into well
Remediation duration	6 day period of injection	approximately 30 days
Remedial costs	\$511,115 site preparation and operation activities, drilling, construction, operations, sampling, pre and post demonstration characterization, demobilization and reporting and project management.	\$562,000 for project management, pre- demonstration characterization, remediation operations and oxidant recirculation, resistivity monitoring, and post demonstration characterization and demobilization
Performance results:	94% destruction of total VOCs,	points where TCE was non-detect but

Parameter	Savannah River Site	Portsmouth Gaseous Diffusion Plant
	95% PCE and 88% TCE	distribution not uniform

3.1 Case Study 1: Fenton's reagent at the Savannah River Site in Aiken, South Carolina.

All italicized text in this section is taken directly from *Innovative Technology* Summary Report DOE/EM-0484 Fenton's Reagent. October 1999).

General operation of the Fenton's reagent application at Savannah River proceeded as follows:

- *i)* The site selected for the demonstration was an area of approximately 50 foot (ft) by 50 ft adjacent to a known source of DNAPL; a small DNAPL plume located below the water table was treated over a 6-day period.
- *ii)* The catalyst solution of 100 parts per million (ppm) ferrous sulfate, pHadjusted with concentrated sulfuric acid, was initially injected into the subsurface to ensure adequate migration into the formation, while the groundwater pH was adjusted to between 4 and 6.
- *iii)* Subsequent injection of the H_2O_2 and catalyst utilized a patented mixing and injection process. Injections were conducted in batch mode with one batch injected per day.
- *iv)* Following 6 days of injection, the site was characterized to determine treatment efficiency.

Average contaminant concentrations in the treatment area groundwater were 119.49 mg/l PCE and 21.31 mg/l TCE before treatment and were reduced to 0.65 mg/l PCE and 0.07 mg/l TCE at completion of treatment. The estimated pre-test mass of DNAPL in the treatment zone was 593 lbs; the estimated post-test mass of DNAPL was 36 lbs.

Average pH was 5.71 before treatment and 2.44 at completion of treatment. Change in pH was due to addition of acid to maintain optimal oxidation conditions and, to some extent, due to production of CO_2 from the oxidation process. After 17 months, pH has risen to 3.4 to 4.0.

TR0132

A.9

Average baseline groundwater temperature in the treatment zone was 19.2°C; this was raised to a maximum of 34.7°C by the oxidation process. Dissolved oxygen concentrations increased from an average of 9.3 mg/l before treatment to 24 mg/l after treatment. Average baseline chloride concentration was 3.61 mg/l; chloride reached a maximum of 24.33 mg/l at the completion of the treatment process. The increase in chloride concentration verified oxidation of PCE and TCE by the peroxide. Hydrogen peroxide concentrations in the monitoring wells ranged from approximately 2 to 5 ppm.

Monitoring of gases in the headspace of monitoring wells for CO_2 , PCE, and TCE during the injection process indicated:

- *i)* Gases were escaping from water in the monitoring wells during injection due to the violent oxidation process.
- *ii)* Carbon dioxide levels in these gases rose to over 3,500 ppmv (ambient CO₂ levels are approximately 300-400 ppmv). Elevated CO₂ levels verified DNAPL oxidation in the subsurface.
- *iii) PCE (from 0 to 190 ppmv) and TCE (0 to 80 ppmv) were evident in the vapor and can be attributed to sparging of water in the wells.*

Lessons learned during the technology application included the following:

Design Issues:

- *i)* The efficiency of the process increases at higher contaminant concentrations and decreases as target treatment levels become more stringent.
- *ii)* Higher H_2O_2 concentrations provide faster reaction times, significantly greater removal of DNAPL type contaminants, but less efficient H_2O_2 use.
- *iii)* Highly alkaline soils may require mineral acid addition to bring the pH into the optimal range.
- iv) Organic carbon content may impact treatment because the hydroxyl radical is relatively nonselective. However, no significant effect was observed with contaminant levels of 500-2000 ppm with total organic carbon of 0.1 to 1.3 (Watts et al. 1994. "On site treatment of contaminated soils using hydrogen peroxide." Project Report T9234-06, Washington State Transport Center, Washington State University).

TR0132

A.10

- v) For in situ groundwater treatment, the number and pattern of injectors and monitoring wells must be designed to ensure maximum coverage of the treatment zone. Because the cost is related to depth (cost per well was approximately \$70/ft) and amount of DNAPL, the number and spacing of the wells becomes critical. The heterogeneity of the subsurface at the site will also control the number and spacing of wells required.
- vi) Duration of operation is not a linear function of volume of DNAPL. Factors affecting the duration of the treatment include: permeability, heterogeneity, and geochemistry of the aquifer.

Implementation Considerations:

- *i)* When implementing in situ oxidation using Fenton's Reagent, general operation considerations include:
 - *pH of the system must be between 3 and 6.*
 - The rate of the reaction increases with increasing temperature (although the efficiency declines above 40 to 50°C
 - For most applications the valence of the iron salts used doesn't matter (+2 versus +3) nor does it matter whether a chloride or sulfate salt of the iron is used, although chlorine salts may generate high rates of chloride during application.
 - Due to oxidation of the subsurface, metals that are mobile under these conditions may be released at some sites. This should be considered during the technology selection process.
- *ii)* Implementation of this technology does not require permanent infrastructure, such as a permanent power source (temporary power is required), permanent water and chemical tanks, etc. Temporary power is required for operation of the system. This is much less expensive for the short duration of operation, typically less than 1 month and in many instances 1 to 2 weeks. Also required is a constant supply of water for process, as well as emergency, purposes. For remote sites where a distribution line with potable water is not available tanks for water storage are appropriate. During the demonstration, approximately 1000 gallons of water per day were used for a 6-day period.

TR0132

A.11

- *iii)* The end products of in situ oxidation are very appealing. No waste is generated from the treatment process, and no material is brought to the surface.
- *iv)* At complex sites in situ oxidation using Fenton's Reagent should be considered in tandem with other technologies. For example, if in situ bioremediation is considered as a polishing step, the pH should be held above 4.0 during the treatment operations.

Needs for Future Development:

- i) The effects on the aquifer geochemistry and microbiology in the treatment zone need to be better understood. Because in situ oxidation is a very robust chemical reaction, a reasonable assumption is that most of the microbial population was destroyed during the reaction. The type of microbial activity that will return to the area and to what extent is not known.
- ii) During the demonstration, the pH dropped dramatically from an average pH of 5.7 before treatment to 2.4 at completion of treatment. Post-test treatment has shown a very slow rebound of the groundwater pH. Three months after completion of the test, the groundwater pH remained at approximately 3.5.

3.2 Case Study 2: Potassium permanganate pilot test conducted at the Portsmouth Gaseous Diffusion Plant in Piketon, Ohio.

All italicized text in the section is taken directly from: Innovative Technology Summary Report DOE/EM-0496 In Situ Chemical Oxidation Using Potassium Permanganate. September 1999.

General operation of the potassium permanganate pilot study to remove TCE DNAPL at the Portsmouth Gaseous Diffusion Plant (PORTS) proceeded as follows:

i) The demonstration at PORTS was implemented using a pair of parallel horizontal wells with 200-foot (ft) screened sections located in a 5-ft thick silty, gravel aquifer within the center of a groundwater plume originating from a known source of DNAPL.

TR0132

A.12

- *ii)* Crystalline KMnO4 was added to groundwater extracted from the upgradient well and re-injected into the downgradient well approximately 90 feet from the extraction well.
- *iii)* Oxidant solution (~2% KMnO4) was recirculated through the horizontal wells for approximately one month.
- *iv)* Subsequent injection of KMnO4 into a nearby vertical well was conducted for 8 days to enhance uniform delivery of the oxidant in the region between the horizontal wells.

Key results of the pilot test include:

- *i)* Lateral and vertical heterogeneities within the aquifer significantly impacted uniform delivery of the oxidant through the horizontal wells.
- *ii)* Significant reductions in TCE were measured in both groundwater and soil samples in areas where the oxidant was delivered. During post-treatment sampling, TCE was not detected (< 5 parts per billion [ppb]) in samples collected from the monitoring wells and soil borings in locations where the oxidant had permeated. However, because oxidant delivery was not uniform, TCE was not reduced to non-detectable levels in all groundwater and soil samples.

Lessons learned during the technology application included the following:

Design Issues

- i) The recirculation concept of introducing permanganate into the subsurface is viable. Oxidant injection without extraction is feasible; however, there is no control in the subsequent movement of the oxidant after its release. Hence, recirculation is likely a preferable mode of operation.
- *ii)* Lateral heterogeneities impact the delivery of oxidants through the horizontal wells; whereas, vertical heterogeneities impact the delivery of oxidants through vertical wells.
- *iii)* If a recirculation approach is used to deliver the oxidant to the subsurface, a system for handling precipitated solids may need to be

TR0132

A.13

incorporated into the treatment system for higher oxidant dosing rates and higher contaminant concentrations.

- *iv)* Higher permanganate concentrations provide faster reaction times, significantly greater removal of DNAPL-type contaminants, but lessefficient oxidant use due to the natural oxidant demand of the subsurface.
- v) Typical treatment ratios for reagent (KMnO₄) to contaminant are greater than 5:1 based on field and laboratory studies. The efficiency of the process increases at higher contaminant concentrations and decreases as target treatment levels become more stringent.
- vi) Organic carbon content may impact treatment because the permanganate is relatively nonselective. However, no significant effect was observed with contaminant levels near 850 mg/l and total organic carbon of 0.1 to 1.3%.
- vii) For in situ groundwater treatment, the number and pattern of injection and extraction wells and monitoring wells must be designed to ensure maximum coverage of the treatment zone. Because the cost is related to depth and amount of DNAPL, the number and spacing of the wells becomes critical.

Implementation Considerations

- *i)* Formation Characteristics:
 - Soil and groundwater pH Permanganate is effective over a pH range of 3 to 12 with an optimum near 7.
 - Soil and groundwater Eh Background redox conditions must be defined to determine potential impacts on speciation and mobility of non-target metals.
 - Soil and groundwater TOC Ambient TOC can exert a demand on oxidant. Low TOC (<0.5%) is preferred to limit such demand or excess reagent will be required. This will increase costs.
 - Soil and groundwater temperature Temperature can impact reaction rates. Extremely low temperatures (e.g., < 10°C) slow reaction rates appreciably; so higher temperatures are preferred.

TR0132

- Soil and groundwater alkalinity and ionic strength High carbonate alkalinity can impact free radical oxidation by scavenging the free radicals produced and limiting oxidation efficiency. High ionic strength (e.g., by salts) can reduce reaction rates.
- *ii)* Contaminant Properties:
 - Type and concentration Applicable to unsaturated halocarbons (PCE, TCE, DCE), aromatics (BTEX), and polyaromatic hydrocarbons (phenols, naphthalene). May mobilize some redox sensitive metals in some settings.
 - Presence of co-contaminants May alter subsurface biogeochemistry and locally mobilize co-contaminants (e.g., redox sensitive metals such as Cr). Manganese oxides may sorb radionuclides.
- *iii)* Treatment Process Characteristics:
 - Delivered oxidant composition High concentrations may be needed to deliver adequate oxidant mass in a limited volume that is advected in the system.
 - Handling and Safety Depending on concentrations and form, permanganate is a strong oxidizer and is incompatible with combustibles. Care is required during handling.
 - *Reactivity and effects on formation matrix Permanganates can lead to some matrix plugging due to precipitation of MnO*₂ solids.
- *iv)* Injection and extraction wells may eventually become clogged from entrained silt, biological growth, mineral precipitates or other factors, but this effect appeared to be minor based on field demonstrations. Well and matrix clogging is expected to be more apparent during applications of oxidant injection/recirculation at higher oxidant concentrations and within areas of suspected DNAPL.

Needs for Future Development

TR0132

A.15

- Natural oxidant demand within a treatment area as related to oxidant dosing requirements needs to be better understood.
- Optimization of the oxidant to contaminant ratio must be further evaluated. While lower oxidant concentrations have been shown to be effective in the field, the residual concentrations may not be sufficient to treat contaminants within lower permeability zones. Higher oxidant concentrations will provide better residual oxidant for diffusion into the lower permeability areas, but may result in less efficient oxidant used (residual oxidant will be consumed by the natural oxidant demand within the matrix).

4 Thermal Technologies

Three case studies of field applications of three variations of the thermal technology are summarized below. The first case study is an outline of the application of a combination of steam flushing and electro-thermal dynamic stripping (ET-DSPTM) completed at the Pinellas Environmental Restoration Project: Northeast Site Area A, Young-Rainey Science, Technology and Research Center in Largo, Florida (U.S. Department of Energy, 2003). The second case study is a summary of an ERH pilot test conducted at Launch Complex 34, Cape Canaveral, Florida (Battelle, 2003). The third case study summarizes the results from an ECH demonstration conducted at the Shell Technology Ventures site in Portland, Indiana (Vinegar *et al.*, 1999). Table A.4 summarizes the site conditions and remedial activities at each of these sites. Sections 4.1 and 4.2 below include summaries of the treatment operation and lessons learned from each application.

A.16

Parameter	Pinellas Northeast Site Area A	Launch Complex 34, Cape Canaveral	Shell Technology Ventures
Technology description	Steam flushing and ET - DSP^{TM}	Electrical Resistive Heating	Electrical Conductive Heating
Soil type	sand; silty clay with some visible gravel	fine to medium sized sand, clay and shell fragments	Fill 1to 7 feet bgs; till to 18 ft bgs; sand and gravel 18-30 ft bgs
Hydraulic conductivity	1X10 ⁻³ cm/s	1.3 to 2.3 ft/day	not provided; effective permeability 2.5 X 10 ⁻⁸ cm/s
Hydraulic gradient	very low	0.00009 to 0.00007 ft/ft	not provided
Depth to groundwater	1 to 6 ft bgs	4-5 ft bgs	22-25 ft bgs
	steam injection rates 100 to 5,000 lbs/hr	not provided	not provided
Well yield	vapor injection rates 1 to 10 scfm per well		
	extraction rates during heating typically 30 gpm		
Contaminants of concern	TCE, cis-1,2-DCE, methylene chloride, toluene, petroleum range organics	ТСЕ	PCE, TCE in 1 st area; 1,1- DCE in 2 nd area
NAPL evidence	soil concentrations: TCE up to 1,000 mg/kg	soil concentrations: TCE greater than 300 mg/kg theoretical threshold for DNAPL	soil concentrations: PCE, TCE, 1,1-DCE, 3,500, 79 and 0.65 ppm, respectively
Volume treated	13,000 yd ³	not provided	6,500 tons
Area treated	10,000 ft ²	75 ft X 50 ft	1^{st} area 7,500 ft ² ; 2 nd area 600 ft ²
Depth treated	ground surface to 35 ft bgs	ground surface to 45 ft bgs	ground surface to 18 ft bgs (1 st area), 11 ft bgs (2 nd area)
DNAPL mass targeted	3,000 lbs	24,889 lbs	not provided

Table A.4: Summary of Thermal Technologies Case Studies

TR0132

	Parameter	Pinellas Northeast Site Area A	Launch Complex 34, Cape Canaveral	Shell Technology Ventures
	Remediation infrastructure	 15 steam injection wells; 28 extraction wells; 21 combined steam-ET-DSPTM; 2 deep ET-DSPTM electrodes 	13 electrodes;12 SVE wells 4-6 ft bgs;	1 st Area:136 heater /suction wells 2 nd Area:18 heater/suction wells dewatering trench in 1 st area to remove excess groundwater Temperature monitoring at 91 locations
Rei	nediation duration	Sept. 26, 2002- Mar. 24, 2003	Aug. 18, 2000-Sept. 19, 2001	Completed during 1997
	Remedial costs	\$3,800,000 including design, permitting, drilling, construction, operations, sampling, waste disposal, demobilization and reporting.	\$613,000 for demonstration and disposal;\$255,000 for pre-demonstration characterization	not provided
	formance results: met remedial goals	\checkmark	~	✓
2)	exceeded remedial goals	\checkmark	~	\checkmark
3)	met MCLs in at least some areas	\checkmark		\checkmark
4)	treatment efficiencies: TCE, etc.	TCE: 99.986% Toluene: 99.888% Methylene chloride: 99.989% Cis-1,2-DCE: 99.852% TPH: 61%	TCE: 97%	lowest treatment efficiency 98.4% for PCE (calculated from soil analyses)
5)	DNAPL mass remaining	1 lb	estimated at 3% of initial mass	mass estimates not provided

TR0132

A.18

4.1 Case Study 1: Steam Flushing and Electro-Thermal Dynamic Stripping Process Applied at Pinellas Northeast Site Area A, Largo, Florida.

General operation of the application of steam flushing and electro-thermal dynamic stripping process proceeded as follows:

- *i)* Hydraulic and pneumatic controls were established by liquid and vapor extraction.
- *ii)* Heating of the perimeter and bottom, while maintaining hydraulic control was achieved one month later.
- *iii)* Heating of all of Area A to the target temperature, using combined steam injection and ET-DSP, was achieved in the following two weeks, while maintaining hydraulic and pneumatic control.
- *iv)* NAPL mass recovery continued under pressure cycling and mass removal optimization conditions for a further 3 months. Pressure cycling continued until recovery entered a "diminishing returns" pattern.
- v) Cool-down and polishing, involving continued vapor and liquid extraction combined with air and cold water injection followed for a one month period.

As seen in Table A.4 above, the remedial action was deemed to have met or exceeded (below MCLs were achieved in many locations) the remedial objectives. The remedial objectives were to meet the cleanup criteria listed in Table 1 below within the specified time period (pp.5-6 and Table 1 of the source document (U.S DOE, 2003):

Chemical	Ground Water Cleanup Goal (μg/L)	Soil Cleanup Goal (µg/kg)
TCE	11,000	20,400
Cis-1,2-DCE	50,000	71,000
Methylene Chloride	20,000	227,000
Toluene	5,500	15,000
TPH - FL-PRO	50,000	2,500,000

TR0132

Hydraulic control of the NAPL and dissolved phase had to be achieved at all times, and the cleanup criteria had to be met for a 24 month period following cessation of remedial activities.

Lessons learned during the technology application included the following:

- *i)* Reduction of the water discharge rate would be achievable by recirculation of treated water, and/or lowering of the water addition rate to the treatment volume.
- *ii)* The extraction wells were observed to rise during curing and operation, leading to well grout seals near the surface requiring repair and maintenance. It would be desirable to prevent this expansion and the resulting asphalt cracking around the wells.
- *iii)* The air stripper had an average treatment efficiency of 73 percent, which is very low. The efficiency could be increased by the use of de-foaming agents, de-scaling agent, or other methods. Routine inspection and maintenance of the air stripper is necessary.
- *iv*) The liquid GAC system experienced operational problems that included fouling by precipitation of inorganics, and potentially biological growth. More focus should be given to prevention of the formation of these. The vapor GAC system should be made more robust for the periods of peak VOC recovery in the V-1 stream. Options include:
 - Design for total of subsurface vapors and air stripper off-gas;
 - Include more GAC vessels or larger vessels; and
 - Automate regeneration based on breakthrough monitoring.
- v) The vapor treatment of vinyl chloride and methylene chloride were problematic. A different vapor treatment system should be considered (*e.g.*, thermal oxidation or catalytic oxidation).
- *vi)* ERT was not effective at monitoring subsurface temperatures. The high dissolved solids in the ground water may have prevented the resistivity effects from temperature from being distinguishable, thus preventing ERT from being effective.

TR0132

A.20

4.2 Case Study 2: ERH Applied at the LC34 site at Cape Canaveral, Florida.

ERH was applied in a pilot study at the site to remove TCE DNAPL. General operation of the application of ERH process proceeded as follows:

- *i)* Pre-demonstration characterization of VOC concentrations in soil inside demonstration area.
- *ii)* Heat application with concurrent SVE for 43 days, followed by a 77 day break in heat application due to hurricane damage to transformer.
- *iii)* Second heat application with concurrent SVE for 98 days. Electrodes are upgraded during this period to enhance power input. This heating period was followed by a 48 day break due to rocket launches.
- *iv)* Third heat application with concurrent SVE for 62 days.
- *v)* Cool down and concurrent SVE for 79 days to evacuate any *in situ* TCE vapors during the cool-down phase.
- *vi)* Post-demonstration soil samples collected to characterize VOC distribution in demonstration area.

The demonstration exceeded the remedial objective of 90% removal of VOC mass in the demonstration area, by achieving an average of 97% removal. However, there was some evidence that VOCs may have migrated outside of the demonstration area and VOC vapors were measured in the ambient air surrounding the demonstration area, indicating that the pneumatic and hydraulic controls for the demonstration were not sufficient to control VOC migration.

Lessons learned during the technology application included the following:

- *i)* Cis-1,2-DCE was generated inside the demonstration area during the ERH application. The optimal enhancement of intrinsic bioremediation requires further study.
- *ii)* There was some evidence that VOCs migrated outside of the demonstration area based upon post-demonstration sampling. A mechanism for channeling vertically migrating vapors through less conductive layers would have been useful in aiding the prevention of VOC migration. In addition, hydraulic and pneumatic control may have also helped.

TR0132

A.21

- *iii)* A novel electrode design, consisting of an electrical cable attached to a ground rod within a graphite fill proved ineffective at the site due to excessive rainfall.
- *iv)* There was some loss of TCE to the ambient air during the heat application, indicating that the SVE system was not sufficiently efficient. The relatively shallow vadose zone may have played a role in vapor losses to the ground surface.

4.3 Case Study 3: ECH Process Applied at Shell Technology Ventures site in Portland, Indiana.

The ECH demonstration was conducted to remove PCE, TCE and 1,1-DCE from two areas at the site. General operation of ECH proceeded as follows:

- *i)* Extraction of any liquids removable via pumping and/or SVE. A dewatering trench was installed after pumping failed to dewater one area of the demonstration area.
- *ii)* Heating of soil for 40-60 days, with concurrent monitoring of *in situ* temperature and pressure distribution, HCl in off-gas (indicating extracted VOCs). Three quarters of the site reached steam superheat temperatures over this period, and all areas exceeded the boiling point of water.
- *iii)* Soil sampling from 50 points to confirm VOC removal in coolest areas prior to shutdown of heating.
- *iv)* Heating was discontinued and soil temperature was monitored for a six month period over which it decreased to below 100 °C.
- *v)* Confirmatory soil sampling.
- *vi)* Resodding of soil to accelerate natural revegetation.

The remedial action was deemed to meet the remedial objectives, with all samples below 0.5 ppm PCE and 0.02 ppm TCE. Soil samples collected along the perimeter of the demonstration area showed that contaminants were not spread outside of the heated area by the heat application.

TR0132

A.22

Lessons learned during the technology application included the following:

- *i)* High groundwater influx led to lower achieved temperature increases even with perimeter drains to control influx. However, steam distillation was sufficient to remove VOCs in this area.
- *ii)* Heating of subsurface resulted in death of ground cover, which required resodding.

5 ZVI Case Studies

One case study each for field applications of zero-valent iron (ZVI) jetted as a slurry with clay and nano scale ZVI mixed into emulsified vegetable oil prior to injection (EZVI) are included below. The ZVI and clay slurry was a full-scale application at DuPont's Kinston plant, Kinston, NC (EPA, In Press). The EZVI case study describes a pilot scale application of EZVI at Launch Complex 34 Cape Canaveral Air Force Station, FL (O'Hara *et al.*, 2003). Table A.5 summarizes the site conditions and remedial activities at the sites. Sections 5.1 and 5.2 below include summaries of the treatment operations and lessons learned from each application.

Table A.5 Summaries of ZVI Case Studies	Table A.5	Summaries	of ZVI	Case Studies
---	-----------	------------------	--------	---------------------

Parameter	Launch Complex 34, Cape Canaveral	Kinston, NC
Technology description	EZVI with nano-scale ZVI	ZVI and clay slurry injection by jetting
Soil type	fine to medium sized sand, clay and shell fragments	15 – 18 ft of sand overlying confining mudstone layer
Hydraulic conductivity	1.3 to 2.3 ft/day	not provided; groundwater velocity 0.05 to 0.1 ft/d
Hydraulic gradient	0.00009 to 0.00007 ft/ft	not provided
Depth to groundwater	4-5 ft bgs	4-5 ft bgs
Well yield	not provided	not provided
Contaminants of	TCE	TCE

Parameter	Launch Complex 34, Cape Canaveral	Kinston, NC
concern	CanaveralKinston, IVCsoil concentrations: TCE greater than 300 mg/kg theoretical threshold for DNAPLsoil concentrations: TCE, from 10 to 100 mg/kg and groundwater 	
NAPL evidence	soil concentrations: TCE greater than 300 mg/kg theoretical threshold for DNAPL	to 100 mg/kg and groundwater
Volume treated	1080 ft ³	not provided
Area treated	9 ft X 15 ft	~30 ft diameter zone
Depth treated	16 to 24 ft bgs	ground surface to 18 ft bgs
DNAPL mass targeted	up to 46 kg	not provided
Remediation Infrastructure	 4 recirculation wells 4 5-point multilevel monitoring wells; 7 monitoring wells 	 core locations monitoring wells 11 treatment (ZVI/Clay) columns emplaced low K cofferdam of jetted
Remediation Duration	EZVI injection was 3 days, post injection monitoring for 6 months	Completed during 1999
Remedial Costs	not provided	not provided

TR0132

A.24

	Parameter	Launch Complex 34, Cape Canaveral	Kinston, NC
Per	formance results:		
1)	met remedial goals	✓	some goals met – mixing not as effective on a particle scale but on inches scale. Jetting appears to have effective in creating sufficient mixing through most of
2)	exceeded remedial goals	Where EZVI was present exceeded goal of 50% mass reduction	source zone
3)	met MCLs in at least some areas		
4)	treatment efficiencies: TCE, etc.	TCE: ~58% based on soil cores ~56% based on mass flux reduction (up to 100% in zones with good EZVI	
	DIVER	distribution	
5)	DNAPL mass remaining	estimated at 42% of initial mass	mass estimates not provided

5.1 Case Study 1: EZVI at LC34 Cape Canaveral Florida.

General operation of the application of EZVI process proceeded as follows:

- *i)* A set of 6 pre-demonstration cores, sampled in 2-ft sections and fully extracted using methanol were collected within the 9.5 x 15 ft pilot test area to estimate DNAPL mass and distribution;
- *ii)* A complete set of groundwater samples from four 5-point multilevel (20 samples) and five fully screen wells collected to get baseline groundwater conditions;
- *iii)* Based on soil and groundwater concentrations the DNAPL was not evenly distributed through the treatment area. The bulk of the DNAPL mass appeared

A.25

to be located in the southwest corner of the pilot test area and at depths of approximately 18 and 24 ft bgs.

- iv) EZVI injected into 8 wells at two depth intervals (20 to 24 ft bgs and 16 to 20 ft bgs) using pressure pulse technology. Pressure pulse technology uses a process of periodic (*e.g.*, 1 pulse per second) large-impulse hydraulic excitations to introduce hydraulic strain energy into the formation to minimize fingering and help fluid flow.
- *v)* Total of 750 gal of EZVI injected into treatment area.
- *vi)* Approximately 1.5 months after injection, a set of interim cores were collected in the test cell to determine distribution of the EZVI and measure degradation.
- *vii)* Groundwater samples were collected weekly from a subset of wells for a period of 4 months.
- *viii)* A set of 6 post-demonstration cores were collected, sampled in 2-ft sections and fully extracted using methanol were collected within the 9.5 x 15 ft pilot test area to estimate post treatment DNAPL mass and distribution.
- *ix)* A set of groundwater samples from all sampling points was collected to evaluate mass flux reduction.

As shown in the table above, the treatment goal of 50% reduction in TCE concentrations was met based on both the soil core results and groundwater results.

Lessons learned during the technology application included the following:

- *i)* EZVI was not evenly distributed , with significant fingering and preferential flow paths;
- *ii)* EZVI appeared to have traveled upwards from the injection depth intervals despite being a DNAPL;
- *iii)* Oil and surfactant in the EZVI appeared to contribute significantly to the degradation observed due to enhanced biodegradation;
- *iv)* Ongoing work on improving delivery mechanisms for the EZVI and investigation of the amount of contribution from biodegradation are needed.

TR0132

A.26

5.2 Case Study 2: In-situ Treatment of a TCE Source Area Using a Jetted Slurry of ZVI and Clay at Dupont plant, Kinston, NC.

General operation of the application of ZVI/Clay slurry process proceeded as follows:

- *i)* slurry consisted of 95% kaolinite and 5% Peerless ZVI on a dry weight basis;
- *ii)* 11 columns were emplaced to depths from 15 to 18 ft bgs and column diameters estimated from 5 to 6 ft placed at centerline distances of 4 to 5 ft.
- *iii)* a low K cofferdam was of jetted ZVI/clay was placed completed around the source area perimeter;
- *iv)* cofferdam jetting centers placed 9 ft part to create interlocking panels;

Some of the remedial goals were met with the treatment (see Table A.5 above). Of the 13 post demonstration samples collected, only 2 showed significant TCE or breakdown products in the post-treatment cores. Direct observation indicated that intimate mixing on a particle scale was not achieved; however, there was slurry on the scale of inches distributed through the soil. Monitoring is ongoing to determine if groundwater concentrations will continue to decline.

6 Multi-Phase Extraction

A case study for a field application of multi-phase extraction (MPE) is summarized below. The case study is an outline of the application of a full scale multiphase extraction application at the 328 Site in Santa Clara, CA (U.S. EPA, 1999). Table A.6 summarizes the site conditions and remedial activities at the site. Section 6.1 below includes summaries of the treatment operation and lessons learned from the application.

Parameter	328 Site
Technology description	Dual Phase Extraction (single pump) with pneumatic fracturing of soil
Soil type	silty clay
Hydraulic conductivity	not provided

Table A.6 Summary of MPE Case Study

TR0132

Parameter	328 Site				
Hydraulic gradient	not provided				
Depth to groundwater	9 ft bgs				
Well yield	not provided				
Contaminants of concern	TCE				
NAPL evidence	TCE concentration in soil 46 mg/kg				
Volume treated	16,000 yd ³				
Area treated	0.5 acre				
Depth treated	0-20 ft bgs				
DNAPL mass targeted	not provided				
Remediation Infrastructure	(#wells, monitoring points, etc.)				
Remediation Duration	November 1996 to October 1998				
Remedial Costs	Design and installation \$300,000 O&M, reporting, analytical \$450,000 Carbon disposal \$100,000				
Performance results:					
1) met remedial goals	\checkmark				
2) exceeded remedial goals					
3) met MCLs in at least some areas					
4) treatment efficiencies: TCE, etc.	Aqueous VOC concentrations decreased from $4,000 \mu g/L$ to $650 \mu g/L$; 1164 lbs TCE mass removed				
5) DNAPL mass remaining	not quantified				

TR0132

A.28

6.1 Case Study 1: 328 Site, Santa Clara, CA

General operation of the application of MPE proceeded as follows:

- *i)* Vacuum application and MPE initiated; clusters of extraction wells operated on a rotating basis.
- *ii)* Pneumatic fracturing of source area wells to improve vapor extraction two months later while cluster operations of extraction wells continued. Fracturing was completed over a 5 month period.
- *iii)* Continued operations focusing on areas of highest VOC concentrations for one year.
- *iv)* Shutdown of MPE system for two months to observe rebound.
- *v)* Restart and continued operation for one month. VOC extraction data suggested that remedial goals had been met.
- *vi)* Confirmation soil sampling one month later. Preparation for system shutdown two months later.

Based upon the post-treatment soil and groundwater data, the remedial action was deemed to have met the remedial objectives. The remedial objectives were to remove the VOC source zone (U.S DOE, 2003).

Lessons learned during the technology application included the following:

- *i)* Pneumatic fracturing was a useful technique for enhancing vapor extraction rates;
- *ii)* Groundwater extraction rates were much higher than anticipated by characterization efforts, necessitating a rotating schedule of well operation to meet the capacity of the groundwater treatment system. This emphasizes the importance of careful characterization and/or pilot testing of the technology for a given site to ensure proper design of the MPE application; and,
- *iii)* The majority of mass extracted using MPE was treated in the vapor stream as opposed to the water stream. VOCs are commonly volatilized from extracted

TR0132

A.29

groundwater during MPE and thus sizing of vapor treatment should be based upon the expected mass of VOCs extracted from both phases.

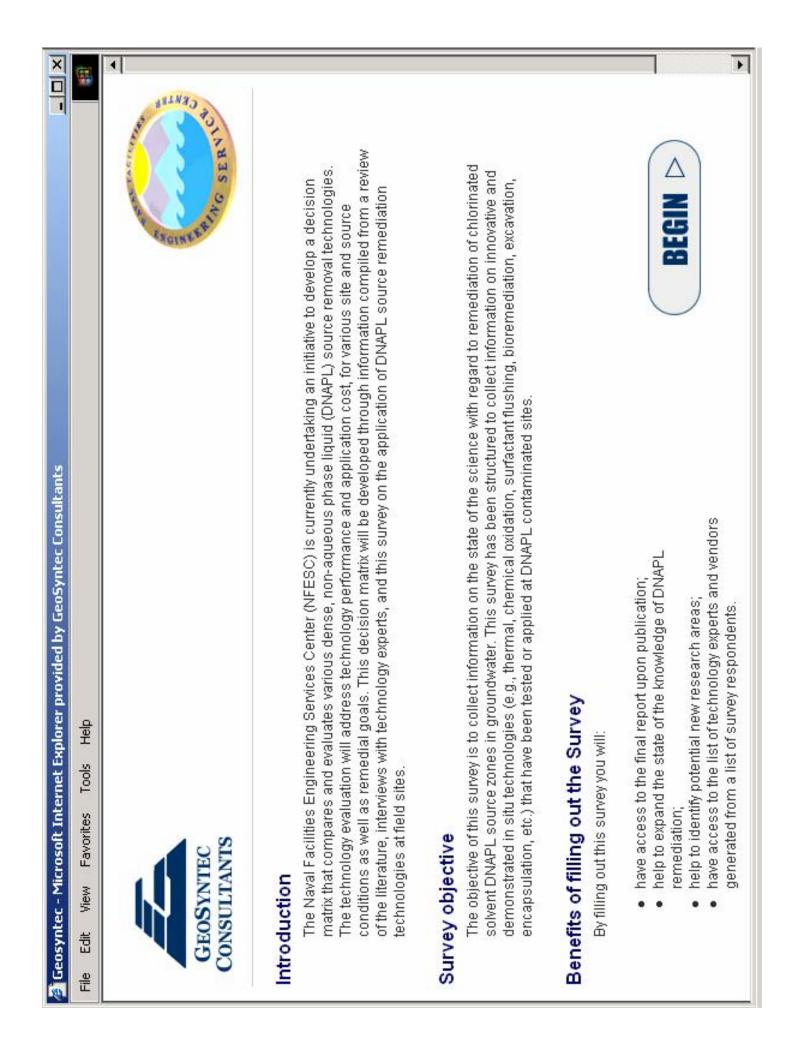
TR0132

A.30

APPENDIX B

Screen Shots of the Web Based Survey

×	-	•							1	F
ants			inant profile, etc.)	assword. On subsequent visits to the survey, you will be asked if you wish to ne opportunity to continue where you left off from previous access events. You data for a given site for up to 30 days. If you wish to change information <u>v@qeosyntec.com</u> .	iple sites, as well as multiple technologies applied at urvey, redundant information (e.g., personal vill not be required to be reentered.	į	ents or concerns, please contact:	Suzanne O'Hara, GeoSyntec Consultants Hydrogeologist Tel. (519) 822-2230 ext. 234 survey@geosyntec.com	LOGIN	o Consultants
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	Favorites Tools Help	ey Instructions The Survey has three components:	personal contact information site specific information (e.g., geology, hydrogeology, contaminant profile, etc.) technology specific information (e.g., performance, costs, etc.)	Begin by entering your email address and a password. On subsequent visits to the survey, you will be asked if you wish to modify information already inputted, or have the opportunity to continue where you left off from previous access events. You will have the opportunity to review and update data for a given site for up to 30 days. If you wish to change information following that time period, please email <u>survey@geosyntec.com</u> .	The survey is set up to allow one person to enter information on multiple sites, as well as multiple technologies applied at each site. To minimize the amount of time required to complete the survey, redundant information (e.g., personal information, and site-specific information for multiple technologies) will not be required to be reentered.	The survey should require approximately 30 to 40 minutes to complete.	Contact Information for Survey Support For further information about the survey, or if you have any questions, comments or concerns, please contact:	Carmen Lebrón, NFESC Julie Konzuk, GeoSyntec Consultants Environmental Engineer Tel. (805) 982-1616 Tel. (519) 822-2230 ext. 241 lebronca@nfesc.navy.mil <u>survey@deosyntec.com</u>		@ Convright 2002 - GenSvoted Consultants
🖉 Geosyntec - Micr	File Edit View F	Survey Instructions The Survey has thre	 per site tech 	Begin by er modify info will have th following th	The survey each site. T information	The survey	Contact Infor			



× □ -	-	R								1	
		ENGRACE CENTRE							days.		
nts				*		eld blank.	LOGIN >		vill be contacted within 3 business	Consultants	
vided by GeoSyntec Consultants			Survey Login To login, please enter your email address and password in the fields below.			a) If this is your first time accessing the survey, please leave the password field blank. You will be able to set your password on the following page.			If you have forgotten your password, please email <u>survey@geosyntec.com</u> and you will be contacted within 3 business days.	© Copyright 2002 - GeoSyntec Consultants	
🖨 Geosyntec - Microsoft Internet Explorer provided	Tools Help		email address and		<mark>*</mark>	 a) If this is your first time accessing the survey, please leave You will be able to set your password on the following page. 		ed by 🏄	word, please email <u>s</u>		
licrosoft Inte	Favorites	VITEC	g in e enter your e	L		r first time ac e to set your (* Mandatory fields are denoted by *	tten your pass		
osyntec - M	Edit View	GEOSYNTEC	Survey Login To login, please en	Email Address:	Password:	f this is your I will be abl		andatory fiel	u have forgo		
E B	File		101 To I	Em	Ъа:	a) Yo(¥ W	lf yo		

×□·		 → 60 	A CONCENTRAL		4 10:44 AM
🖉 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	File Edit View Favorites Tools Help	Address 🛃 https://projects.geosyntec.com/navy_rocs/site_list.asp	GEOSYNTEC CONSULTANTS	Welcome a D: Below are a list of sites associated with your profile. Click here to edit your personal information. Site Name Technology No sites defined DETINE NSTE<	Mstart 🚮 🧔 📓 📷 🖏 😒 » 🔯1 🖓1 🕅 🖓[🖾 T 🗐 G 🕅 A

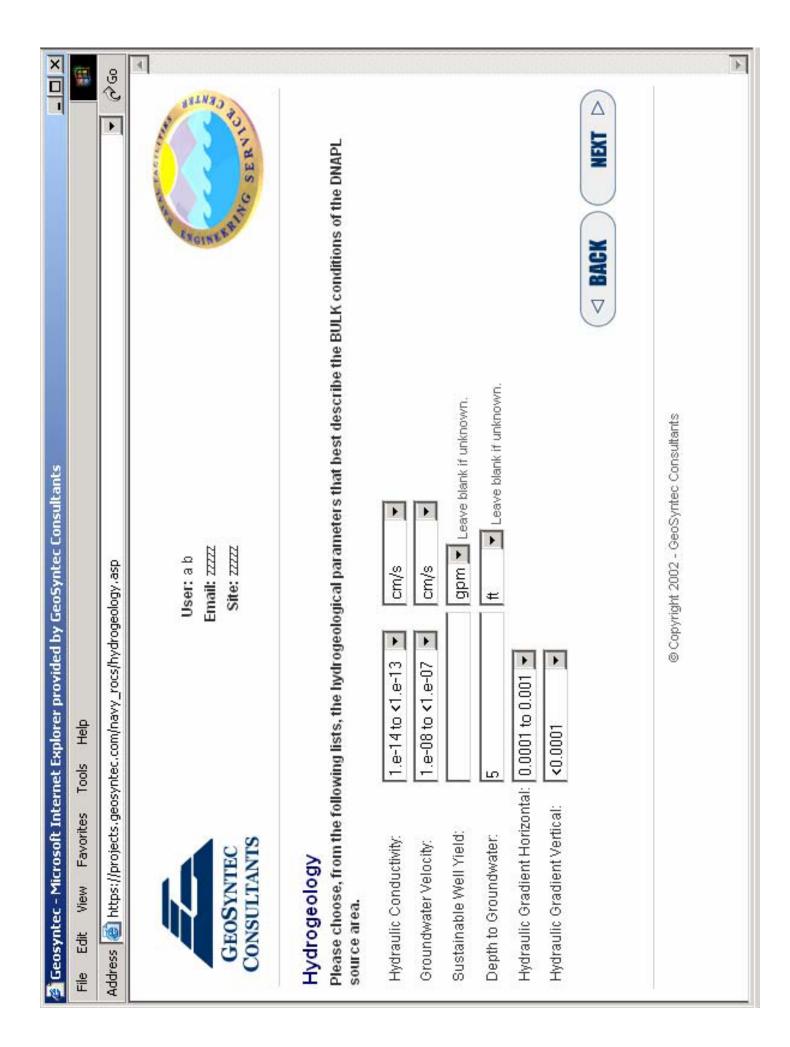
FILE EQIC VIEW FAVORICES LOOIS 1	deb
Address 🛃 https://projects.geosyntec.com/navy_rocs/site.asp	m/navy_rocs/site.asp
Site Information	
Enter as much information as you can about the	an about the site.
Is this site confidential:	Yes C No G
Site Name:	* 22222
	If the site is confidential, please enter a name that will allow you to distinguish it from other sites you may define
Closest City:	22222
Country (and State/Province, if applicable):	Alaska 💌 💉 🔹
Zip Code / Postal Code:	
Is this site represented in the Federal http://www.fitr.gov. O Yes O No O Not sure	Is this site represented in the Federal Remediation Technologies Roundtable database of technology? This can be looked up at <u>http://www.fitr.gov</u> . C Yes
If applicable, please provide the Oper	If applicable, please provide the Operating Unit (OU) or Solid Waste Management Unit (SWMU) name and ID:
OU/SWMU Name:	N/A
OU/SWMU ID:	N/A A/V
🌒 Start 🛛 🛃 🔊 🔊 🖏 🚺 🔇	× Øl Ql Mm. Øl ⊠T @G ØA Om. 4 🗂 📆 6 🕅 20:

	2	S S S S S S S S S S S S S S S S S S S			Technologies will be defined following completion of the site survey			🌾 🗂 📷 🕼 👹 😡 🖓
			Welcome a b! Below are a list of sites associated with your profile. Click <u>here</u> to edit your personal information.	Technology	Edit Site Specifics Technologies will be the site survey		© Copyright 2002 - GeoSyntec Consultants	1.네 쨉에서 (윗6.네 조1.네)(좋)6 [참A.네 [립M.네
tes Tools Help	Address 😂 https://projects.geosyntec.com/navy_rocs/site_list.asp		s associated with your profile. Click		Begin Site Survey		© Copyrigh	W Q S * Q 1.4 Q1.4
File Edit View Favorites	Address (https://projects	GEOSYNTEC CONSULTANTS	Welcome a b! Below are a list of sites	Site Name	77777	DEFINE NEW SITE		🌒 Start 🛛 🛃 🙆 🕅

191	🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants		
1836	File Edit View Favorites Tools Help		
4	Address 🛃 https://projects.geosyntec.com/navy_rocs/geology.asp?action=begin&site_id=53&site_name=zzzz	 ▲ № 	
	GEOSVNTEC GEOSVNTEC Site: ZZZZ Site: ZZZZ	A COLOCIONES	
	Geology Enter relative percentage of media type. If your site has both consolidated and unconsolidated media, then enter the relative percentage of both media in the appropriate box. For fractured media, you are asked to provide information on both the degree of fracturing, as well as the permeability of the matrix.	hen enter the relative ation on both the degree of	
	Consolidated Media (i.e. bedrock) Degree of Fracturing		
	Discretely Fractured: (< 10 conductive fractures)	0%	
	Highly Fractured: (fractures are visible throughout zone of interest, many water conductive fractures) Unfractured:	• %0	
	Matrix Permeability	1	
	Low Permeability Matrix: (crystalline, metamorphic rocks, some limestones, etc.)	- %0	
	High Permeability Matrix: (some limestones, sandstone, etc.)	• %0	
	Unknown:	0%	
_			
	CUnconsolidated Media (i.e. sand, clay, etc.)		
	Low Heterogeneity: (heterogeneity not visible to trained geologist - e.g., beach sands)	• %0	
	High Heterogeneity: (variable soil types, orders of magnitude variation in K, layers, etc.)	0%	
	Unknown:	• %0	
		NEXT >	_

rile cuit view ravorites rouis nep Address 🚳 https://projects.geosyntec.com/navy_rocs/lithology.asp?fractured=True&porous=False		ی ج
Lithology Please choose ONE rock type that best describes the most prevalent geologic material in the source area:	ea:	
Consolidated Media		
C Basaltic (gabbro, peridotite, basalt)		
O Granitic (rhyolite, granite)		
O Intermediate (andesite, diorite)		
O Other igneous rock		
Metamorphic		
O Metashale (slate, phyllite, schist or gneiss)		
O Metabasalt (greenschist, amphibolite, granulite)		
C Other metamorphic rock		
Sedimentary		
C Limestone (Karstic)		
O Limestone (Non-Karstic)		
C Sandstone		
O Shale		
O Other sedimentary rock		
Unknown		
C Type unknown		
🗯 start 📔 😹 📓 📷 🖏 🗐 🔕 🗤 🛛 🔕 Li-j 💁 Li-j 🐻 Li-j 🖉 Li-j 🖉 Li-j 🖉 Li-j 🖉 Li-j 🖉 Ga. 🖄 Ai-j 📴 Mi-j		10:53 4





🗿 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	Syntec Consultants
File Edit View Favorites Tools Help	
Address 🛃 https://projects.geosyntec.com/navy_rocs/source_zone.asp	tasp رکاری در ا
GEOSYNTEC GEOSYNTEC CONSULTANTS	User: a b Email: zzzz Site: zzzz
Source Zone Please choose from the following lists for the parameters th	eters that best describe the source zone conceptual model.
DNAPL Composition - Estimate of the percentage of each type of cl contaminant at your site if a sample of DNAPL has been analyzed:	DNAPL Composition - Estimate of the percentage of each type of chlorinated solvents that best describes the contaminant at your site if a sample of DNAPL has been analyzed:
(ethylenes) trachloroethene, cis 1-2-dichlorethene,	8
Chlorinated Ethanes (i.e. Trichloroethane, 1,1,2,2-tetrachloroethane) Chlorinated Methanes (i.e. chloroform, dichloromethane)	8 8
Chlorobenzenes (i.e. hexachlorobenzene, dichlorobenzene) Other:	8 8
 Dissolved Contaminant Profile Estimate of the maximum concentration of each chlorinated solvent that occurs within the defined source. If value is unknown, please leave the field blank. 	d solvent Choose applicable concentration ranges from the following lease effectiveness:

🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	ultants		× □ -
File Edit View Favorites Tools Help			(H)
Address 🙆 https://projects.geosyntec.com/navy_rocs/source_zone.asp		F	৫ ^৯ Go
Dissolved Contaminant Profile Estimate of the maximum concentration of each chlorinated solvent that occurs within the defined source. If value is unknown, please leave the field blank.	 Other Groundwater Geochemistry Choose applicable concentration r list of chemicals that may be releva effectiveness: 	Other Groundwater Geochemistry Choose applicable concentration ranges from the following list of chemicals that may be relevan in terms of technology effectiveness:	•
Select Units: [mg/L]	Nitrate: Nitrite: Sulfate:	>1 and <10 mg/L Unknown >10 and <100 mg/l	
tetrachloroethene (perchloroethylene): trichloroethene (trichloroethylene): 1,1-dichloroethene (1,1-dichloroethylene):	Sulfide: Phosphate: Calcium:		
cis-1,2-dichloroethene (cis-1,2-dichloroethylene): [Carbonate: Morente:	>10 and <100 mg/L Unknown	1
Chlorinated Ethanes: hexachloroethane:	mercuny. Arsenic: Hexavalent Chromium:	<1 mg/L <1 mg/L Unknown	
pentachloroethane: 1,1,1,2-tetrachloroethane: 1,1,2,2-tetrachloroethane:	Hydrogen: Dissolved Iron: Total Iron:	Unknown	
1,1,1-trichloroethane:	Dissolved Manganese: Total Manganese:	Unknown	Þ

🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	GeoSyntec Consultants			× □ -
File Edit View Favorites Tools Help				-
Address 🛃 https://projects.geosyntec.com/navy_rocs/source_zone.asp	_zone.asp		•	ළ
1,1,2-trichloroethane:	Total Manganese:	Unknown	Þ	•
1,2-dichloroethane (ethylene dichloride):	Fraction of Organic Carbon:	Unknown	Þ	
1,1-dichloroethane:	Biological Oxygen	Unknown	Þ	
chloroethane:	Dissolved Organic Carbon:	Unknown	•	
Chlorinated Methanes:	Dissolved Oxygen:	Unknown	•	
tetrachloromethane (carbon tetrachloride):	Hq	Unknown 🔸	1	
trichloromethane (chloroform):	Oxidation-Reduction	LInknown	•	
dichloromethane (methylene chloride):	Potential:			
chloromethane:	Ethane;	Unknown Unknown	•	
Chlorinated Benzenes:	Ethene:	Unknown		
hexachlorobenzene:		-		
pentachlorobenzene:				
1,2,3,4-tetrachlorobenzene:				
1,2,3,5-tetrachlorobenzene:				
1,2,4,5-tetrachlorobenzene:				1
1,2,3-trichlorobenzene:				
1,2,4-trichlorobenzene:				
1,3,5-trichlorobenzene:				Þ

×	-	.8		٦
× □ -		¢60		
		•		
			NEXT	
			BAG	
			BACK	
			67	
			© Copyright 2002 - GeoSyntec Consultants	
ints			S S	
isulta			Synteo	
ic Cor				
Synte		asp.	502	
Geo		_zone		
ed by		source		
rovid		rocs/		
irer p	de S	/navy.		
Explo	Help	c.com		
irnet	Tools	synte		
t Inte	rites	ts.geo	Zene: Zene:	
rosof	Favorites	projec	roben roben nzene ene: ene: ene:	
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	View	Address 🛃 https://projects.geosyntec.com/navy_rocs/source_zone.asp	1,2,3,4-tetrachlorobenzene: 1,2,3,5-tetrachlorobenzene: 1,2,4,5-tetrachlorobenzene: 1,2,4-trichlorobenzene: 1,2,4-trichlorobenzene: 1,3,5-trichlorobenzene: 1,3-dichlorobenzene: 1,4-dichlorobenzene: 1,4-dichlorobenzene: 0ther (specify contaminant):	
rntec		포	1,2,3,4-tetrachlo 1,2,3,5-tetrachlo 1,2,4,5-tetrachlo 1,2,4-trichlorobe 1,2,4-trichlorobe 1,3-dichloroben 1,3-dichloroben 1,4-dichloroben	
soa5	e Edit	ress	1,2,3, 1,2,3, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,4, 1,2,3, 1,2,4, 1,2,3, 1,2,4, 1,2,3, 1,2,4, 1,2,3, 1,2,4, 1,2,3, 1,2,4,1,2,4,1,4,1,4,1,4,1,4,1,4,1,4,1,4,	
1	File	Add		

×			1					Þ
		C. C. E. NTER	s with					
		SERVICA CANTER	For site fc.) as					
		AL ENCINE SEIT	activities. ne, mass e					o 🖸 Irregular
			Dense Non-Aqueous Phase Liquid (DNAPL) Description Please provide the following information on the DNAPL distribution and mass prior to the onset of remedial activities. For sites with multiple DNAPL source zones, please provide the TOTAL value for all source zones (e.g., areal extent, volume, mass etc.) as appropriate.					C Rectangular
			r to the onse s (e.g., areal	F			clay)	C C
ltants			ion mass prio urce zone	m2	m3	• cm	k matrix or	$\mathbf{C}^{\mathbf{C}}$ Inverted triangle
c Consul		88	script ion and i or all so				iers (roc	0 Inverte
r GeoSynte		User: a b Email: <u>2222</u> Site: <u>2222</u>	APL) De 'L distribut TAL value f				l into low k layers (rock matrix or clay) d fractures	
🗿 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants			Dense Non-Aqueous Phase Liquid (DNAPL) Description Please provide the following information on the DNAPL distribution and mas multiple DNAPL source zones, please provide the TOTAL value for all source appropriate.	Unknown	Unknown	Unknown	 In pools Residual Sorbed Diffused in Dead-end 	Dnknown
Explorer	ls Help		nase Li ormation lease pro					f the face
t Internet	ites Tools		eous Pl owing info e zones, p	What is the total areal extent of DNAPL distribution?	What is the total volume of impacted soil (pre-remediation) containing DNAPL?	What is the maximum depth of the DNAPL?	How is the DNAPL distributed in the subsurface: (choose all that apply)	What was the estimated shape of the DNAPL distribution in the subsurface used in determining treatment geometry:
Microsof	View Favorites	GEOSYNTEC	on-Aque de the foll PL source	ital areal e	ital volum ediation) (iaximum (NAPL dist lat apply)	What was the estimated shape DNAPL distribution in the subs used in determining treatment geometry:
osyntec -	Edit Vie	GEOSYNTEC	Dense No Please provi multiple DNA appropriate.	What is the to distribution?	What is the to soil (pre-rem DNAPL?	What is the m DNAPL?	How is the DNAPL dis subsurface: (choose all that apply)	What was the DNAPL distrii used in deter geometry:
🧟 Ge	File		De Plea muth appl	Wh: disti	Vh soil DN/	λην DN7	Hov sub (chc	Vh: VND geo

Help
Tools
Favorites
View
Edit
File

Enter the total number of sampling locations used to locate the DNAPL source area(s) for each of the following techniques:

IIIy-Screened Monitoring Wells: 10 ested Monitoring Wells: 0 epth-discrete sampling (e.g. 0 :oprobe, waterloo profiler, etc.): 2 oil sampling: 2 her sampling method (describe):			
Nested Monitoring Wells: 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	Nested Monitoring Wells: 0 Depth-discrete sampling (e.g. geoprobe, waterloo profiler, etc.): 0 Soil sampling: 2 Other sampling method (describe): 0	Fully-Screened Monitoring Wells:	10
Depth-discrete sampling (e.g. 0 Depth-discrete sampling (e.g. 3eoprobe, waterloo profiler, etc.): 2 Soil sampling: 2 Dther sampling method (describe): Dther sampling method (de	Depth-discrete sampling (e.g. 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	Vested Monitoring Wells:	0
Soil sampling: 2 2 Other sampling method (describe):	Soil sampling: 2 2 Other sampling method (describe):	Jepth-discrete sampling (e.g. Jeoprobe, waterloo profiler, etc.):	0
Other sampling method (describe):	Other sampling method (describe):	soil sampling:	2
		Other sampling method (describe):	

How was DNAPL presence determined? (check all that apply and enter total number of sampling location or points used):

Site history:		
Known spill(s):	٢	
Inferred from dissolved contaminant concentrations:		Detected using interface probes Enter number used: 12
Direct observation of DNAPL in soil sample:		Detected using Ribbon NAPL samplers (e.g. FLuTEs TM) Enter number used: 0
Extracted from monitoring wells:	٢	
Partitioning tracers:		

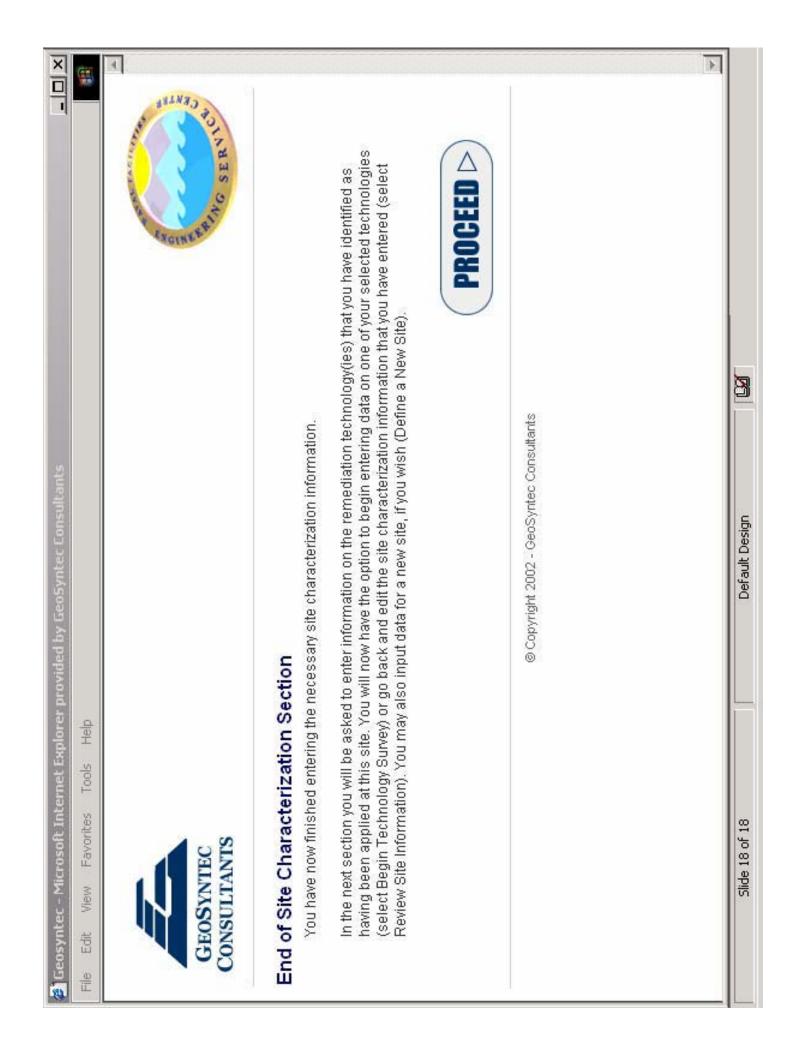
×	9	•									[
	***		4	F	apply and enter									
					How was the total (pre-remediation) DNAPL mass present in the subsurface estimated? (check all that apply and enter number of sampling location or points used):									△ BACK
					urface estimated									
by GeoSyntec Consultants					sent in the subs									
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoS					How was the total (pre-remediation) DNAPL mass pre number of sampling location or points used):								Unknown	
plore	Help				ation) r poir	_		_		_	_			
rnet Ex	Tools		5		smedia tion o			ins in	ements				*	
oft Inte	Favorites		Other method of determination (describe):		(pre-re ng loca			Calculated from concentrations in groundwater or soil:	Calculated from flux measurements:			ibe):	Enter the estimated total (pre- remediation) DNAPL mass:	
Micros	w Fav	arcio.	of dete		e total amplii		×	m con()r soil:	m flux i	acers:	<u></u>	Other method (describe):	nated t DNAPL	
ntec -	it View	r arouning vacers. Not estimated:	nethod be):		vas the er of s	story:	spill(s)	ated fro water c	ated fro	ning tra	imated	nethod	he estir ation) [
🖨 Geosy	File Edit	Not estimated:	Other meth (describe):		How w numbe	Site history:	Known spill(s):	Calculated from con groundwater or soil:	Calcul	Partitioning tracers:	Not estimated:	Other n	Enter the estimated total (pr remediation) DNAPL mass:	

×	GT	4]	F.
plorer provided by GeoSyntec Consultants		User: a b Email: zzzz Site: zzzz		Facility operations constrained site access Presence of surface infrastructure impacted ability to estimate DNAPL mass or effectively conduct remediation	oice, design or effectiveness me-frame	BAGK NEXT >	© Copyright 2002 - GeoSyntec Consultants	
🚰 Geosyntec - Microsoft Internet Explorer provided b	File Edit View Favorites Tools Help	GEOSYNTEC CONSULTANTS	Impacts to Remediation Which of the following factors impacted remediation (check all that apply)?	 Facility operations constrained site access Presence of surface infrastructure impacted ability 	Proximity to surface water impacted technology choice, design or effectiveness Regulatory pressures impacted clean-up criteria/time-frame	BudgetRemediation cost		

×	-					T
		S STRATER NO				
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	Help	User: a b Email: <u>zzzz</u> Site: <u>zzzz</u>	Technology Please choose all the technologies you would like to contribute information about: Excavation			
🗿 Geosyntec - Microsoft Internet E	File Edit View Favorites Tools	GEOSYNTEC CONSULTANTS	Technology Please choose all the technologie Excavation	Bioremediation	Chemical Oxidation Chersulfate C Persultate C Permanganate Fenton's Reagent C Ozone C Other	Thermal ChemOxTechnology ChemOxTechnology Cheam Steam Steam Three phase heating Resistive heating Other

×	ġ.	•						•

eosyntec - Microsoft Internet Ex	File Edit View Favorites Tools Help	Surfactant Flushing	Cosolvent Flushing □	Dual Phase □ Water/DNAPL extraction (i.e, extraction of DNAPL directly) □ DNAPL/air extraction (i.e, DNAPL vapor extraction without lowering the watertable) □ Water/air extraction (i.e., DNAPL vapor extraction through lowering of the watertable)	ZVI/nano-scale iron Czero-valent iron Nano-scale iron	Other		© Copyright 2002 - GeoSyntec Consultants



×	4		F
	SER LICA CENTER	Begin Technology Survey Survey	
		rsonal information. Technology Chemical Oxidation - Permanganate ZVI/hano-scale iron - Zero- valent iron sultants	
eoSyntec Consultants		ofile. Click here to edit your personal i Techn Burvey Edit Site Specifics Chemi Perma ZVI/na Copyright 2002 - GeoSyntec Consultants	
Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants File Edit View Favorites Tools Help		Welcome a b! Below are a list of sites associated with your profile. Click here to edit your per sonal information. Site Name Technology ZZZZ Review Site Survey Edit Site Specifics ZZZZ Review Site Survey Chemical Oxidation ZZZZ Review Site Survey Chemical Oxidation ZZZZ Review Site Survey Chemical Oxidation ZERNENSITE Other Site Specifics Chemical Oxidation DEFINE NEW SITE Ochristian Survey Consultant OFFINE NEW SITE Scopyright 2002 - GeoSyntec Consultants Scopyright 2002 - GeoSyntec Consultants	
tec - Microsoft Internet E View Favorites Tools	GEOSYNTEC	Welcome a b! Below are a list of sites association is the state association is the site has a state as a state	
Geosynt File Edit	CONS	Welcome Below are a Site Name TITT	

Geosyntec - Microsoft Internet E File Edit View Favorites Tools	xplorer provided by GeoSyntec Consultants Help	× 🛛 -
GEOSYNTEC CONSULTANTS	User: a b Email: zzzz Site: zzzz Site: zzzz Technology: Chemical Oxidation - Permanganate	•
Technology Selection What were your reasons for ch	Technology Selection What were your reasons for choosing this technology (select all that apply)?	
Cost of technology application	tion	
Technology perceived to proprior	Technology perceived to provide a good chance at remedial success	
Remediation timeframe was	as a factor (e.g., wanted a technology that was more likely to provide remediation in a shorter timeframe)	
Regulatory pressures guided	ed technology choice	
Impact of remedial technolo	Impact of remedial technology on surface water was minimal	
Impacts of remedial techno	Impacts of remedial technology on secondary water quality was minimal	
🔲 Technology application was	Technology application was perceived to reduce risk to the environment	
Technology application was	Technology application was thought to have a lasting impact on the water quality	
Health and Safety concerns guided technology choice	s guided technology choice	F
Technical implementability	Technical implementability of the remedial technology was thought to be appropriate	
 Legal pressures guided technology choice 	chnology choice	
Pressure from various stake	ke holders (e.g., site owners, past and current, regulators, community, etc.) guided technology choice	
Other:		_
	© Copyright 2002 - GeoSyntec Consultants	Þ
Slide 21 of 21	Default Design	

×	-	1				Þ
by GeoSyntec Consultants		User: a b Email: zzzz Site: zzzz Site: zzzz Technology: Chemical Oxidation - Permanganate		 C Pilot test remedial activities ongoing C Pilot test remedial activities completed C Pilot test remedial activities completed, post-pilot monitoring on-going C Full-scale remedial activities completed C Full-scale remedial activities completed, post-treatment monitoring on-going C Full-scale remedial activities completed, post-treatment monitoring on-going 	CK NEXT >	© Copyright 2002 - GeoSyntec Consultants
🚰 Geosyntec - Microsoft Internet Explorer provided b	Help	User: a b Email: <u>zzzz</u> Site: <u>zzzz</u> echnology: Cherr	ediation?	ing pleted pleted, post-pilot r going npleted npleted, post-treat	■ BACK	Ö ©
ernet Ex	Tools	F	the remo	ities ongo ities com vities con- vities on- vities con-		
crosoft Int	Favorites	IEC	Remediation Stage Currently, at what stage is the remediation?	 C Pilot test remedial activities ongoing C Pilot test remedial activities completed C Pilot test remedial activities completed, C Full-scale remedial activities on-going C Full-scale remedial activities completed C Full-scale remedial activities completed 		
ec - Mi	View	GEOSYNTEC	liation , at wh	est rem est rem est rem cale rei cale rei cale rei		
eosynt	Edit	GEC	emec	Pilott Pilott Pilott Full-s Full-s Full-s Full-s		
	File	0.0	۲ ک			

arce and Monitoring In technology performance (check all that apply)? Intoring wells Is (MLCs) in monitoring wells Is (MLCs) in monitoring wells Is (MLCs) in monitoring wells Is (MLCs) in monitoring wells Is (Inclained and Table)? Is (Inclained and how long have they been on-going to date? In observed to date (if applicable)?	💁 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	×
	Help	1641
In technology performance (check all that apply)? Intorting wells Is (MLCs) in monitoring wells as discharge) as discharge) It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date? It mathement activities and how long have they been on-going to date?	Methods of Assessing Performance and Monitoring	•
onitoring wells Is (MLCs) in monitoring wells ss discharge) state: To Date: To Date:	How are you planning to assess the interim technology performance (check all that apply)?	
Is (MLCs) in monitoring wells ss discharge) It reatment activities and how long have they been on-going to date?	Concentration reduction in specific monitoring wells	
Is (MLCs) in monitoring wells ss discharge) It reatment activities and how long have they been on-going to date?		
ss discharge)	Achieve maximum contamination levels (MLCs) in monitoring wells	
treatment activities and how long have they been on-going to date?	Reduction in plume mass flux (or mass discharge)	
I treatment activities and how long have they been on-going to date? vs vs vs vs vs vs in observed to date (if applicable)?		
I treatment activities and how long have they been on-going to date?	Production of degradation by-products	
It reatment activities and how long have they been on-going to date?		
I treatment activities and how long have they been on-going to date?		
I treatment activities and how long have they been on-going to date?		
days	What is the estimated duration of the field treatment activities and how long have they been on-going to date?	
en observed to date (if applicable)?	To Date: days	
	How much decrease in mass flux has been observed to date (if applicable)? 81-100% 💌	
	 	•
)/ NEXI	

	G
VIEW FAVORICES LOOIS MEID	A second s
What was the technology performance measurement based on (check all that apply)?	•
Concentration reduction in specific monitoring wells	
Reduction in soil concentrations	
Achieved maximum contamination levels (MLCs) in monitoring wells	
Reduction in plume mass flux (or mass discharge)	
Reduction in plume size	
Production of degradation by-products	
Total Mass removed	
Mass remaining	
Will not measure	
What was the duration of the field treatment activities?	
days 🔰	
Was the technology considered successful?	
How much decrease in mass flux was observed (if applicable)?	
Unknown <	
Currently at what stage is the post-treatment monitoring?	
Not Conducted 🕶	F

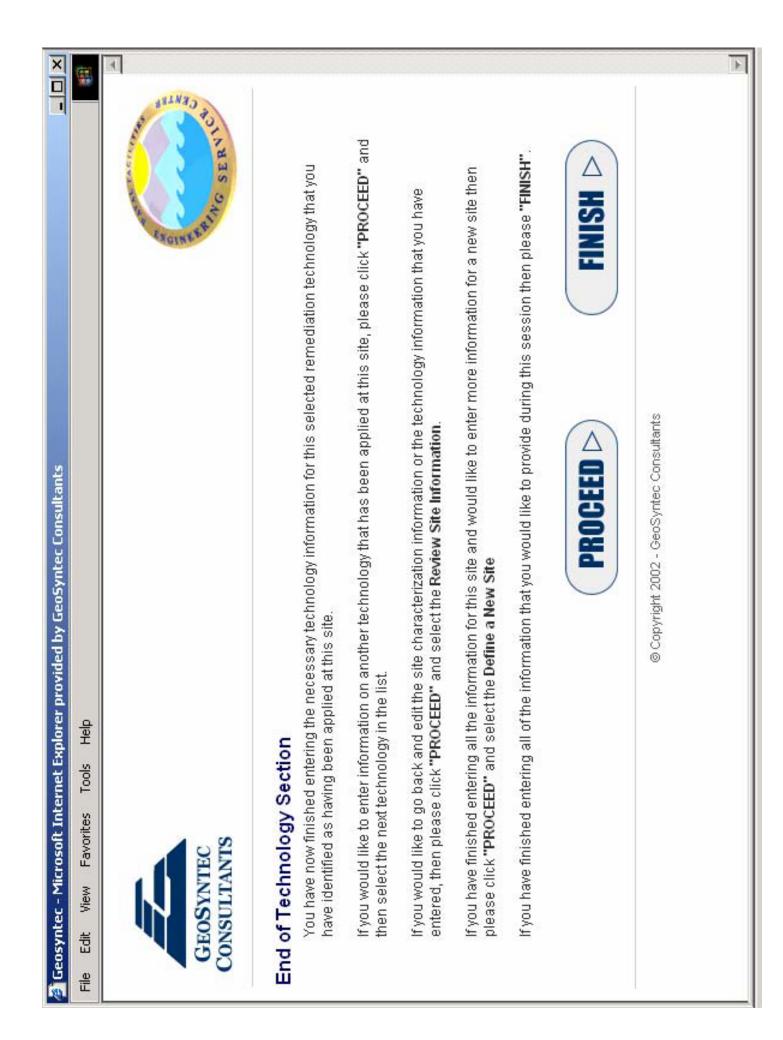
×	giner	•				
🏄 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	orites Tools Help	Was the technology considered successful? Not yet evaluated ▼ How much decrease in mass flux was observed (if applicable)?	ti it	Did post-treatment monitoring indicate that an increase in concentrations (rebound) of dissolved chlorinated solvents was occuring? Not Applicable	© Copyright 2002 - GeoSyntec Consultants	
dicrosoft Int	/ Favorites	iology consi Lated ◀	ation was th	nent monito	► P	
osyntec - N	Edit View	Was the technology co Not yet evaluated 🗸 How much decrease i Unknown 💙	Currently at what st Not Conducted Over what duration	Did post-treatment occuring? Not Applicable	Not Estimated	
ē) Ge	File	P P S		Nha Did	Z	

	С.	•																		
																			•	
		it annivi?	- fadda w																	
ultants		(check all tha						0:												
by GeoSyntec Consultants		en observed						olease specify												
		What impacts on secondary groundwater guality have been observed (check all that apph/?						Elevation in concentration of other dissolved metals (please specify):												
xplorer p	Help	ndwater (ation	other diss				tivity								
ernet E	Tools		2		eration	ation	e gener	ition of o				conduct	anges	solids	٦L	ame	lume			
🚰 Geosyntec - Microsoft Internet Explorer provided	Favorites	i seconda		neration	Hydrogen sulfide generation	Dissolved iron generation	Dissolved manganese generation	concentra	Increased BOD levels	6	Reduction of porosity	Changes in hydraulic conductivity	Groundwater color changes	Increased dissolved solids	Redistribution of DNAPL	Increased source volume	Decreased source volume	Other (please specify):		
ntec - Mi	t View	nnacts of	0.000	Methane generation	/drogen s	ssolved i	ssolved r	evation in	creased E	pH changes	eduction (nanges in	oundwati	creased c	edistributi	creased s	ecreased	her (plea:		
🖨 Geosy	File Edit	What in	l	ž	f D	D			<u>ء</u> _	± □	Ē	ō	õ	ء ا	æ L	ء ا	õ D	ō		

×□	-	•									F
		BALLNAD HOLLARD	zones (check all that apply)?			4					
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	Edit View Favorites Tools Help	GEOSYNTEC GEOSYNTEC CONSULTANTS Technology: Chemical Oxidation - Permanganate	Effectiveness of Technology In general, how would you rate the effectiveness of this technology in remediating DNAPL source zones (check all that apply)?	Able to achieve treatment goals at a reasonable cost	Technology is limited to certain site conditions (i.e. depth of contaminant, geology) Specify:		Technology requires further development before it can be routinely applied	Reasonable ease of implementation	Reasonable ease of use	Reasonable ease of design	Adequate number of trained vendors to implement technology
	File		Ш								L

e it can be routinely applied ent technology Iness Iness Iness Iness	(ite	œ	1000	•							F
by GeoSyntec Consultants e it can be routinely applied ent technology iness intert to apply this particular technology. Itant to apply this particular technology.			**								
c - Microsoft Jr View Favorite: nology requires onable ease of ate number of factors that mij y: id recommend	ew Favorites Tools Help				Reasonable ease of use	Reasonable ease of design	Adequate number of trained vendors to implement technology	Other factors that might affect technology usefulness Specify:	I would recommend the following vendor/consultant to apply this particular technology:	BACK NEXT	© Copyright 2002 - GeoSyntec Consultants
Edit v Techno Reasol Adequa Specify:	Edit	Edit	Edit	Techi Reas	Reas	Reas	Adeq	Other Specif	Iwon		
	e Li	Eile	File								

×	-	A	Þ	1
		EXCINE ENDER	an but not site	
			BAGK the diation but BAGK S BAGK	
			tc.) 0-10% (c) 0-10\% (c) 0	3
		e	ementar plans, e ttants	
聋 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants		User: a b Email: <u>zzzz</u> Site: <u>zzzz</u> Technology: Chemical Oxidation - Permanganate	Cost Breakdown of Technology What was the total cost of remediation for the site including the design and implementation of remediation but not site characterization? Imaching the site including the design and implementation of remediation but not site characterization? Imaching the site including the design and implementation of remediation but not site characterization? Imaching the site including the design and implementation of remediation but not site characterization? Imaching the site including the design and implementation of remediation but not site characterization? Imaching the site including the instrumentation? Image: Set is electron donor/acceptors, oxidant, iron? Image: Set is electron donor/acceptors, oxidant, iron? Image: Treatment operation (annual costs for operator, analytical, reporting, etc.) Image: Dost-treatment monitoring (sampling, analytical, reporting, etc.)	Default Design
xplorer	Help	E	nolog iation fo s, piping s for ope ing, ana	
ternet E	Tools		f Techr I remedia USD vells, on donord Lal costs	
osoft Int	Favorites	EC	Cost Breakdown of Technology What was the total cost of remediation for characterization? Professional costs (design, construction su Treatment infrastructure (e.g. wells, piping, Consumables (e.g. electron donor/acceptor Treatment operation (annual costs for opera Post-treatment monitoring (sampling, analy	Slide 28 of 28
ec - Micr	View	GEOSYNTEC	Cost Breakdo What was the tota Characterization? Professional cost Treatment infrastr Treatment operati Post-treatment m	Slide
eosynte	Edit	GEO	ost BI hat was rofessio reatmen reatmen onsums ost-treat	
	File		U 35L - FOF-	



N	œ	•							F
🚰 Geosyntec - Microsoft Internet Explorer provided by GeoSyntec Consultants	ools Help		k You! We would like to thank you for your time and effort in answering this survey and hope that you will find the Review of DNAPL Source Zone Remediation Technologies Report useful.	A link to the Adobe pdf version of the report will be provided at <u>https://projects.geosyntec.com/NAVY_ROCS/</u> when the report is completed in Fall 2003. All survey participants will be emailed a reminder of the location of the final report when the report has been posted.	If you would like to access the survey at another time within the next 30 days, you may go to <u>https://projects.geosyntec.com/NAVY_ROCS/login.asp</u> and reenter your email address and password. This will allow you to access any of the information that you have already entered, as well as enter new data on a new site or new technology applied at one of your previous sites.	tions or Comments Please feel free to contact us about the survey if:	you have any further information to add that we did not address; you have any other key issues that you feel people should know about this technology that are not addressed; and/or you have any comments about or questions.	, NFESC Julie Konzuk, GeoSyntec Consultants Suzanne O'Hara, GeoSyntec Consultants Engineer Environmental Engineer Hydrogeologist 1616 Tel. (519) 822-2230 ext. 241 Tel. (519) 822-2230 ext. 234 co.navy.mil survey@geosyntec.com	© Copyright 2002 - GeoSyntec Consultants
crosoft Internet Explor	Favorites Tools Help	ey	d like to thank you for you one Remediation Techn	A link to the Adobe pdf version of report is completed in Fall 2003. the report has been posted.	If you would like to access the survey <u>https://projects.geosyntec.com/NAVY</u> to access any of the information that applied at one of your previous sites.	Questions or Comments Please feel free to contact us abo	you have any further information to ad you have any other key issues that you and/or you have any comments about or que	Carmen Lebrón, NFESC Environmental Engineer Tel. (805) 982-1616 Iebronca@nfesc.navy.mil	
🎒 Geosyntec - Mid	File Edit View	End of Survey	Thank You! We would Source Zo	A link to th report is o the report	If you wou <u>https://prc</u> to access applied a:	Questions o Please fe	• • •		

APPENDIX C

Detailed Survey Data Summary

APPENDIX C TABLE OF CONTENTS

APP	ENDIX C: DETAILED SURVEY DATA SUMMARY 1
1	AQUIFER GEOLOGY/LITHOLOGY 2
2	HYDROGEOLOGY
3	DNAPL SOURCE ZONE CHARACTERISTICS 11
4	REMEDIAL TECHNOLOGIES
5	COST AND TREATMENT DURATION DATA 28
6	METHODS OF ASSESSING PERFORMANCE AND MONITORING 32
7	TREATMENT EFFECTIVENESS
8	DATA GAPS
9	SURVEY PARTICIPANT LIST 40

APPENDIX C: DETAILED SURVEY DATA SUMMARY

This Appendix presents a detailed summary of the survey questions and the data collected. Appendix D presents the interpretation of the data in terms of the impact of various site factors and technology choice on the remedial cost, duration and success. Tables are embedded within the text below. Figures may be found at the end of the appendix.

The following caveats are implicit in the data summary provided below:

- *i)* The data is summarized separately for sites where the remedial technology was specified *versus* sites with unspecified technologies. The data for sites with unspecified technologies are presented separately because it is unclear whether these surveys were incomplete due to a lack of respondent interest, or to inappropriateness of data (*e.g.*, remedial activities targeted dissolved phase, as opposed to DNAPL).
- *ii)* Variations in the number of total data available for each survey question arise due to the following:
 - one survey respondent can enter data for multiple sites;
 - one individual site location can have data for multiple technology applications; and
 - few survey questions required answers before allowing the respondent to proceed forward through the survey, resulting in variable totals per question.
- *iii)* For ease of discussion, the term "site" is used here to refer to a unique combination of site location and technology application. One site location may have multiple technology applications, either used sequentially or targeting different portions of the DNAPL source area(s); however, each technology application is treated as a separate site.
- *iv)* Given the limited amount of available data on field applications of DNAPL remediation, the data collected may not be a statistically significant representation of all field applications.

1 Aquifer Geology/Lithology

Survey respondents were asked to specify the following:

- *i)* Type of geologic material present in the DNAPL source area (*i.e.*, unconsolidated versus consolidated media);
- *ii)* Classify the geological material by degree of fracturing and rock matrix permeability (consolidated material only) or degree of heterogeneity (unconsolidated material only); and
- *iii)* Predominant lithology of the aquifer media within the DNAPL source zone.

See below for a summary of the data collected.

Q: Enter the relative percentage of aquifer media type.

Survey respondents were asked to provide information on all types of geologic media present at their site (*i.e.*, consolidated versus unconsolidated), and to categorize the geologic media by the degree of fracturing and rock matrix permeability (consolidated media only) or degree of heterogeneity (unconsolidated media only). The respondent was asked to specify what relative portion (*i.e.*, percentage) of their site contained media with that description; more than one media type/description could be chosen for each site. Table C.1 provides a summary of the data that was entered into the survey database, broken down by sites with a remedial technology specified by the survey respondent and those without where the survey respondent did not complete the survey. Table D.17 in Appendix D provides a more detailed breakdown of the data, including the range of percentages and subdivided by technology (see Appendix D). In total, data was entered for 152 sites, of which 54 sites had both consolidated and unconsolidated media, 8 sites had only consolidated media, and 90 sites had only unconsolidated media.

Aquifer Media Type	Media Description	Descriptive Parameter	Sites with Technology Specified	Sites with Unspecified Technology	Total Sites*
		Discretely Fractured	24	5	29 (46.8%)
	Degree of Fracturing	Highly Fractured	20	4	24 (38.7%)
ated	P	Unfractured	22	9	31 (50.0%)
Consolidated		TOTAL SITES	48	14	62
Cons		Low	28	5	33 (53.2%)
Ŭ	Rock Matrix	High	25	11	36 (58.1%)
	Permeability	Unknown	4	2	6 (9.7%)
		TOTAL SITES	48	14	62
ed		Low	47	18	65 (45.1%)
Unconsolidated		High	92	24	116 (80.6%)
	Heterogeneity	Unknown	14	0	14 (9.7%)
Unc		TOTAL SITES	127	17	144

Table C.1: Summary of the breakdown of the aquifer type.

* - respondents were asked to provide information on all types of geologic media at the site; for sites with multiple geologic media present, it is possible to have answers that add to up to > 100%.

From Table C.1, it is evident that for the sites with consolidated material, there were approximately equal numbers of sites with discretely fractured and highly fractured bedrock, with equally distributed low and high rock matrix permeabilities. For the majority of the sites with unconsolidated material, the soil was highly heterogeneous.

Q: Choose one rock/soil type that best describes the predominant geologic material in the source area.

The lithology choices that were available are listed in Table C.2 below, along with a summary of the number of sites with each lithology where DNAPL remediation was attempted. In total, lithology was specified for 144 sites, of which 117 of these sites had a technology associated with them and the remaining 27 sites were from incomplete survey entries where no technology was specified. Of the 144 total sites, only 16 (11.1%) of the sites had predominantly consolidated media in the source area. The majority (128

or 88.9%) of the sites were located in unconsolidated media. Figure C.1 illustrates the breakdown of the sites that had a technology specified into consolidated (13 of the 117 sites) and unconsolidated (104 of the 117 sites) media, as well as the distribution of the subcategories within each. Table C.2 below and Table D.18 in Appendix D provide more detailed breakdowns of the data, with Table D.18 including the range of percentages and subdivided by technology (see Appendix D).

Of the 16 total consolidated media sites, the majority of the DNAPL remediation was attempted at sites having sedimentary rock (mostly shale, some karstic and non-karstic limestone) as the most prevalent geologic material in the source area (Table C.2). Conversely, DNAPL remediation was attempted at only one site with igneous (granitic) bedrock.

Of the 128 sites where remediation was undertaken in unconsolidated material, the majority of the sites had sand as the most prevalent geologic material in the source area, silt and clay being the next most prevalent (Table C.2). Only a few sites had gravel, till or unknown predominant soil types.

Aquifer Media Type	Lithology		Sites with Unspecified Technology	Sites with Specified Technology	Total Sites
		Metabasalt	0	1	1 (50.0%)
	Metamorphic	Metashale	0	0	0 (0.0%)
		Other	0	1	1 (50.0%)
		TOTAL	0	2	2 (12.5%)
		Shale	1	5	6 (54.5%)
		Sandstone	0	0	0 (0.0%)
-	Sedimentary	Limestone (non-karstic)	1	0	1 (9.1%)
Consolidated	Sedimentary	Limestone (karstic)	0	3	3 (27.3%)
osu		Other	0	1	1 (9.1%)
CO		TOTAL	2	9	11 (68.8%)
		Granitic	0	1	1 (100%)
		Basaltic	0	0	0 (0.0%)
	Igneous	Intermediate	0	0	0 (0.0%)
		Other	0	0	0 (0.0%)
		TOTAL	0	1	1 (6.3%)
	Unknown	TOTAL	1	1	2 (12.5%)
	TO	ΓAL	3	13	16 (11.1%)
	Gra	ivel	0	8	8 (6.3%)
g	Sa	nd	12	45	57 (44.5%)
lated	Till		1	7	8 (6.3%)
solic	Silt Clay Unknown TOTAL		7	22	29 (22.7%)
con			4	18	22 (17.2%)
Un			0	4	4 (3.1%)
			24	104	128 (88.9%)

Table C.2: Summary of the breakdown of the predominant geologic material in the source area.

2 Hydrogeology

Survey respondents were asked to enter data describing the following hydrogeological parameters:

- *i)* Hydraulic conductivity;
- *ii)* Groundwater velocity;
- *iii)* Sustainable well yield;
- *iv)* Depth to groundwater; and
- *v)* Horizontal and vertical hydraulic gradients.

Information for 137 sites was entered into the survey database; of these, 118 sites had a remedial technology specified while 19 did not.

Q: Choose the hydrogeological parameters that best describe the BULK conditions within the DNAPL source area.

Table C.3 summarizes the hydraulic conductivity data entered into the survey database. Survey respondents were asked to enter hydraulic conductivity data as order of magnitude ranges; the data have been grouped into three categories representing low, medium and high hydraulic conductivities for ease of discussion and analysis. Overall, of the 137 sites in the survey database with hydraulic conductivity data, 8 (5.8%) sites have hydraulic conductivities of less than 10^{-5} ft/day, 49 (35.8%) sites have hydraulic conductivities between 10^{-5} ft/day and 10 ft/day, and 23 (16.7%) sites have high hydraulic conductivities (greater than 10 ft/day). It is interesting to note that of the 137 responses to this question, 57 (41.6%) of the sites had an unknown hydraulic conductivity, a key parameter when evaluating site data.

Hydraulic Conductivity	•		Total Sites
< 10 ⁻⁵ ft/day	0	8	8 (5.8%)
10^{-5} to 10 ft/day	9	40	49 (35.8%)
> 10 ft/day	2	21	23 (16.7%)
Unknown	8	49	57 (41.6%)
TOTAL	19	118	137

Table C.3: Summary of the breakdown of hydraulic conductivity data.

Table C.4 provides a summary of the groundwater velocity data entered into the survey database. Survey respondents were asked to choose from order of magnitude ranges; the data have been grouped into three ranges representing low, medium and high groundwater velocities for ease of analysis and discussion. Overall, of the 137 sites in the survey database with groundwater velocity data, 3 (2.2%) sites have low groundwater velocities (< 10^{-3} ft/day), 40 (29.2%) sites have groundwater velocities (> 1 ft/day). Again, of the 118 responses, 79 (57.7%) of the sites had unknown groundwater velocities.

Groundwater Velocity	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
< 0.001 ft/day	0	3	3 (2.2%)
0.001 to 1 ft/day	7	33	40 (29.2%)
> 1 ft/day	1	14	15 (10.9%)
Unknown	11	68	79 (57.7%)
TOTAL	19	118	137

Table C.4: Summary of the breakdown of groundwater velocity data.

Table C.5 presents a summary of the sustainable well yield data collected from the survey. The survey respondents were asked to enter a numerical value for this site characteristic; the data in Table C.5 has been grouped into order of magnitude ranges for ease of presentation. Overall, 49 sites had well yield data entered into the survey, while 88 sites did not have well yield data specified (*i.e.*, answer was left blank if unknown or

unanswered). Of the 49 sites in the survey database with sustainable well yield data, DNAPL remediation has been attempted at only a few (4 or 8.2%) sites with low (≤ 1 gallon per minute) sustainable well yields. The majority of the sites where DNAPL remediation has been attempted have sustainable well yields between 1 and 250 gpm. There was one site where the sustainable well yield was greater than 1,000 gpm.

Sustainable Well Yield	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
< 1 gpm	0	4	4 (2.9%)
1 to 10 gpm	3	12	15 (10.9%)
10 to 100 gpm	2	17	19 (13.9%)
100 to 1,000 gpm	0	10	10 (7.3%)
> 1,000 gpm	0	1	1 (0.8%)
Unspecified	14	74	88 (64.2%)
TOTAL	19	118	137

Table C.5: Summary of the breakdown of sustainable well yield data.

Table C.6 presents a summary of the depth to groundwater data collected from the survey. The survey respondents were asked to enter a numerical value for this site characteristic; the data in Table C.5 has been grouped into ranges for ease of presentation. Overall, 106 sites had depth to groundwater data entered into the survey, while 31 sites did not have groundwater depth specified (*i.e.*, answer was left blank if unknown or unanswered). Of the 106 sites in the survey database with depth to groundwater data, the majority of the sites where DNAPL remediation has been attempted have watertable depths less than 50 ft below ground surface (bgs). There was one site where the depth to groundwater was greater than 200 ft bgs.

Depth to Groundwater	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
< 10 ft bgs	8	42	50 (36.5%)
10 to 20 ft bgs	1	28	29 (21.2%)
20 to 50 ft bgs	2	16	18 (13.1%)
50 to 100 ft bgs	2	3	5 (3.6%)
100 to 200 ft bgs	1	2	3 (2.2%)
> 200 ft bgs	0	1	1 (0.7%)
Unspecified	5	26	31 (22.6%)
TOTAL	19	118	137

Table C.6: Summary of the breakdown of depth to groundwater data.

Table C.7 provides a summary of the horizontal and vertical hydraulic gradient data entered into the survey database. Survey respondents were asked to choose from order of magnitude ranges, which are listed in Table C.7. Overall, of the 137 sites in the survey database with hydraulic gradient data, the majority (37 or 27%) had horizontal gradients within the range of 0.001 to 0.1 ft/ft. A significant number (19 or 14.8%) had flatter gradients (*i.e.*, < 0.001 ft/ft). The majority of the vertical gradients were small, indicating a predominance of horizontal groundwater flow at most DNAPL impacted sites surveyed. Of the 137 sites, a large proportion had unknown hydraulic gradients, with slightly more sites having known horizontal rather than vertical hydraulic gradients.

Direction of Gradient	Hydraulic Gradient	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
	< 0.0001	1	3	4 (2.9%)
	0.0001 to 0.001	4	11	15 (10.9%)
	0.001 to 0.01	2	20	22 (16.1%)
Horizontal	0.01 to 0.1	1	14	15 (10.9%)
	0.1 to 1	0	3	3 (2.2%)
	Unknown	11	67	78 (56.9%)
	TOTAL	19	118	137
	< 0.0001	0	7	7 (5.1%)
	0.0001 to 0.001	1	18	19 (13.9%)
	0.001 to 0.01	0	0	0 (0.0%)
Vertical	0.01 to 0.1	1	1	2 (1.5%)
	0.1 to 1	0	0	0 (0.0%)
	Unknown	17	92	109 (79.6%)
	TOTAL	19	118	137

Table C.7: Summary of the breakdown of hydraulic gradient data.

3 DNAPL Source Zone Characteristics

A series of questions were asked in the survey pertaining directly to the DNAPL source zone. These included the following:

- *i)* Source area chemistry (*i.e.*, DNAPL and dissolved phase contaminant profiles, other groundwater geochemistry);
- *ii)* Size parameters of the DNAPL impacted zone (*i.e.*, areal extent, volume, maximum depth of the DNAPL, DNAPL distribution and shape); and
- *iii)* DNAPL mass estimation (*i.e.*, methods of determining presence and mass, and an estimation of pre-remediation DNAPL mass).

See below for a summary of the data.

Q: Estimate the percentage of each type of chlorinated solvents that best describes the contaminant at your site if a sample of DNAPL has been analyzed.

Table C.8 provides a summary of the breakdown of the DNAPL composition data entered into the survey. Figure C.2 illustrates the breakdown of the DNAPL composition for sites that had an associated technology. Since some of the sites have multicomponent groundwater contaminants, the total percentage can be greater than 100%. The majority (68%) of the sites were impacted with primarily chlorinated ethenes; a smaller percentage had mixed DNAPL compositions (29%). Four sites did not have chlorinated ethenes, of which three of the sites had DNAPL consisting solely of chloroethanes and one site had both chloromethanes and chlorobenzenes.

DNAPL Component	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
Chloroethenes (e.g., PCE, TCE)	6 (100.0%)	71 (94.7%)	77 (95.1%)
Chloroethanes (<i>e.g.</i> , 1,2-DCA, 1,1,2,2-PCA)	2 (33.3%)	23 (30.7%)	26 (32.1%)
Chloromethanes (e.g., DCM)	2 (33.3%)	12 (16.0%)	14 (17.3%)
Chlorobenzenes (e.g., TCB)	1 (16.7%)	7 (9.3%)	8 (9.9%)
Total Sites with DNAPL Composition Data	6	75	81

Table C.8:	Summary of DNAPL Composition
	Summary of Divisi L composition

Q: Estimate the MAXIMUM dissolved phase concentration of each chlorinated solvent that occurs within the defined source.

The maximum dissolved phase data was used to determine the overall distribution of groundwater contaminants grouped in terms of chloroethenes, chloroethanes, chloroethanes, and chlorobenzenes, as well as the maximum, median and average concentration of the dissolved contaminants. Table C.9 and Figure C.3 provide a summary of the distribution of groundwater chemistry data. Since some of the sites have multicomponent groundwater contaminants, the total percentage can be greater than 100%. As expected, the groundwater contaminant data is similar to the breakdown of the DNAPL composition data. Based on these numbers it can be seen that the majority (96%) of the sites were impacted with chlorinated ethenes and that a smaller percentage had mixed groundwater contaminant plumes.

Dissolved Phase Component	Total Sites With Unspecified Technology	Total Sites With Technology Specified	Total Sites
Chloroethenes (e.g., PCE, TCE)	10 (90.9%)	66 (97.0%)	76 (96.2%)
Chloroethanes (<i>e.g.</i> , 1,2-DCA, 1,1,2,2-PCA)	4 (36.4%)	19 (28.0%)	23 (29.1%)
Chloromethanes (e.g., DCM)	2 (18.2%)	10 (15.0%)	12 (15.2%)
Chlorobenzenes (e.g., TCB)	1 (9.1%)	6 (9.0%)	7 (8.9%)
Total Sites with Groundwater Contamination Data	11	68	79

Table C.9: Summary of Dissolved Phase Composition

Table C.10 lists the main groundwater contaminant concentrations entered into the survey. Contaminant concentrations range from parts per billion (ppb) to the single-component solubility (S) of the majority of the contaminants. The single-component solubility represents the maximum theoretical achievable dissolved concentration of the contaminant. The following conclusions can be drawn from this data:

i) The higher chlorinated solvents (*e.g.*, PCE, TCE, 1,1,1,2-PCA, 1,1,1-TCA, DCM) were present at a larger proportion to their theoretical concentrations. The average concentration of these contaminants was >14% of the solubility, with the maximum concentration of each >15%S.

ii) The lower chlorinated solvents (*e.g.*, cis-1,2-DCE, 1,1-DCE, CA, CM), with the exception of 1,2-DCA, were present at a much lower percentage of the theoretical maximum, with average concentrations <12%S, and maximum concentrations <13%S.

Contami	inant	Minimum (mg/L)	Maximum (mg/L)	Median (mg/L)	Average (mg/L)	Solubility (mg/L)
	PCE	0.1 (<0.1%)*	220 (93%)	44 (19%)	63 (27%)	237 ^a
	TCE	1 (<0.1%)	1400 (101%)	100 (7%)	268 (29%)	1385 ^a
Chloroethenes	1,1-DCE	0.03 (<0.1%)	50 (2%)	25 (1%)	25 (1%)	2250 ^b
	cis-1,2- DCE	0.10 (<0.1%)	940 (13%)	9 (0.1%)	126 (2%)	6996 ^a
	1,1,1,2- PCA	150 (14%)	160 (15%)	155 (14%)	155 (14%)	1100 ^e
	1,1,1-TCA	2.1 (0.2%)	2000 (154%)	59 (5%)	529 (41%)	1300 ^a
Chloroethanes	1,1,2-TCA	0.054 (<0.1%)	50 (1%)	25 (0.6%)	25 (0.6%)	4400 ^a
	1,2-DCA	10 (0.1%)	3140 (37%)	10 (0.1%)	1053 (12%)	8500 ^a
	CA	0.01 (<0.1%)	18 (0.3%)	18 (0.3%)	12 (0.2%)	5710 ^c
Chloromethanes	DCM	10 (<0.1%)	12000 (60%)	1000 (5%)	5027 (25%)	20000 ^a
Chloromethanes	СМ	1.8 (<0.1%)	130 (0.2%)	100 (0.2%)	83 (0.2%)	53000 ^f
	1,2,3-TCB	0.2 (<0.1%)	0.2 (<0.1%)	0.2 (<0.1%)	0.2 (<0.1%)	18 ^d
	1,2,4-TCB	4.78 (15%)	50 (160%)	27 (86%)	27 (86%)	31.3 ^d
Chlorobenzenes	1,2-DCB	3.33 (2%)	1000 (640%)	502 (322%)	502 (322%)	156 ^b
	1,3-DCB	1 (0.8%)	100 (75%)	51 (38%)	51 (38%)	133 ^d
	СВ	200 (40%)	200 (40%)	200 (40%)	200 (40%)	500 ^a

Table C.10: Summary of Maximum Groundwater Contaminant Concentrations

Single-Component Solubility values from:

a Pankow and Cherry (1996)

b Environmental Quality Management, 1998

c Mackay and Shiu (1977)

d Banerjee (1984)

e Howard (1990)

f http://chemfinder.cambridgesoft.com/ (visited on Feb 19, 2004)

* Percentages shown are the percentage of the single-component solubility, which represents the maximum theoretical achievable dissolved-phase concentration.

Q: Choose applicable concentration ranges from the following list of groundwater geochemistry parameters that may be relevant in terms of technology effectiveness.

Table D.26 (in Appendix D) summarizes the groundwater geochemistry data collected with the survey, ranging from inorganic parameters (*e.g.*, nitrate, sulfate); metals (*e.g.*, total and dissolved iron); field parameters (*e.g.*, pH, ORP, DO); organic parameters (*e.g.*, f_{oc} , BOD, DOC); and dissolved hydrocarbon gases (methane, ethene, and ethane).

Q: Provide the following information on the DNAPL distribution and mass prior to the onset of remedial activities. For sites with multiple DNAPL source zones, please provide the TOTAL value for all source zones (e.g., areal extent, volume, mass etc.) as appropriate.

Survey respondents were asked to specify the following information:

- *i)* Total areal extent of DNAPL distribution;
- *ii)* Total volume of impacted soil (pre-remediation) containing DNAPL;
- *iii)* Maximum depth of the DNAPL;
- *iv)* Distribution of DNAPL throughout the subsurface (*e.g.*, pools, residual, *etc.*); and
- *v)* Estimated shape of the DNAPL distribution used in determining treatment geometry.

Table C.11 summarizes the distribution of responses to the areal extent of DNAPL. Of the 123 sites with responses to this question, 48 (39%) said that the areal extent was unknown. Although the responses ranged from 0.001 ft² to over 1,000,000 ft², the majority (31%) of the sites had a DNAPL areal extent of 10,000 to 100,000 ft². The accuracy of the two sites with areas less than 1 ft² could not be independently verified. Sites with areal extents less than 100 ft² were generally technology demonstrations.

Areal Extent of DNAPL Impacted Zone	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
0.001 to 0.01 ft ²	0 (0.0%)	1 (0.8%) ^a	1 (0.8%)
$0.01 \text{ to } 0.1 \text{ ft}^2$	0 (0.0%)	1 (0.8%) ^a	1 (0.8%)
1 to 10 ft^2	0 (0.0%)	4 (3.4%)	4 (3.3%)
10 to 100 ft ²	0 (0.0%)	5 (4.2%)	5 (4.1%)
100 to 1,000 ft ²	0 (0.0%)	11 (9.3%)	11 (8.9%)
1,000 to 10,000 ft ²	0 (0.0%)	10 (8.5%)	10 (8.1%)
10,000 to 100,000 ft ²	0 (0.0%)	23 (19.5%)	23 (18.7%)
100,000 to 1,000,000 ft ²	0 (0.0%)	11 (9.3%)	11 (8.9%)
> 1,000,000 ft ²	1 (20.0%)	8 (6.8%)	9 (7.3%)
Unknown	4 (80.0%)	44 (37.3%)	48 (39.0%)
Total Number of Sites	5	118	123

Table C.11: Summary of Areal Extent of DNAPL Impacted Zone

^aThese values were inputted into the survey; their accuracy is unknown.

Table C.12 summarizes the distribution of responses to the volume of the DNAPL impacted zone. Of the 123 sites with responses to this question, 55 (46.6%) said that the volume was unknown. Although the responses ranged from 1 ft³ to over 1,000,000 ft³, the majority of the sites had a DNAPL impacted volume of greater than 100,000 ft³.

Volume of DNAPL Impacted Zone	Number of Sites With Unspecified Technology	Number of Sites With Specified Technology	Total Sites
1 to 10 ft^3	0 (0.0%)	3 (2.5%)	3 (2.4%)
$10 \text{ to } 100 \text{ ft}^3$	0 (0.0%)	5 (4.2%)	5 (4.1%)
100 to 1,000 ft ³	0 (0.0%)	2 (1.7%)	2 (1.6%)
1,000 to 10,000 ft ³	0 (0.0%)	5 (4.2%)	5 (4.1%)
10,000 to 100,000 ft ³	0 (0.0%)	6 (5.1%)	6 (5.0%)
100,000 to 1,000,000 ft ³	0 (0.0%)	24 (20.3%)	24 (20.2%)
> 1,000,000 ft ³	0 (0.0%)	18 (15.3%)	18 (15.2%)
Unknown	5 (100.0%)	55 (46.6%)	60 (48.8%)
Total Number of Sites	5	118	123

Table C.12: Summary of Volume of DNAPL Impacted Zone

Table C.13 summarizes the distribution of responses to the maximum depth of DNAPL. Of the 118 sites with responses to this question, 43 (36%) said that the maximum depth was unknown. Although the responses ranged from 1 ft to 10,000 ft below ground surface (bgs), the majority of the sites had a DNAPL depth between 10 and 100 ft bgs.

Table C.13: Summary of Maximum DNAPL Depth

Maximum Depth of DNAPL (bgs)	Number of Sites With Unspecified Technology	Number of Sites With Specified Technology	Total Sites
1 to 10 ft	0 (0.0%)	9 (7.6%)	9 (7.3%)
10 to 100 ft	2 (40.0%)	57 (48.3%)	59 (48.0%)
100 to 1,000 ft	0 (0.0%)	8 (6.8%)	8 (6.5%)
1,000 to 10,000 ft	0 (0.0%)	1 (0.8%)	1 (0.8%)
Unknown	3 (60.0%)	43 (36.4%)	46 (37.4%)
Total Number of Sites	5	118	123

One of the questions in the survey asked how the DNAPL was distributed in the subsurface, with the option of specifying DNAPL present in pools, as residual, sorbed to the soil particles, diffused into low K layers (rock matrix or clay) or in dead-end fractures. The respondents were asked to choose all of the parameters that applied to their site, resulting in greater than 100% totals. Table C.14 summarizes the DNAPL distribution results. Of the 179 responses to this question, from a total of 75 sites, the majority of the sites had residual (83%) or sorbed (61%) DNAPL. Of the remainder of the responses, 44% of the sites had pooled DNAPL, 40% had DNAPL diffused into low K layers and only 11% had DNAPL trapped in dead-end fractures.

Distribution of DNAPL in Subsurface	Number of Sites With Unspecified Technology	Number of Sites With Specified Technology	Total Sites
Dead-end fractures	0 (0.0%)	8 (10.7%)	8 (9.3%)
Diffused into low permeability layers/rock matrix	1 (9.1%)	30 (40.0%)	31 (36.0%)
Pools	2 (18.2%)	33 (44.0%)	35 (40.7%)
Residual	5 (45.5%)	62 (82.7%)	67 (77.9%)
Sorbed	3 (27.3%)	46 (61.3%)	49 (57.0%)
Total Number of Sites	11	75	86

 Table C.14:
 Summary of DNAPL Distribution in Subsurface

Table C.15 summarizes the survey responses to the shape of the DNAPL distribution in the subsurface. Of the 118 responses, 53 (45%) said that the shape of the DNAPL was unknown. Of the sites where an estimate of the shape of the DNAPL distribution was available, the majority of the sites (36 or 31%) were believed to be irregular in shape.

Shape of DNAPL Distribution in Subsurface	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
Cylindrical	0 (0.0%)	15 (12.7%)	15 (12.2%)
Irregular	0 (0.0%)	36 (30.5%)	36 (29.3%)
Rectangular	0 (0.0%)	13 (11.0%)	13 (10.6%)
Triangular	0 (0.0%)	1 (0.8%)	1 (0.8%)
Unknown	5 (100.0%)	53 (44.9%)	58 (47.2%)
Total Number of Sites	5	118	123

Table C.15: Summary of the Shape of DNAPL Distribution in the Subsurface

Q: Enter the total number of sampling locations used to locate the DNAPL source area(s).

Table C.16 provides a summary of the number of DNAPL sampling locations and the type of methods used to sample for DNAPL. Of a total of 123 sites with sampling location data in the database, 52 of these sites (4 had unspecified technologies) did not specify any sampling method. Fully-screened monitoring well and soil samples were the most frequently used methods (employed at 41% and 37% of sites respectively). However, depth-discrete sampling and nested monitoring wells were used at nearly as many sites (28% and 32% of sites respectively).

Not included in Table C.16 is a count of the sites where other methods were used, since not all users specified the number of sampling locations where the other method was used. Specified other methods for DNAPL source location included the following:

- *i)* Subsampling of soil cores;
- *ii)* Multilevel wells with discrete sampling points;
- *iii)* Cone penetrometer (CPT) with discrete sampling and CPT-ROST(rapid optical screening tool);
- *iv)* hydropunch, simulprobe, multi-level sampler, instrumented membranes, mud rotary, ribbon samplers;
- *v)* open borehole wells; and

vi) downstream spring outflow.

Sampling Method	Number of Sampling Locations	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
Fully Screened	0^{a}	4	69	73 (59.3%)
	0 to 10	0	21	21 (17.1%)
Monitoring	10 to 50	0	20	20 (16.3%)
Wells	50 to 100	1	5	6 (4.9%)
	>100	0	3	3 (2.4%)
	0^{a}	5	79	84 (68.3%)
Nested	0 to 10	0	11	11 (8.9%)
Monitoring Wells	10 to 50	0	25	25 (20.3%)
vv chis	50 to 100	0	3	3 (2.4%)
	>100	0	0	0 (0.0%)
	0^{a}	4	85	89 (72.4%)
Depth-	0 to 10	0	5	5 (4.1%)
Discrete	10 to 50	0	18	18 (14.6%)
Sampling	50 to 100	1	4	5 (4.1%)
	>100	0	6	6 (4.9%)
	0^{a}	4	73	77 (62.6%)
Soil Sampling	0 to 10	0	2	2 (1.6%)
	10 to 50	0	25	25 (20.3%)
	50 to 100	1	12	13 (10.6%)
	>100	0	6	6 (4.9%)
Total Nun	iber of Sites	5	118	123

TableC.16: Number of Sampling Methods Used to Locate DNAPL Source Area(s)

a,"0" was the default answer so if a respondent did not answer this question the value displayed will be "0".

Q: How was the presence of DNAPL determined?

Table C.17 provides a summary of the methods used to identify DNAPL presence. Of a total of 122 sites with sampling location data in the database, 2 of these sites (5 had unspecified technologies) did not estimate the presence of DNAPL. DNAPL presence was most frequently inferred from groundwater chemistry and site history (employed at 57% and 52% of sites respectively). However, known spills and direct observation of DNAPL in wells and soil samples were used at a large number of sites as well (26%, 25% and 30% of sites respectively).

Not included in Table C.17 is a list of other methods used to determine DNAPL presence. Specified other methods for determining DNAPL presence included the following (presented below as stated by survey respondents):

- *i)* degradation, odor, and staining of dedicated sampling;
- *ii)* Research project with a controlled release, thus the exact volume of DNAPL was known;
- *iii)* Combination of geoprobe-based dissolved phase groundwater sampling and soil sampling with source area limits inferred from observed TCE distribution;
- *iv)* Originally inferred from soil sample results (partitioning calculations);
- *v)* Detected in groundwater samples from temporary screened (but not sandpacked) well points;
- *vi)* 20 membrane interface probe locations;
- *vii)* Understanding of the nature of Hazardous Waste migration. If more than 72 barrels of NAPL are disposed of into a site (one truck load) the amount above 72 barrels must be present as NAPL. The flow of GW is slow and the concentrations of NAPLs in GW are small in the time since the large scale production of substituted hydrocarbons very little dissolution can occur. The rest has to be undissolved NAPL.
- *viii)* Soil concentrations converted to equilibrium porewater and compared to solubility limits; and,
- ix) A number of surface soil removal actions for VOCs + PCBs.

Method	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
Site history	1	62	63 (51.6%)
Known spill(s)	0	32	32 (26.2%)
Inferred from dissolved contaminant concentrations	1	68	69 (56.6%)
Direct observation of DNAPL in soil sample	0	37	37 (30.3%)
Extracted from monitoring wells	1	30	31 (25.4%)
Membrane Interface Probe	0	10	10 (8.2%)
Ribbon NAPL Samplers/FLuTEs	0	2	2 (1.6%)
Partitioning tracers	0	5	5 (4.1%)
Presence not estimated	0	2	2 (1.6%)
Total Number of Sites	5	117	122

Table C.17: Number of Methods Used to Determine DNAPL Presence

^a"0" was the default answer so if a respondent did not answer this question the value displayed will be "0".

Q: How was the total (pre-remediation) DNAPL mass present in the subsurface estimated?

Table C.18 provides a summary of the methods used to identify DNAPL presence. Of a total of 122 sites with sampling location data in the database, 24% of these sites did not estimate the pre-remediation DNAPL mass. DNAPL mass was most frequently estimated from groundwater or soil contaminant concentrations (34% sites).

Not included in Table C.18 is a list of other methods used to estimate DNAPL mass. Specified other methods for DNAPL mass estimates included the following (presented as stated by survey respondents):

i) Interpolation of CPT/ROST response;

- *ii)* Distribution is too heterogeneous to achieve any degree of confidence. The estimate is considered highly unreliable;
- *iii)* Area and depth of the two migrating DNAPL plumes (not pools but moving highly saturated NAPL masses) were measured by wells (NAPL mass constituted 99.97% of characterized mass). records of wastes placed into the site. also the area and thickness and average concentration of the dilute solute plumes were measured and along with the partitioning to the aquifer solids was used to assess the mass in the APL partition (0.03% of the characterized mass);
- *iv)* Groundwater modeling (MODFLOW) was used to back-calculate the amount of source present. Current VOC concentrations present in down-gradient monitoring wells were used to 'calibrate' the model and estimate the amount of source present;

Method	Number of Sites With Unspecified Technology	Number of Sites With Technology Specified	Total Sites
Site History	1	29	30 (24.6%)
Known Spill(s)	0	17	17 (13.9%)
Calculated from Groundwater or Soil Contaminant Concentrations	0	41	41 (33.6%)
Calculated from Mass Flux Measurements	0	4	4 (3.3%)
Partitioning Tracers	0	2	2 (1.6%)
Not Estimated	1	28	29 (23.8%)
Total Number of Sites	5	117	122

Table C.18: Number of Methods Used to Estimate DNAPL Mass

^a"0" was the default answer so if a respondent did not answer this question the value displayed will be "0".

4 Remedial Technologies

Survey respondents were asked to specify the technology that was applied at their site and the stage of remediation, and provide reasons for their choice. The data collected are described below.

Q: Choose the remedial technology applied at this site.

Remedial technologies were specified for 118 of the sites entered into the survey. For purposes of analysis, remedial technologies that employed similar principles for DNAPL removal were combined under one "primary technology" (*e.g.*, chemical oxidation using permanganate or Fenton's reagent were defined under the same primary technology of chemical oxidation). Table C.19 summarizes the primary technologies that were specified by respondents to the survey (listed in order of most frequently used to least), and provides the breakdown of the number of sites where each primary technology was applied. Figure C.4 illustrates the distribution of technologies.

Bioremediation, thermal and chemical oxidation were the most often applied technologies (total 77%). Surfactant flushing and ZVI technologies were the least applied technologies (total 8.5%). Information was also provided for dual-phase extraction, excavation, pump and treat, and 6 undefined other technologies. No information was collected in the survey on sites where co-solvent flushing was applied.

Technology	Number of Sites Where Technology was Applied
Thermal	27 (22.9%)
Bioremediation	25 (21.2%)
Chemical Oxidation	25 (21.2%)
Dual-Phase Extraction	13 (11.0%)
Excavation	11 (9.3%)
Other	7 (5.9%)
Zero-Valent Iron	6 (5.1%)
Surfactant Flushing	4 (3.4%)
Co-Solvent Flushing	0 (0.0%)
TOTAL	118

Table C.19: Summary of Technologies Selected in the Survey

Table C.20 provides a summary of the breakdown of the primary technologies thermal, chemical oxidation, dual-phase extraction, other, and zero-valent iron into "secondary" technology descriptions that represent variations of the primary technology. Of the thermal technologies, resistive heating (six- and three-phase) and steam flushing were the most frequently applied, whereas data was input into the survey for only one site

each of conductive heating and low-temperature six-phase heating. Of the chemical oxidation technologies, permanganate appears to be the most frequently chosen oxidant, while data on only one ozone site was entered into the survey. There does not appear to be a preferred method in what medium the DNAPL is extracted with the dual-phase extraction approaches. Of the zero-valent iron variations, non-specified use of ZVI was the most frequently applied alternative. EZVI using nano-scale iron is still in the development stage with only one pilot test completed to date.

The survey respondents were given the option of choosing "other" for a technology selection and providing a description of that technology. Unfortunately, due to an error in the posting of our web site, the link to the descriptions of the "other" technology was broken. All of the users who selected an "other" technology were contacted and asked to provide a description of the technology. However, six users did not reply to the request for additional information; these technology descriptions are listed in the database as "other – undefined". For those "other" technologies that were defined, if the technology descriptions, then the technology was grouped under that primary technology description (*e.g.*, other – conductive heating becomes thermal – conductive heating).

Primary Technology	Secondary Technology	Number of Sites Where Technology was Applied
	Six-Phase Heating	10
	Steam	8
Thermal	Three-Phase Heating	5
Therman	Resistive Heating	2
	Conductive Heating	1
	Low-Temp Six-Phase	1
	Permanganate	15
Chemical Oxidation	Fenton's Reagent	9
	Ozone	1
	Water/DNAPL	5
Dual-Phase Extraction	Water/air	5
LAttaction	DNAPL/air	3
Other	Undefined	6
Oulei	Pump and Treat	1
	Zero-Valent Iron	4
Zero-Valent Iron	EZVI/Nano-Scale Iron	1
	ZVI/Clay Source Treatment	1

Table C.20: Breakdown of Primary Technologies

Q: State your reasons for choosing this technology.

Survey respondents were given a list of criteria to choose from that outlined their reasons for selecting a particular technology. Table C.21 contains a listing of these criteria, along with a summary of the number of times the criteria were chosen. The most frequently chosen criteria were: (*i*) remedial cost (62%); (*ii*) the perception of a good chance at remedial success (80%); (*iii*) appropriate technical implementability (41%); (*iv*) a lasting impact on the water quality (37%); and (*v*) remedial timeframe was a factor (35%).

Criteria	Frequency of Choice
Cost of technology application	52 (61.9%)
Technology perceived to provide a good chance at remedial success	67 (79.8%)
Remediation timeframe was a factor	29 (34.5%)
Regulatory pressures guided technology choice	7 (8.3%)
Impact of remedial technology on surface water is minimal	7 (8.3%)
Impact of remedial technology on secondary groundwater quality is minimal	9 (10.7%)
Technology application was perceived to reduce overall risk to environment	17 (20.2%)
Technology application was thought to have a lasting impact on the water quality	31 (36.9%)
Health and safety concerns guided technology choice	11 (13.1%)
Technical implementability of the remedial technology was thought to be appropriate	34 (40.5%)
Legal pressures guided technology choice	4 (4.8%)
Pressure from various stakeholders guided technology choice	11 (13.1%)
Other	12 (14.3%)
TOTAL SITES WITH DATA	84

Specified "other" reasons included the following:

- *i)* Six sites were technology demonstrations (two were bioremediation applications, three were chemical oxidation using permanganate, and one was emulsified nano-scale ZVI);
- *ii)* One ZVI project was a field research project of soil mixing with an iron and bentonite slurry conducted at the University of Waterloo (Waterloo, Ontario, Canada);
- *iii)* Bioremediation was chosen at two sites where the remediation occurred inside of drycleaners;

- *iv)* Excavation was chosen at one site with low permeability soil (silt), which rendered any *in situ* approach infeasible. The respondent adds, "We needed to get to MCLs in the sand and gravel immediately below the silt layer containing the DNAPL.
- *v)* Low temperature six-phase heating was chosen at one site over six phase heating due to the proximity to underground utilities.
- *vi)* For one site where the technology was undefined, the respondent states, "Ideal site to try innovative technology"; and
- *vii)* At one site, steam flushing was chosen because steam was readily available at the site, and there was no cost to tap into the steam source.

Q: Specify the current stage of remediation.

Survey respondents were asked to specify the stage of remediation for each site (*i.e.*, pilot *versus* full-scale, and completed *versus* ongoing). Table C.22 lists the distribution of treatment stage. Of 80 total DNAPL treatment attempts, 31 were full-scale applications and 49 were pilot tests. Data was collected for only 2 full-scale systems and 33 pilot tests where the remediation was considered to be complete and post-treatment monitoring was not on-going.

Stage of Remediation	Scale of Treatment		
	Full-Scale Application	Pilot Test	
Remediation completed	2	33	
Remediation completed, post- treatment monitoring ongoing	12	6	
Remedial activities on-going	17	10	
TOTAL	31 (38.8%)	49 (61.3%)	

Table C.22:	Breakdown	of Sites by	Stage of R	emediation	and Scale of	Treatment
1 abit C.22.	DICakuowii	Of Shes by	Stage Of Ro		and Scale of	ricatificiti

5 Cost and Treatment Duration Data

Q: What was the duration of field treatment activities to date? For on-going treatment, what is the estimated duration of future field treatment activities?

Table C.23 lists the actual and estimated duration of field treatment activities for both full and pilot scale treatment stages. These durations do not include any post-treatment monitoring. All pilot tests appeared to be completed within 5 years, with the majority (28 of 29 sites) of field treatment activities completed within 2 years. In comparison, only 16 of 22 full-scale applications were completed (or estimated to be completed) within 2 years. Six sites (27%) had longer durations of 5, 10, 15, 30, 50 and 150 years. Section 1.2.2 of Appendix D provides a detailed breakdown and analysis of the data by technology.

Monitoring	Duration in Months	Full-Scale Application	Pilot Test
	< 4	5	13
	4 to 8	4	5
	8 to 16	0	6
Actual	16 to 24	2	1
Duration	24 to 34	0	1
	60	1	1
	120	1	0
	180	1	0
	1 to 10	3	1
	10 to 20	2	1
Estimated Duration	360	1	0
Duration	600	1	0
	1800	1	0
ТОТА	L SITES	22	29

Table C.23: Breakdown of Duration of Field Treatment Activities

Q: Over what duration was post-treatment monitoring conducted/what is the estimated duration (for on-going monitoring)?

Table C.24 lists the duration of post treatment monitoring activities (actual if completed and predicted if ongoing) for both full and pilot scale treatment stages. Post-treatment monitoring was generally terminated within 4 years (29 of 31 sites). One site had monitoring on-going for 10 years.

Monitoring	Duration in Months	Full-Scale Application	Pilot Test
	0	0	2
	0.5 to 5	2	6
	5 to 15	5	5
Post Treatment	15 to 25	1	3
Duration	25 to 35	0	0
	35 to 50	3	0
	50 to 75	1	0
	120	1	0

Table C.24: Breakdown of Post-Treatment Monitoring Duration

Q: What was the total cost of remediation for the site including the design and implementation of remediation but not site characterization?

Table C.25 provides a summary of the DNAPL treatment cost data entered into the survey database. All costs are quoted in US dollars. Figure C.5 is a histogram showing the actual costs entered by the survey respondents for each site. These costs represent the design and implementation of the remediation, but omit the costs required for the initial site characterization. The costs are broken down into the scale of treatment (*i.e.*, full-scale *versus* pilot test) as well as the size of the DNAPL impacted zone. The breakdown by size is less informative for the pilot test applications as the size of the DNAPL impacted zone of the pilot test.

Cost data was provided for 16 sites at which full-scale treatment was either completed (1), treatment complete but post-treatment monitoring on-going (6), or on-going (9). Data was also collected for a total of 31 pilot test, of which the majority of the tests were completed (all but 3 sites). Of the large, full-scale applications, remedial

treatment costs were generally in the millions of dollars, with the largest cost being a \$15M water/DNAPL dual-phase treatment (on-going) at a site with a DNAPL impacted zone that was in the range of 100,000 to 1,000,000 ft². The smallest cost was reported as being \$75K at two sites where full-scale treatment was completed and post-treatment monitoring was ongoing. The sites were a ZVI treatment for a site with a DNAPL impacted zone that was in the range of 100 to 1,000 ft² and a bioremediation treatment with an unknown size. None of the pilot tests had costs greater than \$2M, with the majority being less than \$0.5M. It should be noted that a significant portion of the pilot tests under \$0.5M were small-scale (<1,000 ft²) technology demonstrations/field research projects. If the technology demonstrations are removed from the count, the pilot test costs were generally within the range of \$150,000 to \$2M.

Treatment	Total Cost (USD)	Size of DNAPL Zone ^a			TOTAL
Scale		Small	Large	Unknown	SITES
	<\$0.5M	3	2	2	7
	\$0.5M to \$1M	0	1	0	1
ale	\$1M to \$2M	0	2	0	2
Full-Scale	\$2M to \$4M	0	2	0	2
Ful	\$4M to \$10M	0	2	1	3
	>\$10M	0	1	0	1
	Total	3	10	3	16
	<\$0.5M	10	3	3	16
	\$0.5M to \$1M	1	8	2	11
est	\$1M to \$2M	1	2	1	4
Pilot Test	\$2M to \$4M	0	0	0	0
Pil	\$4M to \$10M	0	0	0	0
	>\$10M	0	0	0	0
	Total	12	13	6	31

 Table C.25:
 Summary of Remedial Cost Data

^aDNAPL zone size classifications were:

small = area < 10,000 ft² and volume < 100,000 ft³

large = area > 10,000 ft^2 and volume > 100,000 ft^3 .

Q: Provide a breakdown of the remedial cost components (Table C.26 provides the criteria presented in the survey).

Survey respondents were asked to provide an approximate breakdown of each significant component (*i.e.*, professional costs, consumables, treatment operation and infrastructure and post-treatment monitoring) of the remedial project relative to the total cost (not including site characterization). Table C.26 provides a summary of the breakdown of the DNAPL treatment cost data entered into the survey database. All data is presented as a percentage of the total actual cost. Figure C.6 is a histogram showing the cost breakdown of the technology applications as a percent of the total costs. The total cost percent varies from 100% because the cost breakdown data was inputted into the survey as a range and the mid point of the ranges were used to calculate the percentages.

Breakdown of Remedial Costs	% of Total Costs	Number of Sites
	0 to 10%	53
	11 to 20%	11
Professional Costs	21 to 30%	8
FIOLESSIONAL COSts	31 to 40%	2
	41 to 50%	2
	51 to 60%	1
	0 to 10%	63
Post Treatment	11 to 20%	7
Monitoring Costs	21 to 30%	6
	31 to 40%	1
	0 to 10%	64
	11 to 20%	4
Consumable Costs	21 to 30%	3
	31 to 40%	4
	51 to 60%	1
	61 to 70%	1

Table C.26: Breakdown of Remedial Cost Data

Breakdown of Remedial Costs	% of Total Costs	Number of Sites
	0 to 10%	43
	11 to 20%	8
	21 to 30%	13
Treatment Operation	31 to 40%	4
Costs	41 to 50%	6
	51 to 60%	1
	61 to 70%	1
	81 to 90%	1
	0 to 10%	50
Treatment Infrastructure Cost	11 to 20%	7
	21 to 30%	8
	31 to 40%	11
	51 to 60%	1

6 Methods of Assessing Performance and Monitoring

Survey respondents were asked to specify how they were planning on evaluating, or had evaluated, the performance of the technology. The data collected are described below.

Q: How are you planning to assess the interim technology performance – or – What was the technology performance measurement based on (check all that apply)?

Depending on the stage of remediation (ongoing *versus* completed), the above question varied; however the list of options was the same for both sets of remedial stages. Table C.27 list the responses and number of sites with responses selected. The most frequently chosen performance metric was the monitoring of concentration reductions in specific monitoring wells (79% of sites). Other frequently chosen metrics include: (*i*) total DNAPL mass removed (36% of sites); and (*ii*) reduction in soil concentrations (36%).

Method of Assessing Performance	Number of Sites
Concentration reduction in specific monitoring wells	63 (79%)
Reduction in soil concentrations	29 (36%)
Achieve MCLs in monitoring wells	11 (14%)
Reduction in plume mass flux (or mass discharge)	18 (23%)
Reduction in plume size	20 (25%)
Production of degradation by-products	22 (28%)
Total mass removed	29 (36%)
Mass remaining	14 (18%)
Will not measure	0 (0%)
TOTAL NUMBER OF SITES	80

 Table C.27:
 Summary of Methods for Assessing Performance

Since asked to check all that apply it is possible to have answers that add to up to > 100%

7 Treatment Effectiveness

Remedial effectiveness can be evaluated using a number of different criteria, including the following:

- *i)* Removal of DNAPL source mass;
- *ii)* Decrease in dissolved phase mass flux from the source area;
- *iii)* Rebound in mass flux; and
- *iv)* Impacts to secondary groundwater quality.

Although survey respondents were asked which of a number of methods they were going to use to evaluate success (Section 6), they were only asked to provide information on

each of the four success criteria listed above. Information on these parameters were collected to rate the overall perceived success of the technology in remediating the DNAPL, and to indicate which external factors might have impacted the remediation process. The responses are discussed below.

Q: What level of source mass removal was achieved?

There were a total of 53 sites with source mass removal data. Table C.28 summarizes the source mass removal data that were specified by respondents to the survey. Of the 53 sites, 62% have not estimated the degree of source mass removal. Responses ranged from 10% to 100% source mass removal, with 16 sites (30%) having greater than 50% mass removal.

Percent Source Mass Removal	Number of Sites
100%	4 (8 %)
>90%	7 (13 %)
80 to 90%	3 (6 %)
50 to 80%	2 (4 %)
25 to 50%	2 (4 %)
10 to 25%	1 (2 %)
<10%	1 (2 %)
Not Estimated	33 (62 %)
Total	53

 Table C.28:
 Summary of Source Mass Removal

Q: How much decrease in mass flux was observed?

There were a total of 80 responses to the question of how much of a decrease in mass flux was observed after remediation. It should be noted that this question was asked whether the technology application was ongoing or completed (decrease observed to date *versus* decrease observed, respectively). In order to evaluate success, data was considered only from the completed technology applications (53 sites). Table C.29 summarizes the mass flux data that were specified by respondents to the survey. Of the 80 responses, 54% said that the mass flux decrease was unknown. Although 6 sites (7.5%) had 0% decrease in mass flux, 21 sites (26%) had 81 to 100% decreases.

Percent Mass Flux Decrease	Number of Sites Completed	Number of Sites Ongoing
81 to 100%	13 (25 %)	8 (30 %)
61 to 80%	5 (9 %)	1 (3.7 %)
41 to 60 %	1 (26 %)	1 (3.7 %)
21 to 40 %	0 (0 %)	1 (3.7 %)
1 to 20 %	0 (0 %)	1 (3.7 %)
0 %	2 (4 %)	4 (15 %)
Not Estimated	32 (60 %)	11 (41 %)
Total	53	27

 Table C.29:
 Summary of Mass Flux Decrease

Q: Currently at what stage is the post-treatment?

For many of the parameters used to evaluate success, a certain level of posttreatment monitoring is necessary to determine whether the level of mass decrease or reduction in mass flux was a permanent change or a temporary change due to things such as dilution, seasonal changes in the groundwater flow direction, desorption or counterdiffusion of contaminants from low permeability layers, *etc.* Table C.30 summarizes the post-treatment monitoring stage data that were specified by respondents to the survey. The data indicates that post-treatment monitoring is being conducted at the majority of sites (72%).

Post – Treatment Monitoring Stage	Number of Sites
Completed	25 (47.2%)
Ongoing	13 (24.5%)
Not Conducted	15 (28.3%)
Total	53

Table C.30: Summary of Post – Treatment Monitoring Stage

Q: Did post-treatment monitoring indicate that an increase in concentrations (rebound) of dissolved chlorinated solvents was occurring?

A total of 53 respondents answered the question of observed rebound. Table C.31 summarizes the rebound data that were specified by respondents to the survey. Rebound was not evaluated at 60% of the sites. Of those sites where rebound was tracked, 67% of the sites had rebound of the dissolved phase concentrations.

Occurrence of Rebound	Number of Sites
Yes	14 (26.4%)
No	7 (13.2%)
Not Evaluated	32 (60.4%)
Total	53

Table C.31: Summary of Rebound Data

Q: What impacts on secondary groundwater quality have been observed?

Table C.32 provides a list of the secondary groundwater quality parameters that the survey respondents had to choose from, of which they were to check all that apply. A total of 35 sites had at least one secondary groundwater quality issue selected for a total of 81 responses. Of these, the most prevalent choice was a decrease in source area volume (34%). None of the sites experienced an increase in source area volume. Changes in groundwater color (23%), elevated concentration of dissolved metals (23%), and methane generation (23%) were also frequent impacts (*i.e.*, occurred at more than 20% of sites).

Secondary Groundwater Quality Indicator	Number of Sites With The Indicator Selected
Methane generation	8 (22.8%)
	· · · ·
Hydrogen sulfide generation	1 (2.9%)
Dissolved iron generation	7 (20.0%)
Dissolved manganese generation	7 (20.0%)
Elevated concentration of other metals	8 (22.9%)
Increased Biochemical Oxygen Demand	4 (11.4%)
Change in pH	7 (20.0%)
Reduced soil porosity	6 (17.1%)
Change in hydraulic conductivity	5 (14.3%)
Change in groundwater color	8 (22.9%)
Increased dissolved solids	3 (8.6%)
Redistribution of DNAPL	4 (11.4%)
Increase in source area volume	0 (0.0%)
Decrease in source area volume	12 (34.3%)
Other	5 (14.3%)
TOTAL SITES WITH SECONDARY GROUNDWATER IMPACTS	35

Table C.32: Summary of impacts to secondary groundwater quality data.

Q: Was this technology considered successful?

One of the questions in the DNAPL survey that was asked of the user was whether the technology application was considered a poor success, a fair success, successful, or not yet evaluated. Table C.33 summarizes the perceived success of the technology applications. Of the sites where remedial success was evaluated, 64% of the remedies were successful, 34% were fairly successful, and 2% performed poorly.

Perceived Success	Number of Sites
Success	28 (52.8%)
Fair Success	15 (28.3%)
Poor Success	1 (1.9%)
Not Yet Evaluated	9 (17.0%)
Total	53

 Table C.33:
 Summary of Perceived Success

Q: Indicate which factors impacted remediation.

Table C.34 provides the list of external factors impacting the remedial process that the survey respondents were asked to choose from. Data from a total of 118 sites was collected (*i.e.*, all sites with specified technologies). The remedial budget and cost was the most frequently chosen factor (61% of sites). The remaining four factors were all chosen with approximately the same frequency (23 to 31% of all sites).

Table C.34:	Summary of Breakdown of External Factors that Impacted the Remedial
Process	

Factors that Impacted the Remedial Process	Number of Sites With The Factor Selected
Facility operations constrained site access	27 (22.9%)
Presence of surface infrastructure impacted ability to estimate DNAPL mass or effectively conduct remediation	28 (23.7%)
Proximity to surface water impacted technology choice, design or effectiveness	35 (29.7%)
Regulatory pressures impacted clean-up criteria/time-frame	37 (31.4%)
Budget / remediation cost	72 (61.0%)
TOTAL SITES WITH FACTORS SELECTED	118

8 Data Gaps

Optimally, a study such as this would include data collected from sites with a wide variety of site characteristics and technologies, enabling statistical analyses of the data to confirm the presence of trends and to determine where correlation exists between factors. However, a minimum amount of data is required to achieve statistically significant conclusions. This minimum was not achieved for a number of reasons for the majority of the parameters analyzed for here.

Although attempts were made to compile a comprehensive database of information pertaining to DNAPL remediation, the data necessary to draw quantitative conclusion is very complex and difficult to collect. The difficulties in collecting this type of information arise from many factors including:

- people do not have a driver to collect or compile the data needed for this type of analysis;
- there are the complications of confidentiality;
- biased views toward a specific technology may exist and without data verification this may not be obvious; and
- it is difficult for people to make the time to respond to this type of survey.

Data deficiencies arise due to a number of factors, including the following:

- *i)* Limited number of survey respondents: The survey was posted on the web page for a total of 6 months. Three rounds of email requests for respondents were sent out to over 700 people, coupled with mailings of marketing materials to over 3,000 people and posting of requests for survey participants on several widely accessed newsletters/webpages. However, only a total of 61 people (plus the 21 published case studies) provided information on technology applications.
- *ii)* **Incomplete survey responses:** Of the 164 DNAPL impacted sites specified in the survey database, only a portion of these provided site characterization data (*i.e.*, geology, hydrogeology, DNAPL source zone parameters), and only 118 of these had a remedial technology specified. Of these 118 sites, 38 sites had little more than site characterization data provided (*i.e.*, no information pertaining to remedial technologies was collected), leaving only 80 sites for which the survey respondent fully completed the survey. Three email

reminders were sent to respondents with incomplete survey data in an attempt to increase the number of completed surveys.

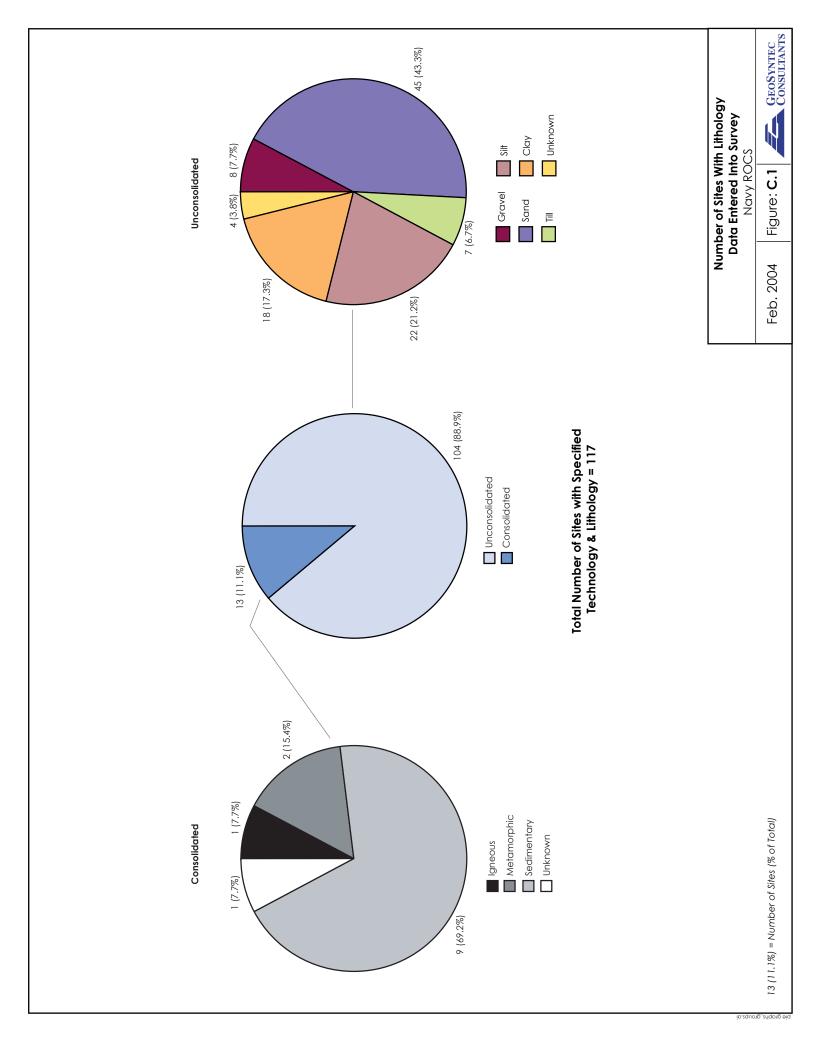
iii) **Insufficient field testing:** There is limited information available on the impact of some factors and technologies, due to a lack of field testing. For example, the use of emulsified nano-scale zero-valent iron is still in the technology development stage, with only one field test completed to date.

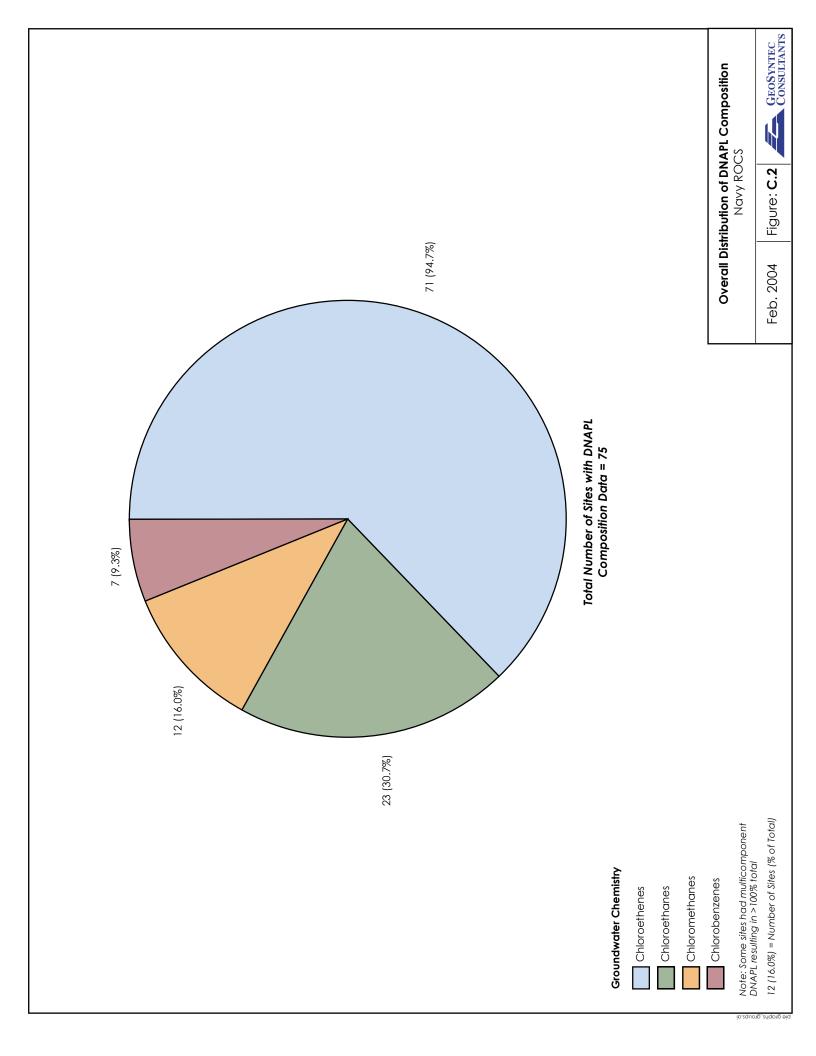
9 Survey Participant List

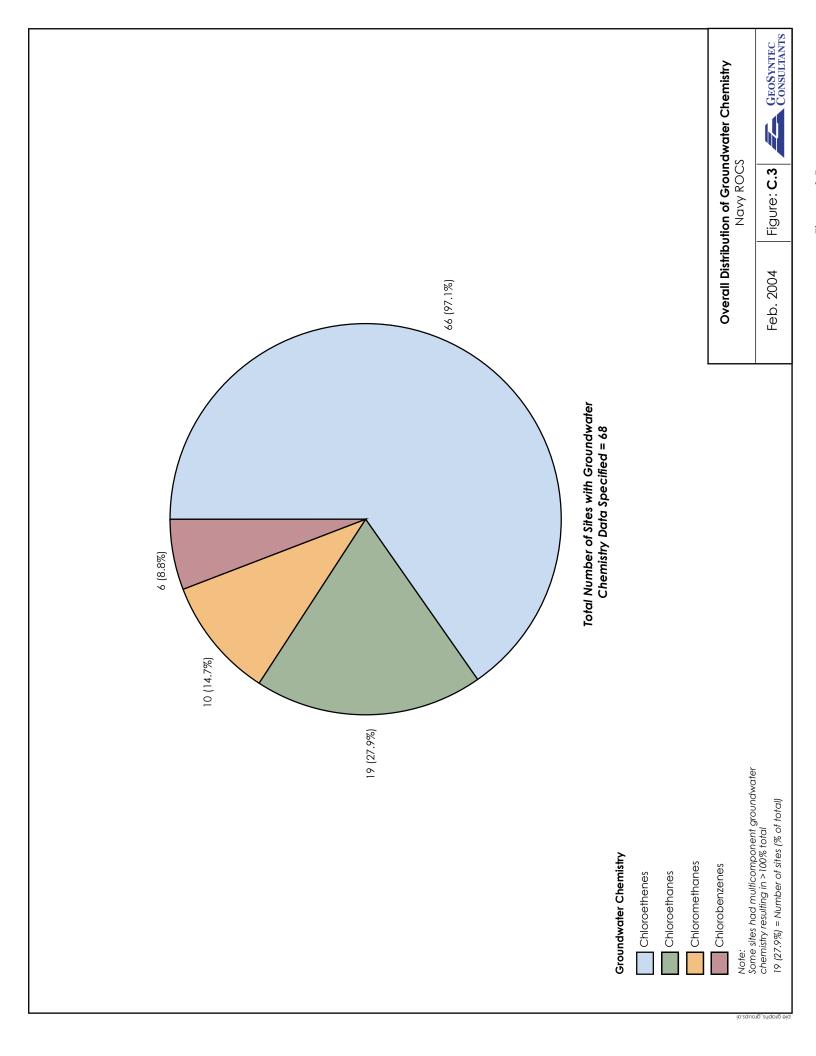
Table C.35 is a list of the 21 published case studies that were entered into the survey by GeoSyntec personnel identifying the site name, location, and technology application. Survey participants were asked if their names and contact information could be included on a list of respondents in the final report as a resource for Navy RPMs, site owners, regulators, etc. considering this technology. Table C.36 is a list of contact information for survey participants, along with the technology that they entered information about, who said that their contact information could be included in the report.

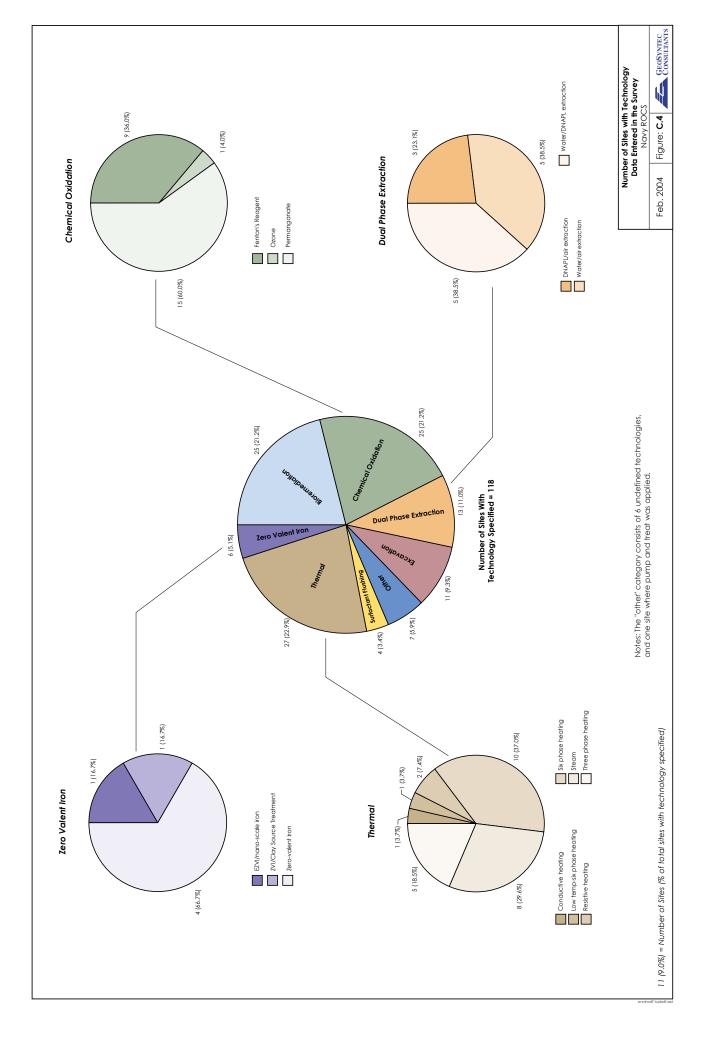
TABLE C.35 LIST OF PUBLISHED CASE STUDIES IN SURVEY Navy Review of Case Studies

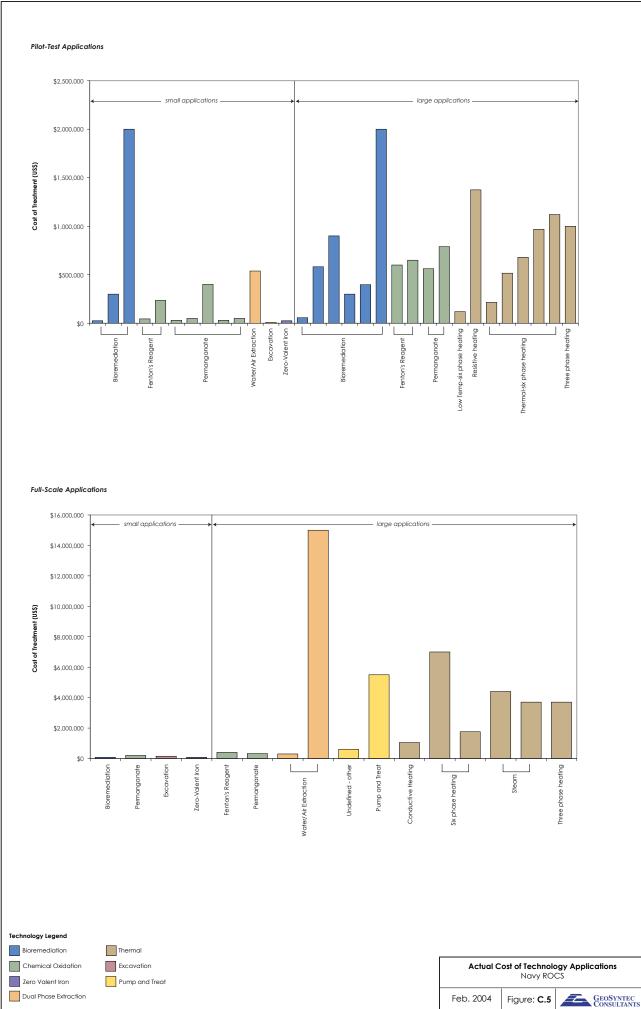
State, Province	Location_Name	Technology
Florida	Pinellas Northeast Site	Bioremediation
Idaho	Idaho National Engineering and Environmental Labor	Bioremediation
New Jersey	Active Industrial Facility	Chemical Oxidation-Fenton's Reagent
Massachusetts	Former News Publisher Facility	Chemical Oxidation-Fenton's Reagent
Maryland	Graces Quarter, APG	Chemical Oxidation-Fenton's Reagent
Ohio	Portsmouth Gaseous Diffusion Plant, X-701B Facilit	Chemical Oxidation-Permanganate
New Hampshire	US Army Cold Regions Research and Engineering Labo	Chemical Oxidation-Permanganate
Florida	Former Dry Cleaning Facility	Chemical Oxidation-Permanganate
California	328 Site, Santa Clara, CA	Dual Phase-Water/air extraction
Virginia	US Defense Supply Center Richmond Acid Neutralizat	Dual Phase-Water/air extraction
North Carolina	Union Chemical Company Superfund Site	Surfactant Flushing
Utah	Operable Unit 2, Hill AFB	Surfactant Flushing
Kentucky	SWMU 91 Treatability Study	Thermal-Resistive heating
Texas	Air Force Plant 4	Thermal-Six phase heating
Oregon	Former Pharmaceutical Manufacturer (ICN)	Thermal-Six phase heating
Florida	Launch Complex 34	Thermal-Six phase heating
Illinois	Former Electronics Manufacturing Facility	Thermal-Six phase heating
Alaska	Poleline Road Disposal Area (PRDA), Operable Unit	Thermal-Six phase heating
South Carolina	Savannah River Site, Area M	Thermal-Six phase heating
Illinois	AG Communication Systems	Thermal-Steam
Ohio	DOE Portsmouth	Thermal-Steam

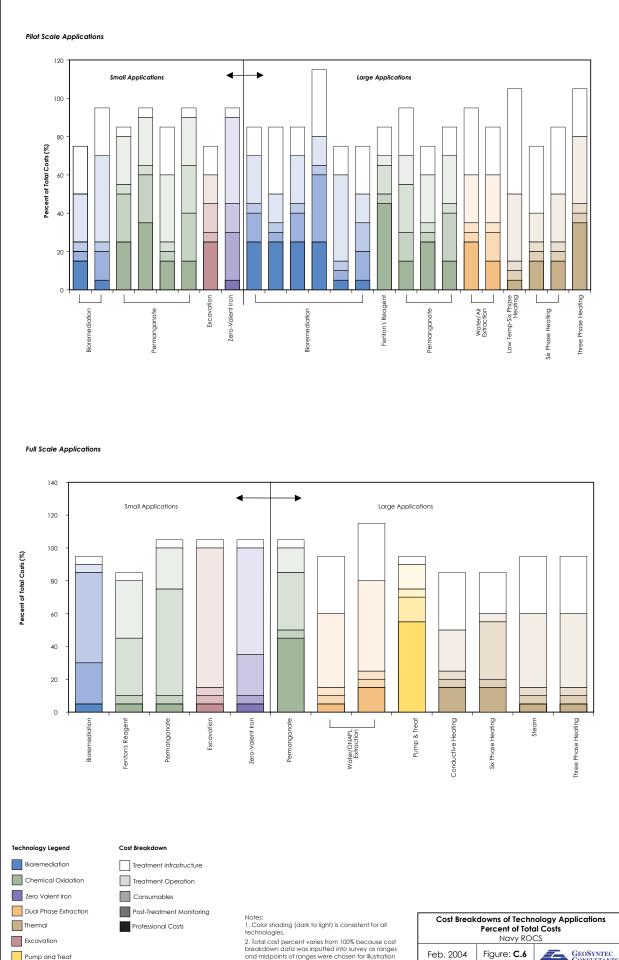












Pump and Treat

2. Total cost percent varies from 100% because cost breakdown data was inputted into survey as ranges and midpoints of ranges were chosen for illustration

Figure: C.6 GEOSYNTEC CONSULTANTS

APPENDIX D

Survey Data Comparison and Interpretation

APPENDIX D TABLE OF CONTENTS

<u>Page</u>

1	Gen	eral Overview of Technology Performance and Effectiveness 1
	1.1	Remedial Technology Selection 1
	1.2	Cost and Treatment Duration
	1.2.1	Treatment Cost
	1.2.2	Treatment Duration
	1.3	Technology Performance Evaluation
	1.3.1	Perception of Technology Effectiveness
	1.3.2	Factors Impacting Remedial Effectiveness
	1.3.3	DNAPL Mass Removal 13
	1.3.4	Mass Flux Decrease 14
	1.3.5	Occurrence of Rebound of Dissolved Phase Concentrations 15
	1.3.6	Achievement of Remedial Goals and Site Closure
	1.3.7	Impacts to Secondary Groundwater Quality 17
	1.3.8	Overall Technology Performance
2	Imp	act of Site Characteristics on Technology Application and Performance 19
	2.1	Aquifer Geology and Lithology
	2.2	Hydrogeology
	2.3	DNAPL Source Zone Characteristics
	2.3.1	Source Zone Geochemistry
	2.3.2	DNAPL Source Zone Areal Extent
	2.3.3	DNAPL Source Zone Volume

2.3.4	Maximum DNAPL Depth	. 30
2.3.5	DNAPL Distribution in Subsurface	. 30
2.3.6	DNAPL Shape in Subsurface	. 31
2.3.7	DNAPL Detection, Sampling and Mass Estimation	. 31

APPENDIX D: SURVEY DATA COMPARISON AND INTERPRETATION

The DNAPL remediation case study data collected with the web-based survey was presented in Section 4 and Appendix C. An interpretation of the data is discussed in the following sections:

- *i)* General overview of technology performance and effectiveness (Section 1); and
- *ii)* Impact of site characteristics on technology application and performance (Section 2).

Smaller tables are embedded directly into the text. The remaining tables and all figures may be found at the end of the appendix.

1 General Overview of Technology Performance and Effectiveness

The technology performance data was analyzed for trends by technology and for other factors that impacted treatment cost, duration and remedial success. The results of the analysis are presented in the following sections:

- *i)* remedial technology selection (Section 1.1);
- *ii)* treatment cost and duration (Section 1.2); and
- *iii)* remedial performance (Section 1.3).

1.1 Remedial Technology Selection

Table D.1 provides a summary of the technology selection criteria broken down by secondary technology. The technology selection criteria that were chosen by survey respondents as one reason for selecting each technology by more than 50% of the survey respondents were as follows (note: no information was provided for ozone oxidation, co-solvent flushing, or ZVI/clay):

- *i)* **Cost of technology application:** all technologies except pump and treat, steam flushing and EZVI.
- *ii)* **Technology perceived to provide a good chance at remedial success:** all technologies except DNAPL/air dual-phase extraction, and pump and treat.

- *iii)* **Remediation timeframe was a factor:** water/DNAPL dual-phase extraction, excavation, conductive and low-temp six-phase heating, and EZVI.
- *iv)* **Regulatory pressures guided technology choice:** pump and treat only.
- *v*) Impact of remedial technology on surface water was minimal: This criteria was not chosen more than 50% of the time for any one technology, but was chosen for at least one site for bioremediation, chemical oxidation and water/DNAPL dual-phase extraction.
- *vi)* Impacts of remedial technology on secondary groundwater quality was minimal: Low-temp six phase heating only.
- *vii)* **Technology application was perceived to reduce risk to the environment:** water/DNAPL dual-phase extraction, excavation, and conductive and low-temp six-phase heating.
- *viii)* Technology application was thought to have a lasting impact on the water quality: bioremediation, DNAPL/air extraction, surfactant flushing, and resistive and three-phase heating.
- *ix)* **Health and safety concerns guided technology choice:** This criteria was not chosen more than 50% of the time for any one technology, but was chosen for at least one site for bioremediation, chemical oxidation, and water/air and DNAPL/air dual-phase extraction.
- *x)* Technical implementability of the remedial technology was thought to be appropriate: Fenton's reagent, water/air and DNAPL/air dual-phase extraction, surfactant flushing and three-phase heating.
- *xi)* Legal pressures guided technology choice: Pump and treat only.
- *xii)* **Pressure from various stakeholders guided technology choice:** pump and treat and resistive heating only.

Other reasons for technology selection included (stated as entered into the survey):

- *i)* Bioremediation, Chemical Oxidation, EZVI The site was ideal/or chosen for technology demonstrations;
- *ii)* Excavation Low permeability of soil (silt) rendered any *in situ* approach infeasible. We needed to get to MCLs in the sand and gravel immediately below the silt layer containing the DNAPL; and

iii) Low-temp six-phase heating was chosen at one site over higher-temp sixphase thermal methods due to proximity to underground utilities – are considering chemical oxidation or other technology instead.

1.2 Cost and Treatment Duration

1.2.1 Treatment Cost

The treatment cost data was analyzed for the following:

- *i)* Trends in the treatment costs, by comparing total costs, the cost breakdown, and approximate unit cost comparisons; and
- *ii)* The impact of size of the DNAPL impacted zone and scale of treatment on the costs.

Optimally, the trends in the treatment costs and calculation of unit costs should be performed using the size of the targeted treatment zone. However, this data was not collected with the survey; the assumption was made, for purposes of analysis and categorization, that the targeted treatment zone and the DNAPL impacted zone were of similar orders of magnitude. Unit costs for full-scale applications were thus calculated using the size of the DNAPL impacted zone. Unit costs for pilot tests, however, were not calculated as it is likely that the targeted treatment area of the pilot test is significantly smaller than the DNAPL impacted zone of the entire site. To determine the impact of the size of the DNAPL impacted zone on the costs, the data (pilot- and full-scale) was sorted into small applications, which corresponded to areas < 10,000 ft² and volumes < 100,000 ft³.

The treatment cost data is summarized or illustrated in tables and figures as follows:

- *i)* Table D.2 provides a breakdown of the total treatment cost data by secondary technology, the size of the DNAPL impacted zone, and the scale of treatment (*e.g.*, full-scale *versus* pilot test);
- *ii)* Table D.3 summarizes the cost breakdown data, categorized by secondary technology and the scale of treatment;
- *iii)* Figure D.1 illustrates the distribution of the actual costs across all sites (by the scale of treatment and size of the DNAPL impacted zone);
- *iv)* Figure D.2 shows the cost breakdown by percent of costs, again sorted by the scale of treatment and the size of the DNAPL impacted zone;

- *v)* Figure D.3 is similar to Figure D.2, but with the cost breakdown as actual costs;
- *vi)* Figure D.4 shows unit costs by areal extent of the DNAPL impacted zone for the full-scale applications only; and
- *vii)* Figure D.5 shows a similar figure to Figure D.4, but with unit costs by volume of the DNAPL impacted zone. Figure D.6 shows the unit cost comparison by DNAPL mass removed.

As discussed in Appendix C, the smaller applications were generally less than US\$0.5M, with the exception of one bioremediation pilot test. The costs of the pilot tests were similar within size categories, and did not seem to be impacted by technology. However, at full-scale, the cost of the dual-phase extraction, pump and treat and thermal applications were generally significantly more than that of the sites where bioremediation and chemical oxidation were applied (Figure D.1 and Table D.2).

In terms of the breakdown of costs, the following trends appear to be present (Table D.3 and Figures D.2 and D.3):

- *i)* Treatment infrastructure appears to represent a larger proportion of the costs of larger treatment applications than smaller ones for both pilot- and full-scale applications.
- *ii)* Consumables (*e.g.*, electron donor, iron, oxidant, surfactant, *etc.*) tend to represent a larger proportion of the cost of full-scale applications in comparison to pilot tests;
- *iii)* Professional costs appear to comprise a larger portion of pilot test applications than full-scale; and
- *iv)* Treatment operation represents a significant portion of costs for all technologies.

Unit costs were calculated by volume and area of the DNAPL-impacted zone for full-scale applications only. Again, unit costs are only calculated for the full-scale applications since the targeted treatment size of the pilot tests were likely to be less than the size of the DNAPL impacted zone, thus affecting the accuracy of the calculation. The unit costs are shown as ranges as the survey respondents were asked to specify the volume, area and DNAPL mass data from a list of order of magnitude ranges. Order of magnitude ranges were specified since, due to the difficulty in obtaining accurate volume, area and DNAPL mass data, any data provided were not likely to be more accurate than that.

Figures D.4 and D.5 show the unit cost variation for the full-scale sites with both cost and area, and cost and volume data provided. The unit costs are affected by the size of the DNAPL impacted zone, with smaller sites having much larger unit costs. It is difficult to draw any meaningful conclusions between technologies, given the lack of data. Very little data was available where both DNAPL mass removal and cost data were provided, as a result, no trends by mass removed are discernible (see Figure D.6).

1.2.2 Treatment Duration

Figures D.7 and D.8 illustrate the range in treatment duration for pilot test and fullscale applications respectively. The data shown in the histograms are data from individual sites. The treatment durations shown on Figures D.7 and D.8 are broken down by the field activity duration (actual to date and estimated future activity) and the estimated/actual duration of the post-treatment monitoring program. Therefore, these durations represent the time required until no further remedial action with that particular technology is required on site.

Of the limited amount of pilot-test data available, the bioremediation sites appeared to require the longest pilot tests, although the range in duration was large (2 months to 4 years; Figure D.7). There did not appear to be any trends in the relative duration of the post-treatment monitoring versus the treatment activities. There was also limited fullscale data collected (see Figure D.8). Note the change in scale on the vertical axis (duration in months). The one conclusion that can be drawn from this comparison is the significant difference between the duration of the remedial technologies that employ *in* situ destruction or enhancement of the DNAPL mobilization/flushing mechanisms (e.g., bioremediation, chemical oxidation, excavation, thermal and ZVI technologies) versus that of the technologies that rely on flushing as the principal DNAPL removal mechanism (*i.e.*, pump and treat and dual-phase extraction). The expected treatment durations of the three sites using water/DNAPL dual-phase extraction and pump and treat were 45 and 75 years (dual-phase) to 158 years (pump and treat). The remainder of the technologies had expected durations of less than 4 years. It should be noted that pump and treat has been used on that site for five years to date, while the two dual-phase extraction sites have been in operation for 10 and 15 years, and are still operating. No trends in the relative proportion of post-treatment monitoring *versus* treatment activity duration are apparent, except perhaps for a longer duration of post-treatment monitoring for full-scale versus pilot test applications.

Optimally, treatment duration should be analyzed for correlations to the following:

- *i)* the size of the targeted treatment area, as the size of the application will impact the treatment implementation and operation;
- *ii)* DNAPL distribution (*e.g.*, pools, residual, *etc.*);

- *iii)* the cost of the treatment; and
- *iv)* remedial success.

However, insufficient data was collected in order to do so (see Appendix C - Section 8 and Section 6 of the report).

1.3 Technology Performance Evaluation

Several remedial effectiveness criteria were evaluated to determine what impact technology choice and site characteristics may have on the degree of success of remediation. General technology performance criteria included the following:

- *i)* the perception of technology effectiveness and performance (Section 1.3.1); and
- *ii)* factors impacting remedial performance (Section 1.3.2).

More specific technology performance criteria were also evaluated, including the following:

- *i)* extent of DNAPL mass removal (Section 1.3.3);
- *ii)* decrease of mass flux (Section 1.3.4);
- *iii)* occurrence of rebound of dissolved phase concentrations (Section 1.3.5);
- *iv)* achievement of remedial goals and/or site closure (Section 1.3.6); and
- *v)* impacts to secondary groundwater quality (section 1.3.7).

DNAPL remediation technologies may be effective in terms of only one or more of the above criteria; however, the most effective technology would meet all criteria. Therefore, each technology was also evaluated in how well it met all criteria (Section 1.3.8).

1.3.1 Perception of Technology Effectiveness

Survey respondents were asked to rate the technology performance at their site. Figure D.9 provides a summary of the perceived success sorted by primary technology. Table D.4 provides a breakdown of the perceived success data by secondary technology as well as the role of the survey respondent in the groundwater community. The following conclusions can be drawn:

- i) Of the sites where success was evaluated, more than 50% of the sites were rated successful for bioremediation (78% of 9 sites), chemical oxidation using Fenton's reagent (63% of 8 sites), excavation (75% of 4 sites), ZVI (67% of 3 sites), and thermal (100% of 4 sites of six-phase heating, 50% of 2 sites of steam flushing, and 100% of 1 site each of conductive and low-temperature six-phase heating) technologies.
- *ii)* Only bioremediation, excavation and thermal technologies (six-phase, conductive and low-temperature six-phase heating) were rated successful for more than 75% of the sites.
- *iii)* Technologies that were predominantly rated as having a fair success were chemical oxidation using permanganate (57% of 7 sites), surfactant flushing (100% of 2 sites), the thermal technology resistive heating (1 site), and EZVI (1 site).
- *iv)* The three sites using dual-phase water/air extraction had ratings of good, fair and poor successes.

There was not enough data to discern any correlation between role in the environmental community and perception of success.

Survey respondents were also asked to rate their technology according to a list of effectiveness criteria, which are listed in Table D.5. Table D.5 also summarizes the responses according to the respondent's role in the environmental community. The following can be concluded from the distribution of responses (effectiveness criteria is listed in bold print, with the results after):

- *i*) Able to achieve treatment goals at a reasonable cost: Regulators (100% of 2) and vendors (83% of 6) agreed most strongly with this statement. Site owners (25% of 4) agreed the least with this statement. Consultants (52% of 33) and DoD RPMs (54% of 13) were fairly neutral.
- *ii)* Technology is limited to certain site conditions: Site owners agreed most strongly with this statement (75% of 4). Vendors agreed the least strongly (17% of 6). Consultants (36% of 33), regulators (50% of 2) and DoD RPMs (46% of 13)_were fairly neutral.
- *iii)* **Technology requires further development before it can be routinely applied:** In general, few respondents agreed with this statement. Vendors agreed the most frequently at 33% of 6 respondents.

- *iv)* Reasonable ease of implementation: Regulators and site owners all (100% of 6) agreed with this statement. Vendors, consultants, and DoD RPMs were neutral (50%, 49% and 46% respectively).
- *Reasonable ease of use:* Again, regulators and site owners agreed 100% with this statement. Vendors, consultants, and DoD RPMs were neutral (50%, 55%, and 69% respectively).
- vi) **Reasonable ease of design:** The responses were similar to the previous two, with the exception of fewer vendors agreeing with this statement (33% of 6).
- *vii)* Adequate number of trained vendors to implement technology: No vendors agreed with this statement, but 100% of the regulators agreed. A large proportion of site owners also agreed.

Table D.6 provides a breakdown of the responses by secondary technology. Table D.7 contains a listing of all technologies that had greater than 50% agreement with each statement. The following conclusions can be drawn from this data:

- *i)* Technologies that were generally considered reasonable to implement and use and were able to achieve the remedial goals at a reasonable cost were bioremediation, chemical oxidation using permanganate, excavation, conductive and low-temp six-phase heating and ZVI.
- *ii)* Pump and treat and EZVI were not generally rated as able to achieve remedial goals at reasonable costs, but were frequently chosen for their ease of use and design.
- *iii)* The ZVI technologies were the only technologies that were thought to need further development prior to routine use.
- *iv*) Technologies that were generally thought to be more limited by particular site conditions included chemical oxidation using permanganate, DNAPL/air dual-phase extraction, excavation, surfactant flushing, ZVI, and resistive and low-temperature six-phase heating. Section 1.3.2 provides more detail on the specific limitations.

Technology			Technology is limited to certain site conditions	Technology requires further development before routine application	Reasonable ease of implementation	Reasonable ease of use	Reasonable ease of design	Adequate number of trained vendors	TOTAL SITES
		Able to achieve treatment goals at a reasonable cost	Ţ	dev	R			Ψ	
Bioreme	ediation	\checkmark			\checkmark	\checkmark	\checkmark		17
Chemical Oxidation	Fenton's Reagent	\checkmark							9
Chemical Oxidation	Permanganate	\checkmark	\checkmark		\checkmark	\checkmark	\checkmark		13
	Water/air				\checkmark	\checkmark	\checkmark	\checkmark	6
Dual-Phase Extraction	Water/DNAPL				\checkmark	\checkmark	\checkmark	\checkmark	3
	DNAPL/air		\checkmark						1
Excav	ation	\checkmark	\checkmark		\checkmark				5
Pump an	d Treat				\checkmark	\checkmark			1
Surfactant	Flushing	\checkmark	\checkmark						2
	Conductive	\checkmark				\checkmark	\checkmark		1
	Low-Temp six- Phase	\checkmark	\checkmark		\checkmark	\checkmark	\checkmark		1
Thermal	Resistive	\checkmark	\checkmark						1
	Six-Phase	\checkmark							9
	Three-Phase								3
	Steam Flushing	\checkmark							5
	EZVI			\checkmark		\checkmark	\checkmark		1
ZVI	ZVI/Clay	\checkmark	\checkmark		\checkmark	\checkmark	\checkmark		1
	ZVI	\checkmark	\checkmark	\checkmark	\checkmark	\checkmark		\checkmark	2

Table D.7: Summary of Technologies for Which >50% of Survey Respondents Chose the Technology Performance Criteria

1.3.2 Factors Impacting Remedial Effectiveness

Table D.6 provides a listing of specific limitations provided by survey respondents for each technology. These limitations can be summarized as follows:

- *i)* heterogeneity impacts the effective delivery of amendments used for bioremediation, chemical oxidation (both Fenton's reagent and permanganate), and surfactant flushing;
- *ii)* bioremediation requires a particular microbial community; however, bioaugmentation with the appropriate microbial community can be done to compensate;
- *iii)* high carbonate content causes a more rigorous reaction and higher reagent consumption for Fenton's reagent;
- *iv)* permanganate degrades only a limited range of DNAPLs degradable, the presence of DNAPL pools can impact remedial effectiveness, and the precipitation of manganese dioxide can cause pore plugging and reduction of soil permeability when using permanganate;
- *v)* the aquifer must be dewatered adequately for DNAPL/air extraction;
- *vi)* water/air extraction is applicable to sites with low transmissivity and permeability, with no possibility for short-circuiting; extreme heterogeneity requires careful engineering;
- *vii)* extraction limited to shallow overburden only;
- *viii)* pump and treat systems must be compatible with other on-site systems, and containment must be achieved;
- *ix)* low-temperature six-phase heating is applicable only for DNAPL pools and for depths that can be economically serviced by hand bailing;
- *x)* resistive and six-phase heating are impacted by soil conductivity;
- *xi)* cost is significant for the thermal technologies six- and three-phase heating, and there is a shortage of qualified vendors;
- *xii)* proximity of underground utilities/infrastructure are problematic, and DNAPL present in narrow "chimney" configurations at depths beyond 50 ft bgs may present significant application and cost challenges for six-phase heating; and

- xiii) testing of steam flushing in fractures has been inconclusive; and
- *xiv)* ZVI with clay is applicable only in unconsolidated materials that are accessible to soil mixing or jet grouting equipment. The construction contractor must know how to undertake soil mixing.

Table D.8 provides a breakdown of external factors that may impact remedial effectiveness. Table D.9 summarizes the data by presenting the technologies that had each external factor chosen by survey respondents at least 50% of the time. No trends are apparent, other than that most technologies were impacted by budget/remediation costs.

			1				T
Techno	Facility operations constrained site access	Presence of surface infrastructure impacted ability to estimate or remediate DNAPL mass	Proximity to surface water impacted technology choice, design or effectiveness	Regulatory pressures impacted clean- up criteria/time-frame	Budget/remediation cost	TOTAL SITES	
Bioremed	liation	\checkmark				\checkmark	25
Chemical Oxidation	Fenton's Reagent					\checkmark	9
	Permanganate					\checkmark	15
	Water/air					\checkmark	5
Dual-Phase Extraction	Water/DNAPL					\checkmark	5
	DNAPL/air	\checkmark		\checkmark			3
Excava	tion			\checkmark	\checkmark	\checkmark	11
Pump and	l Treat				\checkmark		1
Surfactant]	Flushing						4
	Conductive					\checkmark	1
	Low-Temp six- Phase				\checkmark		1
Thermal	Resistive	\checkmark	\checkmark	\checkmark			2
	Six-Phase					\checkmark	10
	Three-Phase			\checkmark			5
	Steam Flushing				\checkmark	\checkmark	8
	EZVI		\checkmark			\checkmark	1
ZVI	ZVI/Clay		\checkmark				1
	ZVI	\checkmark	\checkmark				4

Table D.9: Summary of Technologies that had >50% of Survey Respondents Selecting Each External Factor That Impacted Remedial Effectiveness

1.3.3 DNAPL Mass Removal

Figure D.10 and Table D.10 summarize the DNAPL mass removal by primary and secondary technologies respectively. There were a total of 20 sites with source mass removal data. Table D.11 summarizes the distribution of mass removal effectiveness by technology. The majority (70%) of sites had >80% mass removal. Only four sites achieved 100% mass removal (one excavation pilot test, one six-phase and one conductive heating [both full-scale], and one chemical oxidation pilot test using permanganate). DNAPL/air dual-phase extraction performed poorly with less than 10% mass removal. The low-temperature six-phase heating pilot test also performed poorly in terms of mass removal; however, it should be noted that this pilot test was terminated prematurely in order to focus on another remedial technology. Therefore, the extent of DNAPL mass removal does not necessarily reflect what may have been achieved if the pilot test had continued.

Table D.10 also contains a listing of other relevant site characteristics. Of the four sites where 100% source mass removal was believed to be achieved, rebound was not evaluated at two (50%) of the sites; the DNAPL was present only as residual in three (75%) of the sites and had an unknown distribution at the fourth. Two (50%) of the sites were full-scale treatments (one with post-treatment monitoring ongoing) and two (50%) of the sites were pilot-scale treatments.

Techno	Technology		25 to 80% Mass Removal	< 25% Mass Removal	TOTAL SITES
Bioremediation		1 (100%)			1
Chamical Oridation	Fenton's Reagent	4 (100%)			4
Chemical Oxidation	Permanganate	2 (50%)	2 (50%)		4
Dual-Phase Extraction DNAPL/air				1 (100%)	1
Excava	Excavation				3
Surfactant I	Flushing	1 (100%)			1
	Conductive	1 (100%)			1
Thermal	Low-Temp six- Phase			1 (100%)	1
	Six-Phase	1 (100%)			1
	Steam Flushing		1 (100%)		1
ZVI	EZVI		1 (100%)		1
TOTALS	SITES	14 (70%)	4 (20%)	2 (10%)	20

Table D.11: Summary of DNAPL Mass Removal Effectiveness

1.3.4 Mass Flux Decrease

Figure D.11 and Table D.12 summarize the mass flux decrease by primary and secondary technologies respectively; Table D.12 also contains a listing of other site characteristics. Table D.13 summarizes the distribution of mass flux decrease by technology. There were a total of 21 sites with mass flux decrease data (Table D.13). The majority (62%) of sites had a mass flux decrease greater than 80%. In comparison, 86% of sites had a mass flux decrease greater than 50%. The only technologies that did not achieve >80% mass flux decrease in at least half of the sites were excavation and EZVI. Two sites had no mass flux decreases (chemical oxidation using permanganate was applied at one, excavation the other). Both of these technologies were targeting DNAPL in a predominantly fractured bedrock environment. The excavation was ongoing. Of these 13 sites with >80% decrease, rebound was evaluated at only three (23%) sites (two with no rebound, one with rebound), the DNAPL distribution was unknown for 8 (62%) sites and present as residual at 4 (31%) sites and residual and sorbed at 1 (8%) site.

Tashn	logy		Observed Ma	ss Flux Decrea	ase	TOTAL	
Techno	biogy	0%	41 to 60%	61 to 80%	81 to 100%	SITES	
Bioremediation				2 (50%)	2 (50%)	4	
Chemical Oxidation	Fenton's Reagent				2 (100%)	2	
Chemical Oxidation	Permanganate	1 (25%)		1 (25%)	2 (50%)	4	
Excavation		1 (50%)		1 (50%)		2	
Surfactant	Flushing			1 (50%)	1 (50%)	2	
Thermal	Six-Phase				5 (100%)	5	
Therman	Steam Flushing				1 (100%)	1	
ZVI	EZVI		1 (100%)			1	
TOTAL	SITES	2	1	5	13	21	

Table D.13: Summary of Observed Mass Flux Decrease

1.3.5 Occurrence of Rebound of Dissolved Phase Concentrations

Figure D.12 and Table D.14 summarize the rebound occurrence by primary and secondary technologies respectively. Table D.15 summarizes the distribution of rebound occurrence by secondary technology. There were a total of 21 sites with rebound data (Table D.15). The majority (67%) of sites did not have rebound occur. The technologies where rebound did occur were permanganate (83% of 6 sites), excavation (50% of 2 sites), and low-temperature six-phase heating (100% of 1 sites). Again, it should be emphasized that the low-temperature six-phase heating site was a pilot test that was terminated prematurely; therefore, the occurrence of rebound does not necessarily reflect the result of a completed application

Table D.14 contains a listing of other related site characteristics along with the rebound and technology data. Of the seven sites with rebound, four had DNAPL present in pools and two had DNAPL diffused into low K layers. However, of the 14 sites with no observed rebound, three sites had DNAPL in pools and four had DNAPL diffused into low K layers so DNAPL distribution may not be a good indicator on its own for evaluating the difficulty of achieving success.

Te		Rebound	Occurred	TOTAL
Ie	chnology	yes	no	SITES
Bior	emediation		3 (100%)	3
Chemical	Fenton's Reagent		4 (100%)	4
Oxidation	Permanganate	5 (83%)	1 (17%)	6
Ex	Excavation		1 (50%)	2
	Conductive		1 (100%)	1
	Low-Temp six- Phase	1 (100%)		1
Thermal	Resistive			
	Six-Phase		1 (100%)	1
	Three-Phase			
	Steam Flushing		1 (100%)	1
71/1	EZVI		1 (100%)	1
ZVI	ZVI		1 (100%)	1
тот	AL SITES	7	14	21

Table D.15: Summary of Occurrence of Rebound of Dissolved Phase

 Concentrations

1.3.6 Achievement of Remedial Goals and Site Closure

Survey respondents were only asked to evaluate success based on source mass removal, mass flux decrease, rebound and perceived success. Subsequent to the closure of the survey, an email request was sent out to the 53 participants who provided success data asking them the following questions:

- *i)* State whether they achieved dissolved phase concentrations reduced to less than maximum concentration levels (MCLs) upon completion of the remediation scheme.
- *ii)* If MCLs were achieved, state the stage of treatment (*i.e.*, pilot test versus full-scale) and the length of post treatment monitoring.
- *iii)* If MCLs were not achieved, were remedial goals met?
- *iv)* If MCLs were not achieved, but regulatory closure was, state how regulatory closure of the site was achieved.

In total, 8 sites had responses to these questions. Although out of the 53 sites, 4 sites were believed to have had 100% source mass removal and 13 sites had greater than 81% mass flux reduction, only one site claimed to have achieved MCLs after remediation. The site where MCLs were achieved was an excavation application with silt as the dominant lithology; however, the site did not achieve regulatory goals with respect to cis 1,2-DCE which has an MCL of 70 ppb and a site clean up goal of 5 ppb. The post remediation monitoring is ongoing, several years after the completion of remediation.

As a number of the sites were pilot scale demonstrations, achieving MCLs and site closure would not be expected at that stage of remediation. Of the 8 responses to the questions about MCLs, five stated that the pilot scale remediation attempts did meet the project goals, which were set above MCLs. Of these, two were bioremediation applications, one was a dual phase application, one was a permanganate application that is now going to full scale and one is an application of six phase heating that is now going to full scale and one was a full scale application which did not achieve remedial goals or MCLs and one was a full scale application of bioremediation were MCLs and regulatory closure are expected to be achieved within three years.

1.3.7 Impacts to Secondary Groundwater Quality

Figure D.13 summarizes the responses to observed secondary groundwater quality impacts sorted on a per primary technology basis. Table D.16 summarizes secondary groundwater quality impacts by secondary technology. If we consider only the criteria that were selected for at least 50% of sites per technology, the following conclusions can be drawn:

- *i)* **Generation/mobilization of undesirable compounds** (*e.g.*, methane, hydrogen sulfide, dissolved iron, manganese and other metals mobilization): Primarily impacted bioremediation, chemical oxidation using permanganate (manganese generation, dissolved metals, permanganate, chloride) and Fenton's reagent (metals mobilization, with hexavalent chromium observed at one site), EZVI (methane generation due to stimulated bioactivity), and dual-phase extraction.
- *ii)* Changes in groundwater aesthetic parameters (*e.g.*, increased biochemical oxygen demand, pH change, increased dissolved solids and change in groundwater color): Primarily impacted bioremediation, and chemical oxidation using permanganate (change in groundwater color only).
- *iii)* Changes in soil hydraulic properties (*e.g.*, changes in hydraulic conductivity and reduced soil porosity): Primarily occurred during chemical oxidation using Fenton's reagent, conductive heating, and ZVI.

iv) Changes in DNAPL distribution (*i.e.*, decrease/increase in source area volume and redistribution of DNAPL): Redistribution of DNAPL occurred with Fenton's reagent, water/DNAPL dual-phase extraction, and excavation. No technology application caused an increase in the source area. Decreases in source area volume were primarily observed for all dual-phase extraction methods, excavation, ZVI, three-phase heating, and steam flushing.

In general, bioremediation, chemical oxidation (particularly permanganate) and dual phase extraction had the most secondary groundwater issues identified, although some of them were noted to be temporary (*i.e.*, groundwater color changes with the application of permanganate).

1.3.8 Overall Technology Performance

It is possible for remediation at a site to meet one or more of the success criteria, yet fail to meet other criteria. For example, a large amount of DNAPL mass may be removed from the subsurface, but rebound may occur, coupled with insignificant mass flux decrease. The optimal result from remedial activities would be a large amount of DNAPL mass removed, significant mass flux reduction, and no post-treatment rebound of dissolved phase concentrations. To evaluate the level of success that was achieved for various combinations of site conditions and technologies, the sites were sorted according to varying degrees of each success criteria. Sites with "unknown" responses to any of the success criteria eliminated the site from consideration. The following results were found:

- *i)* Greater than 80% mass removal AND a greater than 61% reduction in mass flux AND no observed rebound AND a perceived success: Only two sites met the most stringent of the success criteria. Bioremediation and chemical oxidation (permanganate) were applied at these sites. Both sites had sandy aquifers and were pilot-scale applications.
- 80% mass removal AND a greater than 61% reduction in flux AND perceived a success: Four sites met this criteria, with the technologies being chemical oxidation (permanganate two sites), bioremediation (one site) and excavation (one site). All four sites had sandy aquifers. All but the site with excavation were pilot-scale applications.
- *iii)* 80% mass removal AND a greater than 61% reduction: Six sites in total met this criteria, with the technologies applied being chemical oxidation (permanganate two sites), bioremediation (one site), excavation (one site), surfactant flushing (one site) and thermal six-phase heating (one site). All sites except for the surfactant flush had sandy

aquifers; the surfactant flush was conducted in a clay aquifer. Only the excavation and the thermal six phase heating were full-scale applications, the remaining were pilot-scale.

- 80% mass removal AND no observed rebound AND a perceived success: Seven sites met this criteria. These sites include the three chemical oxidation sites (one permanganate, two Fenton's reagent), one bioremediation, two thermal applications (steam and conductive heating) and one excavation. If the perceived success criteria is removed from this evaluation, another chemical oxidation application (Fenton's reagent) is included. The sites were all unconsolidated, and the lithologies comprised three sand, three clay and one silt sites. Five of the sites were completed full scale applications and two were pilot scale applications.
- W) Greater than 61% reduction in flux AND no observed rebound AND a perceived success: Two bioremediation, one chemical oxidation (permanganate), and one thermal (six phase heating) sites met this criteria (four in total). Two of the sites had sandy aquifers, the other two had clay. Two of the applications were full scale and two were pilot scale.

The two predominant trends that appear to be consistent throughout all sites that met all degrees of success criteria were:

- *i)* Not one of the "successful" sites had remediation applied in unconsolidated media. The prevalent lithology was sand, with some sites with predominantly clay.
- *ii)* The majority of the "successful" sites were applied at the pilot-scale. Only excavation and thermal technologies met even the least stringent of the criteria for full-scale applications.

2 Impact of Site Characteristics on Technology Application and Performance

The data was also analyzed to determine the impact of the following site characteristics on the technology application and performance:

- *i*) aquifer geology and lithology (Section 2.1);
- *ii)* hydrogeology (Section 2.2);
- *iii)* DNAPL source zone characteristics (Section 2.3); and

iv) DNAPL detection, sampling and mass estimation (Section 2.4).

Optimally, a full analysis would also include looking at the impact of these site characteristics on remedial costs and treatment duration. Unfortunately, insufficient cost and treatment duration data was collected to come to any meaningful conclusions (see Section 1.2); therefore, remedial costs and treatment duration are not discussed here.

2.1 Aquifer Geology and Lithology

Figure D.14 provides a breakdown of predominant lithology distribution in the DNAPL source zone by technology. Table D.17 provides a more detailed breakdown of the data by lithology sub-category (*e.g.*, metashale, limestone, sand, *etc.*) and secondary technology. In general, data was entered into the survey for only unconsolidated geologic media sites for the technologies dual phase extraction, pump and treat, and surfactant flushing. Similarly, data was collected for only unconsolidated sites for the technologies excavation, ZVI and thermal, with the exception of 1 site each where they were applied at a site with sedimentary rock. Both bioremediation and chemical oxidation were applied at the largest range of geologic media types, including both consolidated and unconsolidated media.

Table D.18 provides a detailed breakdown of the geology data (*i.e.*, degree of fracturing and rock matrix permeability for consolidated media, and degree of heterogeneity for unconsolidated media). Based on this data, all technologies except for pump and treat, some of the thermal technologies (*i.e.*, conductive heating, low-temp six-phase and six-phase heating), and the EZVI and ZVI/clay combination were performed at sites with at least some bedrock. It should be noted, however, that data was collected for only one site for each of these technologies except for six-phase heating. Again, bioremediation and chemical oxidation using permanganate have been applied over the widest variation in geological conditions.

Tables D.19 through to D.22 outline the breakdown of the various remedial success parameters by the predominant lithology in the source area. A number of interesting trends arise from these tables. For example, sites with consolidated media were generally described as follows:

- *i)* no successful fractured bedrock sites are reported (Table D.19);
- *ii)* no one entered data of estimates of the DNAPL source mass removal at a fractured bedrock site (Table D.20);
- *iii)* neither of the fractured bedrock sites had any reduction in mass flux (Table D.21); and

iv) the occurrence of post-treatment rebound was unknown (Table D.22).

In comparison, sites with predominantly unconsolidated media could be described as follows:

- *i)* two-thirds of 43 sites with unconsolidated media were perceived to be successfully remediated, with the remaining almost one-third perceived to be fairly successful (one poor success as well; Table D.19);
- 55% of 20 sites had a >90% reduction in the DNAPL source mass, with the majority of the remainder experiencing at least a 50% reduction (Table D.20);
- *iii*) the majority (68% of 19 sites) of the sites had a mass flux reduction of 80 to 100%, with only one site experiencing less than a 40% decrease (Table D.21); and
- *iv)* two-thirds of 20 sites did not experience post-treatment rebound (Table D.23).

As mentioned in Section 8 of Appendix C, these conclusions are based on a limited data set. Although attempts were made to compile a comprehensive database of information pertaining to DNAPL remediation, the data necessary to draw quantitative conclusion is very complex and difficult to collect.

Geologic	Predominan	t Lithology in	Number	of Sites W	ith Perceiv	ed Success
Media	Source Area		Poor	Fair	Success	Not Yet Evaluated
		Metabasalt				
	Metamorphic	Other Metamorphic				1
ted		Shale				
Consolidated	Sedimentary	Limestone (karstic)		1		
Con		Other Sedimentary				1
	Igneous	Granitic				
	Unknown					
	Gravel			2	1	2
ted	Sa	and		6	13	1
lida	Т	ïill			1	1
Unconsolidated	S	ilt	1	2	4	2
Unc	C	lay		2	8	1
	Unk	nown		2	1	

 Table D.19:
 Trends in Perceived Success With Lithology in Source Area

			Nu	mber	[•] of Si M	tes W lass R			f Sou	rce
Geologic Media	Predominant Lithology in Source Area		Unknown	< 10%	10 to 25%	25 to 50%	50 to 80%	80 to 90%	%06 <	100%
		Metabasalt								
	Metamorphic	Other Metamorphic	1							
ted		Shale								
Isolida	Consolidated Consolidated Consolidated	Limestone (karstic)	1							
Cor		Other Sedimentary								
	Igneous	Granitic								
	Unk	nown								
	Gra	avel	4							
ted	Sa	und	9			2	2	1	3	3
olida	Т	'ill	2							
Unconsolidated	S	ilt	2	1	1			2	1	
Unc	Clay		5						3	1
	Unk	nown	3							

Table D.20: Trends in Source Mass Removal With Lithology in Source Area

			Nur			Vith Le eduction	vel of N n	lass
Geologic Media			Unknown	%0	0 to 40%	40 to 60%	60 to 80%	80 to 100%
		Metabasalt						
	Metamorphic	Other Metamorphic		1				
ited		Shale						
Consolidated	Sedimentary	Limestone (karstic)		1				
Cor		Other Sedimentary						
	Igneous	Granitic						
	Unk	nown						
	Gra	avel	3					1
ted	Sa	and	11			1	3	5
olida	Т	ʻill	1					1
Unconsolidated	Silt		5				1	
Unc	C	lay	5				1	3
	Unk	nown						3

 Table D.21:
 Trends in Mass Flux Reduction With Lithology in Source Area

Geologic		t Lithology in		of Sites Wit	h Rebound
Media	Sourc	e Area	Unknown	Yes	No
		Metabasalt			
	Metamorphic	Metamorphic	1		
ted		Shale			
Consolidated	Sedimentary	Limestone (karstic)	1		
Con		Other Sedimentary			
	Igneous	Granitic			
	Unknown				
	Gr	avel	2	1	1
ted	Sa	and	10	4	6
lida	Т	fill	2		
Unconsolidated	S	Silt Clay		1	2
Unc	C				5
	Unk	nown	3		

Table D.22: Trends in Rebound With Lithology in Source Area

2.2 Hydrogeology

The breakdown of hydrogeological parameters by primary technology is illustrated in several figures as follows:

- *i*) Figure D.15 summarizes the hydraulic conductivity;
- *ii)* Figure D.16 summarizes the groundwater velocity; and
- *iii)* Figure D.17 summarizes the sustainable well yield.

Table D.23 provides a more detailed breakdown of the hydraulic conductivity, depth to groundwater, groundwater velocity, hydraulic gradient and sustainable well yield data by secondary technology.

The most prevalent hydrogeological characteristics of the sites at which DNAPL remediation has been attempted can be described as follows (Section 2, Appendix C):

- *i*) hydraulic conductivity (K) between 10⁻⁵ ft/day and 10 ft/day (49 of 80 sites, 61%);
- *ii)* groundwater velocity (v) between 10^{-3} and 1 ft/day (40 of 58 sites, 69%);
- *iii)* sustainable well yield between 1 and 100 gpm (29 of 49 sites, 59%);
- *iv)* depth to groundwater less than 50 ft below ground surface (bgs; 97 of 106 sites, 92%); and
- v) horizontal gradients within the range of 0.001 to 0.1 ft/ft. (37 of 59 sites, 63%), and small (<0.001 ft/ft) vertical gradients, indicating a predominance of horizontal groundwater flow at most DNAPL impacted sites surveyed.

Each remedial technology was applied at sites that generally fit into the above categories, with the following notable exceptions (Table D.23):

- *i)* **Bioremediation:** Also applied at a number of sites with highly conductive and permeable aquifers (*e.g.*, 4 of 7 sites had sustainable well yields greater than 100 gpm, with one site above 1,000 gpm; and 4 of 14 sites had v>1 ft/day).
- *ii)* Chemical oxidation using Fenton's reagent and permanganate: The *Fenton's reagent* sites could all be categorized as above. Notable exceptions for the *permanganate* sites include the application at: (*i*) larger depths (*e.g.*, 2 of 13 sites with groundwater depths between 50 to 200 ft bgs); and (*ii*) sites with permeable and conductive aquifers (*e.g.*, 1 of 6 sites with v>1 ft/day; 3 of 10 sites with K>10 ft/day; 2 of 7 sites with a horizontal hydraulic gradient between 0.1 and 1 ft/ft; and 1 of 6 sites with a sustainable well yield between 100 to 1,000 gpm).
- *iii)* **Dual-phase extraction:** *Water/air* and *water/DNAPL* extraction were also employed at lower conductivity sites (*e.g.*, both *water/air* and *water/DNAPL* had one site with a sustainable well yield <1 gpm; *water/DNAPL* had one site with a horizontal gradient between 0.0001 to 0.001 ft/ft). *DNAPL/air*, in comparison, was applied at more conductive sites (*e.g.*, one site with K>10 ft/day, and one site with a sustainable well yield between 100 to 1,000 gpm).

- *iv)* **Excavation:** Excavation was also applied at 1 (of 9) sites with a deeper groundwater table (50 to 100 ft bgs), and both low permeability (*e.g.*, one site each with $K < 10^{-5}$ ft/day and sustainable well yield <1 gpm) and high permeability aquifers (*e.g.*, two [of 6] sites with K > 10 ft/day and 1 [of 6] sites with a sustainable well yield between 100 to 1,000 gpm).
- *v)* **Surfactant flushing:** Also applied at sites with higher conductivity aquifers (*e.g.*, one [of 3] sites had *K*>10 ft/day, and one [of 2] sites with sustainable well yields between 100 and 1,000 gpm).
- vi) Thermal technologies: Resistive, six-phase, and three-phase heating, as well as steam flushing have all been applied at sites with more permeable aquifers (e.g., sustainable well yields as high as 100 to 1,000 gpm, and v>1 ft/day at 1 site each for three-phase heating and steam flushing, K>10 ft/day at one to two sites each for all four technologies).
- *vii)* **ZVI:** EZVI was applied at a high conductivity site (K>10 ft/day), while ZVI was applied at one low conductivity site (K<10-5 ft/day).

Table D.24 outlines the breakdown of the perceived remedial success by the hydrogeology parameter. No trends in the data are discernible, likely due more to a lack of data than a lack of impact of the hydrogeology on the remedial success. The breakdown of the remainder of the success parameters is not shown, since the data available for these factors are even fewer in number.

		Number	r of Sites Wi	th Perceived	Success
Hydrogeolo	ogy Parameters	Poor	Fair	Success	Not Yet Evaluated
	< 50 ft		8	21	3
Depth to Groundwater	50 to 200 ft		1	1	
	> 200 ft				
	< 0.001 ft/day		1	2	
Groundwater Velocity	0.001 to 1 ft/day		4	8	
	> 1 ft/day			1	
	< 10 ⁻⁵ ft/day			4	
Hydraulic Conductivity	10 ⁻⁵ to 10 ft/day		4	9	2
Conductivity	> 10 ft/day		1	2	
Horizontal	< 0.001		1	3	
Hydraulic	0.001 to 0.1		1	5	1
Gradient	> 0.1		1	1	
	< 10 gpm		2	2	
Sustainable Well Yield	10 to 100 gpm		1		1
	> 100 gpm				

Table D.24: Trends in Perceived Success With Hydrogeology Parameters

2.3 DNAPL Source Zone Characteristics

The DNAPL source zone characteristics, which included DNAPL composition, size of the source zone (areal extent, volume and depth), shape of the source zone, and the DNAPL distribution in the subsurface, were evaluated on a technology specific basis.

2.3.1 Source Zone Geochemistry

The distribution of DNAPL composition was evaluated for each of the technologies. Data was collected from a total of 91 sites. Figure D.18 summarizes the DNAPL composition distribution by technology. Bioremediation, thermal, excavation and dual phase extraction were applied at sites contaminated with all four groups of DNAPLs. ZVI and surfactant flushing were applied at sites with only chloroethene contamination. Chemical oxidation was applied at sites with chloroethenes, chloroethanes, and chloromethanes as components of the DNAPL.

The distribution of groundwater chemistry was also evaluated for each of the The breakdown by technology varies slightly from the groundwater technologies. chemistry breakdown. This could be due to different components in the dissolved phase then in the DNAPL phase or it could also be due to the respondents answering "unknown" to one of the questions and providing an answer to the other, changing the percentages when broken down by technology. There are a total of 80 sites with groundwater chemistry data. Figure D.19 summarizes the groundwater chemistry Bioremediation, chemical oxidation and dual phase distribution by technology. extraction were applied at sites contaminated with all four groups of contaminants. ZVI and surfactant flushing (only one application with groundwater chemistry data) were applied at sites with only chloroethene contamination. Thermal and excavation were applied at sites with chloroethenes, chloroethanes, and chloromethanes. The two sites without chloroethenes were sites impacted by chloroethanes and at each of these sites remediation was undertaken by chemical oxidation using Fenton's Reagent.

There are no obvious trends in the success data in terms of the DNAPL or groundwater chemistry data (Table D.25).

Table D.26contains a detailed breakdown of the groundwater geochemistry (non-VOC data) by secondary technology. Little geochemistry data was provided; therefore, no discernible trends can be seen.

2.3.2 DNAPL Source Zone Areal Extent

Figure D.20 summarizes the distribution in DNAPL areal extent by technology. On a per technology basis, bioremediation and chemical oxidation have been applied at the largest range in areal extent. Of the 7 sites with areal extents of source zones greater than 1,000,000 ft², three of the sites were remediated using excavation, two with dual phase extraction and one each with bioremediation and thermal.

There are no obvious trends in the success data in terms of the areal extent. Of the four sites with an areal extent of 1 to 10 ft, all were considered successful. Of those with data, there was >90% mass removal and 81 to 100% mass flux decrease. However, at all the other areal extent ranges, remediation was considered either successful or a fair success. For the site with the largest area that had success data (100,000 to <1,000,000 ft²), the application was considered successful although there was only >10 to 25% mass removal and rebound did occur (Table D.27). It should be noted; however, that at this site the low-temperature six-phase heating demonstration was terminated before completion in order to focus on another remediation technology application at the site.

2.3.3 DNAPL Source Zone Volume

Figure D.21 summarizes the distribution in DNAPL volume data by technology. On a per technology basis, chemical oxidation and excavation have been applied at sites with the largest range in volumes of DNAPL impacted soil, followed by bioremediation and dual phase extraction. Based on the data provided from the survey, thermal treatments have been mainly applied at sites with greater than 100,000 ft³ volume of DNAPL impacted soils. In contrast, of the two surfactant flushing sites with a known volume of DNAPL impacted soil, both were at sites with between 10 and 100 ft³.

There were no trends in the success data with respect to DNAPL volume.

2.3.4 Maximum DNAPL Depth

Figure D.22 summarizes the distribution in maximum DNAPL depth data by technology. On a per technology basis, bioremediation has been applied at sites with the greatest depths of DNAPL distribution with three sites having depths between 100 and 1000 ft bgs and one site having a DNAPL depth greater than 1,000 ft bgs. Chemical oxidation, thermal, and excavation have been applied at sites with DNAPL depths between 1 and 1,000 ft bgs.

There are no trends in the success data with respect to maximum DNAPL depth (Table D.28) with the exception that for all but three sites with success data, the maximum DNAPL depth is <100 ft. Of those three sites one site had DNAPL between 100 and 1000 ft and was considered a success with >90% mass removal using a thermal (steam) application, one was considered a fair success with 81 to100% mass flux reduction using bioremediation, and at the third site, a chemical oxidation using permanganate site with DNAPL at a depth of 100 to 1000 ft, success has not yet been evaluated; however, to date there has been 0% mass flux reduction.

2.3.5 DNAPL Distribution in Subsurface

Figure D.23 summarizes the range in DNAPL distribution in the subsurface by technology. On a per technology basis, surfactant flushing was applied at sites with DNAPL present as either pools or at residual saturation, where as bioremediation, chemical oxidation, dual phase extraction and excavation were attempted at sites with DNAPL distributed in all of the classifications.

Table D.29 summarizes the remediation success factors for the various DNAPL distribution parameters. There are no obvious trends with respect to the DNAPL distribution data.

2.3.6 DNAPL Shape in Subsurface

Figure D.24 summarizes the range in DNAPL source zone shape in the subsurface by technology. This question was asked to get an idea of the level of characterization that had been conducted on the source zones. On a per technology basis, surfactant flushing was applied at sites with DNAPL present as either pools or at residual saturation, where as bioremediation, chemical oxidation, dual phase extraction and excavation were attempted at sites with DNAPL distributed in all of the classifications.

2.3.7 DNAPL Detection, Sampling and Mass Estimation

Table D.30 provides a summary of the relative frequency that each sampling method is used at a site, for only the sites where at least one method was specified. The relative percentage is described by the maximum, minimum, average and median usage percentages. No one method was used at all sites. Only 14 of 123 sites used only one sampling method, of which 8 sites used only fully-screened monitoring wells; 3 sites used only nested monitoring wells; 3 sites used only soil sampling; and 0 sites used only depth-discrete sampling. The median and average relative frequencies indicate that soil sampling is the most frequently chosen sampling method.

Also included in Table D.30 is a description of the number of sampling locations used at each site, also described by the maximum, minimum, average and median number of sampling locations. As can be seen by these numbers, large numbers of sampling locations have been used at some sites; however, the average numbers are generally much smaller.

Frequency	Sampling Method	Frequ	ency of Use of	Sampling M	ethods
Metric	Sampning Wiethou	Minimum	Maximum	Median	Average
	Fully-Screened Monitoring Wells	0.0%	100.0%	22.7%	29.5%
Relative Percentage of Use on Each Site	Nested Monitoring Wells	0.0%	100.0%	2.8%	15.0%
	Depth-Discrete Samples	0.0%	98.5%	0.0%	19.6%
	Soil Samples	0.0%	100.0%	44.4%	35.8%
	Fully-Screened Monitoring Wells	0	700	6	36
Overall	Nested Monitoring Wells	0	50	3	10
Usage	Depth-Discrete Samples	0	1300	0	73
	Soil Samples	0	1500	20	71

 Table D.30:
 Number of Sampling Locations Used to Locate the DNAPL Source Area(s)

Table D.31 outlines the correlation between the various remedial success parameters and the sampling method used to detect and locate the DNAPL source area. While it should be noted that there was minimal data available to analyze, there are some interesting trends that are illustrated in this table, including:

- *i)* at the sites where the treatment was perceived to be successful, the average method for locating the DNAPL source area consisted of an approximately even mixture of fully-screened monitoring wells, nested monitoring wells, depth-discrete samples and soil samples. Conversely, the less successful sites predominantly used fully-screened monitoring wells on average;
- *ii)* the sites most successful at removing DNAPL mass from the source area tended to have soil samples being the predominant method for DNAPL detection. In comparison, the sites where less DNAPL mass was removed tended to rely more heavily on fully-screened monitoring wells; and
- *iii)* There does not appear to be any trend in the mass flux reduction or rebound occurrence with DNAPL sampling method.

		Average Per	cent of Sampl	ing Locatior	ıs per Site		
Success Metric	Criteria	Fully- Screened Monitoring Wells	Nested Monitoring Wells	Depth- Discrete Sampling	Soil Samples	Total Sites	
	Poor					0	
Perceived Success	Fair	61.9%	8.6%	6.9%	22.6%	5	
	Success	26.8%	22.4%	23.9%	26.9%	19	
	<10%					0	
DNAPL	10 to 25%	1.1%	0.0%	93.3%	5.6%	1	
	25 to 50%	64.4%	4.8%	1.9%	28.8%	2	
Mass	50 to 80%	64.7%	33.3%	0.0%	0.0%	1	
Reduction	80 to 90%	6.9%	5.0%	30.3%	57.8%	2	
	>90%	3.7%	17.0%	31.5%	47.8%	3	
	100%	33.7%	12.5%	12.5%	41.3%	2	
	0%					0	
	0 to 40%					0	
Mass Flux Reduction	40 to 60%	28.8%	9.6%	3.8%	57.7%	1	
1.coulouron	60 to 80%	50.0%	5.0%	15.0%	30.0%	2	
	80 to 100%	54.5%	0.0%	3.7%	41.8%	3	
Rebound	Yes	35.8%	8.7%	26.9%	28.7%	5	
Occurrence	No	32.5%	7.6%	14.7%	45.2%	8	

Table D.31: Correlation Between Treatment Success and DNAPL Sampling Method

Note: Blank spaces denote a lack of data

TABLE D.1: SUMMARY OF TECHNOLOGY SELECTION CRITERIA DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Naiy ROCS

						N	MBER OF SIT.	ES THAT MA	TCH THE TEC	TOLOGY STATEMENT	SELECTION	NUMBER OF SITES THAT MATCH THE TECHNOLOGY SELECTION CRITERIA PER TECHNOLOGY	3 TECHNOLO	GY							
TECHNOLOGY SELECTION	BIOREMEDIATION	CHEMI	CHEMICAL OXIDATION		co-	1-JAUL-F	DUAL-PHASE EXTRACTION		EXCAVATION	OTHER		SURFACTANT			THERMAL	١L			ZERO-VAL	ZERO-VALENT IRON	TOTAL
CRITERIA		Fenton's Reagent	Permanganate	Ozone	SOLVENT FLUSHING	Water/Air V	Water/DNAPL DNAPL/Air		_	Undefined	Pump and Treat	FLUSHING	Conductive Low-Temp Heating Six-Phase		Resistive Heating	Six-Phase Heating	Three- Phase Heating	Steam Flushing	Emulsified With Clay ZVI only	Clay ZVI	
Cost of technology application	16 (88.9%)	5 (62.5%)	8 (57.1%)			4 (66.7%)	2 (66.7%)	1 (100.0%)	5 (83.3%)	1 (50.0%)		1 (50.0%)	1 (100.0%)	1 (100.0%) 1 (100.0%)		5 (55.6%)	1 (33.3%)			1 (50.0%))%) 52 (61.9 %)
Technology perceived to provide a good chance at remedial success	15 (83.3%)	8 (100.0%)	9 (64.3%)			4 (66.7%)	3 (100.0%)		4 (66.7%)	1 (50.0%)		2 (100.0%)	1 (100.0%)	1 (100.0%) 1 (100.0%)		8 (88.9%) 3	3 (100.0%) 4 (80.0%)		1 (100.0%)	2 (100.0%)	0%) 67 (79.8%)
Remediation timeframe was a factor	6 (33.3%)	3 (37.5%)	6 (42.9%)			2 (33.3%)	2 (66.7%)		3 (50.0%)				1 (100.0%)	1 (100.0%)		2 (22.2%)		2 (40.0%) 1 (100.0%)	(100.0%)		29 (34.5%)
Regulatory pressures guided technology choice	1 (5.6%)		3 (21.4%)			1 (16.7%)	1 (33.3%)				1 (100.0%)										7 (8.3%)
Impact of remedial technology on surface water was minimal	3 (16.7%)	1 (12.5%)	2 (14.3%)				1 (33.3%)														7 (8.3%)
Impacts of remedial technology on secondary groundwater quality was minimal	3 (16.7%)		2 (14.3%)			1 (16.7%)	1 (33.3%)		1 (16.7%)					1 (100.0%)							9 (10.7%)
Technology application was perceived to reduce risk to the environment	4 (22.2%)	1 (12.5%)	5 (35.7%)			2 (33.3%)	3 (100.0%)	1 (100.0%)	3 (50.0%)				1 (100.0%)	1 (100.0%)		1 (11.1%)	1 (33.3%)				17 (20.2%)
Technology application was thought to have a lasting impact on the water quality	9 (50.0%)	3 (37.5%)	3 (21.4%)			2 (33.3%)		3 (300.0%)	1 (16.7%)	1 (50.0%)		1 (50.0%)		1	(100.0%)	1 (100.0%) 1 (11.1%) 2 (66.7%) 1 (20.0%)	2 (66.7%) 1	(20.0%)			31 (36.9%)
Health and safety concerns guided technology choice	6 (33.3%)	1 (12.5%)	2 (14.3%)			1 (16.7%)		1 (100.0%)													11 (13.1%)
Technical implementability of the remedial technology was thought to be appropriate	8 (44.4%)	5 (62.5%)	5 (35.7%)			3 (50.0%)	1 (33.3%)	2 (200.0%)	1 (16.7%)	1 (50.0%)		1 (50.0%)					2 (66.7%) 2 (40.0%)	(40.0%)			34 (40.5%)
Legal pressures guided technology choice			2 (14.3%)					1 (100.0%)			1 (100.0%)										4 (4.8%)
Pressure from various stakeholders guided technology choice	3 (16.7%)		3 (21.4%)			1 (16.7%)	1 (33.3%)	1 (100.0%)			1 (100.0%)				1 (100.0%)						11 (13.1%)
Other ^a	4 (22.2%)		3 (21.4%)						1 (16.7%)	1 (50.0%)							_	1 (20.0%)	1 (100.0%)	1 (50.0%))%) 12 (14.3 <i>%</i>)
TOTAL SITES WITH DATA	18	8	14	0	0	9	3	1	6	2	-	2	-	-	1	6	3	5	1	0 2	84

Nets Black sports dicate no survey respondent selected the corresponding factors for that technology application. "Other reasons for technology selection included the following: Black sports are technology demonstration projects, and now were performed inside dry cleaners Cherrical condition or work created on the domonstration projects, and now were performed inside dry cleaners Cherrical condition or work concerned in the domonstration projects, and now were performed inside dry cleaners Cherrical condition or work are choology demonstrations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situations Excavation - Low permeability of said (sith) tradeed any in-situation Excavation - Low permeability of said (sith) tradeed any in-situation Steam fusition - Line steam (so on site in succes) Excavation - Line steam steak) variable at the site. No cost to application steam succes. "EXViruanes are into - East section of Science Thesis. Son and benconte sharty. Referenced work by Sharon Wadey UW Masters Thesis, 2002. Other references also available." ZVI - University of Waterion Masters of Science Thesis. Son Science Thesis.

TABLE D.2: SUMMARY OF REMEDIAL COST DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY, SCALE OF REMEDIATION AND TREATMENT SIZE NAVY ROCS

TOTAL STIFS	WITH COST	DATA	3	0	0	0	0	0	3	2		2	2	5	-	10	2	0	0	0	_	0	3	16	4	-	-	0	0	0	12	3	8	2	0	0	0	13	3	2	-	0	0	0	9	31
NOG	KON	ZVI only	-						1							0							0	-	-	-					1							0							0	1
1 149 1 1 1 1	ZERO-VALENT IRON	Emulsified With Clay ZVI only							0							0							0	0							0							•							0	0
2500	ZERO	Emulsified							0							0							0	0							0							•							0	0
		Steam Flushing							0				-	-		2							0	2							0							•							0	0
	8	Three- Phase Heating							0				-			1							0	_							0		-					-							0	1
	IMAL	Resistive Six-Phase Heating Heating							0			-		-		2							0	2							0	1	9	-				w							0	5
TERMONT	THEN	p Resistive Heating							0							0							•	•							0	_		-				-							0	1
ĸ		Conductive Low-Temp Heating Six-Phase							0							0							•	0							0	1						-							0	-
CHNOLOG		Conductiv Heating							0			-				1							0	-							0							•							0	0
NUMBER OF SITES THAT COULD BE DESCRIBED BY REMEDIAL COST PER TECHNOLOGY	SURFACTANT	FLUSHING							0							0							0	0							0							0							0	0
REMEDIAI		Pump and Treat							0							0					-		1	-							0							•							0	0
CRIBED BY	OTHER	Undefined							0	_	-					1							0	_							0							•							0	0
ULD BE DESC	EXCAVATION	-	1						1							0							0	-	-	-					1							•							0	-
THAT CO		NAPL/Air							0							0							0	0							0							0							0	0
MBER OF SU	DUAL-PHASE EXTRACTION	Water/Air Water/DNAPL DNAPL/Air				-			0	1	_				1	1	_						0	_							0		_	_	_			0						_	0	0
IN 1711N	HA-TVND	Vater/Air W				_		-	0	_	_					0	_			-			0	0		-					1		_	_	_			0				-		_	0	1
00	9	SOLVENT FLUSHING V							0							0							0	0							0							•							0	0
-		Ozone F							0							0							0	0							0							0							0	0
	CHEMICAL OXIDATION	Permanganate	1						1	1	_					1							0	2	~	, ,					5		-					-		-					1	7
10 INALIO	CHEMICA	Fenton's P. Reagent				_			0		_					0	-						_	_	-	-					1		2	_	_			7	1					_	1	4
NOW FINANCIA	BIOREMEDIATION								0							0	-						-	-	,	4	-				3	1	-					7	2	-	-				4	6
7		COSTS (USD)	< \$0.5M	\$0.5 to \$1M	\$1M to \$2M	S2M to \$4M	S4M to \$10M	> \$10M	Total	<\$0.5M	\$0.5 to \$1M	\$1M to \$2M	S2M to S4M	S4M to \$10M	> \$10M	Total	<\$0.5M	\$0.5 to \$1M	\$1M to \$2M	S2M to S4M	S4M to \$10M	> \$10M	Total		- 60 EM	SO 5 to \$1M	SIM to S2M	S2M to S4M	S4M to S10M	> \$10M	Total	<\$0.5M	\$0.5 to \$1M	S1M to S2M	S2M to S4M	S4M to \$10M	> \$10M	Total	<\$0.5M	S0.5 to S1M	SIM to \$2M	S2M to \$4M	S4M to \$10M	> \$10M	Total	
SIZE OF R				St		Small \$2	\$4 [.]				St	\$1	Large S2	S4.				S	SI	Unknown S2			L	TOTAL		8	S	Small \$2			L		St		Large S2	S4.				1×	SI	Unknown \$2				TOTAL
	SCALE OF	FREATMENT								1			Full-Scale											1												Pilot-Scale			1							

Notes "DNAPL zone sizes are classified as follows: "DNAPL zone sizes are classified as follows: "Small: area < 10,000 ft², volume > 100,000 ft² Large: area > 10,000 ft², volume > 100,000 ft² Blank spaces denote no data entered for that parameter.

TABLE D.3: SUMMARY OF REMEDIAL COST DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY, SCALE OF REMEDIATION AND TREATMENT SIZE Nary ROCS

		-																					
COST	COST	SCALE OF	BIOREMEDIATION		CHEMICAL OXIDATION		co-	DUAL-P	DUAL-PHASE EXTRACTION		EXCAVATION	OTE	OTHER S	SURFACTANT			THERMAL	IVI			ZERO-VALENT IRON	T IRON	WITH COST
CATEGORY BF	BREAKDOWN	TREATMENT		Fenton's Reagent	Permanganate Ozone		SOLVENT FLUSHING	Water/Air W	ater/Air Water/DNAPL DNAPL/Air	DNAPL/Air		Undefined	Pump and Treat	FLUSHING	Conductive Heating	Low-Temp Six-Phase	Resistive Six-Phase Heating Heating		Three- Phase F	Steam Flushing	Emulsified With Clay ZVI only	ay ZVI only	BREAKDOWN DATA
		Full-Scale	3	4	3			_	2		3	-						-	e 	4		-	24
	0 to 10%	Pilot Test	9	3	2			-		1	1			2		1	_	5	-	-		-	29
		Total	12	7	5	0	0	2	2	1	4	1	0	2	0	1	1	6	2	5	0 0	2	53
		Full-Scale	0		0			-	1		0				1			1		0			3
	10 to 20%	Pilot Test	-	0	4			-		0	0			0			0	2	0	0			×
	<u>ı </u>	Total	-	0	4	0	0	-	1	0	0	0	0	0	1	0	0	3	0	0	0 0	0	Π
		Full-Scale																					0
	20 to 30%	Pilot Test	4	0	2			_		0	-			0			0	0	0	0			~
	L	Total	4	0	2	0	0	_	0	0	_	0	0	0	0	0	0	0	0	0	0 0	0	8
<u> </u>		Full-Scale																					0
	30 to 40%	Pilot Test			-				ľ									t	-				0
		Total	0	0	-	0	0	•	0	0	0	0	0	0	0	0	0	0	-	0	0	•	2
1	T	Eull-Scale			-	,	,	,	,	,	,	,	,		,	,	ļ	,		,		, ,	•
a	40 to 50%	Pilot Test		-													ſ						-
Costs		Total	0	_	-	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	2
<u> </u>		Eull Coolo							•	•			-				,		,	,		•	-
	50 to 60%	Full-State Dilot Tast							Ť				-					t					
		Total	0		U		0				U		-	•		•	U			•	0	•	-
1	I	Fold Colo	-		•	•	•	-	-	•	-	•	-	•		•		•		•		•	
	60 to 70%	Full-Scale Dilot Tast							T			T						t					
		Total	U				U		U		U		•	•		ď				•	0	•	
		Evil Coolo		•	•	-	•		-					•		•		-	•	•		•	
	70 to 80%	Pilot Test							Ì			t						t	+				
		Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
L		Full-Scale																					0
	80 to 90%	Pilot Test																					0
	I	Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0
		Full-Scale	2	4	4			-	3		3	-			1			2	-	4		1	27
	0 to 10%	Pilot Test	6	4	4			2		_	2			2		-	-	7	2	-			36
	ı	Total	П	8	8	0	0	3	3	-	5	-	0	2	1	1	-	6	3	s S	0 0	1	63
		Full-Scale											1										1
	10 to 20%	Pilot Test	4		-			-															9
	L	Total	4	0	-	0	0	_	0	0	0	0	-	0	0	0	0	0	0	0	0 0	0	7
<u>I</u>		Full-Scale	-																				1
	20 to 30%	Pilot Test			4																	-	S
	L	Total	_	0	4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	-	9
L		Full-Scale																					0
	30 to 40%	Pilot Test	-																				-
		Total	1	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	1
		Full-Scale						-										_					0
Post-1 reatment	40 to 50%	Pilot Test																					0
	ı	Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0
		Full-Scale																					0
	50 to 60%	Pilot Test																					0
	1	Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0
		Full-Scale						-	-			-						_	_				0
	60 to 70%	Pilot Test															-						0
	1	Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0
L		Full-Scale																					0
	70 to 80%	Pilot Test																					•
	ı	Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0
1		Full-Scale																					0
	80 to 90%									-										-			
	0.01 00 00	Pilot test	_					t	t									+	+				0

TABLE D.3: SUMMARY OF REMEDIAL COST DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY, SCALE OF REMEDIATION AND TREATMENT SIZE Nary ROCS

MEMON: MEANON: MEANON: <th< th=""><th>E</th><th>COST</th><th>SCALE OF</th><th>BIOREMEDIATION</th><th></th><th>CAL OXIDATIO</th><th></th><th>co-</th><th>DUAL-P</th><th>HASE EXTRA</th><th></th><th>XCAVATION</th><th>OTH</th><th></th><th>AURFACTANT</th><th></th><th></th><th>THERMAL</th><th></th><th></th><th>ZERO-VA</th><th>ZERO-VALENT IRON</th><th>WITH COST</th></th<>	E	COST	SCALE OF	BIOREMEDIATION		CAL OXIDATIO		co-	DUAL-P	HASE EXTRA		XCAVATION	OTH		AURFACTANT			THERMAL			ZERO-VA	ZERO-VALENT IRON	WITH COST
Hole File File <th< th=""><th>ORY</th><th>BREAKDOWN</th><th>N TREATMENT</th><th></th><th></th><th>Permanganate</th><th></th><th></th><th>Water/Air V.</th><th>Vater/DNAPL</th><th>DNAPL/Air</th><th></th><th></th><th></th><th></th><th>Conductive I Heating</th><th>ow-Temp F</th><th>Resistive Six-Phase Heating Heating</th><th></th><th>Three-Steam Phase Flushing</th><th>g Emulsified With Clay ZVI only</th><th>th Clay ZVI o</th><th></th></th<>	ORY	BREAKDOWN	N TREATMENT			Permanganate			Water/Air V.	Vater/DNAPL	DNAPL/Air					Conductive I Heating	ow-Temp F	Resistive Six-Phase Heating Heating		Three-Steam Phase Flushing	g Emulsified With Clay ZVI only	th Clay ZVI o	
Motor Motor <th< td=""><td></td><td></td><td>Full-Scale</td><td>2</td><td>3</td><td>2</td><td></td><td></td><td>1</td><td>3</td><td></td><td>3</td><td></td><td>1</td><td></td><td>1</td><td></td><td></td><td>1</td><td>4</td><td></td><td></td><td>22</td></th<>			Full-Scale	2	3	2			1	3		3		1		1			1	4			22
Notion 1 0 <td></td> <td>0 to 10%</td> <td>Pilot Test</td> <td>13</td> <td>3</td> <td>7</td> <td></td> <td></td> <td>3</td> <td></td> <td>-</td> <td>1</td> <td></td> <td></td> <td>2</td> <td></td> <td>1</td> <td>1</td> <td></td> <td></td> <td></td> <td></td> <td>42</td>		0 to 10%	Pilot Test	13	3	7			3		-	1			2		1	1					42
Matrix I <td></td> <td></td> <td>Total</td> <td>15</td> <td>6</td> <td>6</td> <td>0</td> <td>0</td> <td>4</td> <td>3</td> <td>-</td> <td>4</td> <td>0</td> <td>-</td> <td>2</td> <td>1</td> <td>-</td> <td>1</td> <td>8 3</td> <td>ŝ</td> <td>0</td> <td>0 0</td> <td>64</td>			Total	15	6	6	0	0	4	3	-	4	0	-	2	1	-	1	8 3	ŝ	0	0 0	64
Model Matrix I <thi< td=""><td></td><td></td><td>Full-Scale</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>+</td><td></td><td></td><td></td><td></td><td>0</td></thi<>			Full-Scale															+					0
Model Model I		10 to 20%	Pilot Test		_		•	•		4		_		4	•	4		4		•	4		4
Market Funder Market F			Total		_	0	•	0	•	0	0	-	0	0	0	0	0	0	0	0	0	0	4.
International conditional condi		20 to 30%	Full-Scale Pilot Test			,																-	
Hole Indication Indication <td></td> <td></td> <td>Total</td> <td>0</td> <td>0</td> <td>- 1</td> <td>0</td> <td>0 0</td> <td>0</td> <td>0</td> <td>0</td> <td>1 60</td>			Total	0	0	- 1	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0	0	1 60
Modelly Tendent Tendent <t< td=""><td></td><td></td><td>Full-Scale</td><td></td><td>-</td><td>_</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>•</td><td></td><td></td><td></td><td></td><td>•</td><td></td><td>4</td></t<>			Full-Scale		-	_										•					•		4
		30 to 40%	Pilot Test																				• •
			Total	0	-	-	0	0	0	0	0	0	-	0	0	0	0	0	1	0	0	0 0	4
Hondrise Total I <t< td=""><td></td><td></td><td>Full-Scale</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>H</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>0</td></t<>			Full-Scale											H									0
Model Table To <	bles		Pilot Test		•	¢	•	ľ			4		•	•	4		•	•			4		•
			Total	0	•	•	-	•	•	0	0	0	•	0	0	0	•	0	0 0	0	0	0 0	•
$ \ \ \ \ \ \ \ \ \ \ \ \ \ $		20 40 60%	Full-Scale Pilot Tast	-																			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Total	-	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	-
M0.0% Find team model			Full-Scale			_																	-
		60 to 70%	Pilot Test	_			Ì											_	_	_		_	0
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $			Total	0	0	-	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0	0 0	-
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		70 to 80%	Full-Scale Pilot Test																				0
			Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0	0 0	•
M0.09% Flot rest 0			Full-Scale																_				0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		80 to 90%	Pilot Test															_	_	_		_	0
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$			Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0		0	0 0	0
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		0.40.109/	Full-Scale Dilot Tast	9	m 4	64 6				-	-	- 7	-					-	1	m -			81 55
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		0/ 01 01 0	Total	•	t r	1	•	e	- (-		- *	ŀ	0	16	•	0			- 7	0	0	43
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Full-Scale				,	,				2		1									6
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10 to 20%	Pilot Test	9		_						-							-				0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Total	3	0	2	0	0	0	0	0	_	0	-	0	0	0	0	1 0	0	0	0 0	∞
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Full-Scale			-										1							2
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		20 to 30%	Pilot Test	3		5	-	ľ	2	1	,			ľ	,				_		,	,	=
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			Total	9	0	6	0	0	2	0	0	0	0	0	0	1	0	0	1	0	0	0 0	13
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		30 to 40%	Full-Scale Pilot Test		-	-											-		-				- "
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $			Total	0	-	_	•	0	0	0	0	0	0	0	0	0		0	0	0	0	0 0	4
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Ħ		Full-Scale				,			-		2		,		2				-			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	5	40 to 50%	Pilot Test	2																		-	e
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Total	2	0	0	0	0	0	1	0	0	0	0	0	0	0	0	0 1	1	0	0 1	9
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			Full-Scale							-									_	_			
Image: constraint of the second of the se		50 to 60%	Pilot Test		•	d	4	•			4	•	•	4	4	4	•			•	4	•	0.
Plantane Plantane Plantane 0			L 0131 Eull-Scala				•	-		-	•		•	•	0	•	•	•		•	•		
Test 0		60 to 70%	Pilot Test																			-	- 0
Full-State Full-State Full-State 0 PilotTest 0 Full-State 0			Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0	0	-
PilotTest PilotTest PilotTest 0 <td></td> <td></td> <td>Full-Scale</td> <td></td> <td></td> <td></td> <td>ſ</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td>t</td> <td></td> <td></td> <td></td> <td></td> <td>_</td> <td>_</td> <td></td> <td></td> <td></td> <td>0</td>			Full-Scale				ſ						t					_	_				0
Total 0 <td></td> <td>70 to 80%</td> <td>Pilot Test</td> <td></td> <td>0</td>		70 to 80%	Pilot Test																				0
Full-Scale Pilot Test 0 0 0			Total	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0 0	0	0	0 0	0
Titot test 0 0 0 0		7000 07 00	Full-Scale				T					-	+					+					- <
		** ** ** **	Total	0	0	0	0	0	0	0	0	-	0	0	0	0	0	0	0	0	0	0 0	> -

TABLE D.3: SUMMARY OF REMEDIAL COST DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY, SCALE OF REMEDIATION AND TREATMENT SIZE Navy ROCS

	COST	EGORY 1				<u> </u>									Twentment	Infractment					<u> </u>								
	COST	BREAKDOWN		0 to 10%			10 to 20%			20 to 30%			30 to 40%			40 to 50%			50 to 60%			60 to 70%			70 to 80%			80 to 90%	
	SCALE OF	CATEGORY BREAKDOWN TREATMENT	Full-Scale	Pilot Test	Total	Full-Scale	Pilot Test	Total	Full-Scale	Pilot Test	Total	Full-Scale	Pilot Test	Total	Full-Scale	Pilot Test	Total	Full-Scale	Pilot Test										
	BIOREMEDIATION		3	9	6		3	3		3	3		2	2			0			0			0			0			
		Fenton's Reagent	4	3	7		1	-			0			0			0			0			0			0			
	CHEMICAL OXIDATION	Permanganate Ozone	4	5	6		2	2		2	2			0			0			0			0			0			
					0 0			0 0			0 0			0 0			0 0			0 0			0 0			0 0			
NU			-	-	2			•		-	1		-	1			0			0			0			0			
MBER OF SIT.	DUAL-PHASE EXTRACTION	/Air Water/DN	-		-			0			0	2		2			0			0			0			0			
ES THAT COUL	XTRACTION	Water/Air Water/DNAPL DNAPL/Air		-	-			0			0			0			0			0			0			0			
D BE DESCRIBE	EXCAVATION	ŗ	3	_	4		-	1			0			0			0			0			0			0			
D BY COST BI		Undefined Pump and Treat			0			0			0	1		1			0			0			0			0			
REAKDOWN	OTHER SU		-		-	-		0	-		0			0	-		0			0	-		0			0			
NUMBER OF SITES THAT COULD BE DESCRIBED BY COST BREAKDOWN PARAMETER PER TECHNOLOGY	SURFACTANT	FLUSHING		2	2			0			0			0			0			0			0			0			
ER TECHNOL		Conductive Low-Temp Resistive Six-Phase Heating Six-Phase Heating Heating		-	0			0			0	-	-	-			0			0		-	0			0			
OGY	F	Low-Temp Resis Six-Phase Heat			0			0 0			9			0 0			0 0		1	1 6			0 0			0 0			
	THERMAL	Resistive Six-Phase Heating Heating	-	5	9			0 0			0		7	2			0 0			0			0 0			0 0			
		se Three- g Phase Heating		-	-			0		-	-	-		-			0			0			•			0			
		Steam Flushing	3	-	4			0			0	-		-			0			0			0			0			
	ZERO-VALENT IRON	Emulsified With Clay ZVI only			0 0			0 0			0 0			0 0			0 0			0 0			0 0			0 0			
	NT IRON	lay ZVI only	1	-	2			0			0			0			0	_		0			0			0			
TOTAL SITES	WITH COST		22	28	50	0	7	7	0	7	4	9	5	11	0	0	0	0	1	1	0	•	0	0	0	0	0	0	

Notes "DNAPL zone sizes are classified as follows: "and: area < (10,000 ft" volume < 100,000 ft" Large: area > 100,000 ft" volume > 100,000 ft Blank spaces denote no data entered for that parameter.

Technology	Role in Groundwater Community	Perceived Success
	Consultant	Success
Bioremediation	Consultant	Success
	DOD RPM	Success
	DOD RPM	Success
	Published Case Studies	Fair Success
	Published Case Studies	Fair Success
	Consultant	Fair Success
	DOD RPM	Success
	Published Case Studies	Fair Success
Chemical Oxidation-Fenton's Reagent	Published Case Studies	Success
	Vendor	Success
	Vendor	Success
	Vendor	Success
	Consultant	Fair Success
	Consultant	Success
	Consultant	Success
Chemical Oxidation-Permanganate	Consultant	Success
	DOD RPM	Fair Success
	Published Case Studies	Fair Success
	Published Case Studies	Fair Success
	DOD RPM	Success
Dual Phase-Water/air extraction	Published Case Studies	Fair Success
	DOD RPM	Poor success
	Consultant	Success
Excavation	Consultant	Success
Excavation	Consultant	Success
		Fair Success
Surfactant Flushing	Published Case Studies	Fair Success
Gunaetant i fushing	Published Case Studies	Fair Success
Thermal-Resistive heating	Published Case Studies	Fair Success
	DOD RPM	Success
Thermal-Six phase heating	Published Case Studies	Success
mormal expluse heating	Published Case Studies	Success
	Published Case Studies	Success
Thermal-Steam	Consultant	Success
Thermal - Conductive Heating	Published Case Studies DOD RPM	Fair Success
ő		Success
Thermal - Low temp-six phase heating	Consultant	Success
ZVI - EZVI/nano-scale iron	Consultant	Fair Success
ZVI -Zero-valent iron	Site Owner/RPM	Success
	Vendor	Success

TABLE D.4: SUMMARY OF PERCEIVED SUCCESS SORTED BY TECHNOLOGY Navy ROCS

TABLE D.5: SUMMARY OF TECHNOLOGY PERFORMANCE CRITERIA COLLECTED WITH THE SURVEY SORTED BY SURVEY RESPONDENT ROLE IN ENVIRONMENTAL COMMUNITY Navy ROCS

TECHNOLOGY EFFECTIVENESS		RO	LE IN ENVIRONM	IENTAL REMEDI	ROLE IN ENVIRONMENTAL REMEDIATION COMMUNITY	Y		
CRITERIA	CONSULTANT	REGULATOR	VENDOR	DOD RPM ^a	SITE OWNER	ACADEMIC	PUBLISHED CASE STUDIES	TOTAL
Able to achieve treatment goals at a reasonable cost	17 (51.5%)	2 (100.0%)	5 (83.3%)	7 (53.8%)	1 (25.0%)		16 (72.7%)	48 (60.0%)
Technology is limited to certain site conditions	12 (36.4%)	1 (50.0%)	1 (16.7%)	6 (46.2%)	3 (75.0%)		5 (22.7%)	28 (35.0%)
Technology requires further development before it can be routinely applied	4 (12.1%)	0 (0.0%)	2 (33.3%)	1 (7.7%)	0 (0.0%)		1 (4.5%)	8 (10.0%)
Reasonable ease of implementation	16 (48.5%)	2 (100.0%)	3 (50.0%)	6 (46.2%)	4~(100.0%)		2 (9.1%)	33 (41.3%)
Reasonable ease of use	18 (54.5%)	2 (100.0%)	3 (50.0%)	9 (69.2%)	4~(100.0%)		1 (4.5%)	37 (46.3%)
Reasonable ease of design	19 (57.6%)	2 (100.0%)	2 (33.3%)	8 (61.5%)	4~(100.0%)		2 (9.1%)	37 (46.3%)
Adequate number of trained vendors to implement technology	10(30.3%)	2 (100.0%)	0 (0.0%)	4 (30.8%)	3 (75.0%)		0(0.0%)	19 (23.8%)
Other factors that might affect technology usefulness	2 (6.1%)	1 (50.0%)	1 (16.7%)	5 (38.5%)	0 (0.0%)		0(0.0%)	9 (11.3%)
TOTAL RESPONDENTS PER CATEGORY	33	2	9	13	4	0	22	80

Notes Blank spaces indicate no sites were impacted by the corresponding criteria.

TABLE D.6: SUMMARY OF TECHNOLOGY PERFORMANCE CRITERIA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

тесн	INOLOGY	ROLE IN ENVIRONMENTAL COMMUNITY	Able to achieve treatment goals at a reasonable cost	Technology is limited to certain site conditions ^b	Technology requires further development before it can be routinely applied	Reasonable ease of implementation	Reasonable case of use	Reasonable ease of design	Adequate number of trained vendors to implement technology	Other factors that might affect technology usefulness ^{de}	TOTAL SITES WITH TECHNOLOGY DATA
BIOREN	MEDIATION	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies	6 3 1 2	2 3	3	5 3 1	6 3 1	7	3		17
	Fenton's Reagent ^e	Total Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	12 (70.6%) 1 4 3 8 (88.9%)	5 (29.4%) 1 1 2 (22.2%)	4 (23.5%) 0 (0.0%)	10 (58.8%)	11 (64.7%) 1 1 2 (22.2%)	9 (52.9%) 1 1 2 (22.2%)	6 (35.3%) 1 1 (11.1%)	0 (0.0%)	9
CHEMICAL OXIDATION	Permanganate	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	4 1 2 7 (53.8%)	5 1 6 (46.2%)	1 (7.7%)	7 1 8 (61.5%)	7 1 8 (61.5%)	7 1 8 (61.5%)	4 (30.8%)	0 (0.0%)	13
	Ozone	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies									0
CO-SOLVE	ENT FLUSHING	Total Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0
	Water/Air	Consultant DOD RPM ⁴ Regulator Site Owner Vendor Published Case Studies Total	1 (16.7%)	1 1 2 (33,3%)	0 (0.0%)	1 2 3 (50.0%)	1 2 3 (50.0%)	1 2 3 (50.0%)	1 2 3 (50.0%)	1 1 1 (16.7%)	6
DUAL-PHASE EXTRACTION	Water/DNAPL	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	1 (33.3%)	1 (33.3%)	0 (0.0%)	1 1 2 (66.7%)	1 1 2 (66.7%)	1 1 2 (66.7%)	1 1 2 (66.7%)	1 (33.3%)	3
	DNAPL/Air	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	0 (0.0%)	1 (100.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	1
EXC	AVATION	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	2 1 3 (60.0%)	2 1 3 (60.0%)	0 (0.0%)	1 1 3 (60.0%)	1 1 2 (40.0%)	1 1 2 (40.0%)	1 1 2 (40.0%)	0 (0.0%)	5

TABLE D.6: SUMMARY OF TECHNOLOGY PERFORMANCE CRITERIA COLLECTED WITH THE SURVEY SORTED BY TECHN	OLOGY
Navy ROCS	

TEC	HNOLOGY	ROLE IN ENVIRONMENTAL COMMUNITY	Able to achieve treatment goals at a reasonable cost	Technology is limited to certain site conditions ^b	Technology requires further development before it can be routinely applied	Reasonable ease of implementation	Reasonable ease of use	Reasonable case of design	Adequate number of trained vendors to implement technology	Other factors that might affect technology usefulness ^{de}	TOTAL SITES WITH TECHNOLOGY DATA
	Undefined	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	1
OTHER	Pump and Treat	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	0 (0.0%)	0 (0.0%)	0 (0.0%)	1 (100.0%)	1	0 (0.0%)	0 (0.0%)	1 1 1 (100.0%)	1
SURFACT	ANT FLUSHING	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	l 1 (50.0%)	l 1 (50.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	2
	Conductive Heating	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies	1				1	1		1	1
	Low-Temp Six-Phase	Total Consultant DOD RPM ⁴ Regulator Site Owner Vendor Published Case Studies Total	1 (100.0%) 1 1 (100.0%)	0 (0.0%) 1 1 1 1 (100.0%)	0 (0.0%)	0 (0.0%) 1 1 1 1 (100.0%)	1 (100.0%) 1 1 (100.0%)	1 (100.0%) 1 1 (100.0%)	0 (0.0%)	1 (100.0%) 1 1 1 1 (100.0%)	1
	Resistive Heating	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	1 (100.0%)	1 1 (100.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	1
THERMAL	Six-Phase Heating	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	1 1 6 (66.7%)	1 1 2 4 (44,4%)	1 1 (11.1%)	1 (11.1%)	1 (11.1%)	1 (11.1%)	0 (0.0%)	1 1 2 (22.2%)	9
	Three-Phase Heating	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	1 (33.3%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	0 (0.0%)	1 (33.3%)	0 (0.0%)	1 1 1 (33.3%)	3
	Steam Flushing	Consultant DOD RPM ^a Regulator Site Owner Vendor Published Case Studies Total	2 4 (80.0%)	0 (0.0%)	0 (0.0%)	1 2 (40.0%)	1 (20.0%)	l 1 2 (40.0%)	1 (20.0%)	1 (33.3 76) 1 1 (20.0%)	5

TABLE D.6: SUMMARY OF TECHNOLOGY PERFORMANCE CRITERIA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

тесн	INOLOGY	ROLE IN ENVIRONMENTAL COMMUNITY	Able to achieve treatment goals at a reasonable cost	Technology is limited to certain site conditions ^b	Technology requires further development before it can be routinely applied	Reasonable case of implementation	Reasonable case of use	Reasonable ease of design	Adequate number of trained vendors to implement technology	Other factors that might affect technology usefulness ^{de}	TOTAL SITES WITH TECHNOLOGY DATA
		Consultant			1		1	1		1	
		DOD RPM ^a									
		Regulator									
	Emulsified	Site Owner									1
		Vendor									
		Published Case Studies									
		Total	0 (0.0%)	0 (0.0%)	1 (100.0%)	0 (0.0%)	1 (100.0%)	1 (100.0%)	0 (0.0%)	1 (100.0%)	
		Consultant									
		DOD RPM ^a									
ZERO-VALENT		Regulator									
IRON	With Clay	Site Owner	1	1		1	1	1		1	1
indiv		Vendor									
		Published Case Studies									
		Total	1 (100.0%)	1 (100.0%)	0 (0.0%)	1 (100.0%)	1 (100.0%)	1 (100.0%)	0 (0.0%)	1 (100.0%)	
		Consultant									
		DOD RPM ^a									
		Regulator									
	ZVI only	Site Owner	1	1		1	1	1			2
		Vendor	1		1	1	1	1		1	
		Published Case Studies									
		Total	2 (100.0%)	1 (50.0%)	1 (50.0%)	2 (100.0%)	2 (100.0%)	2 (100.0%)	0 (0.0%)	1 (50.0%)	

Notes

^aDOD RPM refers to Department of Defense remedial project manager.

^bSpecified site conditions for which technology may be limited included the following: *Bioremediation:* "Technology is limited by low permeability of till material that is the source area. We are hoping it will prove effective in the overlying fill aquifer."

"Presence of DHE organisms or bioaugmentation required." "The effectiveness of In-Situ Reactive Zone (IRZ) technology using soluble carbohydrates on pooled DNAPL has not been proven, although it has been shown to be applicable to sorbed or residual DNAPL. Optimal ranges for site characteristics are discussed in "Technical Protocol for Using Soluble Carbohydrates to Enhance Reductive Dechlorination of Chlorinated"

"primarily limited to source areas where the appropriate redox conditions can be promoted" *Chemical oxidation, Fentor's reagent:* "high carbonate content causes more vigorous reaction and higher reagent consumption "permeabilities < 1 x 10-6 cm/sec"

Chemical oxidation, permanganate: "Highly sensitive to ability to deliver reagent to subsurface and achieve good mixing" "In situ chemical oxidation using permanganate is more effective for remediation of dissolved contaminant and residual DNAPL. The remediation of pooled DNAPL has also been achieved in test facilited with technology we recently developed to treat the Mn oxide precipitates."

ability to deliver oxidant

"The distribution of the DNAPL within the site geology plays a key role in the application of this technology. In addition, permanganate is only applicable to some types of DNAPL "primarily limited to smaller source areas where DNAPL is present at residual non-wetting phase saturation; readily applied in more permeable geology; fractured environments

DNAPL/air dual-phase extraction: "site soils could not be dewatered adequately to get air into soil" Water/air dual-phase extraction: "Site soils could not be dewatered adequately to get air into soil" Water/air dual-phase extraction: "Vacuum-enhanced recovery (VER) is applicable to sites with generally low transmissivity (<500 gpd/ft), permeability of 1e-3 to 1e-6, and no potential for short-circuiting. Extreme heterogeneity requires careful engineering design."

Solution and the second and the s

"shallow overburdenplume only. less practical with depth" Surfactant flushing: "permeability contrasts lead to difficulty in surfactant distribution"

Surjection/instance_productions can be only in an accurate databased Low-temp six-phase hearing: "Applicable only when DNAPL is found pooled." Resistive hearing: "lectrical conductivity of soil" Six-phase hearing: "Low heat 6 phase was tried (max of 70 degrees C) because of the close proximity of underground utilities. May have to try chemical oxidation or some other technology at the pilot study site."

"soil conductivity"

"High treatment costs per unit volume. Justified when difficult geology and/or restricted timelines are encountered. DNAPL in narrow "chimney" configuration at depths beyond 50 feet may present significant application and cost challenges.

Zero-valent iron: "Unconsolidated materials accessible to soil mixing equipment or jet grouting equipment."

⁶One duplicate site is included in the Fenton's reagent data, where a vendor and DOD RPM both entered data. The only differences between their replies to this question were the vendor did not select "ease of use" or "ease of design" as applying to this technology.

^dSpecified "other" factors that may impact technology usefulness included the following:

Chemical oxidation, permeasured content of oxidation by-product Mn oxide can cause pore plugging and reduction of permeability. Technology in treating the Mn oxide precipitates has been developed."

Low-temp six-phase heating: "Limited to depths that can economically be serviced by hand bailing"

Pump and treat: "containment, compatibility with adjacent PRPs' syste

Six-phase heating: "Technology is expensive" "shortage of qualified vendors" Three-phase heating: "Cost is significant"

Steam flushing: "Testing in fractures has been inconclusive at other sites."

Emulsified nano-scale ZVI: "further work needs to be conducted on refining the injection of the EZVI into the treatment zone"

Zero-valent iron: "Chlorinated compounds amenable to treatment with ZVI"

"Construction contractor must know how to undertake soil mixing adequately."

⁶Other comments not specific to impact on technology usefulness included the following: *Water/air dual-phase extraction*: "VER has been used successfully at low-permeability sites that are not amenable to conventional P&T/vapor extraction techniques. To overcome air and groundwater flow restrictions of low-permeability formations, high vacuums are created at a well by liquid-ring or other specialty pumps. The high vacuum creates a much greater driving force for airflow in the unsaturated zone. Combined with gravity, this increases the rate of groundwater and/or NAPL recovery, and the size of the capture zone. In addition, in highly adsorptive silt and clay formations, subsurface oxygen levels are increased, enhancing naturally-occurring biodegradation. Water/DNAPL dual-phase extraction: "It is currently used at over 40000 sites. It is the first step of Triple Train. It is the cheapest removal unit operation of the three in Triple Train

(20 dollar/gal). It only works with the NAPL saturations above residual. Second unit operation of Tri [comment was too long and remainder is missing] Conductive heating: "Even an area where treatment temperatures never exceeded 212 F was treated. This suggests that the remedial goals could have been achieved at lower cost."

TABLE D.8: SUMMARY OF EXTERNAL FACTORS IMPACTING REMEDIAL EFFECTIVENESS / TECHNOLOGY CHOICE DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY MAY ROCS Nay ROCS

EXTERNAL FACTORS THAT							NUMBER OF	SITES THAT	COULD BE DE	SCRIBED BY	' EXTERNAL	NUMBER OF SITES THAT COULD BE DESCRIBED BY EXTERNAL FACTORS PER TECHNOLOGY	3 TECHNOL	0GY								TOTAL WITH
IMPACTED REMEDIAL CHOICE /	BIOREMEDIATION	CHEM	CHEMICAL OXIDATION		CO-SOLVENT	1-TVDQ	PHASE EXTRACTION	CTION	EXCAVATION	OTH	OTHER	SURFACTANT			THERMAL	MAL			ZERO-VALENT IRON	NT IRON	UNSPECIFIED	UNSPECIFIED TECHNOLOGY
EFFECTIVENESS		Fenton's Reagent	Permanganate Ozone		FLUSHING	Water/Air W	Water/DNAPL DNAPL/Air	DNAPL/Air		Undefined	Pump and Treat	Undefined Pump and FLUSHING Conductive Low-Temp Resistive Six-Phase Three-Phase Steam Treat Heating Six-Phase Heating Heating Flushing	Conductive Heating	Low-Temp Six-Phase	Resistive Heating	Six-Phase Heating	Three-Phase Heating		Emulsified With Clay ZVI only	Jay ZVI only		SPECIFIED
Facility operations constrained site access	11 (44.0%)		5 (33.3%)				1 (20.0%)	2 (66.7%)	3 (27.3%)	1 (16.7%)							1 (20.0%)	1 (12.5%)		2 (50.0%)		27 (22.9%)
Presence of surface infrastructure impacted ability to estimate DNAPL mass or effectively conduct remediation	9 (36.0%)	2 (22.2%)	5 (33.3%)	1 (100.0%)			1 (20.0%)	1 (33.3%)	2 (18.2%)							1 (10.0%)	1 (20.0%)	1 (12.5%)	1 (100.0%) 1 (100.0%) 2 (50.0%)	0%) 2 (50.0%)		28 (23.7%)
Proximity to surface water impacted technology choice, design or effectiveness	9 (36.0%)		3 (20.0%)				1 (20.0%)	2 (66.7%)	6 (54.5%)	3 (50.0%)		1 (25.0%)			1 (50.0%)	(50.0%) 2 (20.0%)	3 (60.0%)	2 (25.0%)			2 (40.0%)	35 (29.7%)
Regulatory pressures impacted clean-up criteria/time-frame	7 (28.0%)		5 (33.3%)	1 (100.0%)		1 (20.0%)	1 (20.0%)		9 (81.8%)	3 (50.0%) 1 (100.0%)	1 (100.0%)			1 (100.0%)		1 (50.0%) 2 (20.0%)	2 (40.0%)	4 (50.0%)				37 (31,4%)
Budget / remediation cost	17 (68.0%)	5 (55.6%)	9 (60.0%) 1 (100.0%)	1 (100.0%)		3 (60.0%)	4 (80.0%)	1 (33.3%)	8 (72.7%)	3 (50.0%)		1 (25.0%)	1 (100.0%)		2 (100.0%)	2 (100.0%) 7 (70.0%) 2 (40.0%) 5 (62.5%)	2 (40.0%)	5 (62.5%)	1 (100.0%)	1 (25.0%)	1 (20.0%)	72 (61.0%)
TOTAL SITES WITH DATA	25	6	15	1	0	5	5	3	Π	9	6 1	4	1	1 1 2		10	5	8	1 1 4	4	s	118

Antes Notes Blank spaces indicate no survey respondent selected the corresponding factors for that technology applicatio

01.0GV	
7 TECHN	
CD BY	
SORTE	
ZAL	
REMOV	
ASS	
ARV OF SOURCE MASS REMO	
O A)
SUMMAR	
TARLE D.10:	

Lithology Dead-end L Lithology fractures L Sand Clay Sand L Sand Sand Sand Sand Clay Sand Sand Sand Sand Clay Sand Sand Sand Sand Sand Sand	Navy ROCS						I		DNAPI Distribution	hution				
Dead-end Dirtused into Dirtused into Ead-end Dirtused into Image Image X Pilot test completed X Image X X Y Pilot test completed Image X X X Pilot test completed Image X X X Pilot test completed Image X X X Full-scale completed Image X X X					ŀ					0,000	I			
Joigy fractures Low K Layers In pools Residual Storbed Rage Image: Image	Mass Flux	Mass Flux	Mass Flux			Perceived		Dead-end	Diffused Into					Post Treatment
Image: constraint of the stand of	Site_ID Technology Decrease Rebound	Decrease		Rebour	pc	Success	Lithology	fractures	Low K Layers	In pools	Residual	Sorbed	Stage	Monitoring
Image: constraint of the stand of	248 Chemical Oxidation-Permanganate 81-100% No	81-100%		٥N	-		Sand				Х		Pilot test completed	
Clay Clay Full-scale completed Sand X X Pilot test completed Clay X X Pilot test completed Sand X X Pilot test completed Sand X X X Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed Sand X X X Pilot test completed Sand X X X Pilot test completed Sand X X Y Pilot test completed Sand X X Pilot test completed Sand X X Y Pilot test completed Sand X X Y Pilot test comp	218 Thermal-Six phase heating 81-100% Unknown	81-100%	_	Unkno		Not yet evaluated	Sand						Full-scale completed	On going
SandNameNameNamePilot test completedClayNameXXPilot test completedSandXXXPilot test completedSandXXXPilot test completedSandXXXXPilot test completedSandXXXXPilot test completedSandXXXXFull-scale completedClayXXXYFull-scale completedSandXXXFull-scale completedSandXXXPilot test completedSandXXXPilot test completedSandXXXYPilot test completedSandXXXXPilot test completedSandXXXXPilot test completedSandXXXXPilot test completed <t< td=""><td>144 Thermal- Conductive Heating Unknown No</td><td>Unknown</td><td></td><td>٥N</td><td></td><td></td><td>Clay</td><td></td><td></td><td></td><td>×</td><td></td><td>Full-scale completed</td><td></td></t<>	144 Thermal- Conductive Heating Unknown No	Unknown		٥N			Clay				×		Full-scale completed	
ClayClayXPilot test completedSandXXXPilot test completedSandXXXXPilot test completedSitXXXXFull-scale completedSitXXXXYClayXXXXFull-scale completedSandXXXXFull-scale completedClayXXXXFull-scale completedSandXXYPilot test completedSandXXXYPilot test completedSandXXXYPilot test completedSandXXXXPilot test completedSandXXXYPilot test completedSandXXXXPilot test completed <t< td=""><td>46 Excavation Unknown Unknown</td><td>Unknown</td><td>_</td><td>Unkn</td><td>own</td><td></td><td>Sand</td><td></td><td></td><td></td><td>×</td><td></td><td>Pilot test completed</td><td></td></t<>	46 Excavation Unknown Unknown	Unknown	_	Unkn	own		Sand				×		Pilot test completed	
Sand Sand X X Pliot test completed Sand X X X X Pliot test completed Sand X X X X Pliot test completed Sand X X X X Pliot test completed Clay X X X Full-scale completed Clay X X X Full-scale completed Sand X X Y Full-scale completed Sand X X X Pliot test completed Sand X X <td< td=""><td>235 Surfactant Flushing 81-100% Unk</td><td>81-100%</td><td>_</td><td>Unk</td><td>Unknown</td><td>Fair Success</td><td>Clay</td><td></td><td></td><td></td><td>Х</td><td></td><td>Pilot test completed</td><td></td></td<>	235 Surfactant Flushing 81-100% Unk	81-100%	_	Unk	Unknown	Fair Success	Clay				Х		Pilot test completed	
Sand Early Pilot test completed Slit X X X Full-scale completed Clay X X X Full-scale completed Clay X X X Full-scale completed Clay X X Full-scale completed Clay X X Full-scale completed Sand X X Pliot test completed Sand X X Y Pliot test completed Sand X X X Pliot test completed Sand X X Y Pliot test completed Sand X X X Pliot test completed Silt X X X Pliot test completed Silt	250 Chemical Oxidation-Permanganate 81-100% Y	81-100%		~	Yes		Sand				×	×	Pilot test completed	
Silt X X X Full-scale completed Clay X X X Full-scale completed Clay X X X Full-scale completed Clay X X X Full-scale completed Sand X X X Full-scale completed Sand X X X Full-scale completed Silt X X X Full-scale completed Silt X X Y Full-scale completed Silt X X X Full-scale completed Sand X X X Pilot test completed	197 Bioremediation 61-80% N	61-80%	-	2	No		Sand						Pilot test completed	On-going
Clay X X X Full-scale completed Clay X X X Full-scale completed Sand X X Y Full-scale completed Silt X X Y Pilot test completed Silt X X X Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed <td>100 Excavation D</td> <td>Unknown</td> <td></td> <td>2</td> <td>No</td> <td></td> <td>Silt</td> <td></td> <td>×</td> <td>×</td> <td>×</td> <td>×</td> <td>Full-scale completed</td> <td>On-going</td>	100 Excavation D	Unknown		2	No		Silt		×	×	×	×	Full-scale completed	On-going
Clay X X Full-scale completed Sand X X Full-scale completed Sand X X Full-scale completed Sit X X Full-scale completed Sit X X X Full-scale completed Sit X X X Full-scale completed Sit X X Y Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed	245 Chemical Oxidation-Fenton's Reagent Unknown N	Unknown		z	No		Clay		×		×	×	Full-scale completed	On-going
Sand X Full-scale completed Sand X X X Sand X X Full-scale completed Sitt X X Y Sitt X X Pilot test completed Sitt X X Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed	153 Thermal-Steam Unknown N	Unknown		Z	No		Clay		×				Full-scale completed	On-going
Sand X X Full-scale completed Slitt X X Full-scale completed Slitt Y X Full-scale completed Slitt X Y Full-scale completed Sand X Y Full-scale completed Sand X X Y Pliot test completed Sand X X X Pliot test completed Silt X X X Pliot test completed	117 Chemical Oxidation-Fenton's Reagent Unknown N	Unknown		Z	No		Sand						Full-scale completed	On-going
Silt X X Pilot test completed Silt X X Full-scale completed Sand X X Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed Silt X X X Pilot test completed	155 Excavation 61-80% Y	61-80%		У	Yes		Sand		×		Х	×	Full-scale completed	On-going
Silt Eull-scale completed Sand X X Pilot test completed Sand X X Y Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed	45 Chemical Oxidation-Fenton's Reagent Unknown	Unknown	-		No	Fair Success	Silt				×		Pilot test completed	
Sand X X Pilot test completed Sand X X X Pilot test completed Silt X X X Pilot test completed Silt X X X Pilot test completed Silt X X X Pilot test completed	186 Chemical Oxidation-Fenton's Reagent Unknown Un	: Unknown	-	Π	Jnknown		Silt						Full-scale completed	
Sand X X X Pilot test completed Silt X X X Pilot test completed Silt X X X Pilot test completed	237 Thermal-Steam 81-100% Un	81-100%	_	'n	Jnknown	Fair Success	Sand				×		Pilot test completed	
Sand X X X Pilot test completed Sand X X X Pilot test completed Sitt X X X Pilot test completed Sitt X X X Pilot test completed Sitt X X X Pilot test completed	249 Chemical Oxidation-Permanganate Unknown	Unknown		ĺ	Yes	Fair Success	Sand			×	×	×	Pilot test completed	
Sand X X Pilot test completed Silt X X X Pilot test completed Silt X X X Pilot test completed	181 Chemical Oxidation-Permanganate 61-80% Y	61-80%		7	Yes	Fair Success	Sand			×	×	×	Pilot test completed	
Silt X X Pilot test completed Silt X X X Pilot test completed	253ZVI - EZVI/nano-scale iron 41-60%	41-60%		_	No	Fair Success	Sand			Х	Х		Pilot test completed	
Silt X X X X X Pliot test completed	241 Thermal - Low temp-six phase heating Unknown	Unknown			Yes		Silt			Х	Х	Х	Pilot test completed	
	244 Dual Phase-DNAPL/air extraction Unknown Un	Unknown		۱Ч	Unknown	Poor Success	Silt		×	×	×	×	Pilot test completed	On going

TABLE D.12: SUMMARY OF MASS FLUX DECREASE SORTED BY TECHNOLOGY Navy ROCS

Source Mass Technology Source Mass Mass Removal Mass Ferevoid Mass Perceived Nucleation-Permanganatic Nucleation-Permanganatic Perceived Nucleation-Permanganatic Diffused Intro Low K Diffused Intro Inspect Nucleation-Permanganatic Perceived Nucleation-Permanganatic Perceived Low K Diffused Intro Nucleation-Permanganatic Perceived Nucleation-Permanganatic Perceived Low K Diffused Intro Nucleation-Permanganatic Perceived Low K Perceived Low K Perceived Low K Perceived Nucleation-Permanganatic Perceived Low K Perceived Low								DNAP	DNAPL Distribution	ion			
Technology Mass Perceived Lithology Fractures Low K Residual Sortices Lithology Fractures Low K Residual Sortices Lithology Fractures Lithology K	Mass		Source					Diffused Into					
100%NoSuccessSandSand X X 100%UnknownNot yt evaluatedSandSand X X X >90%UnknownFair SuccessClay X X X X >90%YesSuccessSand X X X X >90%YesSuccessSand X X X X >90%VesSuccessSand X X X X >90%UnknownNoSuccessUnknown X X X X UnknownUnknownEair SuccessUnknown X X X X UnknownUnknownUnknownEair SuccessUnknown X X X X UnknownUnknownNot yet evaluatedGravelUnknown X X X X UnknownUnknownNot yet evaluatedTill X X X X UnknownUnknownNot yet evaluatedTill X <t< th=""><th>Flux Decrease</th><th></th><th>Mass Removal</th><th>Rebound</th><th>Perceived Success</th><th>Lithology</th><th>Dead-end Fractures</th><th>Low K Llayers</th><th>ln pools</th><th>Residual</th><th>Sorbed</th><th>Stage</th><th>Post Treatment Monitoring</th></t<>	Flux Decrease		Mass Removal	Rebound	Perceived Success	Lithology	Dead-end Fractures	Low K Llayers	ln pools	Residual	Sorbed	Stage	Post Treatment Monitoring
10% Unknown Natyte valuated Sand Sand <th></th> <th>Chemical Oxidation-Permanganate</th> <th>100%</th> <th>No</th> <th>Success</th> <th>Sand</th> <th></th> <th></th> <th></th> <th>×</th> <th></th> <th>Pilot test completed</th> <th></th>		Chemical Oxidation-Permanganate	100%	No	Success	Sand				×		Pilot test completed	
00/06 Unknown Notyet evenanee Oatus Notyet evenanee >90% Yes Success Sand No X X >90% Yes Success Sand X X X >90% Yes Success Sand X X X >90% Ves Success Unknown Nenown Fair Success Unknown X X X Unknown Unknown Fair Success Unknown X X X X Unknown Unknown Netwown Fair Success Unknown X X X Unknown Unknown Netwown Netwown Netwown X X X Unknown Unknown Success Unknown X X X X Unknown Unknown Success Sand T X X X X Unknown Unknown Success Sand T		Thomal Siv above booting	1000/		Not vict evictinated							Eull coolo completed	
>00% Vmmon Call		Surfactant Flushing	%00I >00%		Fair Surress	Clav				×		Pilot test completed	
>50 < 80% Unknown Fair Success Sand N X Unknown No Success Clay N X X Unknown Unknown Eair Success Unknown N X X Unknown Unknown Eair Success Unknown N X X Unknown Unknown N Fair Success Unknown X X Unknown Unknown Notyte evaluated Gravel A X X Unknown Unknown Notyte evaluated Gravel X X X Unknown Unknown Notyte evaluated Till X X X Unknown Unknown Success Unknown X X X S0% No Success Sand X X X S0% No Success Sand X X X S0% Yes Fair Success Sand X X X S0% Yes Fair Success Sand X X X S0% Yes Fair Success Sand X X X S0% Yes Fai		Chemical Oxidation-Permanganate	%06<	Yes	Success	Sand				××	×	Pilot test completed	
UnknownNoSuccessClayOlay<		Thermal-Steam	>50 <80%	Unknown	Fair Success	Sand				×		Pilot test completed	
		Thermal-Six phase heating	Unknown	No	Success	Clay						Pilot test completed	
UnknownUnknownFair SuccessUnknownUnknownEair SuccessUnknownUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUUU	81-100%	Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	Fair Success	Unknown						Pilot test completed	
BioremediationUnknownUnknownEair SuccessSandIIIIChemical Oxidation-Fentors ReagentUnknownUnknownNot yet evaluatedGravelGravelIII		Bioremediation	Unknown	Unknown	Fair Success	Unknown	_					Pilot test completed	
Chemical Oxidation-Fentoris Reagent Unknown Not yet evaluated Gravel		Bioremediation	Unknown	Unknown	Fair Success	Sand						Pilot test completed	
Themal-Six phase heating Unknown Unknown Unknown Unknown Unknown Unknown Unknown Unknown Not yet evaluated Till D <thd< th=""> <thd< th=""> D <</thd<></thd<>		Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	Not yet evaluated	Gravel	-					Pilot test completed	
Themal-Six phase heating Unknown Unknown Not yet evaluated Till Image		Thermal-Six phase heating	Unknown	Unknown	Success	Unknown	-					Pilot test completed	
Thermal-Six phase heating Unknown Unknown Success Clay Image Image X X Bioremediation >90% No Success Sand P X X X Excavation >80% Yes Success Sand X X X X Excavation >80<690%		Thermal-Six phase heating	Unknown	Unknown	Not yet evaluated	III						Pilot test completed	
Bioremediation >90% No Success Sand Image: Constraint of the second of the s		Thermal-Six phase heating	Unknown	Unknown	Success	Clay				×		Pilot test completed	
Excavation >80 < 40% Yes Success Sand X X X X X Chemical Oxidation-Permanganate >25 < 50%		Bioremediation	%06<	No	Success	Sand						Pilot test completed	On going
Chemical Oxidation-Permanganate >25 550% Yes Fair Success Sand X X X X Bioremediation Unknown No Success Clay M X X X X Surfactant Flushing Unknown Unknown No Success Clay M X X X X ZVI - EZVI/Inano-scale ion >25 <50%		Excavation	>80 <90%	Yes	Success	Sand	_	×		×	×	Full-scale completed	On going
Bioremediation Unknown No Success Clay Image: Clay <td>61 900/</td> <td>Chemical Oxidation-Permanganate</td> <td>>25 <50%</td> <td>Yes</td> <td>Fair Success</td> <td>Sand</td> <td>_</td> <td></td> <td>×</td> <td>×</td> <td>×</td> <td>Pilot test completed</td> <td></td>	61 900/	Chemical Oxidation-Permanganate	>25 <50%	Yes	Fair Success	Sand	_		×	×	×	Pilot test completed	
Unknown Unknown Fair Success Silt X >25 <50%	% 00-10	Bioremediation	Unknown	No	Success	Clay	_					Full-scale completed	On going
Name Name Name Name Name Name >25 <50%		Surfactant Flushing	linknown	l lakoova	Fair Success	tio			×			Pilot test completed	
nganate Unknown Unknown Not yet evaluated Metamorphic The Netamorphic X X X X X X X X X X X X X X X X X X X	41-60%	ZVI - EZVI/nano-scale iron	>25 <50%	No	Fair Success	Sand			< ×	×		Pilot test completed	
Excavation Inknown Late Success Limestone X X X	700	Chemical Oxidation-Permanganate	Unknown	Unknown	Not yet evaluated	Metamorphic						Pilot test completed	
	80	Excavation	Unknown	Unknown	Fair Success	Limestone	×	×		×	×	Full-scale completed	On going

TABLE D.14: SUMMARY OF REBOUND SORTED BY TECHNOLOGY Navy ROCS

							DNAPL	DNAPL Distribution	ç			
												Post
		Source Mass	Mass Flux	Perceived		Dead-end	Diffused into					Treatment
Rebound	Technology	Removal	Decrease	Success	Lithology	fractures	Low K Layers In pools	In pools	Residual	Sorbed	Stage	Monitoring
	Chemical Oxidation-Permanganate	%06<	81-100%	Success	Sand				×	X	Pilot test completed	
	Chemical Oxidation-Permanganate	Unknown	Unknown	Success	Shale	×	×	×	×	×	Pilot test completed	
	Excavation	>80 <90%	61-80%	Success	Sand		×		×	×	Full-scale completed	On going
Yes	Chemical Oxidation-Permanganate	>50 <80%	Unknown	Fair Success	Sand			×	×	×	Pilot test completed	
	Chemical Oxidation-Permanganate	>25 <50%	61-80%	Fair Success	Sand			×	×	×	Pilot test completed	
	Thermal - Low temp-six phase heating	>10 <25%	Unknown	Success	Silt			×	×	×	Pilot test completed	
	Chemical Oxidation-Permanganate	Unknown	Unknown	Fair Success	Gravel						Pilot test completed	
	Chemical Oxidation-Permanganate	100%	81-100%	Success	Sand				×		Pilot test completed	
	Other - Conductive Heating	100%	Unknown	Success	Clay				×		Full-scale completed	
	Bioremediation	>00%	61-80%	Success	Sand						Pilot test completed	On going
	Excavation	>00%	Unknown	Success	Silt		×	×	×	×	Full-scale completed	On going
	Chemical Oxidation-Fenton's Reagent	>00%	Unknown	Success	Clay		×		×	×	Full-scale completed	On going
	Thermal-Steam	>00%	Unknown	Success	Clay		×				Full-scale completed	On going
Q	Chemical Oxidation-Fenton's Reagent	>00%	Unknown	Success	Sand						Full-scale completed	On going
2	Chemical Oxidation-Fenton's Reagent	>80 <90%	Unknown	Fair Success	Silt				×		Pilot test completed	
	ZVI - EZVI/nano-scale iron	>25 <50%	41-60%	Fair Success	Sand			×	×		Pilot test completed	
	Thermal-Six phase heating	Unknown	81-100%	Success	Clay						Pilot test completed	
	Bioremediation	Unknown	61-80%	Success	Clay						Full-scale completed	On going
	Bioremediation	Unknown	Unknown	Success	Sand		×	×	×	×	Pilot test completed	On going
	Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	Success	Gravel						Pilot test completed	
	ZVI -Zero-valent iron	Unknown	Unknown	Success	Sand				×	×	Pilot test completed	

TABLE D.16: SUMMARY OF IMPACTS TO SECONDARY GROUNDWATER QUALITY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Nary ROCS

					INUM	SER OF SITE	NUMBER OF SITES THAT COULD BE DESCRIBED BY SECONDARY GROUNDWATER QUALITY PARAMETER PER TECHNOLOGY	D BE DESCE	UBED BY SEC	ONDARY G	ROUNDWATE	R QUALITY P	ARAMETER PI	ER TECHNOLA	ADC						
IMPACTS TO SECONDARY	BIOREMEDIATION	CHEMI	CHEMICAL OXIDATION	z	сo	DUAL-P	DUAL-PHASE EXTRACTION	I NOIL:	EXCAVATION		OTHER S	SURFACTANT			THERMAL			_	ZERO-VALENT IRON	T IRON	
GROUNDWATER QUALITY		Fenton's Reagent	Permanganate Ozone		SOLVENT FLUSHING	Water/Air W	Water/DNAPL DNAPL/Air)NAPL/Air		Undefined	Undefined Pump and Treat	FLUSHING	Conductive Low-Temp Heating Six-Phase		Resistive Six Heating H	Six-Phase P Heating H	Three-Stu Phase Flus Heating	Steam Flushing Em	Emulsified With Clay ZVI only	ay ZVI only	TOTAL
Methane generation	7 (63.6%)																	1.(1	1 (100.0%)		8 (21.6%)
Hydrogen sulfide generation	1 (9.1%)																				1 (2.7%)
Dissolved iron generation	5 (45.5%)					1 (33.3%)	1 (50.0%)														7 (18.9%)
Dissolved manganese generation			5 (55.6%)			1 (33.3%)	1 (50.0%)														7 (18.9%)
Elevated concentration of other metals ^c		1 (100%)	7 (77.8%)																		8 (21.6%)
Increased BOD ^a	4 (36.4%)																				0 (0.0%)
Change in pH	6 (54.5%)		1 (11.1%)																		7 (18.9%)
Reduced soil porosity	2 (18.2%)		1 (11.1%)															1.(1	1 (100.0%)	2 (100.0%)) 6 (16.2%)
Change in hydraulic conductivity		1(100.0%)	1 (11.1%)										1 (100.0%)							2 (100.0%	5 (13.5%)
Change in groundwater color			7 (77.8%)				1 (50.0%)														8 (21.6%)
Increased dissolved solids	2 (18.2%)		1 (11.1%)																		3 (8.1%)
Redistribution of DNAPL		1 (100.0%)	1 (11.1%)				1 (50.0%)		1 (50.0%)												4 (10.8%)
Increase in source area volume							_						_	_	_	_	_	_	_		0 (0.0%)
Decrease in source area volume	2 (18.2%)		1 (11.1%)			2 (66.7%)	1 (50.0%)	1 (100.0%)	2 (100.0%)				_	_	_	1.(1	1 (100.0%) 1 (100.0%)	(%0.00	_	1 (50.0%)	12 (32.4%)
Other ^b	1 (9.1%)		2 (22.2%)				1 (50%)							1 (100%)	1(1 (100%)		1(1 (100%)		7 (18.9%)
TOTAL SITES WITH DATA	п	-	6	0	0		2	1	2	0	0	0	-	-	0	1	1	1	1 0	2	37

Nat Billal, spaces indicate no sites were impacted by the corresponding criteria. ¹ Biordenical cosygen demand (BOD). ¹ Norder "response included the following: Biorenordiation - One survey respondent specified the following Chemical oxygen demand (BOD). ¹ Biorenordiation - One survey respondent specified the following Chemical oxydent specified the following Biorenordiation - One survey respondent specified the following Chemical oxydent specified the following Data Phase settation of Vater. And DNAPL, exponse included "None contaminant runs kinds the theorem is inpact to color. ¹ Data Phase settation of Vater and DNAPL, exponse included "None contaminant runs and discontinued in order to focus criterion and a fifterent technology." ¹ Contenting permanganate - Response included "None contaminant runs and discontinued in order to focus criterion and a fifterent technology." ² Specified devarde concentrations in match induction and and the following. ² Specified devarde concentrations in match included "Phase and DNAPL component." ² Specified devarde concentrations in match induction and a different technology." ³ Specified devarde concentrations in match induction and a profession permonangiant. ⁴ Specified devarde concentrations in match induction and a profession permonangiant. ⁴ Specified devarde concentrations in match induction and a profession permonangiant (3), and transient increases of some metals (unspecified, 1 site).

TR0132\Report\Final\ Table D.16 - Secondary gw impacts

TABLE D.18: SUMMARY OF GEOLOGY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Nay ROCS

7	FLUSHING Water/Air Water/DNAPL DNAPL/Air
-	
-	
- -	- -
5	5
-	-
- •	
4	4
_	
3 5 7	3 5 7
	1
-	-
1	-
1 2 2	
7 0 -	4 0 0
-	-
-	-
0 2 2	0 2 2
0	•
с. С	С
-	-
0 1	0 1
	-
-	-
-	
3 3	3 3
s S	ы С
-	-
0 1	0 1
5 7	5 7

Notes *Ranges represent the portion of the site aquifer that can be described with the related parameter. Blank spaces denote no data entered for that parameter.

TR0132/Report/Final/ Table D.18 - Geology

TABLE D.23: SUMMARY OF HYDROGEOLOGY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

							~	NUMBER OF SITES THAT COULD BE DESCRIBED BY HYDROGEOLOGY PARAMETER PER TECHNOLOGY	TTES THAT C	OULD BE DES	CRIBED BY .	HYDROGEOU	JOGY PARAM	ETER PER	ECHINULUU	_							TOTAL SITES
		BIOREMEDIATION	CHEMIC	CHEMICAL OXIDATION		-0-	DUAL-P	DUAL-PHASE EXTRACTION		EXCAVATION	OTHER		SURFACTANT			THERMAL	2			ZERO-VALENT IRON	NT IRON	UNSPECIFIED	WITH
HYDROGEOLO(HYDROGEOLOGY PARAMETERS		Fenton's Reagent	Permanganate	Ozone	SOLVENT FLUSHING V	Water/Air	Water/Air Water/DNAPL DNAPL/Air			Undefined	Pump and Treat		Conductive Low-Temp Heating Six-Phase		Resistive Six-Phase Heating Heating		Three-Si Phase Flu Heating Flu	Steam Flushing Emul	Isified With C	Emulsified With Clay ZVI only	TECHNOLOGY	SI
	< 10 ft	6	2	5	-		3	4	1	4	_	-		-	1		2	3	1	1 1	3	8	42
	10 to 20 ft	7	7	5					-	ю	4		_			-	7	_	-			-	28
	20 to 50 ft	5		_			_		-	_	_		-	_	-	-		-	2		-	2	16
Depth to	50 to 100 ft			-						-									-			2	3
Groundwater	100 to 200 ft			-													_					-	2
	> 200 ft	-																					1
	Unspecified	3	s	2			-	-		2	-		2			_	5		3			2	26
	TOTAL	25	6	15	1	0	5	S	3	11	9	1	4	1	1	2	10	5	8 1	1 1	4	19	118
	< 0.001 ft/day				F														1		1		3
	0.001 to 1 ft/day	10	-	5				-	-	2	2		2			-	-	3	3	-	-	7	33
Groundwater	> 1 fh/day	4		-				-		3	-					_	_	_	-	_		-	14
V elocity	Unknown	11	8	9	1		5	3	2	6	3	1	2	1	1	1	8	1	3		2	11	68
	TOTAL	25	6	15	1	0	5	5	3	11	9	1	4	1	1	2	10	5	8	1 1	4	19	118
	< 10 ^{°5} ft/day	1								1	2			1			1		1		1		8
Hydraulic	10 ⁻⁵ to 10 ft/day	6	3	7			-	2	2	3	2	1	2			1		3	2	-	-	6	40
Conductivity	> 10 ft/day	7		3					-	2			-				-	5	2	_		2	21
	Unknown	8	6	5	1		4	3		5	2		1		-	-	8		3		2	8	49
	TOTAL	25	6	15	1	0	5	5	3	11	9	1	4	1	1	2	10	5	8	1 1	4	19	118
	< 0.0001	2									H					-		$\left \right $			-	1	3
	0.0001 to 0.001	2		_				-	-	-			-		_	-	_	5	1	_		4	11
Horizontal	0.001 to 0.01	4	-	2						-	5	-			_	_	-	5	3	-	-	2	20
Hydraulic	0.01 to 0.1	5		e				-		7	-		-						-			-	14
Gradient	0.1 to 1			5						-													3
	Unknown	12	8	8	-		5	3	2	6	2		2	1	1	-	9	1	3		2	11	67
	TOTAL	25	6	15	1	0	5	5	3	11	6	1	4	1	1	2	10	5	8	1 1	4	19	118
	< 0.0001	2		-						2									1		-		7
	0.0001 to 0.001	3							-	3	5		-			_	_		3			-	18
Vertical Hydraulic	c 0.001 to 0.01		-				+				+			T		+	+	+	+			-	0 -
Gradient	0.1410		-													+		+				-	
	Unknown	20	~	14	-		5	s	2	9	4	-	e	-	-	-	6	5	4	-		17	92
	TOTAL	25	6	15	1	0	s	ŝ	3	11	9	1	4	1	1	2	10	5	8	1	4	19	118
	<1 gpm				F		1	1		1	1												4
	1 to 10 gpm	2	-	9				2		-	_		-		_	_	_	_	-	_	_	e	12
Suctoinable Well				7			-		-	3	-	-				_	-	5	2		-	2	17
SUSTAILIADIC WEIL Viold				-					-	-	-		-				+	_	1				10
nari	> 1000 gpm	- :		-			-	-		-	,				-	_	-		-		-	:	- 1
	Unspecified	18	~	6	_		3	2	_	so :	6		2	_	_	_	6	_	4		3	14	74
	TOTAL	25	9	15	-	0	2	5	3	11	9	1	4	1	1	2	10	5	8	-	4	19	118

Notes Blank spaces denote no data entered for that parameter.

>0 - 10% 0% %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 >0 - 10% 0% 0% 0% 0% >20 - 30% %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 %0 2 20 - 10% 0% 0% 0% 0% 0% 0% 0% 0% 10% 0% 0% 0% 0% 0% >70 - 80% 0% 0% 0% 0% >70 - 80% >20 - 30% >10 - 20% %0 >90 - <100% 100% 0% >20 - 30% 100% >70 - 80% 100% 100% >10 - 20% 100% 100% 100% 100% 100% >10 - 20% >70 - 80% >80 -90% >70 - 80% 100% 100% 100% 100% 100% 100% 100% 100% 100% 100% 100% 0% 100% 100% 100% %00I Excavation Excavation Excavation Other - Conductive Heating Other - Low temp-six phase heating Thermal-Six phase heating Chemical Oxidation-Fenton's Reagent Bioremediation Chemical Oxidation-Fenton's Reagen No ZVI/nano-scale iron-Zero-valent iron Unknown ZVI/nano-scale iron-Zero-valent iron Chemical Oxidation-Fenton's Reager Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Chemical Oxidation-Permanganate Dual Phase-DNAPL/air extraction Dual Phase-Water/air extraction Dual Phase-Water/air extraction Thermal-Six phase heating Thermal-Six phase heating Thermal-Six phase heating Other - EZVI/nano-scale iron Thermal-Resistive heating Surfactant Flushing Surfactant Flushing Bioremediation Bioremediation Bioremediation Bioremediation Bioremediation team Thermal-Steam Bioremediation Bioremediation Technology Therma Unknown Yes Unknown No No Unknown Unknown Unknown Unknown Unknown No Unknown Yes No Unknown Unknown Perceived Success Source mass removal Mass Flux Decrease Rebound >90% 61-80% No Unknown Jnknown Unknown Unknown Unknown Unknown Unknown Unknown Jnknown Unknown Unknown Yes Yes No No No ٩N No Yes No No Yes ٩N No 61-80% Unknown Unknown Unknown Unknown Unknown Unknown 81-100% 81-100% 81-100% Unknown Unknown 81-100% 81-100% 81-100% 81-100% 81-100% 81-100% Unknown Unknown Unknown 81-100% 81-100% Unknown 61-80% Unknown Unknown 61-80% Unknown 0% 81-100% 61-80% Unknown Unknown Unknown Unknown Unknown Unknown Unknown Unknown 41-60% 81-100% 100% >10 <25% Unknown Unknown Unknown Unknown >90% Unknown Unknown Unknown >80 <90% Unknown >50 <80% >25 <50% Unknown Unknown Unknown Unknown Unknown Vnknown >90% 100% >90% >80 <90% Unknown Unknown Unknown Unknown >50 <80% >80 <90% Unknown >25 <50% Unknown Unknown Unknown Unknown %06< 100% ×06< %06< <10% Navy ROCS Fair Success Poor Success Success

TABLE D.25: SUMMARY OF SUCCESS DATA SORTED BY DNAPL COMPOSITION

TR0132\Report\Final\ Table D.25 - Success by DNAPL comp

%0 %0

%0 %0

100%

%0

⁻enton's Reager

themical Oxidation-F

Jnknown

>40 - 50% 0%

>40 - 50% 100%

Chemical Oxidation-Permanganate Thermal-Six phase heating Thermal-Six phase heating

Unknown Unknown Unknown

81-100% 0% 81-100% 81-100%

Unknown Unknown Unknown 100%

Not yet evaluated

TABLE D.26: SUMMARY OF GROUNDWATER GEOCHEMISTRY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

									NUMBER	JF SITES THAT	COULD BE DE	SCRIBED BY	GEOCHEMIST	RY PARAME	TER PER TECH						1		TOTAL SIT
	atsuvanjo:		BIOREMEDIATION		EMICAL OXI	NOITAO	co-		T-PHASE EXT	RACTION	EXCAVATION	OTH	R SUR	FACTANT	-	THE	RMAL			ZERO-VALE	INT IRON	UNSPECIFIEI	D
	OCHEMIST	ky pakameteks		Fenton's Reagent					r Water/DNAP	L DNAPL/Air		Undefined	ump and FI Treat	USHING C	onductive Low- Heating Six-F	Temp Resistiv	e Six-Phase Heating			ulsified With 6	Clay ZVI on		Y TECHNOLO SPECIFIE
		< 1 mg/L 1 += 10 == -1	s 5	-					-		-							-	1	1		- 3	12
		1 to 10 mg/L 10 to 100 mg/L								-	-			_				-	5		-	-	1 0
	Nitrate	100 to 1,000 mg/L	-																				-
		>1,000 mg/L			-																		0,0
		I Independent	- 12	=	- 11	-	-	v			9	4	-		-	,	=	"	~	-	"	0	7 9
		Unknown < 1 mo/1.	4	-	2	-		0	4	0	-	0	_	c	_	7	=	- -	o –	-		r c	00 6
		1 to 10 mg/L	_							-				_							•	1	, s
		10 to 100 mg/L																					•
	Nitrite	100 to 1,000 mg/L																					•
		>1,000 mg/L																					•
		Non-detect	20	:	- :						:	,		,		•	:		,	•	•	- :	e :
		Unknown	70	=	9	-	0	^	0	2	=	0	_	3	_	7	=	n	9	_	4	0	=
		1 to 10 mo/L	v								-										-	,	
		10 to 100 mo/L			-						-								-	_	-	4 -	~ ∝
	ulfate	100 to 1.000 mg/L							-			-									-	. 0	
Number Numer Numer Numer <th></th> <th>>1.000 mg/L</th> <td></td> <td>-</td> <td></td> <td>-</td> <td>6</td>		>1.000 mg/L		-																		-	6
		Non-detect																					•
		Unknown	14	11	15	1	0	5	4	4	10	5	1	4	1	1 2	11	5	7	0 1	3	7	105
		< 1 mg/L	-		_	_				_		-			_	_	_	_	_	_	-	-	3
		1 to 10 mg/L	-		_	_				_	-				_	_	_	_		_	_	2	2
		10 to 100 mg/L																					•
Not with the second s	lfide	100 to 1,000 mg/L																					•
Methode Mathode Mathode <t< td=""><th></th><th>>1,000 mg/L</th><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>_</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>0</td></t<>		>1,000 mg/L									_												0
$ = \left[$		Non-detect	- 2	2		-	<				=		-		-	,	=		•	-	-	2	121
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			+7	71	CT	-		0	-	t	=	n	-	t	-	7	-	0	0	-	t -	- 10	171
1000000000000000000000000000000000000		1 to 10 mg/L	-																		-	-	10
1000.0001 1000.0001 <t< td=""><th></th><th>10 to 100 mg/L</th><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>2</td></t<>		10 to 100 mg/L																					2
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	sphate	100 to 1,000 mg/L																					•
Networks: 31 16 1 0 3 4 1 4 1 4 1 Networks: 1		>1,000 mg/L																					0
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		Non-detect	2	2	1	-				,	2				-	,	:		a	-	•	:	- ?
In b (Nmit, 1) (0.00mit, 1) (0.00m		< 1 ma/l	47	12	01	-		0	0	t	71	>	-	t	-	7	-	0	•	-	t	-	±71 0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		1 to 10 mg/L.			-																		-
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		10 to 100 mg/L	2																		-	2	6
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	cium	100 to 1,000 mg/L	-	-	-				-		-		-					-	2			2	6
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		>1,000 mg/L																					0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Non-detect									_												0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Unknown	24	=	14	-	0	5	4	4	=	9	0	4	-	1 2	=	4	9	1	4	6	115
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		< 1 mg/L 1 to 10 mg/l	-																	_	-		- ~
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10 to 100 mg/L	_		5				-		_		-					-	2			2	6
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	nesium	100 to 1,000 mg/L		-																		-	-
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		>1,000 mg/L																					0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Non-detect				-																2	0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Unknown	25	П	14	-	0	S	4	4	=	9	0	4	_	1 2	Π	4	9	0	4	10	115
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		< 1 mg/L							-		_								_			-	- 7
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		1 to 10 mg/L							-												-	,	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	onate	100 to 1,000 mg/L		-																		-	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		>1,000 mg/L																					0
1 million 1		Non-detect	30	=	91	-	4	v	-		=	3	-	4	-	,	=	v	r	-	-	d	- <u>-</u>
1 10 (1002L) 1 10 (1002L) 1 10 (1002L) 1 10 (1002L) 1 10 (1002L) 1 10 (1002L) 1 10 (1002L) 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		UIIKHOWII	1	=	01			n	4	+ -	-	•	_	t -	1	7		0 -	- c	1	+ +	4	171
100 100 mgL 100 100 mgL 100 100 100 mgL 100 100 mgL 100 100 100 mgL 100 100 mgL 100 100 100 100 100 mgL 100 100 mgL 100 100 100 100 100 mgL 100 100 mgL 100 100 100 100 100 100 100 mgL 100 100 mgL 100 100 100 100 100 100 100 100 100 100		1 to 10 mo/L	-							-	-	-		-				-	4		-	-	• -
1000 (M0mgL 1000 (M0mgL 1000 (M0mgL 1000 (MmgL 1000		10 to 100 mg/L																					•
	rcury	100 to 1,000 mg/L				_									_	_		_		_	_		•
12 16 1 0 5 3 11 5 0 3 1 1 2 1 4 6 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1 1 4 1		>1,000 mg/L	-		+	+																	•
		I henowin	25	1	16	-	-	*	v	5	=	v		5	-	·	=	4	y	-	4	1	7

TR0132\Report\Final\ table D.26 - groundwater geochemistry

TABLE D.26: SUMMARY OF GROUNDWATER GEOCHEMISTRY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

	CHEMISTR		THE PARTY NAMES OF TAXABLE PARTY.		01 00 00 X 10 X	1000	0.0	1.11	CONTRACTOR OF A DESCRIPTION OF A DESCRIP	1.0000	A REAL PROPERTY OF A REAL PROPERTY.	0.000	1	and a second second			A REAL PROPERTY.			ľ		1000		2
	TTOTTOTTO /		BIOREMEDIATION		ICAL OXIDA		co-		-PHASE EXTR	ACTION	EXCAVATION	ΠO	HER S	URFACTANT			THERM	- 1-	- 1-		ERO-VALE	NTIRON	UNSPECIFIE	MITH
		A LANAMETERS		Fenton's Reagent	Permangana	Ozone	SOLVENT		Water/DNAPL	DNAPL/Air			Pump and Treat	FLUSHING	Conductive Heating	Low-Temp Six-Phase	Resistive 5 Heating				fied With C	lay ZVI on		Y TECHNOLOGY SPECIFIED
		< 1 mg/L	-	-				_	-		-		-						, i			-	3	7
		1 to 10 mg/L	-							-		-		1					-					، و
		10 to 100 mg/L																+						
	NI Sellic	100 I0 1,000 IIB/L																+						
		Non-detect	-																				-	-
		Unknown	24	=	16	-	0	s	4	6	=	5	0		-	-	2	=		-	-	4	6	114
		< 1 mg/L		-							-		,				1	:				-	-	6
		1 to 10 mg/L																						0
		10 to 100 mg/L																						0
	xavalent	100 to 1.000 mg/L																						0
	romium	>1,000 mg/L																						•
1 1		Non-detect			-											ŀ								6
		Unknown	25	=	15	-	•	2	~	e	=	9	-	e	-	-	2	=		-	-	4	12	117
		< 0.1 mg/L	en														1					-	6	4
		0.1 to 0.5 me/L														ŀ								•
		0.5 to 1 mg/L																						0
Note: Note: <th< td=""><td>drogen</td><td>1 to 1.5 mg/L</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>0</td></th<>	drogen	1 to 1.5 mg/L																						0
Nervent Biology Normal Biology Normal		- 15 mol																						
		A to the total																						
		Indiana	24	2	2	-				-	5		-	-	-		,	:	+		•	•	:	2
1 1		Unknown	24	12	10	-	0	^	<u>.</u>	4	12	0	-	4	_	_	7	=	-		-	4.	= .	124
10 10 1		< 1 mg/L	4	_					-		_									-		-	_	<u>e</u> .
00.0000000000000000000000000000000000		1 to 10 mg/L	-		-								-									-		4
Normatic		10 to 100 mg/L	5							-		-		_					-				2	-
1 1	ved Iron	100 to 1,000 mg/L	-					-										-	_		_	_		
Network Network <t< td=""><td></td><td>>1,000 mg/L</td><td></td><td></td><td></td><td></td><td></td><td>-</td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>-</td><td>_</td><td></td><td>_</td><td>_</td><td></td><td>0</td></t<>		>1,000 mg/L						-										-	_		_	_		0
1 1		Non-detect						-										-			_	_		0
10.1008L 1<		Unknown	19	11	15	1	0	5	4	3	11	5	0	3	-	-	2	11		_	1	3	6	106
1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 2 1 1 2 1		< 1 mg/L	3		-						-								-		_	-		2
Net of the field of t		1 to 10 mg/L	3	-	2				-		-									-			-	6
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		10 to 100 mg/L	-							-			-	_					+				-	~
Number Number	al Iron	100 to 1,000 mg/L	_																			-		7
Without Minimum 1 1 0 0 1 0 0 1 0 1 0 0 1		~1,000 mg/L																t						•
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		I labored	01	=	2	-		v		,	9	9	-	,	-	-	,	=	+		-	~	=	2
		UIIKIIUWII	6	-	-	-	D		+ -	0	- 12	-	-	n	-	-	4	=			-	n -		102
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		1 to 10 mo/I	7 -		-				-	-	-		-	-				+	╞	-		-	n -	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10.10.10.11	-							-				-					╎			-	-	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		100 to 1000 mg/L	-																			-		4 0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		-1 000			-																			-
		~1,000 mg/L			-													t						- <
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Non-detect		9	:			,			:	,						:	+		•		0	0 99
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		UIKIIOWI	52	71	+1	-		0	+ -	^	-	0		c	-	-	7	-			-	^ •	6	601
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		< 1 mg/L 1 to 10 m o/T	~ -		-				-	-	-		-	-					-		-	-	7	8 4
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10 10 10 100 100 10	-		-					-				-				t	-			-		
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$, and a second		-															t				-		4 0
	lauganese																	+						
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Non-detact																						
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $		I htmaxim	66	12	15	-		v	4	"	=	4	-		-	-	,	=	+	-	-	"	=	51
	I	/ 10 ⁴ leaflea	77	-	3						-	,	,	'n			4	:		-		'n	:	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10 ⁴ to 10 ³ Lo/Lo																				-	,	1 -
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		10 10 10 NB/NB																+				-	4	-
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	on Organic-		-	_									-						-					0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	rbon				-																			-
Non-deter Non-deter <t< td=""><td></td><td>>10⁻¹ kg/kg</td><td>-</td><td></td><td></td><td></td><td>_</td><td></td><td></td><td>1</td><td></td><td></td><td></td><td>1</td><td></td><td></td><td></td><td></td><td>-</td><td></td><td></td><td></td><td></td><td>\$</td></t<>		>10 ⁻¹ kg/kg	-				_			1				1					-					\$
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		Non-detect				_													_		_	_		0
1 1 1 1 1 1 1 1 1 10/mgL 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		Unknown	25	=	15	-	0	5	5	3	П	9	0	3	-	-	2	11		1	1	4	11	114
1000%L 1 1 1 1 1 1000%L 1 1		< 1 mg/L	_														+				-	-		7
1000 ungL 1000 ugL 1000 ugU 1000 ugL 1000 ugU 1000 ugL 1000 ugL		1 to 10 mg/L	-						-	-				_				+	-					0
Non-Weil																		+						•
Titon mgic 1 2 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 1 2 1 <th1< th=""> <th1< th=""> <th1< th=""> <th1< t<="" td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td>T</td><td></td><td></td><td></td><td>+</td><td>+</td><td>-</td><td></td></th1<></th1<></th1<></th1<>																	T				+	+	-	
	(inc)	Alon dataot	-															+						
		I Infmontin	24	12	16	-		v	4		12	9				Ì								-

TR0132\Report\Final\ table D.26 - groundwater geochemistry

Page 2 of 3

TABLE D.26: SUMMARY OF GROUNDWATER GEOCHEMISTRY DATA COLLECTED WITH THE SURVEY SORTED BY TECHNOLOGY Navy ROCS

								NUMBER OF	SITES THAT	COULD BE DE	SCRIBED B	Y GEOCHEM	NUMBER OF SITES THAT COULD BE DESCRIBED BY GEOCHEMISTRY PARAMETER PER TECHNOLOGY	ETER PER	TECHNOLO	AC							TOTAL STEES
		BIOREMEDIATION		CHEMICAL OXIDATION	NOL	c0-	DUAL	DUAL-PHASE EXTRA	EXTRACTION	EXCAVATION		OTHER S	SURFACTANT			THERMAL			⊢	ZERO-VAI	ZERO-VALENT IRON	UNSPECIFIED	
GEOCHEMIST	GEOCHEMISTRY PARAMETERS		Fenton's Reagent	Permanganate	te Ozone	SOLVENT	Water/Air	Water/Air Water/DNAPL DNAPL/Air	DNAPL/Air		Undefined	Pump and Treat	FLUSHING	Conductive Heating	Conductive Low-Temp Heating Six-Phase	Resistive Six-Phase Heating Heating		Three- Phase Heating	Steam Flushing	ulsified	Emulsified With Clay ZVI only		SIS
	< 1 mg/L	-							1	_			1				·	-	1	-	-		7
	1 to 10 mg/L	2	-					_													_	-	4
Dissolved																	+	+			-	-	2
Organic Carbon	=																	+					_
(DOC)	>1,000 mg/L																						0
	Uhknown	22	=	16	-	0	v	4	"	12	9	-		-	-	2	=	4	2	•	"	=	114
	< 1 mg/L	.0	-	-						-	-		b						- 7			-	13
	1 to 5 mg/L	2		_				_		2											-		12
	1	-							-				-					_	-			3	5
Dissolved Oxygen																					_		0
	Non-detect					_																	0
	Unknown	16	11	14	1	0	5	4	3	9	5	1	3	1	1	2	11	3	5	0	1 3	9	99
	< 4																						0
	4 to 6	4								-										_	1 4	4	10
	6 to 7	7	2	2		_	_	-	1	1	-		-					2	2		-		21
μd	7 to 8	2		-				-		-									-	_	_	3	7
	8 to 9																						0
	> 9	-																					-
	Unknown	13	10	13	-	0	5	3	3	9	5	-	3	-	-	2	11	3	5	0	0 0	6	89
	<-150 mV	4									-						+	+			2	-	7
Oxidation-	-150 to -50 mV	4		-													+	_	1	-	-	-	6
Reduction	-50 to 0 mV	-	-							2													4
Potential (ORP)				-				-								+						_	2
		9														_				_	-		4
	Unknown	15	=	14	-	0	2	4	4	10	5	-	4	_	_	2	=	4	7	0	1	6	102
	< 1 mg/L		-						-	-			-				+	_	-	_		-	14
	1 to 10 mg/L	7															+	+				-	7
Mathema	10 to 100 mg/L	-													t	t	+					-	- <
Memane	100 to 200 mg/L														t	t	+						
	> 200 mg/L Non-detect			-														+				-	-
	Unknown	17	=	15	-	0	5	2		11	9	-	3	-	-	2	=	4	7	0	1 5	10	110
	< 1 mg/L	4	1							-													9
	1 to 10 mg/L																					-	0
	10 to 20 mg/L																			_	_		0
Ethane	20 to 30 mg/L																+	+					0
	30 to 40 mg/L																+	+					0.
	A 40 mg/L Non-detect	- 6		-														+				-	4
	Unknown	19	1	15	-	0	S	5	4	11	9	-	4	-	-	2	=	~	~	-	1 5	. 11	117
	< 1 mg/L	4								-								$\left \right $					9
	1 to 10 mg/L	2																					2
	10 to 100 mg/L	_																					_
Ethene	100 to 1,000 mg/L		-													+		+					_ <
	1/000 to 2/000 m - 1														t	t	+						
	> 2,000 mg/L			-							-					+		+				,	
	I Jnknown	7	- 01	-	-	0	~	5	4	1	- ~	-	4	-	-	2	=	~	8	0	1	7	ر 113
TOTAL SITES WITH	S WITH			2		,		,			,					4		,	2	,		:	
GEOCHEMISTRY DATA	STRY DATA	27	12	16	-	0	ŝ	ŝ	4	12	9	1	4	-	-	7	=	ŝ	œ	-	1 5	13	128

Notes Blank spaces denote no data entered for that parameter.

TABLE D.27: SUMMARY OF SUCCESS DATA COLLECTED WITH THE SURVEY SORTED BY AREAL EXTENT	
TABLE D.27: SUMMARY (Navy ROCS

Areal Extent (ft ²)	Technology	Lithology	Source mass removal	Mass Flux Decrease	Rebound	Perceived Success
	Chemical Oxidation-Permanganate	Sand	100%	81-100%	No	Success
1 45 / 10	Excavation	Sand	100%	Unknown	Unknown	Success
	Chemical Oxidation-Permanganate	Sand	>90%	81-100%	Yes	Success
	Bioremediation	Sand	Unknown	Unknown	Unknown	Success
	Chemical Oxidation-Permanganate	Sand	>50 <80%	Unknown	Yes	Fair Success
10 to < 100	Chemical Oxidation-Permanganate	Sand	>25 <50%	61-80%	Yes	Fair Success
	ZVI/nano-scale iron-Zero-valent iron	Sand	Unknown	Unknown	No	Success
	Excavation	Sand	>80 <90%	61-80%	Yes	Success
100 to /1 000	ZVI - EZVI/nano-scale iron	Sand	>25 <50%	41-60%	No	Fair Success
100 10 11 000	ZVI/nano-scale iron-Zero-valent iron	Sand	Unknown	Unknown	Unknown	Success
	Thermal-Resistive heating	Clay	Unknown	Unknown	Unknown	Fair Success
	Thermal - Conductive Heating	Clay	100%	Unknown	No	Success
1 000 to <10 000	Chemical Oxidation-Fenton's Reagent	Gravel	Unknown	Unknown	No	Success
1,000 10 ~10,000	Bioremediation	Sand	Unknown	Unknown	Unknown	Success
	Dual Phase-Water/air extraction	Gravel	Unknown	Unknown	Unknown	Fair Success
	Excavation	Silt	>00%	Unknown	No	Success
	Chemical Oxidation-Fenton's Reagent	Clay	>90%	Unknown	No	Success
	Thermal-Steam	Clay	>90%	Unknown	No	Success
	Chemical Oxidation-Fenton's Reagent	Silt	>80 <90%	Unknown	No	Fair Success
	Thermal-Steam	Sand	>50 <80%	81-100%	Unknown	Fair Success
	Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	81-100%	Unknown	Fair Success
10,000 to $<100,000$	Bioremediation	Unknown	Unknown	81-100%	Unknown	Fair Success
	Bioremediation	Sand	Unknown	Unknown	No	Success
	Dual Phase-Water/air extraction	Till	Unknown	Unknown	Unknown	Success
	Bioremediation	Clay	Unknown	Unknown	Unknown	Success
	Thermal-Six phase heating	Silt	Unknown	Unknown	Unknown	Success
	Chemical Oxidation-Permanganate	Gravel	Unknown	Unknown	Yes	Fair Success
	Dual Phase-DNAPL/air extraction	Silt	<10%	Unknown	Unknown	Poor Success
100,000 to <1,000,000	Thermal - Low temp-six phase heating	Silt	>10 <25%	Unknown	Yes	Success

TABLE D.28: SUMMARY OF SUCCESS DATA COLLECTED WITH THE SURVEY SORTED BY DNAPL DEPTH	
TABLE D.28: SUMMARY	Navy ROCS

DNAPL Depth (ft)	Technology	Lithology	Source mass removal	Mass Flux Decrease	Rebound	Perceived Success
	Chemical Oxidation-Permanganate	Gravel	Unknown	Unknown	Yes	Fair Success
1 to <10	Chemical Oxidation-Permanganate	Sand	100%	81-100%	No	Success
	Thermal-Steam	Sand	>50 <80%	81-100%	Unknown	Fair Success
	Bioremediation	Sand	Unknown	Unknown	Unknown	Success
	Bioremediation	Sand	Unknown	Unknown	Unknown	Success
	Bioremediation	Sand	Unknown	Unknown	No	Success
	Chemical Oxidation-Fenton's Reagent	Silt	>80 <90%	Unknown	No	Fair Success
	Chemical Oxidation-Fenton's Reagent	Unknown	Unknown	81-100%	Unknown	Fair Success
	Chemical Oxidation-Fenton's Reagent	Clay	>90%	Unknown	No	Success
	Chemical Oxidation-Permanganate	Sand	>50 <80%	Unknown	Yes	Fair Success
	Chemical Oxidation-Permanganate	Sand	>25 <50%	61-80%	Yes	Fair Success
	Chemical Oxidation-Permanganate	Sand	>90%	81-100%	Yes	Success
10 to <100	Dual Phase-Water/air extraction	Gravel	Unknown	Unknown	Unknown	Fair Success
	Excavation	Sand	100%	Unknown	Unknown	Success
	Excavation	Silt	>90%	Unknown	No	Success
	Thermal - Conductive Heating	Clay	100%	Unknown	No	Success
	Thermal - Low temp-six phase heating	Silt	>10 <25%	Unknown	Yes	Success
	Thermal-Resistive heating	Clay	Unknown	Unknown	Unknown	Fair Success
	Thermal-Six phase heating	Silt	Unknown	Unknown	Unknown	Success
	ZVI - EZVI/nano-scale iron	Sand	>25 <50%	41-60%	No	Fair Success
	ZVI/nano-scale iron-Zero-valent iron	Sand	Unknown	Unknown	No	Success
	ZVI/nano-scale iron-Zero-valent iron	Sand	Unknown	Unknown	Unknown	Success
100 to < 1,000	Thermal-Steam	Clay	>00%	Unknown	No	Success
1,000 to $<10,000$	Bioremediation	Unknown	Unknown	81-100%	Unknown	Fair Success
	Bioremediation	Clay	Unknown	Unknown	Unknown	Success
TInknown	Chemical Oxidation-Fenton's Reagent	Gravel	Unknown	Unknown	No	Success
	Dual Phase-Water/air extraction	Till	Unknown	Unknown	Unknown	Success
	Excavation	Sand	>80 <90%	61-80%	Yes	Success

TABLE D.29: SUMMARY OF SUCCESS DATA COLLECTED WITH THE SURVEY SORTED BY DNAPL DISTRIBUTION Navy ROCS

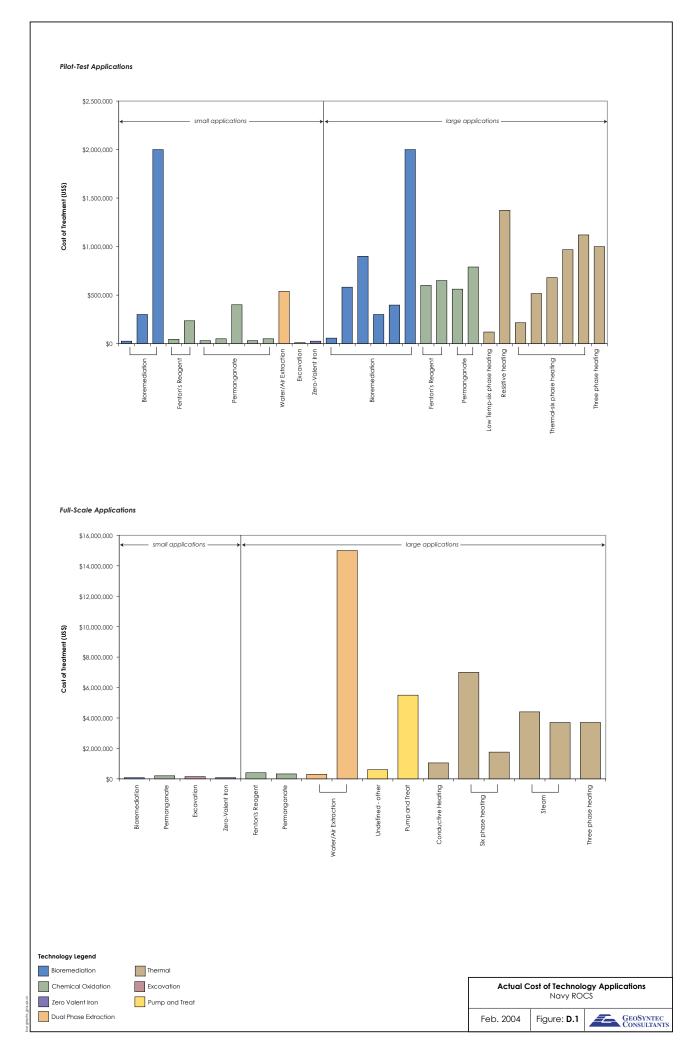
Pools
Ξ.
Ę
DNA
with]
Sites
10

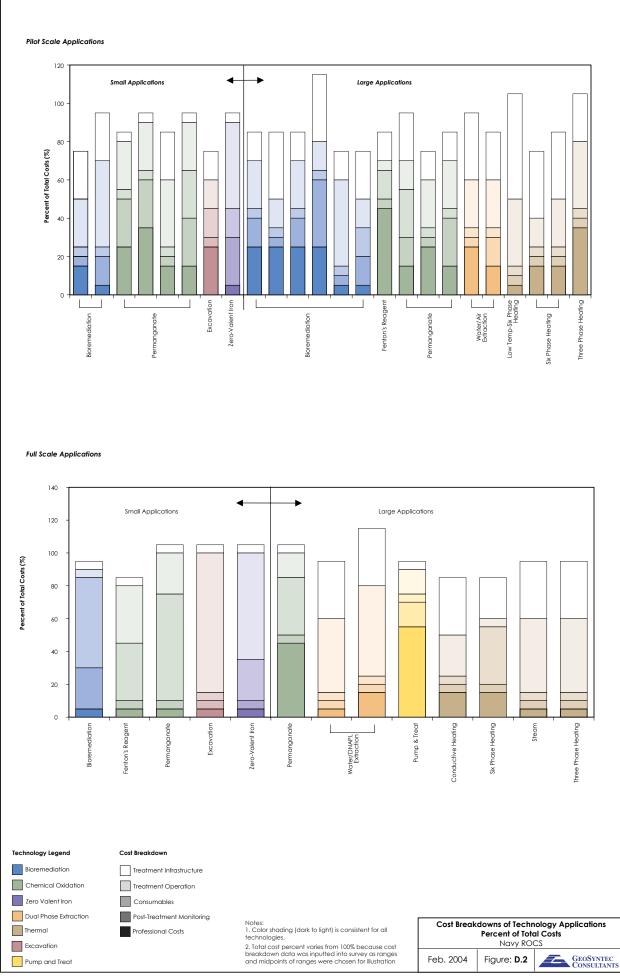
10 Sites with DNAPL in Pools						DNAPL Distribution in Subsurface	ribution in S	Jubsurface			
Technology	Source Mass Removal	Mass Flux Decrease	Rebound	Perceived Success	Dead-end Fractures	Diffused Into Low K Layers	In pools	Residual	Sorbed	Treatment Stage	Post Treatment Monitoring
Excavation	>06<	Unknown	No	Success		Х	Х	х	Х	Full-scale completed	On going
Chemical Oxidation-Permanganate	>50 <80%	Unknown	Yes	Fair Success			x	x	x	Pilot test completed	
Chemical Oxidation-Permanganate	>25 <50%	61-80%	Yes	Fair Success			x	x	x	Pilot test completed	
ZVI - EZVI/nano-scale iron	>25 <50%	41-60%	No	Fair Success			x	x		Pilot test completed	
Thermal - Low temp-six phase heating	>10 <25%	Unknown	Yes	Success			x	x	x	Pilot test completed	
Dual Phase-DNAPL/air extraction	<10%	Unknown	Unknown	Poor Success		x	x	x	Х	Pilot test completed	On going
Surfactant Flushing	Unknown	61-80%	Unknown	Fair Success			x			Pilot test completed	
Bioremediation	Unknown	Unknown	No	Success		x	x	x	x	Pilot test completed	On going
Thermal-Three phase heating	Unknown	Unknown	Unknown	Not Yet Evaluated		x	x	x	Х	Pilot test completed	
Thermal-Six phase heating	Unknown	Unknown	Unknown	Success			х	x		Pilot test completed	On going

One site with DNAPL trapped in dead-end fractures

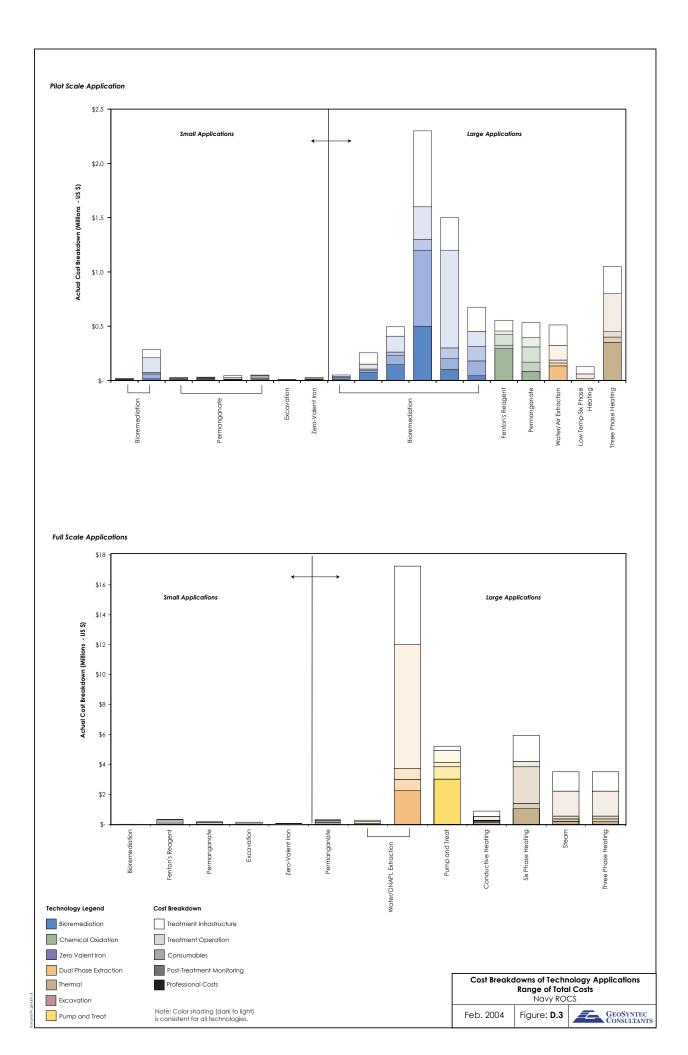
ass Mat	One site with DNAPL trapped in dead-end fractures Technology Removal
uwot	0% Unknown

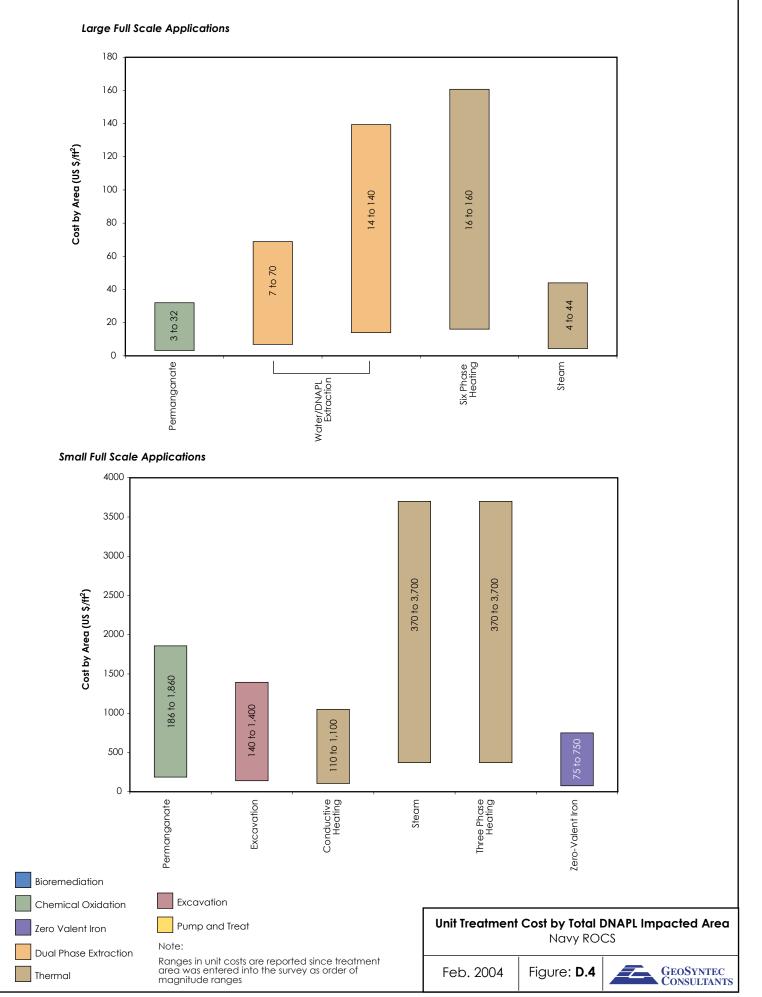
Post Treatment Monitoring On going Full-scale completed Pilot test completed Treatment Stage Full-scale completed Full-scale completed Full-scale completed Full-scale completed Pilot test completed Sorbed $\times | \times$ × × $|\times|$ × × $|\mathbf{x}|\mathbf{x}|$ × **DNAPL** Distribution in Subsurface Residual $\times | \times$ × × × × × × × × In pools × $\times | \times$ × Diffused Into Low K Layers × × $| \times |$ × × $|\times|$ × × \mathbf{x} × × Dead-end Fractures × **Perceived Success** Not Yet Evaluated Poor Success Fair Success Success Success Success Success Success Success Success Success Rebound Unknown Unknown Unknown Unknown Unknown Unknown No No No Yes No No Unknown Unknown Unknown Unknown **Mass Flux** Unknown Unknown Unknown Unknown Decrease Unknown 61-80% 0%0Source Mass >80 <90% Unknown Unknown Unknown Unknown Unknown Unknown Removal >90% <10% >00% >90% 11 Sites with DNAPL Diffused into Low K Layers Chemical Oxidation-Fenton's Reagent Dual Phase-DNAPL/air extraction Dual Phase-Water/air extraction Thermal-Three phase heating Technology Thermal-Steam Bioremediation Bioremediation Bioremediation Excavation Excavation Excavation

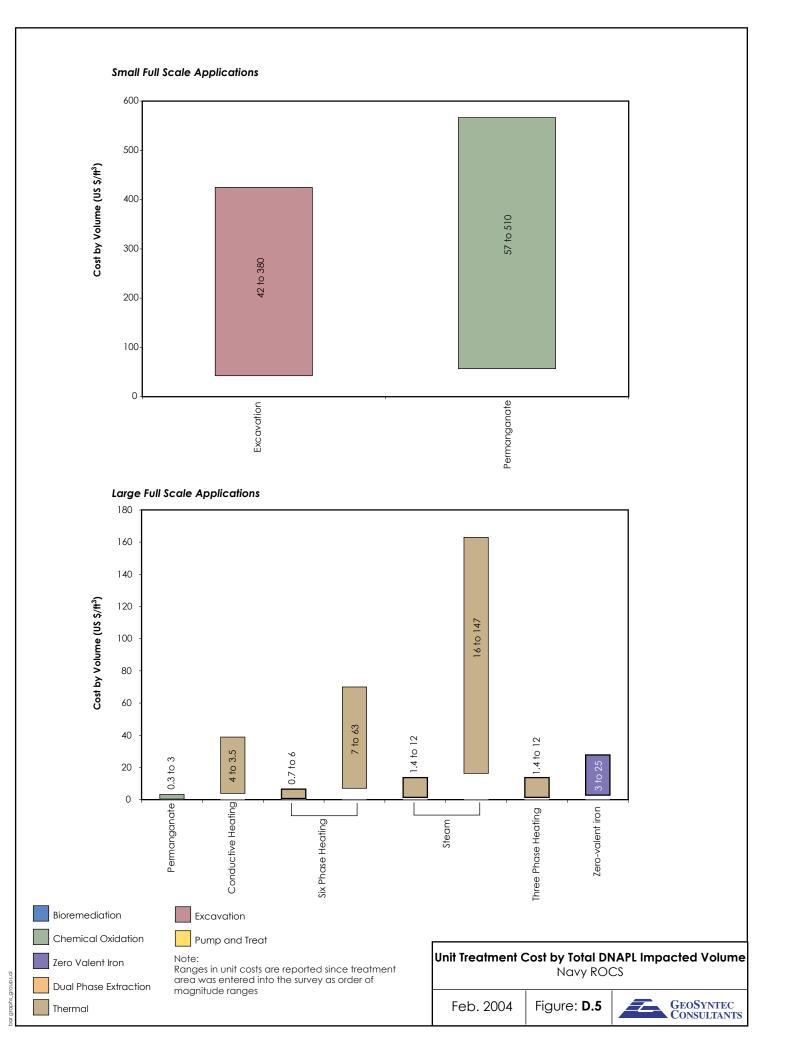


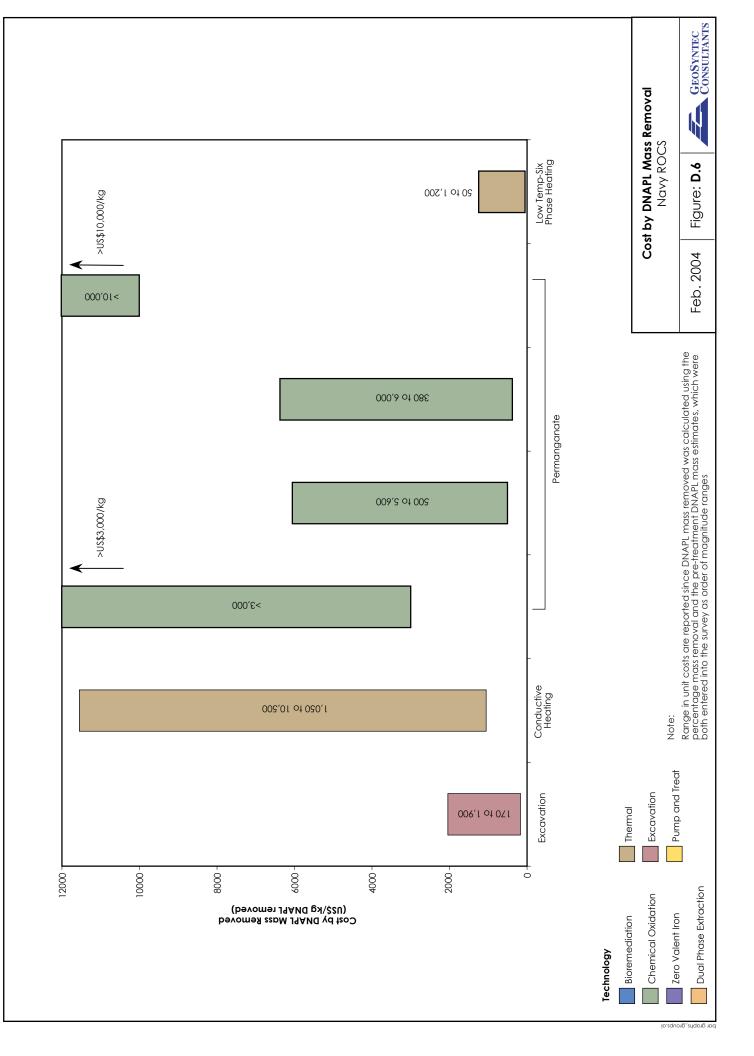


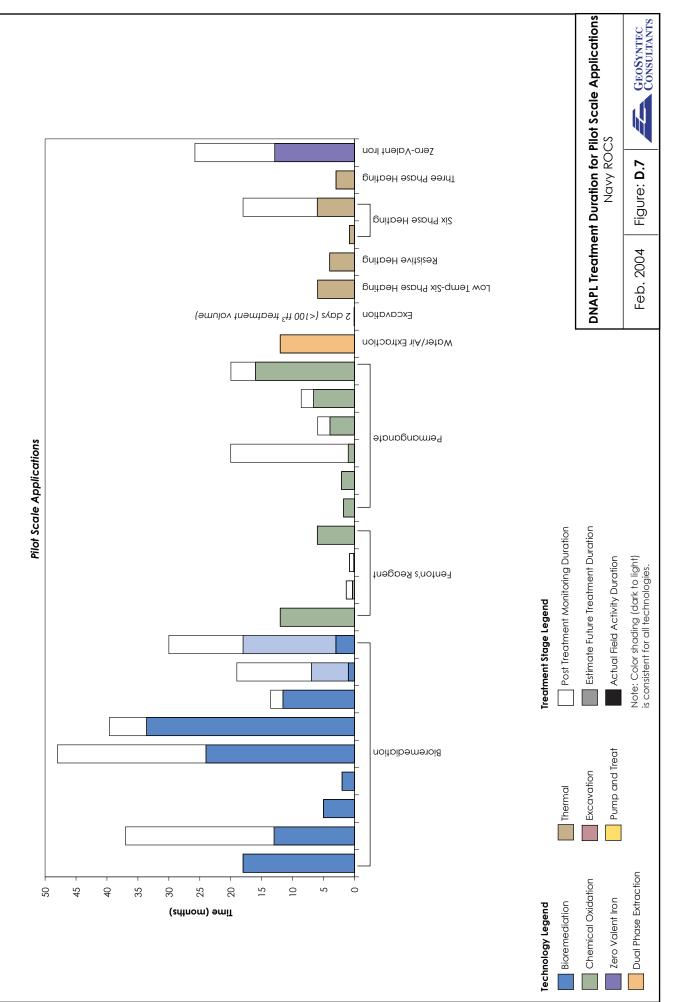
Pump and Treat











par Graphs_groups.ai

